# Investigating novel dyes for indoor DSSCs : study of the devices long-term performance upon bufexamic acid pre-adsorption

<u>B. Charrier a, b, c</u>, M. Rubes a, b, D. Civarellia, V. Challuri c, M. Bonomo a, N. Barbero a, d, F. Sauvage c, e, C. Barolo a, d, f

<sup>a</sup> Univ. Torino, NIS Interdept, Dept Chem, I-10125 Turin, Italy, Univ Torino, INSTM Reference Ctr, I-10125 Turin, TO, Italy <sup>b</sup> Dipartimento di Ingegneria dell'informazioneed Elettrica e Matematica Applicata (DIEM), Univ. Salerno, Via Giovanni Paolo II 123, 84084 Fisciano, Italy <sup>c</sup> *G*-lyte SAS, 15 rue de Baudeloque, 80000 Amiens, France

- <sup>d</sup> Ist Sci Tecnol & Sostenibil Sviluppo Materiali Cer, Via Granarolo 64, I-48018 Faenza, RA, Italy
- <sup>e</sup> CNRS of Amiens, 33 Rue Saint-Leu, 80039 Amiens, France
- <sup>f</sup> Univ Torino, Ctr Interdipatimentale Innovaz ICxT, Lungo dora Siena 100, I-10153, Turin, TO, Italy

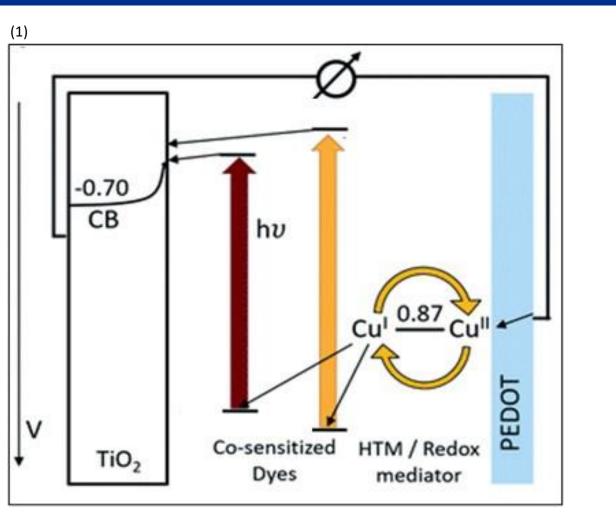
Chimica Divisione di Chimica Organica

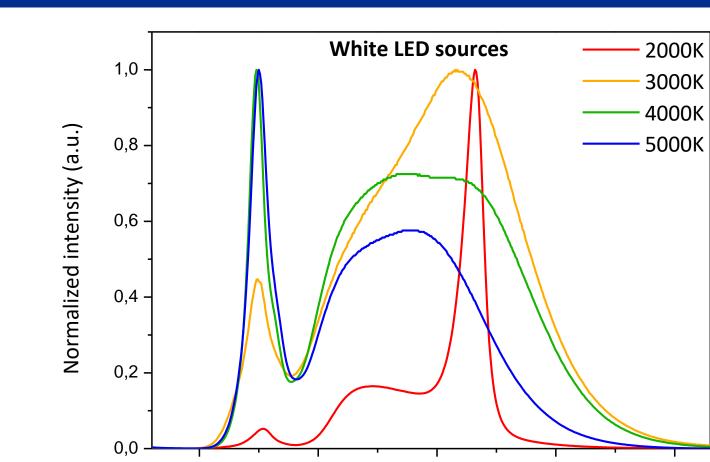
# **Background and approach**

## Introduction

The Internet of Things is growing tremendouly, and the question of powering many small, connected devices has recently arisen.

Unlike other PV technologies, Dye Sensitized Solar Cells (DSSCs) are able to efficiently convert diffuse and weak light, while being industrially produceable. Thus, DSSCs can be designed to convert ambient light and désigne be suitable for individually powering these



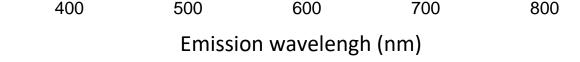


## **Objectives**

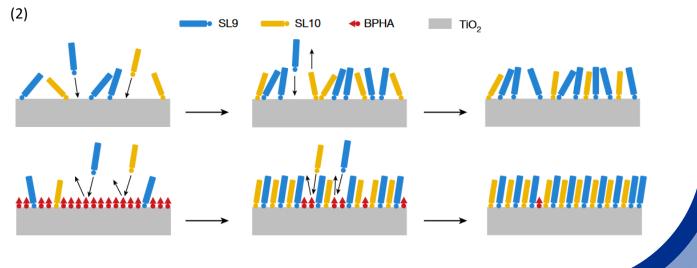
SL9 and SL10 are two recently published dyes<sup>2</sup> that were used in co-sensitized indoor DSSCs. They were integrated in industrially-processed devices and their performance and stability was assessed. On the other hand, the effect of 2-(4-Butoxyphenyl)-N-hydroxyacetamide (BPHA) as a pre-adsorber was investigated.

smart items<sup>1</sup>.

