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# Evolution of chemico-physical characteristics during manufacture and ripening of Castelmagno PDO cheese in wintertime.

Marta Bertolino<sup>1</sup>, Paola Dolci, Manuela Giordano, Luca Rolle, Giuseppe Zeppa
Dipartimento di Valorizzazione e Protezione delle Risorse Agroforestali, Settore di Microbiologia Agraria e
Tecnologie Alimentari, Facoltà di Agraria, Università degli Studi di Torino, Via Leonardo da Vinci 44 Grugliasco
(TO), Italy

8

#### 9 Abstract

Biochemical, volatile and textural profiles during manufacture and ripening were determined in 10 samples of Castelmagno PDO cheese obtained from three different batches in the main artisan 11 cheese plant of Castelmagno PDO production area. At the end of manufacture, samples were 12 13 characterized by a pH of 6.57 and 52.4% moisture content. The HPLC analysis of organic acids 14 and sugars showed the exhaustion of lactose content, while the Urea-PAGE indicated extensive primary proteolysis on both  $\beta$ -casein and  $\alpha_{s1}$ -casein. During ripening, cheeses were characterized 15 16 by high degradation of  $\beta$ -casein and  $\alpha_{s1}$ -casein due to bacterial action. RP-HPLC profiles showed a high production of peptides eluted between 20 and 30 minutes. A total of 92 volatile compounds 17 18 were identified in cheese headspace. Texture profiles showed an increase in hardness, gumminess, 19 chewiness and adhesiveness values as well as a decrease in cohesiveness during ripening.

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email address: marta.bertolino@unito.it

<sup>&</sup>lt;sup>1</sup> Corrisponding author: Tel.: +39-0116708705; fax: +39-0116708549

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 evolution, ripening.

3 1. Introduction

Castelmagno PDO cheese is one of the most important Italian hard cheeses and was given the 4 Protected Denomination of Origin label (PDO) in 1996. It takes its name from the homonymous 5 small town in Piedmont (North West Italy) where it was originally produced. Currently, the 6 7 production area is limited to three municipalities (Castelmagno, Pradleves and Monterosso Grana) in the Province of Cuneo in Piedmont. The cheese is produced by six manufacturers (2 industrial 8 9 and 4 artisanal dairy plants). It is usually made from raw cow milk obtained from two consecutive 10 milkings. The evening milk may be partially skimmed after overnight creaming at 15°C in 11 shallow and large diameter tanks. Such semi-skimmed milk is mixed at a 1:1 ratio with the whole milk collected during the successive morning milking. A small percentage of ewe or goat's milk 12 13 may be added to cow's milk, although such practice is not currently in use. Production technology does not allow the use of starter cultures, so acidification is due to indigenous lactic acid bacteria 14 15 and milk is coagulated with liquid calf rennet at 32-38°C. The curd is transferred to molds and 16 harvested for at least 18 h for complete whey elimination. Then the curd is left at 10°C for a period of 2-4 days under the whey obtained from previous cheesemaking. The curd is then milled, 17 18 dry-salted and strongly pressed. Finally, the cheese is placed in natural caves where ripening takes place at 10-12°C and 85-90% humidity for at least 60 days. The cheese has a cylindrical shape, 19 measuring 12-20 cm high and 15-25 cm in diameter, and weighing 2-7 kg. Penicillium spp. from 20 21 the environment occasionally colonizes the interior part of the cheese during the final phase of ripening. Due to the presence of this colonization, the Castelmagno PDO cheese is usually 22 23 considered a hard blue cheese variety (Ottogalli, 2001; Gobbetti & Di Cagno, 2002; Gobbetti, 24 2004) but nowadays the cheese is marketed before the appearance of mould.

