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# Furoxan nitric oxide donors disperse Pseudomonas aeruginosa biofilms, accelerate growth, and repress pyoverdine production

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1	Furoxan nitric oxide donors disperse Pseudomonas aeruginosa biofilms, accelerate	
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# 19 Key words:

- 20 Furoxans, 1, 2, 5 Oxadiazole N-Oxides, nitric oxide donors, Pseudomonas aeruginosa,
- 21 Biofilms, Pyoverdine, Biofilm Dispersal

22

### 23 Abstract

The use of nitric oxide (NO) as a signal for biofilm dispersal has been shown to increase the 24 susceptibility of many biofilms to antibiotics, promoting their eradication. The delivery of NO 25 to biofilms can be achieved by using NO-donors with different kinetics and properties of NO 26 release that can influence their efficacy as biofilm control agents. In this study, the kinetics of 27 three furoxan-derivatives were evaluated. The effects of these NO-donors, which have an 28 advantageous pharmacological profile of slower onset with an extended duration of action, on 29 Pseudomonas aeruginosa growth, biofilm development and dispersal were also characterized. 30 31 Compound LL4254, which showed a fast rate of NO release, induced biofilm dispersal at 32 approximately 200 µM. While LL4212 and LL4216 have a slower rate of NO release, both compounds could induce biofilm dispersal, under the same treatment conditions, when used at 33 higher concentrations. Further, LL4212 and LL4216 were found to promote P. aeruginosa 34 growth in iron-limited minimal medium, leading to a faster rate of biofilm formation and 35 glucose utilization, and ultimately resulted in early dispersal of biofilm cells through carbon 36 starvation. High concentrations of LL4216 also repressed production of the siderophore 37 pyoverdine by more than 50-fold, via both NO<sub>x</sub>-dependent and NO<sub>x</sub>-independent mechanisms. 38 The effects on growth and pyoverdine levels exerted by the furoxans appeared to be mediated 39 40 by NO-independent mechanisms, suggesting functional activities of furoxans in addition to their release of NO and nitrite. Overall, this study reveals that secondary effects of furoxans 41 42 are important considerations for their use as NO-releasing dispersal agents, and that these 43 compounds could be potentially re-designed as pyoverdine inhibitors.

# 44 Introduction

Bacterial cells growing within biofilms can be up to a thousand times more resistant to antimicrobial agents than their planktonic counterparts, making their eradication very difficult <sup>1, 2</sup>.
This enhanced tolerance to stress, immune defenses and antibiotics may partly explain the
observation that biofilms are associated with up to 80% of all microbial infections and the
majority of chronic and recurrent infections including pneumonia in cystic fibrosis patients,
wound infections and medical implant-associated infections <sup>1, 3, 4</sup>.

51 The insensitivity of biofilms to conventional antimicrobial treatments appears to be multifactorial and involves: (i) a protective barrier of self-produced extracellular polymeric 52 substances (EPS) that can inactivate or reduce diffusion of bactericidal compounds <sup>5</sup>, (ii) 53 enhanced lateral transfer of genes including those from drug resistant and extremely drug 54 resistant (XDR) strains <sup>6</sup> and (iii) the presence of dormant-like persister cells <sup>7</sup>. Given the 55 central role played by biofilms in promoting antibiotic resistance and causing the failure of 56 57 therapeutic treatments, there is an urgent need to develop alternative strategies specifically aimed at biofilm development processes. Towards this end, the dispersal phase of the biofilm 58 59 life cycle has been targeted as a key stage that can be manipulated to control biofilms<sup>8</sup>. During dispersal, bacterial cells transit from a biofilm to a planktonic mode of growth, rendering them 60 more susceptible to various antibiotics 9, 10. Biofilms disperse in response to several 61 environmental cues, such as changes in carbon levels and iron levels <sup>11-13</sup>. In addition, low 62 nanomolar concentrations of the gaseous free radical nitric oxide (NO) were found to induce 63 biofilm dispersal in *Pseudomonas aeruginosa*, an opportunistic pathogen and the model 64 organism for biofilm studies <sup>9</sup>, as well as several other bacterial species <sup>8</sup>. Furthermore, 65 exposure of pre-established biofilms to the NO-donor compound sodium nitroprusside (SNP) 66 increased the susceptibility of biofilm cells towards several antimicrobial agents, including the 67 antibiotics tobramycin and ceftazidime in P. aeruginosa biofilms grown both on abiotic 68 surfaces and in ex vivo sputum samples of CF patients <sup>9, 10</sup>. Recently, the use of NO gas 69 delivered at low dose to CF patients, together with intravenous tobramycin and ceftazidime, 70 reduced P. aeruginosa biofilms and improved lung function when compared to placebo-treated 71 control patients in proof of concept clinical studies <sup>14</sup>. These observations suggest that the use 72 of NO, together with conventional antibiotics, represents a promising alternative for the 73 treatment of chronic biofilm infections. 74

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76 The delivery of NO for medical purposes, in addition to the gaseous form, is typically achieved by using soluble donors. These compounds spontaneously release NO when dissolved in 77 aqueous solutions and the kinetics of release are a function of their donor chemistry. In this 78 79 way, donors can be tuned to optimize the release of NO. Furoxans are heterocyclic compounds 80 containing a 1,2,5-oxadiazole 2-oxide ring and two substituent groups at positions 3 and 4 of the furoxan ring (Figure 1)<sup>15</sup>. These substituents influence several properties of the furoxan 81 82 compound, such as its solubility, the rate of NO production, and whether NO-release is thiolactivated or could occur spontaneously <sup>15-17</sup>. This flexibility allows for the design of furoxan 83 compounds that generate varying fluxes of NO when administered <sup>17</sup>. In general, furoxans have 84 an advantageous pharmacological profile of a slow onset with an extended duration of action, 85 as compared to other NO donors <sup>15, 18</sup>. In turn, furoxans as such, or used in designing NO-donor 86 hybrid drugs, have been observed to be active against a variety of targets and have been 87 assessed for use in cardiovascular diseases, neurological and inflammatory disorders <sup>19-21</sup>. More 88 recently, a few furoxan derivatives were found to inhibit ABC transporters in MDR tumor cells 89 22, 23 90

