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Thermo-elastic behaviour of grossular garnet at high pressures and temperatures

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ABSTRACT

The thermo-elastic behaviour of synthetic single crystals of grossular garnet ($\text{Ca}_3\text{Al}_2\text{Si}_3\text{O}_{12}$) has been studied *in situ* as a function of pressure and temperature separately. The same data collection protocol has been adopted to collect both the pressure-volume (P - V) and temperature-volume (T - V) datasets in order to make the measurements consistent with one another. The consistency between the two datasets allows simultaneous fitting to a single pressure-volume-temperature equation of state (EoS), which was performed with a new fitting utility implemented in the latest version of the program EoSFit7c. The new utility performs fully weighted *simultaneous* fits of the P - V - T and P - K - T data using a thermal pressure Equation of State combined with any PV EoS. Simultaneous refinement of our P - V - T data combined with that of K^T as a function of T allowed us to produce a single P - V - T - K^T equation of state with the following coefficients:

$V_0 = 1664.46(5) \text{ \AA}^3$, $K_{T0} = 166.57(17) \text{ GPa}$ and $K = 4.96(7)$ $\alpha_{(300\text{K}, 1\text{bar})} = 2.09(2) \cdot 10^{-5} \text{ K}^{-1}$ with a refined Einstein temperature (θ_E) of 512K for a Holland-Powell-type thermal pressure model and a Tait 3rd-order EoS. Additionally, thermodynamic properties of grossular have been calculated for the first time from crystal Helmholtz and Gibbs energies, including the contribution from phonons, using density functional theory within the framework of the quasi-harmonic approximation.

Keywords: grossular, high-pressure, high-temperature, diffraction, bulk modulus, P-V-T-K fit, EoSFit

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1. INTRODUCTION

53 Stable over a wide range of pressure and temperature conditions, garnets are among the most
54 abundant phases in the Earth's upper mantle and transition zone. Yet coherent and homogeneous
55 equation of state coefficients to reliably describe their behaviour at simultaneous high pressures and
56 temperatures still have to be determined with sufficient accuracy and precision. Recent papers have
57 shown the importance of a robust and valid description of the elastic behaviour of the most
58 abundant mineral phases for the accurate calculation of host-inclusion entrapment pressures that in
59 turn allow the growth conditions in the Earth to be inferred (e.g. Angel et al. 2014, 2015a, 2015b;
60 Ashley et al. 2015; Milani et al. 2015). However, the determination of a full set of elastic
61 coefficients of a mineral is always challenging because of the limitations in performing
62 simultaneous high-pressure (P) and temperature (T) experiments on one side and because of the
63 lack of consistent measurements performed separately at high- P and high- T . Therefore, a reliable
64 description of the elastic behaviour of minerals at high- P, T with a single set of Equation of State
65 (EoS) coefficients can be extremely challenging.

66 Pressure-volume EoS are well-developed and are capable of reproducing the isothermal
67 volume or density changes of materials to within the experimental uncertainties. A natural approach
68 to describing the P - V - T behaviour of a material is therefore to describe how the volume (V_{0T}), bulk
69 moduli (K_{0T}^1) and its pressure derivatives (K_{0T}') change with increasing temperature (at room
70 pressure), and then use these parameters to calculate the isothermal compression at the temperature
71 of interest. A potential limitation of this approach is that while the variation of K_{0T} with T can be
72 measured, the variation of K_{0T}' with T has been measured very rarely, although it is obvious that it
73 should increase slightly with increasing temperature. As Helffrich and Connolly (2009) pointed out,
74 the common assumption that K_{0T}' does not change with temperature in combination with the
75 approximation that $\partial K_{0T}' / \partial T$ is constant, often leads to the prediction of non-physical negative

¹ Notation: subscripts on K (e.g. $K_{P,T}$) indicate the reference pressure and temperature to which the bulk modulus is referred; superscripts S or T (e.g. K^S or K^T) stand for adiabatic or isothermal quantities.

76 thermal expansion coefficients at reasonably modest pressures for a large number of materials. An
 77 alternative approach is to employ the idea of thermal pressure (e.g. Anderson 1995). Then the total
 78 pressure (P) at a given V and T can be expressed as the sum of two terms:

$$79 \quad P(V, T) = P(V, T_{ref}) + Pth(T)$$

80 The function $P(V, T_{ref})$ is the isothermal equation of state for material at the reference temperature,
 81 up to the volume at P and T_{ref} . As shown in the Figure 1, if $T > T_{ref}$, then $P(V, T_{ref}) < P(V, T)$. The
 82 thermal-pressure $Pth(T)$ is the pressure that would be created by increasing the temperature from
 83 T_{ref} to T at constant volume, which is the isochor of the material passing through the final P, T point
 84 of interest. As shown in Figure 1, the concept of thermal pressure is therefore essentially a way of
 85 calculating $P(V, T)$ along a different path in $P-T$ space. We can consider that the isothermal
 86 compression part that gives us $P(V, T_{ref})$ is well-defined. The slope of the isochor is given by

$$87 \quad \left(\frac{\partial P}{\partial T} \right)_V = \frac{\alpha}{\beta} = \alpha K^T \quad (\text{where } \alpha \text{ is the volume thermal expansion and } \beta \text{ is the compressibility of the}$$

88 material) so the thermal pressure is $Pth = \int_{T_{ref}}^T (\alpha K^T)_V dT$. In order to use a thermal-pressure EoS one

89 then has to know, or to make an assumption, about how the product αK^T varies along the isochor.