Although a little is known about the microbiology of Castelmagno PDO cheese (Dolci, 1 2 Alessandria, Rantsiou, Rolle, Zeppa & Cocolin, 2008; Dolci, Alessandria, Rantsiou, Bertolino & 3 Cocolin, 2010) there are no studies on the technology, gross composition, glycolysis, proteolysis, lipolysis, volatile and textural profiles of this cheese. Therefore, the aim of this research was to 4 determine the biochemical, volatile and textural profile of Castelmagno PDO cheese. Since its 5 production process consists of 4-5 days, it influences the biochemical pathways that determine the 6 7 final characteristics of the cheese. As a consequence, it was also necessary to analyze the samples during Castelmagno PDO cheese manufacture and not only during ripening. 8

9

#### 10 2. Materials and methods

11 2.1. Materials

Samples were taken from three batches of Castelmagno PDO cheese produced in the main 12 13 artisanal dairy plant in the town of Castelmagno (Piedmont, Italy) during the wintertime. The three batches were produced on different consecutive days (A, B, C) by using milk from the same 14 15 farm. The cow milk used in cheese production had a pH of  $6.59 \pm 0.01$ , and contained  $4.51 \pm$ 16 0.17% lactose,  $3.45 \pm 0.07\%$  protein and  $3.40 \pm 0.32\%$  fat. For each batch the milk (A1, B1, C1), the curd after the cut (A2, B2, C2), the curd after 24 hours (A3, B3, C3), the curd after 3 17 18 days under the whey (A4, B4, C4), and the cheese after 3 (A5, B5, C5), 30 (A6, B6, C6), 60 (A7, 19 B7, C7), 90 (A8, B8, C8), 150 (A9, B9, C9) days of ripening were sampled. Samples were 20 transferred to the laboratory in refrigerated conditions and milk was immediately analysed for pH 21 and gross composition. Cheesemaking samples (A2, B2, C2, A3, B3, C3, A4, B4 and C4) were 22 analysed immediately for pH and an aliquot was also frozen and subsequently used for 23 compositional, glycolysis, proteolysis and volatile analysis. Cheese ripening samples (from A3 to C9) were immediately analysed for pH and texture profile and an aliquot was also frozen and
 subsequently used for compositional, glycolysis, proteolysis and volatile analysis.

3

4 *2.2. Methods* 

5

6 2.2.1. Compositional analysis

Milk samples were analysed for lactose, protein and fat content by using a Milko Scan<sup>TN</sup> FT 120
(Foss, Padova, Italy). Cheese samples during manufacturing and ripening were analysed for:
moisture by the oven drying method at 102°C (IDF, 1982), salt by titration with AgNO<sub>3</sub> (IDF, 1988), total protein and pH 4.6-soluble nitrogen by Kjeldhal method (IDF, 1993), and fat by the
FIL-IDF Standard 5A method (1969). The pH was determined with a Portamess 913 pHmeter
(Knick, Berlin, Germany) placing the penetration electrode in contact with the sample mass. All
analyses were performed in triplicate.

14

15 2.2.2. Assessment of proteolysis

16 The pH 4.6-insoluble and -soluble extracts were prepared according to the method of Kuchroo 17 and Fox (1982), which was slightly modified as outlined by Hayaloglu, Guven, Fox, Hannon and 18 McSweeney (2004). Urea-polyacrylamide gel electrophoresis (Urea-PAGE) was performed on 19 the insoluble fraction using a Protean II xi vertical slab-gel unit (Bio-Rad Laboratories Ltd., Watford, UK) according to the method of Shalabi and Fox (1987). The gels were stained directly 20 21 with Coomassie Brilliant Blu G-250 using the method of Blakesley and Boezi (1977) and 22 destained using distilled water. After destaining, gel slabs were digitised by a scanner (Epson 23 Perfection 1650, Seiko Epson Corporation, Nagano, Japan ). Scans of the electrophoretograms 24 were used to quantify bands using densitometric software (Image Master TotalLab 1D Gel analysis v 1.11 software, Nonlinear Dynamics Ltd, Newcastle-upon-Tyne, UK). Similar bands
 were recognized visually as described by McSweeney, Poochet, Fox and Healy (2004) and peak
 volumes of corresponding bands were quantitatively determined.