Despite the many applications of furoxans, they have not been tested for their activity against 91 biofilms. In this study, we evaluated the release of nitrogen oxide species (NO<sub>x</sub>), namely NO 92 93 and nitrite (NO2<sup>-</sup>), from three furoxans, LL4212, LL4216 and LL4254, and studied their effect on P. aeruginosa biofilm development, where they induced biofilm dispersal. LL4212 and 94 95 LL4216 were, additionally found to affect bacterial growth under iron-limited conditions in an NO<sub>x</sub>-independent manner, increasing the rates of glucose utilization and in turn leading to the 96 97 earlier onset of glucose-starvation-induced dispersal of P. aeruginosa biofilms. Further, LL4216 was found to reduce expression of the siderophore pyoverdine in an NO<sub>x</sub>-independent 98 manner. Taken together, the study indicated that, depending on their backbone, furoxans may 99 100 have secondary effects on bacterial growth and is an important consideration for their design and use as NO-releasing agents. 101

102

## **103 Results and Discussions**

#### 104 Kinetics of NO<sub>x</sub> release from LL4254, LL4212 and LL4216

105 The study of the kinetics of  $NO_x$  release from the furoxan compounds indicated that both NO 106 and nitrite ( $NO_2^-$ ) were spontaneously released from the furoxans under physiologically 107 relevant conditions (pH, temperature). The results, expressed as percent mol/mol of NO<sub>x</sub> released with respect to the concentration of furoxan in solution (Figure 2a - c), showed two 108 types of NO<sub>x</sub>-production behavior. For LL4254, NO was the predominant species released, 109 accounting for 82% of the total NO<sub>x</sub> detected (~ 95%) by 24 h (Figure 2a); the liberation 110 occurred very fast, with 80% of the total NO<sub>x</sub> produced within the first 30 min. In contrast, 111 NO<sub>2</sub><sup>-</sup> was the dominant species generated from LL4212 and LL4216 and liberation occurred in 112 an almost linear fashion over a few hours. By 24 h, 83% and 32% of the NO<sub>x</sub> species were 113 detected as NO<sub>2</sub><sup>-</sup> from LL4212 and LL4216, respectively, while only 2.6% and 1.2% were 114 115 measured as NO (Figure 2b, c). An alternative chemiluminescence-based NO detection method, also confirmed the results obtained with the DAN assay (data not shown). 116

Previous studies showed that thiol groups such as those present in L-cysteine (L-cys) can 117 promote NO<sub>x</sub> liberation from furoxan compounds and increase their activity <sup>17</sup>. Since thiol 118 groups are commonly produced by bacteria, in the form of metabolites or proteins, they are 119 120 highly relevant in the context of bacterial infection treatments and their effects on NO<sub>x</sub> release from LL4212, LL4216 and LL4254 were investigated. Exogenous addition of L-cys reduced 121 the extent of NO<sub>\*</sub> liberated from the fast releasing LL4254 but promoted their production from 122 the slower releasing LL4212 and LL4216 (Figure 2d f). At t = 1 h, the presence of L-cys 123 reduced the amount of NO and NO2<sup>-</sup> generated from LL4254 to 4% and 36%, respectively, 124 from an initial 70% and 90% in the absence of L-cys (Figure 2d). At t = 4 h, NO<sub>x</sub> production 125 increased from 26% to 28% in the presence of L-cys for LL4212, and from 8% to 20% for 126 LL4216. NO production from these compounds also increased in the presence of L-cys, from 127 1.8% to 2.2% for LL4212 and from 0.8% to 1.5% for LL4216 (Figure 2 e, f). 128

129 Exogenous addition of L-cys reduced the extent of NO<sub>x</sub> liberated from the fast releasing compound

130 LL4254, promoted their production from the slower releasing LL4216 while did not affected the NO<sub>x</sub>

131 production from LL4212 (Figure 2d - f). At t = 1 h, the presence of L-cys reduced the amount of NO

and  $NO_2^-$  generated from LL4254 to 4% and 36%, respectively, from the initial 70% and 90% in the

absence of L-cys (Figure 2d). At t = 4 h, NO<sub>2</sub><sup>-</sup> production increased from 8% to 20% in the presence

134 of L-cys for LL4216. NO production from LL4216 and LL4212 was not influenced by L-cys (Figure

135 2 e, f).

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- 137

#### 138 Furoxan NO<sub>x</sub>-donors can induce *P. aeruginosa* biofilm dispersal

To determine if the NO-releasing furoxans could induce biofilm dispersal, P. aeruginosa 139 biofilms were treated with 100 µM, 200 µM or 500 µM of the furoxans for 1 h before 140 quantification of biofilm biomass by CV staining. At 200 µM or above, LL4254 reduced the 141 biofilm biomass by >70% compared to untreated control samples (p < 0.0001) (Figure 3a, c). 142 This biomass reduction, which was associated with an increase in OD<sub>600</sub> reading of the 143 supernatant (data not shown), occurred after 1 h treatment, and therefore cannot be linked to 144 growth effects but instead is a clear indication of biofilm dispersal events. These results were 145 in agreement with previous studies, where NO was found to induce P. aeruginosa biofilm 146 dispersal <sup>9, 24</sup>. Dispersal by LL4254 occurred in a concentration-dependent manner. In contrast, 147 LL4212 and LL4216 were unable to induce significant biofilm dispersal at concentrations 148 149 between 100  $\mu$ M and 500  $\mu$ M.

To assess if dispersal was caused by NO or other non-specific interactions between *P*. *aeruginosa* and the LL4254 backbone, the assay was repeated in the presence of the NO scavenger cPTIO. Under the conditions used, cPTIO alone did not have a significant effect on both planktonic and biofilm growth (data not shown). Addition of cPTIO reduced dispersal by LL4254 from an average of 73% to 18%, suggesting that the effect of LL4254 on dispersal was NO-dependent (Figure 3b, d).