In indoor PV, a white LED light source is generally considered. Hence, the photoactive sensitizers that convert light should be able to efficiently absorb the wavelengths emitted by the LEDs, depending on the source temperature.

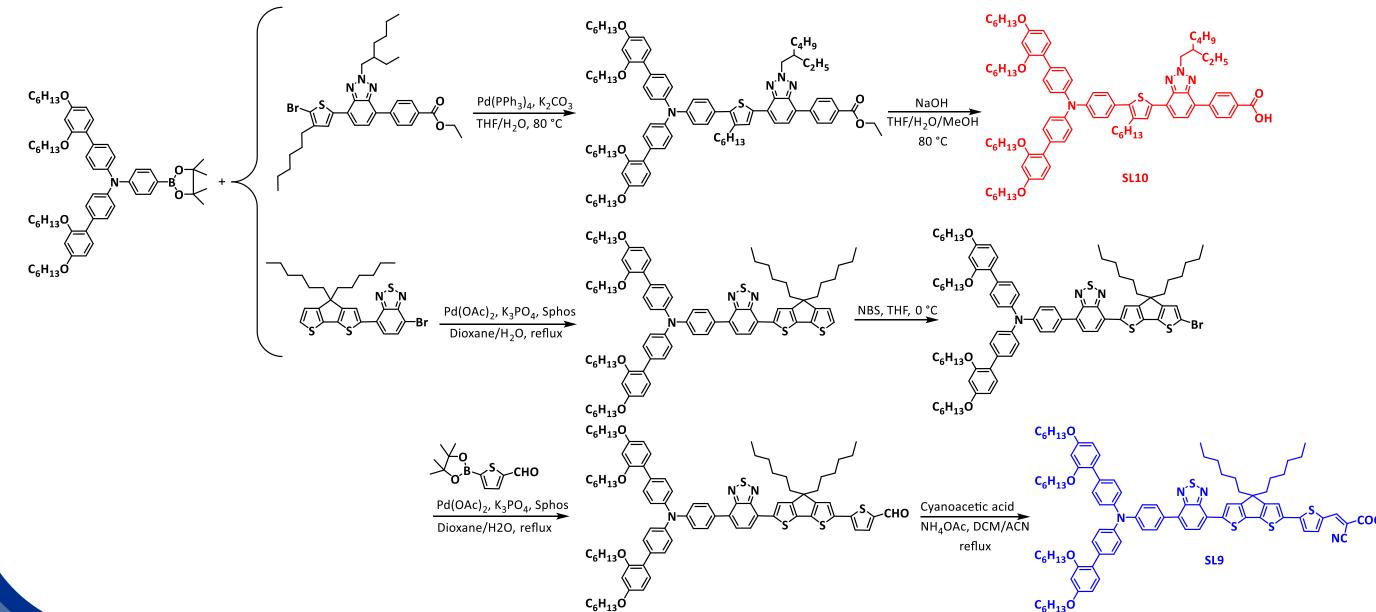


The role of BPHA is to provide better close packing of the sensitizers at the surface of the photoanode, and to potentially prevent dyeelectrolyte electron recombination.<sup>3</sup>