Peptides of the pH 4.6-soluble fraction of cheeses were determined by RP-HPLC using the
method described by Hayaloglu *et al.* (2004) utilising a HPLC system (Thermo Electron
Corporation, Waltham, MA, USA) equipped with a isocratic pump (P1000), and a multiple
autosampler (AS3000) fitted with a 20 µL loop a UV detector (UV100) set at 214 nm.

8 Individual free amino acids (FAA) of the pH 4.6-soluble fractions of cheeses were prepared and
9 analysed according to the method of Bertolino, Zeppa, Gerbi and McSweeney (2008).

10

#### 11 2.2.3. Assessment of organic acid, sugars, diacetyl and acetoin

12 Organic acids (citric, orotic, pyruvic, lactic, oxalic, hippuric, isobutyric, valeric and isovaleric), sugars (lactose, glucose and galactose), diacetyl and acetoin were determined by high 13 14 performance liquid chromatography according to the method of Zeppa and Rolle (2008). Five grams of sample were added to 25 mL of 0.013N H<sub>2</sub>SO<sub>4</sub> (mobile phase) and homogenised for 10 15 min with a Stomacher blender (PBI, Milano, Italy). The extract was subsequently centrifuged for 16 17 5 min at 2500 g and the supernatant was filtered through a PTFE 0.20  $\mu$ m disposable syringe membrane filter (Sartorius AG, Göttingen, Germany). The HPLC system (Thermo Electron 18 Corporation, Waltham, MA, USA) was equipped with an isocratic pump (P1000), a multiple 19 autosampler (AS3000) fitted with a 20 µL loop, a UV detector (UV100) set at 210 and 290 nm 20 21 and a Refractive Index detector (RI-150). The analyses were performed isocratically at 0.8 22 mL/min and 65°C with a  $300 \times 7.8$  mm i.d. cation exchange column (Aminex HPX-87H) 23 equipped with a Cation H<sup>+</sup> Microguard cartridge (Bio-Rad Laboratories, Hercules, CA, USA). Three replicates for each sample were analysed. The data treatments were carried out using the 24

ChromQuestTM chromatography data system (ThermoQuest, Inc., San Jose, CA, USA).
 Analytical grade reagents were used as standards (Sigma-Aldrich Corporation, Milan, Italy).

3

#### 4 2.2.4. Volatile compounds analysis

Grated homogenized sample (5g) was placed in a 40 mL vial fitted with a PTFE silicone septa 5 (Supelco, Bellefonte, PA, USA), through which the SPME syringe needle fitted with a Stable 6 7 Flex 2cm-50/30 µm divinylbenzene-carboxen-polydimethylsiloxane (DVB-CAR-PDMS) fiber (Supelco, Bellefonte, PA, USA) was introduced. The internal standard was methyl nonaoate 8 (Sigma Aldrich) at a final concentration of 80.4 µg/kg in the sample (Katechaki, Panas, Rapti, 9 10 Kandilogiannalis & Koutinas, 2008). The vial was placed in a heat/stir plate at 80°C for 35 min for the absorption phase. After exposure in the headspace (HS), the fibre with the analytes was 11 12 retracted and transferred to the injector, which was operated in the splitless mode at a temperature 13 of 280°C for 4 min. Compound identification was achieved with a Shimadzu GC-17A gas 14 chromatograph (GC) coupled with a Shimazdu QP5000 quadrupole mass spectrometer (Shimadzu Corporation, Kyoto, Japan). The GC was equipped with a DB-wax column (30 m, 0.25 mm i.d., 15 and 0.25 µm film thickness) and a split/splitless injector. The carrier gas was ultrahigh purity 16 17 (99.999%) helium with a flow rate of 1 mL/min. The following column temperature programming sequence was used: an initial temperature of 35°C for 3 min, increased to 110°C at a rate of 18 19 5°C/min, increased to 240°C at a rate of 10°C/min and a final extension at 240°C for 10 min. 20 Mass spectra were recovered in the electron impact mode at an ionisation voltage of 70 eV. The 21 ion source and the interface were maintained at 250°C. Identification was achieved by comparison to standard compounds where available, or/and using the NIST 12 and the NIST 62 22 23 data base (National Institute of Standards and Technology, Gaithersburg MD, USA).