NO release from LL4254 was established as an important factor for inducing biofilm dispersal. 156 As the rate of both NO and NO<sub>2</sub><sup>-</sup> generated from LL4212 and LL4216 was much lower than 157 that of LL4254 under cell-free physiological conditions (Figure 2), it is likely that at the same 158 159 compound concentration, both LL4212 and LL4216 produce a lower amount of NO, This may 160 in turn influence the effective NO concentration perceived by the bacteria at a given time, as NO generated may be consumed by cellular processes, reaction with oxygen, and reactions 161 with other chemical compounds present in the medium <sup>25, 26</sup>, hence limiting the extent of 162 biofilm dispersal. Therefore, the assay was repeated using 3.6 mM LL4212 and LL4216, which 163 correspond to the amount of NOx released from approximately 400 µM and 200 µM LL4254 164 165 within 1 h. The furoxan stocks were solubilized in DMSO, and while at low concentrations ( $\leq$ 166 0.5% v/v) DMSO did not significantly alter *P. aeruginosa* growth or biofilm formation in our assays (data not shown), at a higher level of 3.6% v/v (the amount present when adding 3.6167 168 mM furoxan), DMSO reduced the amount of biofilm formed by approximately 14% (Figure 169 3f). Therefore, samples to which an equivalent volume of DMSO was added, were used as a 170 control for comparison of the extent of biofilm dispersal induced with 3.6 mM furoxan treatments. At 3.6 mM, LL4212 and LL4216 dispersed 48% and 70% of the biofilm biomass 171

respectively (Figure 3e, f). Decrease in biofilm biomass observed was concomitant with an 172 increase in planktonic biomass, thus confirming that the observed effects correlated with 173 dispersal of bacteria from the biofilm. Although LL4216 alone, under physiological, cell-free 174 conditions has a slower rate of NO<sub>x</sub> release than LL4212, in this experiment, LL4216 was 175 found to induce a larger extent of dispersal than LL4212. This is likely due to the presence of 176 nucleophiles, e.g. thiol groups, that are produced or secreted by the bacteria <sup>15-17</sup> that promoted 177 NO<sub>x</sub> release from LL4212, LL4216 and LL4254 to different extents, as supported in cell-free 178 NO kinetics experiments (Figure 2). 179

180

#### Furoxans accelerate *P. aeruginosa* biofilm development and enhance glucose utilization 181

Because furoxans have a slower rate of release and can produce NO<sub>x</sub> over longer periods of 182 time, we then assessed whether these compounds can constantly prevent the switch for 183 attachment of free-floating bacteria, maintain cells in a planktonic mode of growth and thus 184 inhibit biofilm formation over time. Furoxans were added to M9GC medium together with the 185 P. aeruginosa at the time of inoculation. After 6 h, wells that had been treated with LL4212 186 and LL4216 showed a significant concentration-dependent reduction in biofilm biomass, with 187 188 increases in planktonic growth, compared to untreated wells, with 100 µM of LL4216 reducing biofilm biomass by > 70% (p < 0.0001) (Figure 4a, c). 189

To further characterize the impact of furoxans on biofilm formation, biofilm biomass of 190 untreated control groups and groups treated with 500 µM LL4212 or 200 µM LL4216, both of 191 which reduced the biofilm biomass at t = 6 h by > 80%, was quantified over time (Figure 4b). 192 Surprisingly, addition of LL4212 and LL4216 promoted both biofilm formation and planktonic 193 growth over the first 4 h and 3 h, respectively (Figure 4b). Subsequently, the biofilm dispersed 194 while planktonic growth continued to increase. A similar sharp decrease in biofilm biomass 195 during growth had already been observed in *P. aeruginosa* biofilms and was linked to biofilm 196 dispersal due to the sudden depletion of the carbon source and the onset of starvation <sup>27</sup>. The 197 levels of glucose in the biofilm cultures were then determined over time and the extent of 198 biofilm reduction at t = 6 h was consistent with a reduction in glucose concentration (Figure 199 200 4d, e). The effect was the most pronounced with 500 µM LL4216, which led to complete glucose depletion by t = 6 h. 201

202 While NO is known to affect biofilm dispersal, a potential role in regulating glucose utilization and growth has not been observed before. To determine if the effects of LL4212 or LL4216 on 203 glucose consumption were dependent on NO, biofilm prevention experiments were carried out 204 with furoxans added together with the NO scavenger cPTIO, or using NO-depleted furoxans 205 which had been incubated in culture medium for 24 h before inoculating with P. aeruginosa, 206 thus resulting in exhaustion of NO released from the donor compounds. The data show that 207 after 6 h, P. aeruginosa biofilms had dispersed under both conditions (Figure 5a - c), 208 suggesting that the effects of LL4212 and LL4216 on biofilm formation were not due to NO. 209