90 For many materials and P, T ranges of interest (e.g. Figure 1) the temperatures are in the region of
 91 the Debye temperature, so the isochor is linear and thus αK^T is a constant. In this case

$$92 \quad Pth \sim (\alpha K^T)_V (T - T_{ref}).$$

93 A number of other assumptions or models of the behaviour of the material can be made,
 94 including the Debye and similar models which make assumptions about the vibrational and hence
 95 thermal behaviour of materials. Holland and Powell (2011) noted that the form of αK^T against
 96 temperature should have the shape of a heat capacity function and developed an approximation for
 97 the integral of αK^T that involves an Einstein function, as follows:

$$P_{th} = \alpha_0 K_{00} \left(\frac{\theta_E}{\xi_0} \right) \left(\frac{1}{\exp(\theta_E/T) - 1} - \frac{1}{\exp(\theta_E/T_{ref}) - 1} \right)$$

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with ξ_0 being the same as in the thermal expansion equation of Kroll et al. (2012). This thermal-pressure model has the physically-correct properties that the product αK^T becomes constant at high temperatures while it decreases to zero at low temperatures. This means that both the bulk modulus K_{0T} and the thermal expansion become constant at low temperatures, and both have an approximately linear variation with temperature above θ_E . The exact expressions for thermal expansion and bulk modulus as a function of temperature depend on the choice of isothermal equation of state, but at T_{ref} and zero pressure $\alpha = \alpha_0$ (Holland and Powell 2011). In this study we used the Holland and Powell (2011) model of thermal pressure in combination with the Tait isothermal EoS (so as to maintain compatibility with Thermocalc, Holland and Powell 2011) to determine the EoS of grossular ($\text{Ca}_3\text{Al}_2\text{Si}_3\text{O}_{12}$) garnet. We show that the thermal pressure approach allows the full P - V - T EoS to be determined by fitting simultaneously measurements of the volume and bulk modulus over a range of temperature but only at room pressure. Additional separate measurements of the volume compression curve at high- P can be used to provide either an independent test of the EoS parameters, or be used to further constrain them in simultaneous global fits of all available data. This approach has the advantage of not requiring data from simultaneous high- P , high- T measurements which remain challenging with respect to both the control and accurate measurement of P and T , and with respect to the precision in volume measurements. A comparison with previous literature data (e.g. Skinner 1956; Isaak et al. 1992; Thieblot et al. 1998; Kono et al. 2010; Gwanmesia et al. 2014; Du et al. 2015 etc..) is provided together with a further crosscheck of our experimental data against the results obtained by density functional theory (DFT) ab-initio simulation. Thermodynamic properties of grossular have been calculated for the first time from crystal Helmholtz and Gibbs energies, including the contribution from phonons, using density functional theory within the framework of the quasi-harmonic approximation.

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2. METHODS

2.1 *Sample synthesis and characterization*

127 Synthetic single crystals of pure end-member grossular (Gr₁₀₀) garnets were synthesized at
128 high pressures with a multi-anvil apparatus at the Bayerisches Geoinstitut. Stoichiometric mixtures
129 of CaO (prepared by decarbonation of CaCO₃), Al(OH)₃ and SiO₂ were packed into platinum
130 capsules of 3.5 mm length and 2 mm diameter. Samples were synthesized at 6 GPa and 1300°C.
131 The experimental sample was heated in 23 mins and the temperature was kept at 1300°C for 40
132 mins. Each multi-anvil experiment was performed with Cr-doped MgO octahedra of 18 mm edge
133 length combined with tungsten carbide cubes of 11 mm truncation-edge lengths. For all the
134 experiments a graphite heater was employed. Experiments were quenched by cutting the electrical
135 power supply to the furnace. Single-crystals of up to ca. 100 µm in diameter were recovered from
136 the capsules. The compositions of the crystals have been determined by electron microprobe at the
137 University of Padova using a Cameca CAMEBAX-micro operating at 15 nA and 15 kV with
138 standards of andradite for Si, spinel for Al and diopside for Ca. Results from the chemical analyses
139 are reported in Table 1. Two crystals (with size ca. 80 x 30 x 30 µm) were selected for the X-ray
140 diffraction study based on the absence of twinning and visible inclusions and on the quality of their
141 diffraction peak profiles (FWHM not greater than 0.15°).