#### 1 2.2.5. Texture analysis

2 For Texture Analysis, samples were cut with a thin blade in 20 mm squared cubes and immediately analyzed. The TPA test was carried out using a Universal Testing Machines (UTM) 3 TA-XT2i Texture Analyzer<sup>®</sup> (Stable Micro System, UK) equipped with a 25 kg loadcell and 4 HDP/90 platform. Samples were compacted in height to 30% of the original using a crosshead 5 speed of 0.8 mm/s and a P-35 DIA cylinder stainless flat probe (Kapoor, Metzger, Biswas & 6 Muthukummarappan, 2006; Blazquez et al., 2006). Each sample was subjected to a two-cycle 7 compression with 5 s between cycles (Drake, Gerard, Truong & Daubert, 1999). For the 8 9 acquisition of the force-time curve, a Texture Export Exceed software rel. 2.54 (Stable Micro 10 Systems, Godalming, UK) was used. According to Gunesakaran and Mehemet Ak (2003), the 11 following parameters were measured from the force-time curves (Figure 1): hardness (N, as F<sub>1</sub> maximum force), cohesiveness (adimensional, as  $(A_2)/(A_1+A_{1W})$ ), adhesiveness (mJ, as  $A_3$ ), 12 13 gumminess (N, as hardness  $\times$  cohesiveness), springiness (mm, as d<sub>2</sub>), chewiness (mJ, as gumminess  $\times$  springiness) and resilience (adimensional, as  $(A_{1w}/A_1)$ ). For each batch and point of 14 15 ripening, five analyses were performed.

16

#### 17 2.2.6. Statistical analysis

The distribution and the differences in the compositional parameter, organic acids, sugars, diacetyl, acetoin, free amino acids, and textural parameters of Castelmagno PDO samples were analysed using Brown-Forsythe test of homogeneity of variance, ANOVA and the Duncan mean comparison test respectively to underline the normal distribution of the data and differences during the manufacture and ripening of cheeses. Calculation was performed by Statistica 7.0 Software (Statsoft, Tulsa, USA).

#### 2 **3. Results and discussion**

#### 3 *3.1. Compositional analysis*

The average of pH, moisture, salt, fat, protein and pH 4.6-soluble nitrogen contents of 4 Castelmagno PDO samples during manufacturing and after 3, 30, 60, 90, 150 days of ripening are 5 shown in Table 1. The pH of cheeses was between 6.57 to 4.71 during manufacturing and 6 7 between 4.71 to 5.02 during ripening due to the microbial ecosystem evolution as reported by Dolci, Alessandria, Rantsiou, Bertolino & Cocolin, (2010). The pH average value during 8 9 Castelmagno PDO market life (after 60 days of ripening), was 4.94 lower than that reported by 10 Gobbetti and Di Cagno (2002). According to moisture data after 60 days of ripening (period after 11 which the cheese can be sold), Castelmagno PDO cheese can be categorized as a hard cheese with an average value of 35.5% (McSweeney, Ottogalli & Fox, 2004). After 3 days under the whey the 12 13 curd is grinded, pressed and formed with a high loss of whey. As a consequence, samples at 3 14 days of ripening showed a high decrease in moisture content.

Low salt levels found during Castelmagno PDO manufacturing (0.75%) was due to the fact that it is a dry-salted cheese and salt is added during the curd grinding at the end of cheesemaking. During ripening, salt levels increased to an average value of 2.46%, which was in line with data reported by Delforno (1960).

Fat content increased from an average value of 23.2% during the manufacture period to an
average of 31.1% after 60 days. The second one was lower than that reported by Gobbetti (2004)
and Merlo (2001) but within the range reported by Delforno (1960).

Protein content increased from an average value of 19.31 % during manufacture to an average of
26.56 % after 60 days, whilst the pH 4.6-soluble nitrogen rose from 7.13% to 13.65% as a

consequence of the decrease of moisture. Protein content during market life was higher than that
 reported by Gobbetti (2004).