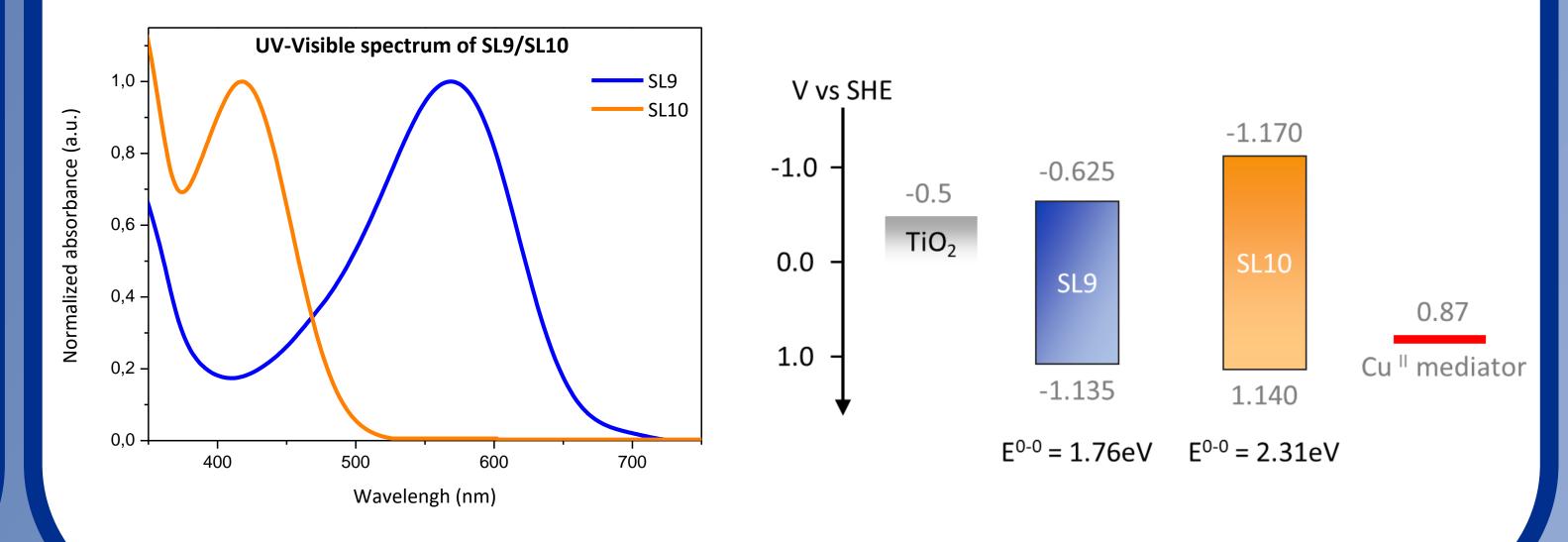
# Synthesis of the sensitizers

SL9 and SL10 were first synthetized over the course of 15 synthetic steps. The so-call push-pull dyes are composed of an electron-rich part, while the second part is composed of several electron withdrawing units, ending with an anchoring acceptor moiety.



Both dyes structures were confirmed by <sup>1</sup>H, <sup>13</sup>C NMR and High Resolution Mass Spectrometry. The optoelectronic properties of SL9 and SL10 were characterized by UV-Visible spectroscopy and Cyclic Voltammetry, to determine their HOMO and LUMO energy levels.

Characterisation



# **Results and conclusions**

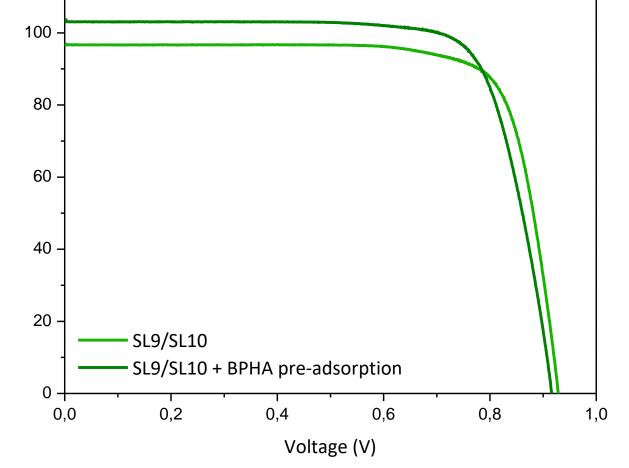
### **Champion cells performance**

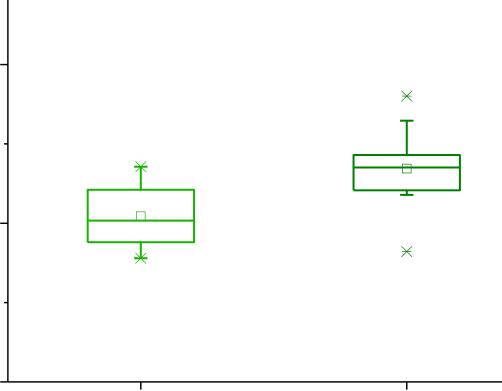
The DSSCs based on SL9 and SL10 were able to reach up to more than **25 %** efficiency under *1000 lux (3000K* white LED light source), achieving performance similar  $\widehat{T}$ to commercial devices. BPHA reliably improved the  $J_{sc}$ ,  $\underline{\mathfrak{F}}$ while no highly detrimental effect was observed.

	V <sub>oc</sub> (V)	J <sub>SC</sub> (μA/cm²)	FF	PCE (%)
SL9/SL10	0,928	96,7	0,784	24,3
SL9/SL10 + BPHA pre-adsorption	0,916	103,1	0,768	25,0

## Sampling

A set of 20 standard and pre-adsorbed SL9/SL10 cells fabricated in order to reliably evaluate the was (%) 20 benefits of BPHA. In average, pre-adsorbing BPHA before sensitization increased the PCE by about 0.7%, while the Fill Factor was also surprisingly enhanced by a few percents.