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2.2 *Thermal expansion behaviour by single crystal X-ray diffraction experiments*

144 The high-temperature single-crystal X-ray diffraction experiments were performed at the
145 University of Pavia with a HUBER four-circle point detector diffractometer operating at 50 kV and
146 30 mA (MoK α radiation) and equipped with a 0.8 mm short collimator and automated with the
147 SINGLE software (Angel and Finger 2011). This diffractometer is equipped with a newly designed

148 micro-furnace controlled by a Eurotherm temperature regulator (Alvaro et al. 2015). For the high- T
149 experiments, one single crystal of grossular was placed inside a thin-walled quartz vial (0.3 mm
150 inner diameter and 26 mm long) and held in position by means of quartz wool. The vial was then
151 mounted on a metal goniometer head on the diffractometer. During the high-temperature
152 experiments the effects of the crystal offsets and the diffractometer aberrations were compensated
153 by using the eight-position centring method (King and Finger 1979) before starting the high- T
154 measurement. The unit-cell parameters (Table 2 and Figure 2) were determined at 23 different
155 temperatures from room- T up to 1020 K using only 4-position centring because of the spatial
156 restrictions described in Alvaro et al. (2015). A further 7 measurements were then made on cooling
157 back to room temperature. Each individual measurement took about 12 hours, and the high-
158 temperature experiment lasted for a total of 14 days. Unconstrained unit-cell parameters confirmed
159 the cubic symmetry within 1 e.s.d. Therefore, only constrained unit-cell parameters obtained by
160 vector least-squares fit (Ralph and Finger 1982) are reported in Table 2. Unit-cell parameters
161 determined on the heating and cooling cycles are indistinguishable.

162 The low-temperature single-crystal X-ray diffraction experiment was performed on the very
163 same crystal used for the high- T experiment still mounted inside its quartz vial. The experiment was
164 carried out at the Institut für Mineralogie und Kristallographie at the University of Wien using a
165 Stoe StadiVari diffractometer equipped with an Incoatec I μ S microsource (Mo-target, with multi-
166 layer mirror optics, operating at 50kV and 0.1mA) and a Pilatus 300K area detector. **Low**
167 **temperatures were** achieved using an Oxford Cryojet system that allows measurements to be
168 performed between 100 and 500 K. Full intensity data collections up to $110^\circ 2\theta$ (with a coverage of
169 99% and redundancy higher than 4) were performed at 10 different temperatures from room- T down
170 to 100 K. Correction of the crystal offsets from the goniometer center were made by first
171 performing a short data collection and adjusting the crystal position to make the frame scale factors
172 flat. Unconstrained unit-cell parameters confirmed the cubic symmetry within 1 e.s.d.; therefore,
173 only constrained unit-cell parameters are reported in Table 2 and Figure 2 together with those

174 obtained at high- T .

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176 *2.3 Compressibility behaviour by single crystal X-ray diffraction experiments*

177 High-pressure single-crystal X-ray diffraction experiments for the grossular garnet were
178 carried out with a Huber four-circle diffractometer at the Bayerisches Geoinstitut. The
179 diffractometer operates at 50 kV and 40 mA (MoK α radiation), is equipped with a point detector
180 and is automated by SINGLE software (Angel and Finger 2011). The sample was loaded in a ETH-
181 type diamond-anvil cell (DAC, Miletich et al. 2000) using a steel gasket, pre-indented to 100 μm
182 thickness and with a hole diameter of 250 μm . Methanol:ethanol mixture 4:1 was used as pressure
183 transmitting medium, which remains hydrostatic up to about ~ 9.5 GPa (Angel et al. 2007; Klotz et
184 al. 2009). A single crystal of quartz was loaded in the DAC together with the sample and used as a
185 pressure standard (Angel et al. 1997). During the centring procedure, the effects of crystal offsets
186 and diffractometer aberrations were eliminated from the refined peak position by the eight-position
187 centring method (King and Finger 1979). Unconstrained unit-cell parameters were obtained by
188 vector least-squares fit (Ralph and Finger 1982) on not less than 20 reflections up to $2\theta = 29^\circ$. The
189 symmetry-unconstrained unit-cell edges show deviations smaller than 1 e.s.d. from the constrained
190 ones thus confirming the cubic symmetry within 1 e.s.d.. Therefore, only the constrained unit-cell
191 parameters have been reported in Table 3 and Figure 3.