These observations of the impact of LL4212 and LL4216 on biofilms did not correlate with a 210 211 typical dispersal response to NO resulting in biofilm prevention. To further elucidate the effect of the furoxans on biofilms, we then examined their impact in our assay on a known marker of 212 biofilm dispersal in *P. aeruginosa*, including when induced by NO, which is a decrease in the 213 production of the iron chelating siderophore pyoverdine <sup>28</sup>. A similar decrease in pyoverdine 214 levels was also previously observed upon dispersal with 200 µM of LL4254 and 3.6 mM of 215 216 LL4212 or 4216 (Figure 3e). The results revealed that in the presence of 500 µM NOexhausted LL4216, pyoverdine levels were reduced by at least 50-fold compared to untreated 217 biofilms, even though there was no free NO present. In contrast, the use of 500 µM of NO-218 depleted LL4212 had no effect on the pyoverdine levels with respect to the untreated controls 219 220 (Figure 5c). To control for a potential direct influence of LL4216 on the pyoverdine fluorescent signal, 500 µM of LL4216 were added to filtered, cell-free M9GC medium collected from 221 microtiter plate cultures grown under the same conditions, which contained pyoverdine. 222 Pyoverdine levels were found to be relatively stable in filtered medium without any compound 223 and LL4216 at 500 µM only slightly decreased pyoverdine relative fluorescent units (RFU) by 224 225 6.3% and 6.9%, after 6 h and 24 h respectively, compared to untreated cell-free solutions. In contrast, when 500 µM LL4216 was added to a non-filtered culture inoculated with P. 226 *aeruginosa* at t = 0 h, pyoverdine expression was > 99% lower than that of the control after 227 both 6 h and 24 h, suggesting that LL4216 actively repressed pyoverdine expression, rather 228 than simply affecting the fluorescence signal from pyoverdine (Figure 5d). Thus, these results 229 230 suggest that the furoxans can reduce pyoverdine synthesis in P. aeruginosa, via both NOindependent (LL4216) and NO-dependent (LL4212) mechanisms. 231

Finally, because another major  $NO_x$  released from LL4212 and LL4216 is  $NO_2^-$ , its effect on biofilm formation in our assay was investigated. As opposed to NO,  $NO_2^-$  is stable in solution and can be added directly to the culture medium When  $100 - 500 \mu$ M of exogenous nitrite were added at t = 0 h in place of LL4212 and LL4216, there was no significant effect on planktonic growth, biofilm formation or pyoverdine production in *P. aeruginosa* (Supplementary figure 3).

Collectively, these results suggest that LL4212 and LL4216 likely increase the growth rates of 238 239 P. aeruginosa through increased rates of glucose metabolism and cause biofilm dispersal via a carbon or glucose starvation-induced response <sup>11, 12</sup>. The use of NO depleted LL4212 or 240 LL4216 in M9GC medium induced the same effects. Further, the addition of exogenous 241 sources of nitrite, the main NO<sub>x</sub> species released from LL4212 and LL4216, did not affect 242 planktonic growth, biofilm formation or pyoverdine production. This indicates that the 243 backbone or by-product of NO<sub>x</sub> released from these two compounds, but not NO or NO<sub>2</sub>-, were 244 responsible for the increased growth. 245

246

#### 247 LL4212 and LL4216 are not utilized as direct carbon or nitrogen sources by *P. aeruginosa*

One possibility to explain the accelerated biofilm formation and increased growth of P. 248 *aeruginosa* in the presence of the furoxans, is that these compounds may be used as carbon or 249 nitrogen sources for metabolism. To determine the impact of the furoxans on metabolism, P. 250 aeruginosa was inoculated in various modified M9 media with or without available carbon or 251 252 nitrogen sources. In M9 medium supplemented with casamino acids and glucose as sources of carbon (M9GC), or M9 medium supplemented with glucose only (M9G), the addition of 253 500 µM LL4212 or LL4216 increased both the growth rates and growth yields of P. 254 aeruginosa (Figure 6a, Supplementary figure 4), which agrees with our previous observations. 255 In contrast, the addition of LL4212 and LL4216 to M9 medium with glucose but without any 256 nitrogen source (M9-N), did not significantly alter growth rates, although there was a slight 257 increase in total growth (Figure S2c). In M9 medium without any carbon source (M9S), P. 258 aeruginosa did not grow at all whether in the absence or presence of LL4212 and LL4216 259 (Figure Supplementary figure 2d). 260

This hence suggests that the compounds are most likely not utilized as a direct source of energy for *P. aeruginosa*, and may influence growth through interfering with processes related to glucose metabolism and increases in ATP production (Supplementary figure 6). This latter point is also supported by the observation that glucose was more rapidly depleted in thepresence of these compounds.

266

# LL4216 inhibits pyoverdine production under both low and high iron conditions in *P. aeruginosa*, but under high iron conditions reduces the growth rate

269 Another nutrient essential for growth in *P. aeruginosa* is iron, which is typically limiting in M9. However, bacteria can acquire essential trace amounts from contaminants in the water or 270 271 components used to prepare the medium, typically by producing siderophores like pyoverdine, which has high iron affinity. Since the furoxans appeared to influence both growth and 272 273 pyoverdine production in *P. aeruginosa*, we assessed if these effects may be associated with 274 iron uptake. As expected, the addition of 500 µM LL4212 and LL4216 to M9GC resulted in increased growth and reduced pyoverdine levels (Figure 6b, c). In contrast, under the same 275 conditions, LL4254 did not affect growth or pyoverdine production. In M9GC medium, the 276 277 increase in growth rate showed a clear concentration dependence for LL4212, while there was only a slight increase in growth rate for LL4216 at 500 µM. With respect to pyoverdine 278 production, LL4212 repressed pyoverdine only when added at 500 µM, while LL4216 showed 279 280 a clear concentration-dependent reduction.

In M9GCFe, P. aeruginosa displayed an increased growth rate of about 8-fold compared to 281 282 M9GC. In M9GCFe, LL4212 addition resulted in a slight reduction in growth rate relative to the control while LL4216 reduced the growth rate in a concentration dependent fashion, with 283 a 38% reduction when added at 500 µM (Figure 6b). The presence of added iron also generally 284 reduced pyoverdine production. While LL4212 only slightly further reduced this production, 285 by 12% when added at the highest concentration tested, the effects were more pronounced with 286 LL4216, which at 500 µM induced a drastic further reduction in pyoverdine to background 287 288 levels. The addition of LL4254 had no influence on growth rate and pyoverdine production in either medium (Figure Supplementary figure 5c, f). Taken together these data suggest that 289 290 LL4212 and LL4216 may have opposite impacts on P. aeruginosa growth depending on the 291 availability of iron, resulting in increased growth under iron limited conditions, while reducing 292 growth when iron is replete.