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- 4

#### 5 *3.2. Assessment of proteolysis*

The data of pH 4.6-SN level in Castelmagno PDO samples are shown in Table 1. During cheese manufacturing pH 4.6-SN decreased as a consequence of its use as a nutritional requirement by LAB (Monnet, Condon, Cogan & Gripon., 1996) and due to its diffusion in whey as a consequence of the attainment of equilibrium of soluble constituents into two solutions as reported for the cheese ripening in brine (Abd El Salam and Alichanidis, 2004). Contrarily, pH 4.6-SN increased during the ripening period due to the breakdown of casein into peptides and amino acids by the action of chymosin, plasmin, and bacteria.

13 Urea-PAGE electrophoretograms of the pH 4.6insoluble fractions of Castelmagno PDO cheese (batch A) during manufacture and ripening are shown in Figure 2. Bands in the 14 15 electrophoretograms and the densitometric analysis (data not reported) showed that the 16 degradation of  $\beta$ - and  $\alpha_{s1}$ -case occurs early, already during cheese manufacturing due to the rennet and plasmin activity effects. However it can be seen that neither  $\beta$ - nor  $\alpha_{s1}$ -casein were 17 totally degraded at the end of ripening but  $\alpha_{s1}$ -case hydrolysis rate was greater than that of 18 19  $\beta$ -casein during all stages of ripening. As a consequence of  $\alpha_{s1}$ -casein hydrolysis, in all electrophoretrogram samples the band corresponding to  $\alpha_{s1}$ -I-casein ( $\alpha_{s1}$ -CN f 24-199) which is 20 21 the first product of rennet action on  $\alpha_{s1}$ -casein, was present; from the third day of ripening other bands corresponding to other peptides appeared (marked as z  $\alpha_{s1}$ -CN ) these are characterised to 22 have faster mobility than  $\alpha_{s1}$ -I-casein ( $\alpha_{s1}$ -CN f 24-199) which are degradation products of  $\alpha_{s1}$ -23

casein due to rennet and indigenous milk proteinases action. From the thirtieth day of ripening, 1 cheeses showed a band corresponding to the peptide  $\alpha_{s1}$ -I-casein ( $\alpha_{s1}$ -CN f 102-199). Concerning 2  $\gamma$ -case ins (the polypeptides produced by the action of plasmin on  $\beta$ -case ins), the  $\gamma_2$ -case in [ $\beta$ -3 case in (f106-209)] was present at the highest concentration followed by  $\gamma_3$ - [ $\beta$ -case in (f108-209)] 4 and  $\gamma_{1-}$  [ $\beta$ -casein (f29-209)] caseins. The RP-HPLC profiles of the pH 4.6-soluble fractions of 5 Castelmagno PDO cheese (batch A) are shown in Figure 3. To compare the chromatographic data 6 obtained by RP-HPLCs, visual identification of similar peaks were evaluated. Common peaks 7 were evident in the region 5-8 min in the chromatograms of all with an increase in concentration 8 9 during ripening. Similar peptides eluted with retention times of 11-18 and 24-29 mins were 10 observed in samples 24 hours after manufacturing until the 150th day of ripening, with increased 11 concentration during ripening. All these regions were composed principally of amino acids and hydrophilic peptides (Gonzalez del Llano, Polo & Ramos, 1995; Pavia, Trujillo, Guamis & 12 13 Ferragut, 2000). However, qualitative and quantitative differences were observed in the region 14 included between 30 and 50 minutes considered to be composed mainly of hydrophobic peptides (Gonzalez del Llano et al., 1995; Pavia et al., 2000). Cheese proteolysis was also monitored by 15 16 determining the levels of individual free amino acids (FAAs). These data are reported in Table 2. 17 Overall, the total concentration of FAAs increased considerably from the end of manufacture 18  $(14.78 \pm 4.89 \text{ mg/g of cheese})$  to the end of ripening  $98.56 \pm 14.13 \text{ mg/g of cheese}$ . Glutamic acid, valine, leucine, phenylalanine and lysine were the FAAs characterised by the highest 19 concentration during cheese manufacturing whilst the most common FAAs during ripening were 20 21 aspartic acid, glutamic acid, valine, leucine and phenialanine. Most of these amino acids were 22 previously found to be present at high concentration in several hard or extra-hard Italian cheese 23 varieties (Resmini, Pellegrino, Hogenboom & Bertuccioli, 1988; Albenzio et al., 2001; Gobbetti, 24 2004).