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193 *2.4 Fitting of the pressure-volume-temperature-modulus data using EoSFit7c*

194 All fits described in this manuscript have been performed with the EoSFit7c program (Angel
195 et al. 2014). Previous versions of this program were able to fit isothermal EoS, thermal expansion
196 models and the thermal-pressure EoS of Holland and Powell (2011) to P - V , V - T or P - V - T data, by
197 the method of least-squares. We have now extended the capabilities of EoSFit-7c to also fit all of
198 these forms of EoS to bulk and linear moduli, either alone or in combination with volume or unit-
199 cell parameter data. Fits are performed by weighted non-linear least squares to minimise the

200 residuals $P_{\text{obs}} - P_{\text{calc}}$, using the effective variance method (Orear 1982) to convert the experimental
 201 uncertainties in volume and bulk moduli, pressure and temperature of each datum into an effective
 202 uncertainty in P (Angel 2000). Input bulk moduli can be either adiabatic (K^S) or isothermal values
 203 (K^T), but in either case they must be Reuss values, corresponding to the response of the material
 204 under hydrostatic pressure. The Voigt and Reuss bulk moduli of a material quantify its volume
 205 response to the application of uniform strain and uniform stress (hydrostatic pressure) respectively.
 206 When uniform stress is applied to a cubic material, it undergoes isotropic (i.e. uniform) strain.
 207 Therefore, the Voigt and Reuss bulk moduli of a cubic material are identical. We remind the reader
 208 that this is not true for materials of lower symmetry and therefore the use of Voigt values, or Voigt-
 209 Reuss-Hill averages, in Equations of state of crystals with symmetry lower than cubic is
 210 inconsistent with the theory underlying equations of state under hydrostatic conditions, and will
 211 lead to incorrect results (e.g. Angel et al. 2009).

212 If the input moduli to EosFit7c are isothermal, no further conversion is made, and the
 213 observational equations for such data in the least-squares are simply the expression for the bulk
 214 modulus in terms of pressure for each EoS. These expressions are direct for invertible EoS such as
 215 the Murnaghan or Tait EoS (Freund and Ingalls 1989; Holland and Powell 2011; Angel et al. 2014).
 216 For non-invertible EoS the procedure used is to calculate the compression ratio V/V_{00} corresponding
 217 to the observed K and then use the P - V expression of the EoS to calculate the corresponding
 218 pressure. If adiabatic moduli are provided as input data, they are converted to isothermal values by
 219 $K^S = (1 + \alpha_V \gamma T) K^T$ where the value of thermal expansion (α_V) is taken from the current EoS at the
 220 pressure and temperature of interest. The Grüneisen parameter γ can be expressed in terms of
 221 measurable quantities, for example $\gamma = \frac{\alpha_V K^S}{C_p} = \frac{\alpha_V K^T}{C_v}$. However, since heat capacity data is not
 222 available at elevated pressures, in EosFit7c we use the simple approximation (Anderson 1996) that

223 $\gamma(P, T) = \gamma_0 \left(\frac{V(P, T)}{V_0} \right)^q$, with the value of q close to 1 for ‘normal’ solids under modest P, T

224 conditions. Only Boehler and Ramakrishnan (1980) and Boehler (1982) appear to have attempted to
 225 measure the value of q , showing that it scales with volume. This means that the Grüneisen
 226 parameter scales with volume, increasing with increasing temperature and decreasing upon
 227 compression, which when considered as a measure of the pressure-dependence of the frequencies of
 228 vibrational modes, is reasonable. This will break down when the material becomes very
 229 anharmonic, near to structural phase transitions, but should not be a problem for materials such as
 230 the grossular garnet in this study. The value of γ_0 (i.e. γ value at the reference conditions) and of q
 231 can be set by the user in Eosfit7c.

232 For linear moduli that describe the variation of the unit-cell parameters with pressure as

233 $M_i^T = -a_i \frac{\partial P}{\partial a_i}$, the procedure is analogous to that used for linear dimensions in EosFit7 (Angel et

234 al., 2014). Internally the linear moduli are converted to volume-like bulk moduli as $K_i = \frac{M_i}{3}$ and

235 treated with volume EoS. For the conversion of adiabatic to linear moduli, we use the relationship

236 between isothermal and adiabatic compressibilities $\beta_i^T = \beta_i^S + \frac{\alpha_i \alpha_V T}{C_p}$ which can be derived from

237 the fundamental relationship between isothermal and adiabatic elastic compliances

238 $s_{ijkl}^T = s_{ijkl}^S + \frac{\alpha_{ij} \alpha_{kl} T}{C_p}$ with the C_p in units of $\text{Jm}^{-3}\text{K}^{-1}$ (Nye 1985). The substitution $\frac{\gamma}{K^S} = \frac{\alpha_V}{C_p}$ gives

239 $\beta_i^T = \beta_i^S + \frac{\alpha_i \gamma T}{K^S} = \beta_i^S \left(1 + \frac{\alpha_i \gamma T \beta_V^S}{\beta_i^S} \right)$ which can be rewritten in terms of moduli as

240 $M_i^S = \left(1 + \alpha_i \gamma T \frac{M_i^S}{K^S} \right) M_i^T$. It is important to notice that for cubic materials $\alpha_i = \alpha_V / 3$ and $\frac{M_i^S}{K^S} = 3$

241 so that only for cubic materials this can be written as $M^S = (1 + \alpha_V \gamma T) M^T$ which is the same

242 relation as for the bulk moduli.