# LL4212 and LL4216 may affect biofilm and planktonic growth via influences on glucose utilization and pyoverdine production in *P. aeruginosa*

295 The decrease in pyoverdine levels upon LL4212 and LL4216 addition may be in part due to NO release, which is known to repress the expression of pyoverdine related genes <sup>24</sup>. Indeed, 296 when NO-depleted LL4212 was used, no repression in pyoverdine production was observed. 297 In contrast, the effect of LL4216 on pyoverdine appeared to be independent of NO, as the use 298 299 of the NO-depleted compound still induced a decrease in pyoverdine levels (Figure 5c). Overall, the results show that decreases in pyoverdine production with LL4212 are predominantly 300 301 mediated by NO while similar decreases in LL4216-treated samples were either due to the presence of NO or mediated directly by the LL4216 backbone. Despite these differences, both 302 303 compounds could promote bacterial growth in P. aeruginosa and no correlation between growth rate and pyoverdine levels was observed (Figure 6b, c). Thus, the furoxans likely exert 304 independent effects on growth and pyoverdine levels, especially in the absence of added iron. 305

306 We initially hypothesized that LL4216 may function as a siderophore in place of pyoverdine. 307 Repression of pyoverdine production by LL4216, while still being able to take up iron, could 308 then direct metabolites towards energy generation and growth away from the costly production of pyoverdine. For example, pyoverdine productionwhich has been previously estimated to 309 require approximately 10% of additional carbon consumption per growth unit in a closely 310 related species, *Pseudomonas putida*, grown under iron limitation <sup>29, 30</sup>. However, when 311 supplemented with ferric chloride, no further increase in growth was observed with addition of 312 500 µM of LL4216, suggesting that LL4216 does not function as an alternative iron scavenger 313 and further supporting that LL4216 exerts growth-related effects independent of pyoverdine 314 315 production and iron acquisition.

316

#### 317 Conclusions

318 The results suggest that furoxans with fast and slow NO release profiles could be used at low 319 and high concentrations, respectively, to promote biofilm dispersal. Slow NO-releasing 320 furoxans were also found to be potent at preventing biofilm formation over longer periods of time, in a similar manner as slow NO-releasing polymers had been previously found to inhibit 321 biofilm formation in *P. aeruginosa*<sup>31</sup>, although a potential role of NO in mediating the effects 322 of furoxans on biofilm prevention could not be clearly determined here. Interestingly, slow 323 NO-releasing furoxans may be useful at higher concentrations to, in one single treatment, 324 induce dispersal of pre-established biofilms and prevent re-formation of a biofilm over 325 extended times. 326

327 In addition to inducing biofilm dispersal via NO, the furoxans LL4212 and LL4216 also influenced growth and pyoverdine production of *P. aeruginosa*. The data suggest that the two 328 furoxans exert effects on growth and pyoverdine levels independently of each other and further 329 work is needed to elucidate these mechanisms. While the increased growth rate would not 330 normally be advantageous from the perspective of pathogen control, if the compounds strongly 331 repress siderophore production, they may facilitate control in the host, where iron is severely 332 limited. In this respect, there may be interest in developing these compounds as pyoverdine 333 inhibitors <sup>32</sup>. This is relevant as *P. aeruginosa* is predominantly found in the lungs of cystic 334 fibrosis patients, with pyoverdine expression being a major factor accounting for *P. aeruginosa* 335 virulence, antibiotic resistance and biofilm maturation <sup>33-35</sup>. Further studies would be required 336 for the successful development of furoxans as dispersal agents or for their use in other 337 338 applications.

### 339 Materials and methods

#### 340 Nitric oxide donors and scavengers

- 341 The furoxans LL4212 (3-((2-(dimethylamino)ethyl)oxy)-4-phenylfuroxan), LL4216 (3-((2-
- aminoethyl)thio)-4-phenylfuroxan) and LL4254 (4-(phenylsulfonyl)-3-((2-
- 343 dimethylanimo)ethyl)thio)furoxan) were synthesized as previously reported by Sorba *et al.* <sup>17</sup>.
- 344 2-(4-Carboxyphenyl)-4,4,5,5-tetramethylimidazoline-1-oxyl-3-oxide potassium salt (carboxy-
- 345 PTIO, cPTIO) was purchased from Sigma Aldrich (# C221).

#### **346** Nitrite measurements by Griess reaction (total NOx evaluation)

The total release of  $NO_x$  was evaluated as nitrite ( $NO_2^{-}$ ) by Griess reaction. Furoxan compounds 347 were incubated at 37°C in 50 mM phosphate buffer, pH 7.4 at 0.1 mM concentration in the 348 absence or in the presence of 0.5 mM L-cysteine (L-cys) (5 times mol/mol excess). At regular 349 intervals, the presence of nitrite in the reaction mixture was determined by the Griess assay: 1 350 mL of the reaction mixture was treated with 250 µL of the Griess reagent (4% w/v 351 sulphanilamide, 0.2 % w/v N-naphthylethylenediamine dihydrochloride, 1.47 M phosphoric 352 353 acid). After 10 min at room temperature, the absorbance was measured at 540 nm. A calibration 354 curve was obtained using standard solutions of sodium nitrite at 10 µM to 80 µM. The yield in nitrite was expressed as percent  $NO_2^-$  (mol/mol, relative to the initial compound concentration) 355 356 ± SEM.

#### 357 NO measurements by DAN (2,3-diaminonaphthalene) method

NO release was quantified using a 2,3-diaminonaphthalene (DAN)-based chemical assay, which is based on the immediate reaction of NO with oxygen (O<sub>2</sub>) to form dinitrogen trioxide (N<sub>2</sub>O<sub>3</sub>), which then reacts with non-fluorescent DAN to form the highly fluorescent 2,3naphthotriazole (NAT) that can be quantified by RP-HPLC. Compounds were incubated at 37°C in 50 mM phosphate buffer, pH 7.4 at 0.1 mM concentration with 0.2 mM DAN in the absence or presence of L-cys at 0.5 mM. At fixed time points, NAT in the reaction mixture was evaluated by HPLC analysis according to previously published protocol <sup>36</sup>.