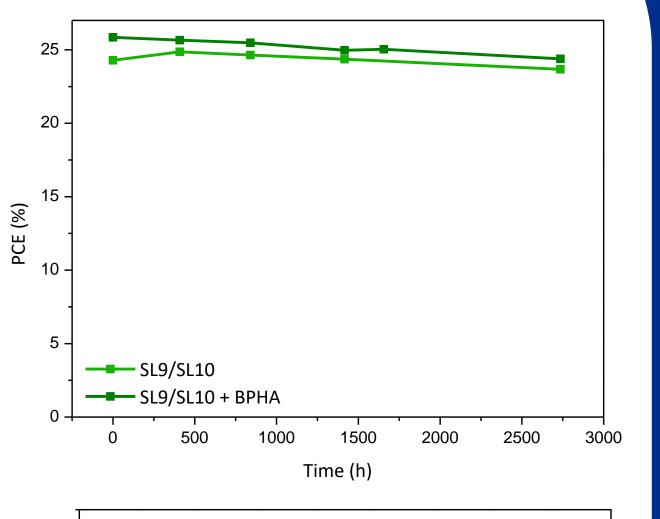
#### **BPHA** pre-adsorption

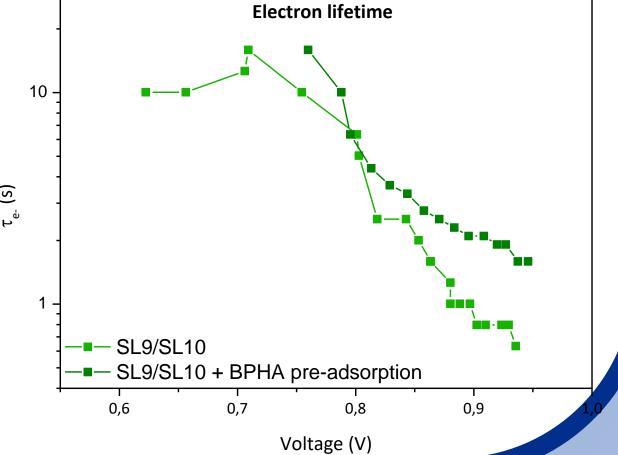
## **Stability**

While kept under ambient conditions, the cells showed good stability with more than **95% of PCE retention** after reaching **3000h**. BPHA did not have any relevent effect on the stability. The same experiments were conducted in harsher conditions (kept @ 60°C, 1sun / 70°C, dark), and **BPHA did not** show any determental effect either.

## **Intensity Modulated Photovoltage** Spectroscopy

In addition to the electron lifetime of the dyes in solution, IMPS was used to evaluate the effect of BPHA on the electron lifetime inside the devices. After 0.8V, the bufexamic acid pre-treatment appears to prevent charge **recombination** at the interface of the photoanodes.



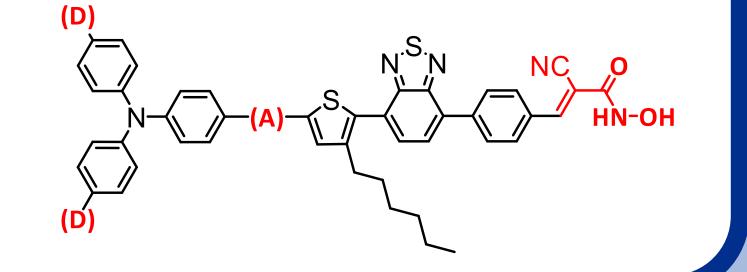


# Prospectives

The synthesis of several SL9-like dyes is currently ongoing. Several strategies are explored to better the spectral response of the dyes and/or to unfavorize charge recombination, namely:

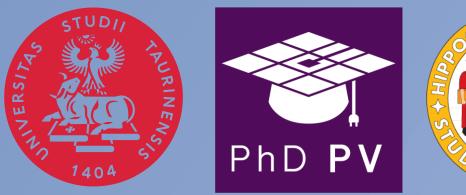
no BPHA

- Modifying the carboxylic acid anchoring acceptor
- Trying alternative triphenylamine-based donor moieties
- Incorporating an extra acceptor group to extend the spectral absorption





References <sup>1</sup> H. Michaels et al., Chem. Sci., **2020**, 11, 2895–2906. <sup>2</sup> Y. Ren *et al., Nature,* **2023**, *613*, 60–65. <sup>3</sup> S. H. Aung *et al., ACS Omega, 2017, 2,* 1820–1825.



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# MOFLES LYTE CITS Università degli Studi di Torino

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