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244 2.5 *Ab-initio DFT (Density Functional Theory)*

245 Ab-initio calculations have been performed in order to crosscheck our experimental results at
246 simultaneous high- P and T . Thermodynamic properties of grossular have been calculated from
247 crystal Helmholtz and Gibbs energies, including the contribution from phonons, using density
248 functional theory (DFT) within the framework of the quasi-harmonic approximation (QHA).
249 Periodic DFT calculations were performed with the plane-wave VASP package (Kresse and
250 Furthmüller 1996) within GGA approximation, with projector-augmented waves (PAW, Blöchl
251 1994) and the PBEsol functional (Perdew et al. 2008). A plane-wave cutoff of 520 eV (1.3 x default
252 VASP value) was used as recommended in the VASP package. Energy convergence was tested for
253 two k-meshes $2 \times 2 \times 2$ and $4 \times 4 \times 4$. Since the denser mesh did not lead to any significant reduction in
254 energy ($\Delta E < 0.001$ eV) the $2 \times 2 \times 2$ mesh was used both for unit cell optimization and subsequent
255 calculation of interatomic forces. The cell was optimized at 17 pressure points within the range [-8
256 to 8 GPa] with step of 1 GPa. The threshold for electronic and ionic convergence was set to 10^{-8} and
257 10^{-6} eV, respectively. The atomic forces were converged to 10^{-5} eV/Å. Phonon frequencies at
258 constant cell volumes were calculated in program PHONOPY (Togo and Tanaka 2015) which takes
259 advantage of the direct method of Parlinsky et al. (1997). The force constants used to construct the
260 dynamical matrices (input for PHONOPY) were calculated in the VASP package using density
261 functional perturbation theory (DFPT, Baroni et al. 2001) and taking into account symmetries of the
262 conventional body-centered cell of the garnet, containing 160 atoms. The results obtained with
263 DFT-GGA commonly overestimate cell volume values of ionic solids by up to 9.5% with respect to
264 experimental results while PBEsol functional brings the discrepancy down to 2% (Perdew et al.
265 2008). In our case, the DFT calculations yielded volume values of 1664.6 \AA^3 (static DFT), 1681.9
266 \AA^3 (zero-point vibration included) and 1687.3 \AA^3 ($T=290$ K) which agree with experimental values
267 of 1664.3 \AA^3 at room P and T , (0.0001 GPa and 300 K) within 0.0 %, 1.1% and 1.4% respectively.
268 This volume overestimation amounts to about -2 GPa, on the pressure scale, with respect to the
269 experimental datum.

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3. RESULTS

3.1 Temperature-volume data

The room- PT unit-cell volumes (V_{00}) of grossular as determined in this study, agree with one another to better than 1 e.s.d. thus confirming the reliability of the unit-cell lattice determination with different instruments and measurement protocols. Therefore, we combined the high- T measurements together with those at low- T in a single dataset without any normalization or re-scaling. The evolution of the unit-cell volume with temperature is reported in Figure 2 and Table 2 where the low- T and the high- T data are reported together. A continuous increase of the unit-cell volume is observed as a function of temperature with no evidence of any irreversible change in the crystal occurred up to the maximum temperature reached in this study. Data collected both with increasing and decreasing temperature overlap within experimental error, thus indicating good experimental reproducibility. The temperature-volume data for the entire range of T studied were fit using the software EoSFit7c (Angel et al. 2014) using a Kroll-type EoS (Kroll et al. 2012) allowing for refinement of the Einstein temperature (θ_E) but not refining K' value that was kept fixed to 5 (i.e. the value from compressibility data, see the following sections). The thermal expansion coefficients obtained were $\alpha_{(298\text{K}, 1\text{bar})} = 2.08(2) \cdot 10^{-5} \text{ K}^{-1}$, $V_{00} = 1664.45(7) \text{ \AA}^3$ with $\theta_E = 515(20) \text{ K}$. We compared our T - V for grossular with those from Isaak (1992) determined by dilatometry. Isaak et al. (1992) measured physical expansions, and converted such values to volume expansions $Y_v(\text{obs})$. Conversion of Isaak et al. (1992) data (taken from their Table 2) to unit-cell volumes has been performed according to $V = V_0(1 + Y_v \text{obs})$. In order to ensure that our measurements and those by Isaak et al. (1992) were on the same volume scale we used our V_{00} at the same reference temperature of 293K used by Isaak et al. (1992). A plot of the two datasets (see Figure 2) shows that they agree within their mutual uncertainties.