- HPLC analyses were performed with a HP 1200 chromatograph system (Agilent Technologies) 365 366 equipped with a quaternary pump (G1311A), a membrane degasser (G1322A), a multiple wavelength UV detector (G1365D) and a fluorescence detector (G1321A) integrated in the 367 HP1200 system. Data analysis was performed using a HP ChemStation system (Agilent 368 Technologies). The sample was eluted on a Zorbax Eclipse XDB-C18 column ( $150 \times 4.6$  mm, 369 5  $\mu$ m; Agilent) with an injection volume of 20  $\mu$ L. The mobile phase consisted of 65% of 15 370 mM potassium phosphate buffer (pH 8.0) and 35% acetonitrile at a flow rate of 1.0 mL min<sup>-1</sup>. 371 The fluorescence signals were obtained using an excitation and emission wavelength of 355 372 and 460 nm, respectively (gain factor = 10). Data analysis was performed by with Agilent 373 ChemStation. The values obtained from integration of the peak of NAT were interpolated in a 374 calibration line, prepared using NaNO<sub>2</sub> (in acidic conditions) as a standard. Briefly sodium 375 nitrite standard solutions were acidified with HCl (pH 2) in the presence of excess DAN (0.2 376 mM). After 10 min, the reaction mixture was diluted in phosphate buffer at pH 7.4 (NO final 377 concentration 1 to 80  $\mu$ M) and analyzed by HPLC. 378
- 379

#### **Bacterial strains and growth conditions**

381 P. aeruginosa PAO1 (ATCC BAA-47) was maintained on agar plates of Luria-Bertani medium with 10 g/L NaCl (LB10) (644520, Difco). Cultures were grown overnight in LB10 medium at 382 37°C with 200 rpm shaking (Infors HT, orbit diameter 25 mm). Overnight PAO1 cultures were 383 subsequently diluted 200 times to an  $OD_{600} = 0.005$  in various culture media depending on the 384 assay, made of M9 salts (M9S) (48 mM Na<sub>2</sub>HPO<sub>4</sub>, 22 mM KH<sub>2</sub>PO<sub>4</sub>, 9 mM NaCl, 19 mM 385 NH<sub>4</sub>Cl, 2 mM MgSO<sub>4</sub>, 0.1 mM CaCl<sub>2</sub>, pH 7.0) supplemented with different carbon and 386 nitrogen sources: M9GC (M9S, 0.04 % w/v glucose; 0.2 % w/v casamino acid), M9G (M9S, 387 0.4 % w/v glucose), M9-N (M9G made without any NH<sub>4</sub>Cl), M9GCFe (M9GC, 3 µM FeCl<sub>3</sub>) 388 or M9GCNO<sub>2</sub> (M9GC, 100 µM to 500 µM KNO<sub>2</sub>) medium. 389

#### **390** Growth studies

Overnight cultures of *P. aeruginosa* were washed three times in M9S made without NH<sub>4</sub>Cl and 391 diluted to an OD<sub>600</sub> of 0.005 in 200 µl of M9GC, M9G, M9-N, M9S, M9GCFe, or M9GCNO<sub>2</sub> 392 393 medium, which were added to each well of a 96 well plate and incubated statically at 37°C for 24 h. Furoxans were added at the time of inoculation (t = 0 h). The growth of *P. aeruginosa* 394 395 was monitored spectroscopically at 600 nm, while pyoverdine production was quantified by measurement of fluorescence intensity (excitation at 398 nm and emission at 460 nm)<sup>37</sup>. All 396 measurements were carried out using a microtiter plate reader (Infinite 200 pro, Tecan). Growth 397 rates were determined by calculating changes in OD<sub>600</sub> over time during exponential growth 398 phase while growth yields refer to OD<sub>600</sub> values recorded at late stationary phase 399

#### 400 **Biofilm assays**

Overnight cultures of *P. aeruginosa* were diluted 1:200 in fresh M9 medium. One mL of the 401 diluted culture was added into a well of a 24-well polystyrene plate (142475, Nunclon), which 402 was subsequently incubated at 37°C with 200 rpm shaking (Infors HT, orbit diameter 25 mm) 403 for no more than 6 h. To assess the effect of furoxan on biofilm dispersal, compounds were 404 405 added into each well after 5 h incubation (t = 5 h) to a final concentration of 100  $\mu$ M, 200  $\mu$ M, 500 µM or 3.6 mM, and the plates were incubated for a further 1 h. To assess the effect of each 406 compound on biofilm formation, the furoxans were added into each well at the time of 407 inoculation (t = 0 h) and biofilms were allowed to form over the next 6 h. For experiments 408 involving NO scavengers, c-PTIO was added into each well to a final concentration of 0.5 mM 409 410 at the same time as furoxans. For experiments involving NO-depleted M9GC medium containing LL4212 or LL4216, 500 µM of each furoxan were first added to M9GC medium 411 and incubated at 37°C for 24 h prior to inoculation with *P. aeruginosa*. Biofilm biomass was 412 quantified by crystal violet (CV) staining as described by Barraud et al. (2014) <sup>38</sup>. Planktonic 413 growth and pyoverdine production were quantified as described above. 414

#### 415 *P. aeruginosa* glucose utilization assay

The utilization of glucose by *P. aeruginosa* in M9 medium was measured using the GO assay kit (GAGO20, Sigma). Experiments were carried out as described in the biofilm assays. Subsequently, the medium from each well was filtered and diluted in ultrapure water to obtain a glucose concentration of approximately between 20 to 80  $\mu$ g mL<sup>-1</sup>. Glucose standards were prepared per the manufacturer's instructions. Each volume of the standard or sample was mixed

- 421 with two volumes of the assay reagent and incubated at 37°C statically for 30 min. The reaction
- 422 was stopped by adding two volumes of  $12 \text{ N H}_2\text{SO}_4$ . Glucose concentrations were determined
- 423 with a microtiter plate reader (absorbance at 540 nm) and interpolating to the standard curve.