#### 2 3.3. Assessment of organic acids, sugars, diacetyl and acetoin composition

Organic acids, sugars, diacetyl and acetoin concentrations of Castelmagno PDO samples are
reported in Table 3.

Lactose metabolism was totally complete at the end of manufacture but already after just 24 hours 5 most of the lactose had been converted into lactate by the growth of starter bacteria or by its loss 6 into whey as reported by McSweeney (2004). Glucose and galactose were also present at very 7 low concentration  $(0.01 \pm 0.01 \text{ mg/g} \text{ of cheese and } 0.11 \pm 0.02 \text{ mg/g} \text{ of cheese respectively})$  and 8 9 they were already absent in curd after 3 days under the whey samples and 3-day-old cheeses 10 respectively due to their use by lactic acid bacteria (LAB) and non starter lactic acid bacteria 11 (NSLAB) as substrate of growth (Michel & Martley, 2001). During ripening, lactic acid was the main organic acid in all samples, representing approximately 95% of the total organic acid 12 13 content in 3-day-old cheeses and 78% of total organic acid content in 150-day-old cheeses. The mean lactic acid concentration during Castelmagno market life was similar to that observed for 14 15 Cheddar and Colby cheeses (Mullin & Emmons, 1997) but higher than that already reported for Castelmagno PDO cheese (Dolci et al, 2008; Zeppa & Rolle, 2008). Citric acid was present with 16 the highest concentration  $(1.32 \pm 0.11 \text{ mg/g of cheese})$  in 24hour-old cheeses; it then decreased to 17  $0.03 \pm 0.02$  mg/g of cheese in 3-day-old cheeses due to its metabolization by Cit<sup>+</sup> strains of LAB 18 19 or NSLABS into acetate, acetoin and diacetyl (McSweeney & Fox, 2004). In particular, all 20 Castelmagno PDO samples demonstrated a higher concentration of diacetyl than acetoin, which 21 can also be derived from the metabolism of pyruvate by NSLAB. The acetic acid concentration 22 increased during manufacture to a final level of  $0.81 \pm 0.08$  mg/g of cheese in samples at 3 days 23 under the whey; it then decreased during the ripening phase. In 150-day-old cheeses it was found 24 at a concentration of  $0.50 \pm 0.03$  mg/g of cheese. Acetate is produced from lactose, lactic acid or

1 citric acid metabolisms or from the catabolism of amino acids. Many authors have reported that 2 its concentration in different PDO cheeses such as Cheddar, Camembert, Beaufourt, Canestrato Pugliese, Murazzano, Raschera, Robiola di Roccaverano and Toma Piemontese ranged from 0.18 3 to 1.89 mg/g of cheese. (Bouzas et al., 1991; Mullin & Emmons, 1997; Faccia, Gambacorta, 4 Lamacchia & Luccia, 2004; Zeppa & Rolle, 2008). The propionic acid concentration increased 5 from manufacture to the end of ripening where it was found to be  $1.43 \pm 0.14$  mg/g of cheese, 6 representing 7.3 % of total organic acid content. The propionic acid is produced from lactic acid 7 8 metabolism by *Propionobacterium* spp. as reported by McSweeney (2004) or from the lypolitic 9 activities of starter and secondary microflora as reported by Collins, McSweeney and Wilkinson 10 (2004). Iso-butyric acid was detected only in samples after 90 days of ripening with a mean 11 concentration of  $0.68 \pm 0.18$  mg/g of cheese. Iso-valeric acid concentration increased during Castelmagno PDO production and in the 150-day-old cheeses it was detected at a concentration of 12 13  $0.69 \pm 0.28$  mg/g of cheese, representing 3.5 % of total organic acid content.