3.2 Temperature-volume-bulk modulus data

296 Isaak et al. (1992) also determined the adiabatic bulk modulus of grossular by RUS
 297 measurements and then used measured heat capacity data and their own thermal expansion data to
 298 also calculate the isothermal bulk modulus at each measurement temperature by means of
 299 $K^S = (1 + \alpha_V \gamma T) K^T$, using the value of the Grüneisen parameter $\gamma = \frac{\alpha_V K^S}{C_p}$. The reliability of the
 300 heat capacity (C_p) data by Isaak et al. (1992) has been confirmed by comparison with the more
 301 recent data by Bosenick et al. (1996) (see Figure 4). The C_p values determined by Isaak et al. (1992)
 302 fall well within the scatter of the data determined by Bosenick et al. (1996) and are also in good
 303 agreement with those we have obtained from the DFT calculations.

304 Using K and V data as a function of T it is obviously possible to refine V_{00} , α_0 , θ_E and K_{00} . But
 305 in addition it is possible and required to refine K' because this affects the results when using the
 306 thermal pressure model as it appears in the isothermal EoS $P(V, T_{ref})$, and at room pressure
 307 $P(V, T_{ref}) = -P_{th}(T)$. Simultaneous refinement of our T - V data and the T - K^S data from Isaak et al.
 308 (1992) for grossular using $\gamma_0 = 1.22$ and $q = 0$ (i.e. $\gamma = \gamma_0$) yielded identical results as to those obtained
 309 from fitting the T - V - K^T data (see Table 4²). This shows that the assumption that γ is constant is a
 310 reasonable approximation for grossular. This observation is also in agreement with Isaak et al.
 311 (1992) who calculated that γ drops by $0.1 (\pm 0.05)$ over the temperature range from 300K to 1200K
 312 and with our results from DFT calculations as shown in Figure 4 according to which γ drops to 1.24
 313 at 650K and then it assumes constant values (e.g. changes of about 2 at the 3rd decimal place) above
 314 650K.

315

316 3.3 Pressure-volume data

317 The unit-cell volume decreases smoothly with increasing pressure, as shown in Figure 3 and
 318 reported in Table 2, up to the maximum hydrostatic pressure reached in this study of ~ 7.5 GPa.

² The Equation of State coefficients reported in Table 4c representing the best fit of our data have been deposited as supplementary material in the form of .eos files (that can be read with EoSFit7c). They are also available directly from the authors.

319 Such pressure covers the stability pressure range for upper mantle garnets. Fit of the P - V data alone
 320 using a 3rd order Tait EoS yielded the following coefficients: $V_{00} = 1664.36(7) \text{ \AA}^3$, $K_{00} =$
 321 $167.45(1.04) \text{ GPa}$ and $K_{00}' = 4.93(31)$ also reported in Table 4a. The goodness of the fit results was
 322 confirmed by the low ΔP_{max} value (calculated as $|P_{\text{obs}} - P_{\text{calc}}|$) of 0.010 GPa. The equation of state
 323 coefficients obtained with this method agree with those obtained by T - V - K fitting to better than 1
 324 e.s.d. thus confirming the reliability of the results from the two fitting approaches.

325

326 3.4 P - V - T - K and PVT Equation of State

327 The remarkably good agreement between the lattice parameters as determined with different
 328 experimental settings and that of the EoS coefficients obtained with different procedures allowed us
 329 to perform a simultaneous refinement of the entire set of EoS coefficients by combining our
 330 pressure-volume and temperature-volume data together with those of temperature-bulk modulus
 331 (Isaak et al. 1992) into a single dataset. Fully weighted simultaneous fit of the P - V - T - K data has
 332 been performed using a thermal pressure Equation of State with a 3rd order Tait EoS, allowing
 333 simultaneous refinement of V_{00} , K_{00} , K_{00}' , α_0 and θ_E . As expected, the resulting coefficients $V_{00} =$
 334 $1664.46(5) \text{ \AA}^3$; $K_{00}=166.57(17) \text{ GPa}$; $K_{00}'=4.96(7)$; $\alpha_{(298\text{K}, 1\text{bar})} = 2.09(2) \cdot 10^{-5} \text{ K}^{-1}$, with $\theta_E = 512(19)\text{K}$
 335 ($\gamma_0=\gamma=1.22$) agree within 2σ to those from P - V and T - V - K separately (see Table 4). At this point it
 336 is important to point out that the main difference between the P - V - T - K and T - V - K fits can be seen in
 337 the description of the bulk moduli as a function of T (Figure 5a). It should be clear from Figure 5a
 338 that while in the thermal regime between room- T and the maximum temperature reached with the
 339 experiments the bulk moduli calculated from the T - V - K and P - V - T - K Equations of State agree
 340 within mutual uncertainties, in the low- T regime much larger **discrepancies** are observed. These
 341 discrepancies are due to the different values of θ_E in the two EoS.