#### 424 Statistics

All statistical tests were carried out using Graphpad Prism 7.0. Results from biofilm and growth assays were analyzed with one-way ANOVA followed by Dunnett's test for multiple comparison ( $\alpha$ = 0.05) against a relevant control group. Geometric means were used in statistical tests involving pyoverdine measurements and analyzed using ANOVA as described above.

430

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- 439

#### 440 Supporting information

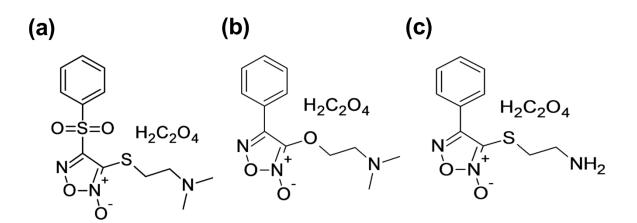
441 *Supporting information available*: This material is available free of charge *via* the internet.

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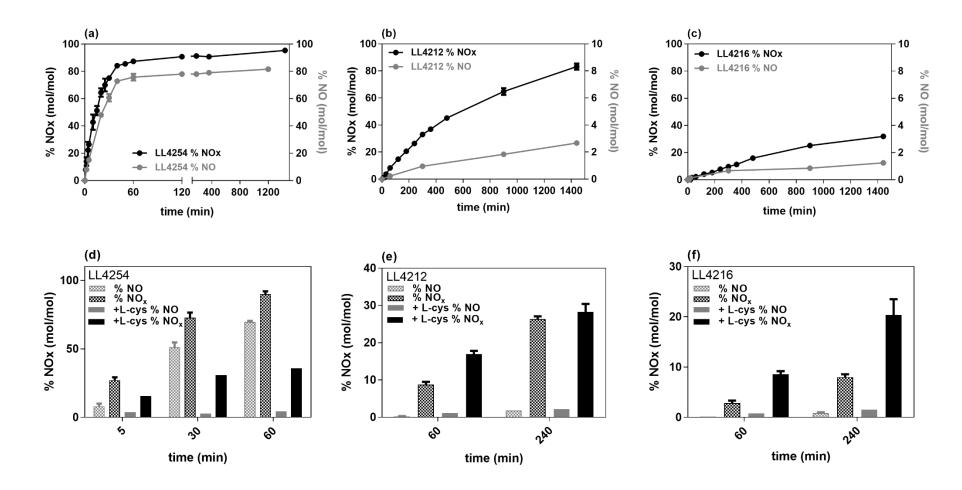
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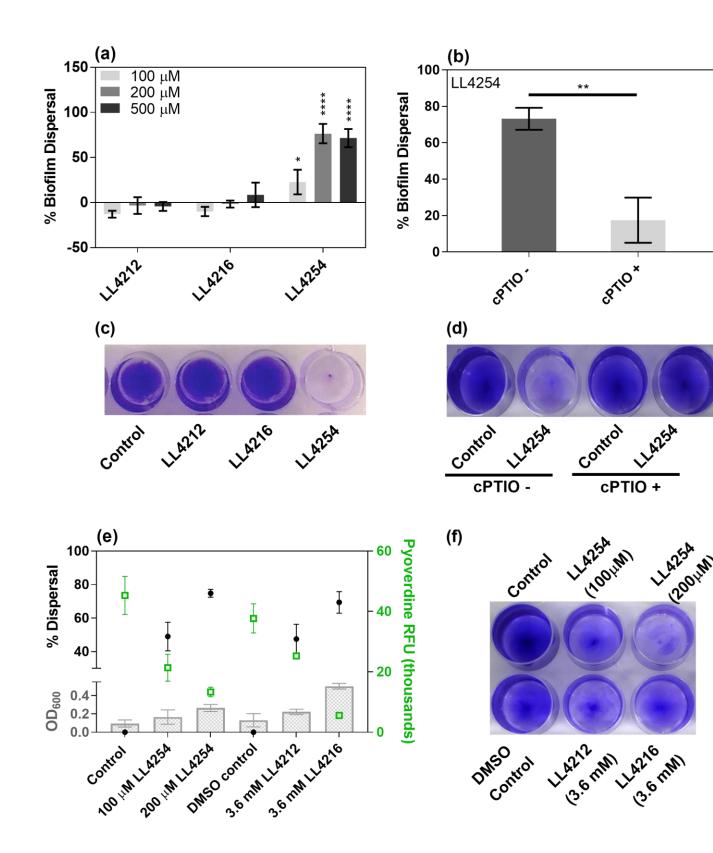
# **Figures**



**Figure 1**. Structures of LL4254 (**a**), LL4212 (**b**) and LL4216 (**c**).

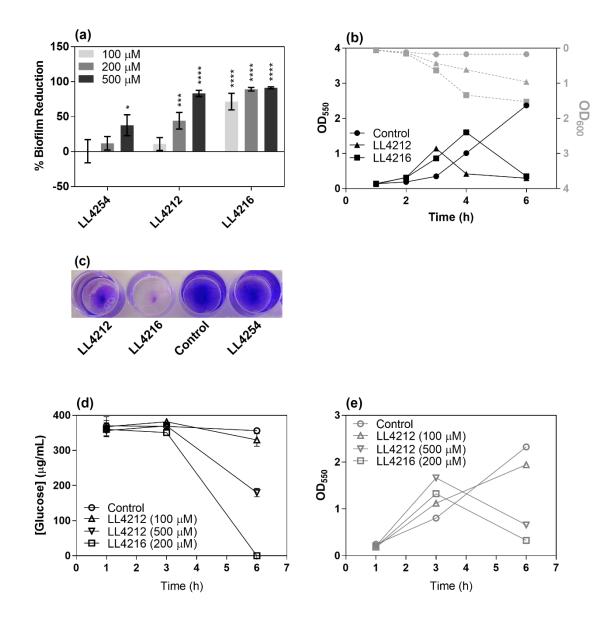


**Figure 2.** NO (gray circle) and NO<sub>x</sub> (NO and NO<sub>2</sub><sup>-</sup>) (black circle) release kinetics of LL4254 (a, d), LL4212 (b, e) and LL4216 (c, f) in phosphate buffer at pH 7.4 in the absence of L-cysteine over time (a - c) and in the presence or absence of L-cysteine at selected time points (d - f). The results are expressed as percent (% mol/mol) of NO or NO<sub>x</sub> released with respect to the quantity of parent furoxan compound. Bars or symbols represent data from three or more replicates and error bars represent standard deviation from the mean.