14

#### 15 *3.4. Volatile compound analysis*

Volatile compounds identified in Castelmagno PDO samples by HS-SPME-GC/MS during 16 manufacturing and ripening are shown in Table 4. A total of 92 compounds were detected: 15 17 18 acids, 28 esters, 13 ketones, 12 aldehydes, 13 alcohols, 3 lactones, 3 hydrocarbons, and 6 19 compounds which could not be classified in these chemical groups. Acids constituted the main 20 chemical class during manufacturing with a mean concentration of 88.61% w/w of total volatile 21 compounds and during ripening (77.92% w/w of total volatile compounds concentration). Acids 22 can originate from three biochemical pathways: lipolysis, proteolysis and glycolysis (Curioni & 23 Bosset, 2002). During manufacturing the most abundant acids were acetic, decanoic, dodecanoic, 24 hexanoic and octanoic acids. The acetic acid increased its concentration from the beginning to the

1 end of manufacture and could have a microbial origin as a product of lactose fermentation due to 2 the growth of lactic and propionic bacteria (McSweeney & Fox, 2004), which are abundant in this 3 cheese as shown by microbiological data (Dolci, Alessandria, Rantsiou, Bertolino & Cocolin, 2010). The other acids were derived from the action of esterases and lipases present in raw milk 4 used for Castelmagno PDO cheesemaking. During cheese ripening, the highest acid 5 concentrations were found for acetic, butyric, decanoic, docecanoic, hexanoic and octanoic acids. 6 7 Esters are important common constituents of the volatile fraction of cheese. Different esters have been reported, such as methyl, ethyl, prophyl and butyl esters as a reaction of free fatty acid with 8 9 ethanol, methanol, propanol and butanol in different cheese varieties (Liu, Holland & Crow, 10 2004). Esters formation is correlated to the growth of lactic acid bacteria and Micrococcaceae 11 (Gripon, Monnet, Lambert & Desmazeaud 1991). In Castelmagno PDO samples, esters concentration represented 1.41% of the total volatile compounds concentration during 12 13 manufacturing and 5.64% during cheese ripening. Ethyl esters were the predominant esters in analysed samples due to the high concentration of ethanol arising from lactose fermentation or 14 15 amino acid catabolism. Among esters during Castelmagno PDO manufacturing, ethyl hexanoate, 16 ethyl octanoate and ethyl decanoate concentrations represented 72.29% w/w of total ester concentration and ethyl butanoate and ethyl acetate represented 8% w/w of total ester 17 18 concentration. The concentration of all esters identified in Castelmagno PDO samples increased 19 during ripening and ethyl hexanoate, octanoate and decanoate represented 74% w/w of total esters 20 concentration during this time. The increase of these esters could be associated to the decrease in 21 corresponding acids. Ethyl hexanoate was also identified as the most abundant ester in other PDO cheeses such as Grana Padano (Moio & Addeo, 1998), Parmigiano Reggiano (Bellesia et al., 22 23 2003) and Pecorino Romano (Di Cagno et al., 2003).

1 Ketones were the second most abundant compounds isolated in Castelmagno PDO samples with a 2 mean percentage of 2.04% (w/w of total volatile compounds concentration) during manufacturing 3 and 8.25% (w/w of total volatile compounds concentration) during ripening. They are formed by 4 enzymatic oxidation of free fatty acids to β-ketoacids and their consequent decarboxilation to 5 ketones. They are very important compounds for dairy products because they have very particular 6 odours and low perception thresholds (McSweeney & Sousa, 2000; McSweeney, 2004). A total 7 of 12 ketones were identified in Castelmagno PDO samples - 2-butanone, 2-pentanone, 2-8 heptanone were the most abundant. Acetoin originates from citrate metabolism as a reduction of 9 diacetyl by the action of lactic acid bacteria (McSweeney & Fox, 2004). The highest 10 concentration of acetoin was detected during cheesemaking at the cut of the curd ( $44.37 \pm 75.69$ 11 µg/kg of cheese);its concentration then decreased until the 90th day of ripening. In 150-day-old 12 cheeses, its concentration was  $3.25 \pm 2.59 \ \mu g/kg$  of cheese and this decrease could be due to its 13 reduction to butanone as reported by Urbach (1993).