342 Simultaneous refinement of the P - V - T data alone yielded the following coefficients $V_{00} =$
 343 $1664.44(6) \text{ \AA}^3$; $K_{00}=166.19(1.25) \text{ GPa}$; $K_{00}'=5.14(40)$; $\alpha_{(298\text{K}, 1\text{bar})} = 2.09(2) \cdot 10^{-5} \text{ K}^{-1}$, with $\theta_E = 512\text{K}$
 344 (not refined). Standard deviations on the bulk modulus (K_{0T}) and its first derivative (K') obtained

345 from the P - V - T fitting are 6-8 times larger than those from the P - V - T - K ones and, when considered,
346 the resulting coefficients from the P - V - T fit agree within 1 sigma with those obtained from the P - V -
347 T - K fits. The P - V - T results are not showed in Figure 5a for sake of clarity, however, the resulting
348 bulk moduli as a function of T agrees with the experimental data within 1 σ .

349 4. DISCUSSION

350 4.1 Thermal expansion: comparison with literature data

351 The measured room temperature-pressure volumes (V_{00}) reported in this study are in good
352 agreement with those previously reported in literature by several authors (see Table 4a and b). So
353 far, only few X-ray diffraction thermal expansion data are available for grossular garnet (e.g.
354 Skinner 1956; Bosenick and Geiger 1997; Thieblot et al. 1998; Du et al. 2015) and most of them
355 have been obtained by means of powder X-ray diffraction. Data from Du et al. (2015) and those by
356 Skinner (1956) have been re-fitted to their entire temperature range 296 K - 859 K (excluding two
357 outliers at 393 K and 477 K) and 284 K - 980 K, respectively. On the other hand, data from Thieblot
358 et al. (1998) have been fitted from 300 K to 1453 K (i.e. well below the decomposition temperature)
359 excluding a few outliers at 607 K, 724 K, and 1241 K. The results of all the extrapolations at 300 K
360 are reported in Table 4b and shown Figure 2. As shown in Table 4b, there are considerable
361 differences between our data and those available in literature. These can be most likely ascribed to
362 the compositional differences between the various samples and/or to the uncertainties given by the
363 different experimental methods. For example, both Skinner (1956) and Thieblot et al. (1998) used
364 grossular-rich samples for their investigation. For sake of comparison, regardless the compositional
365 differences, these literature data have been re-fitted to Fei-type EoS (Fei 1995). Results from the
366 fitting are reported in Table 4b, and Figure 2.

367 As shown in Figure 2, our data for the low- T regime are in good agreement with those
368 reported by Bosenick and Geiger (1997) if their most scattered data (e.g. 195, 220 and 295 K) are
369 not considered.

370 The thermal expansion coefficients at 300 K from both Skinner (1956) and Du et al. (2015)
371 are smaller than those calculated from our fitting. The discrepancies between our dataset and that by
372 Skinner (1956) could be ascribed to the differences in composition as their sample referred to as a
373 ‘grossularite’ has not been properly characterized. The thermal coefficient value obtained fitting the
374 data by Du et al. (2015) is not reliable as demonstrated by the negative thermal expansion at high- T
375 (regardless of the Equation of State model chosen). The same is true also for the data reported by
376 Thieblot et al. (1998), where the thermal expansion value decreases with increasing temperature.
377 Our datasets, ranging from 100 K to about 1000 K, allowed fitting of a Kroll-type EoS (Kroll et al.
378 2012), leading to a much more reliable description of the thermal behaviour of grossular. The
379 reliability of the Einstein temperatures used ($\theta_E=569(13)$ K) have been confirmed by comparison
380 with the saturation temperature (where $T_{\text{sat}} \approx \theta_E/2$) obtained from the fitting of the low- T data alone
381 ($T_{\text{sat}}=270(6)$ K) using a Salje-type EoS (Salje et al. 1991). The agreement within 1 sigma is also
382 achieved when comparing with the θ_E from the P - V - T - K fit [$\theta_E=512(19)$ K, see Table 4c]

383

384 4.2 Compressibility: comparison with literature data

385 The elastic coefficients for grossular have been determined in several studies using different
386 methods (e.g. ultrasonic wave velocity, static compression experiments in diamond-anvil cell, ab-
387 initio calculations) as summarised in Table 4a. For purpose of comparison with the literature data,
388 the adiabatic bulk moduli (K^S) obtained by means of ultrasonic wave velocity techniques have been
389 converted to K^T .