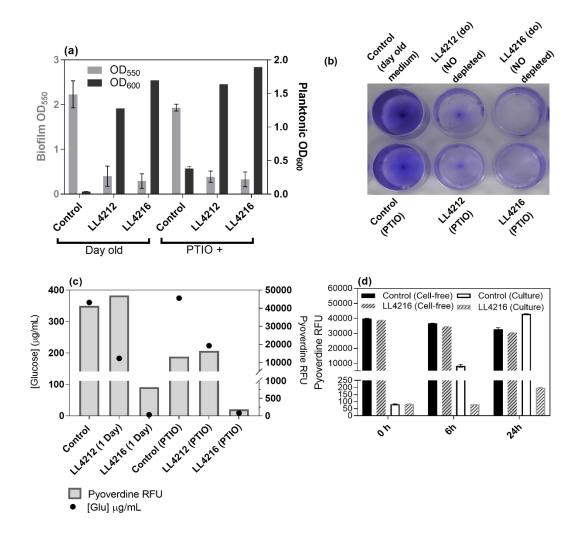
**Figure 3**. Dispersal of *P. aeruginosa* biofilms upon treatment with the furoxans for 1 h (**a**), or upon treatment with 200  $\mu$ M of LL4254 in the presence or absence of 0.5 mM cPTIO (**b**).

The extent of dispersal (black circle) corresponds to an increase in OD<sub>600</sub> (gray bars) and a decrease in pyoverdine fluorescence (green squares) (e). Bars or symbols represent data from three (a, b) or two (e) biological replicates whereas error bars represents the standard deviation from the mean  $-*, p \le 0.05; **, p \le 0.01; ***, p \le 0.001; ****, p \le 0.0001$ . Photographs show CV stains of remaining biofilms following 200 µM of furoxan treatment (c, d) with or without cPTIO (d) or when treated with different concentrations of furoxans (f).

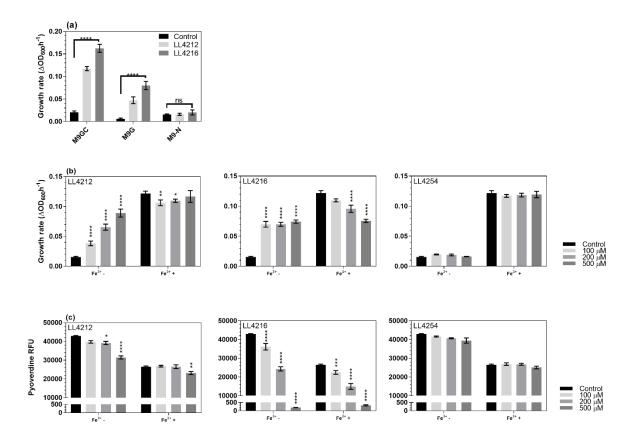


**Figure 4**. Reduction of *P. aeruginosa* biofilms upon treatment with furoxans for 6 h (**a**), with bars representing data from three biological replicates and errors bars representing standard deviation of the mean – \*,  $p \le 0.05$ ; \*\*\*,  $p \le 0.001$ ; \*\*\*\*,  $p \le 0.0001$ . Photographs show CV stains of biofilm remaining after 6 h following 200 µM furoxan treatment (**c**). Biofilm formation and corresponding OD<sub>600</sub> changes over 6 h upon addition of 200 µM of LL4216 and 500 µM LL4212 (**b**). Glucose concentrations were quantified at t = 1 h, 3 h or 6 h after addition of furoxans (**d**), with corresponding changes in CV staining measured (**e**). Symbols

represent data from two biological replicates with error bars showing the standard deviation of the mean in (e), while one representative data set was plotted for (b) and (e).



**Figure 5**. *P. aeruginosa* biofilm reduction upon treatment with 500  $\mu$ M of furoxans in the presence of 0.5 mM cPTIO or NO-depleted furoxans in 1-day-old M9GC medium. Bars of OD<sub>550</sub> measurements represent data from three biological replicates while bars of OD<sub>600</sub> measurements represent data from one experiment. Errors bars represent standard deviation of the mean (**a**). Photograph show CV stains of biofilm remaining after 6 h following furoxan treatment (**b**). Changes in pyoverdine levels and glucose concentrations of samples treated as described in (a), with one representative data set plotted (**c**). Pyoverdine changes over time in cell-free medium or culture in the presence or absence of 500  $\mu$ M LL4216 (**d**).



**Figure 6**. Growth rates in the absence or presence of 500  $\mu$ M of LL4212, LL4216 and LL4254, calculated by changes in OD<sub>600</sub> during the exponential growth phase, in M9 medium supplemented with different carbon or nitrogen sources (**a**). Growth rates (**b**) and pyoverdine fluorescence (**c**) at 24 h of *P. aeruginosa* inoculated statically at 37°C in M9GC (Fe<sup>3+</sup>-) or M9GCFe (Fe<sup>3+</sup>+) medium in the presence of 100 – 500  $\mu$ M of the furoxans. Bars represent data from three biological replicates and errors bars representing standard deviation of the mean. \*,  $p \le 0.05$ ; \*\*,  $p \le 0.01$ ; \*\*\*,  $p \le 0.001$ ; \*\*\*\*,  $p \le 0.001$