390 DFT calculations of a $P(V)$ curve, fitted by a third order Birch-Murnaghan EoS, provide a
391 value of 156 GPa for the bulk modulus K^T at 300K, and a K' of 4.22. The *static* (no zero point and
392 thermal effects included) and the *athermal* (zero point corrected and a temperature of 0K) values
393 are respectively 163.9 and 160.1 GPa (K' fixed at 4.22: the value we obtained at 300K). These
394 results are in line with the *standard* performances of DFT-GGA Hamiltonians like PBEsol, as they
395 generally underestimate bulk moduli by about 10 GPa at room temperature. Another DFT study

396 from the literature (Erba et al. 2014) provided a somewhat higher bulk modulus (171 GPa), but that
397 was a result from a static calculation with a hybrid HF/DFT (B3LYP) Hamiltonian which includes
398 an Hartree-Fock correction to the DFT-GGA electronic exchange functional. Indeed, such
399 correction proved to be crucial to compute bulk moduli (and vibrational frequencies) very close to
400 the experimental data (Prencipe et al. 2009, 2011, 2012, 2014), but it is computationally very costly
401 if a plane-wave basis set is used, as in the present case, and therefore it cannot be afforded. The
402 largest discrepancy in grossular bulk modulus value is with respect to the results obtained by Kawai
403 and Tsuchiya (2012) with a lattice dynamics approach together in combination with a LDA
404 functional. Furthermore, the first derivative of the bulk modulus (K') which we could now reliably
405 refine clearly demonstrates that the correct value must be very close to 5 in agreement with some of
406 the previous studies (see Table 4a).

407 408 4.3 Full elastic behaviour (P - V - T - K): comparison with literature data

409 The new utility available in EoSFit7c allowed us to also apply the same method described in
410 the sections above (P - V - T - K fitting) to the available literature data obtained from independent
411 elastic measurements and PVT datasets (Kono et al. 2010; Gwanmesia et al. 2014). It is remarkable
412 the extra constraint given by the simultaneous weighted fit of the K data (see results in Table 4c)
413 which however applied to the literature dataset does not allow to obtain the same quality as for our
414 elastic coefficients due to the lack of data in the low-temperature regime. This combined with the
415 much lower accuracy and precision that in most cases arises by the limitation of the P - V - T
416 simultaneous measurements (i.e. pressure calibrant, pressure transmitting media etc.) give rise to
417 discrepancies such as those showed in Figure 5b. In this case the discrepancies in bulk modulus
418 values at high- T reach up to 4% (more than 6 GPa at 1400 K). In order to crosscheck the
419 measurements with the DFT results, we performed fitting of the DFT data in the temperature range
420 between 400 and 800K using a thermal pressure equation of state combined with a 3rd order Tait-
421 eos. For purpose of comparison with the other fits, bulk moduli values have been normalized to the

422 bulk moduli extrapolated at 0K from the P-V-T-K fit of the experimental data (Fig. 5a, red line).
423 The trend of the two fits agrees within 1σ thus confirming the reliability of our fitting results via P-
424 V-T-K fits even at the highest temperature where the DFT approach leads to the smallest deviations
425 from the experimental data.

426 4. CONCLUSIONS AND IMPLICATIONS

427 It is obvious that pressure-volume or temperature-volume measurements alone do not
428 provide strong constraints on the cross-terms in P - T space (e.g. $\partial K/\partial T$ or $\partial\alpha/\partial P$) which are
429 fundamental for the description of the elastic properties of a material at simultaneous high- P and
430 high- T . Limitation in experimental measurements and/or in the description of the pressure-volume-
431 temperature data (e.g. incorrect Equations of State, erroneous mis-use of Voigt, Reuss isothermal or
432 adiabatic bulk moduli) has often in the past lead to inaccurate unreliable or even completely wrong
433 results (e.g. negative thermal expansion, erroneous $\partial K/\partial T$ etc...) specially when the EoS are
434 subsequently extrapolated to high- P, T .

435 Pressure-volume EoS are well-developed and, if correctly used, are capable of reproducing
436 the isothermal volume or density changes of materials to within the experimental uncertainties.
437 Previous versions of the EoSFit program were able to fit isothermal EoS, thermal expansion models
438 and the thermal-pressure (Holland and Powell 2011) to P - V , V - T or P - V - T data, by the method of
439 least-squares. We have now extended the capabilities of EoSFit7c to also fit all of these forms of
440 EoS to bulk and linear moduli, either alone or in combination with volume or unit-cell parameter
441 data. With this approach we can now provide a reliable description of the variation of K' with T thus
442 avoiding mistaken extrapolations leading to non-physical behaviour as often occurs for several
443 materials (Helffrich and Connolly 2009).

444 The consistency of the results obtained with this new fitting procedure has been shown by
445 comparison of results from using adiabatic and isothermal moduli. It has also been demonstrated the
446 importance of performing fitting with this procedure in terms of both accuracy and precision of the
447 thermoelastic coefficients when the PT cross-terms ($\partial K/\partial T$ or $\partial\alpha/\partial P$) are extrapolated in PT space

448 (see Figure 5b). Additionally, thermodynamic functions were calculated for grossular using DFT
449 approach within QHA. The obtained results are in fairly good agreement with experiment, when
450 allowing for the known short-comings of the functionals employed in the DFT.

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