14 Aldehydes were present with the highest concentration (4.50% w/w of total volatile compounds 15 concentration) at the end of manufacture of Castelmagno PDO cheeses. This concentration 16 decreased to a mean value of 0.18% w/w of total volatile compounds concentration during 17 ripening because they were rapidly converted to the corresponding alcohols or acids (Lemieux & 18 Simard, 1992). During manufacturing, hexanal, heptanal and 2-nonenal were the aldehydes with 19 the highest concentration and represented 37%, 15% and 30% (w/w) respectively of the total 20 aldehydes concentration of the curd after 3 days under whey. During ripening, the aldehydes with 21 the highest concentration were acetaldehyde, trans 2-hexenal and hexanal with a concentration that represented 45%, 17% and 13% w/w of total aldehydes concentration. Acetaldehyde, which 22 23 represented nearly half the concentration of total aldehydes during ripening, could derive from the breakdown of threonine, from the lactose metabolism, or by the oxidation of ethanol (McSweeney
 & Sousa, 2000).

Alcohols were abundant during Castelmagno PDO manufacturing with a mean percentage of 3 3.56% w/w of total volatiles detected, whilst these levels increased during ripening to a mean 4 percentage of 6.18% (w/w of total volatile compounds concentration). Ethanol was the most 5 6 abundant. It is a product of lactose fermentation or amino acid catabolism and it is the alcohol that 7 contributes to the formation of ethyl esters. Primary alcohols are produced by the reduction of aldehydes derived by the catabolism of the amino acids (Moio & Addeo, 1998) and were present, 8 during manufacture, to a final concentration of 7.11  $\pm$  1.31 µg/kg of cheese and a final 9 concentration at the end of ripening of  $56.87 \pm 15.69 \,\mu\text{g/kg}$  of cheese. Instead, secondary alcohols 10 formed by enzymatic reduction from the corresponding methyl ketones which are produced from 11 12 fatty acids (Collins et al., 2004) were not detected during manufacture but only during ripening. 13 The branched-chain alcohols detected (2-methyl-1-butanol and 3-methyl-1-butanol) derived from the reduction of aldehydes produced from the catabolism of isoleucine and leucine respectively 14 (Yvon & Rijnen, 2001), and were present at a final concentration of 14.79  $\pm$  6.68 and 18.62  $\pm$ 15 9.68 µg/kg of cheese at the end of manufacture and ripening respectively. 16

Among the lactones family, three  $\delta$ -lactones ( $\delta$ -octalactone,  $\delta$ -decalactone and  $\delta$ -dodecalactone) 17 18 were detected. Lactones are cyclic compounds formed by the intramolecular esterification of 19 hydroxyl fatty acids through the loss of water (Molimard & Spinnler, 1996). Lactones were present with a mean percentage of 0.82% w/w of total volatile compounds concentration during 20 21 manufacture of cheese and a mean percentage of 1.33% w/w of total volatile compounds 22 concentration during ripening. Lactones represented only a very small portion (c.a. 0.1% of total 23 volatile compounds concentration) in other cow's milk PDO cheeses like Grana Padano (Moio & 24 Addeo, 1998) but a higher portion in other ewes' milk PDO cheeses like Canestrato Pugliese (c.a 7.8% w/w of total volatile compounds concentration) or Fiore Sardo (c.a. 8.1% of total volatile
 compounds concentration) or Pecorino Romano PDO (c.a. 9.2% w/w of total volatile compounds
 concentration) (Di Cagno *et al.*, 2003).

4

#### 5 *3.5. Textural analysis*

6 The mean values obtained for texture parameters of TPA obtained during ripening are shown in7 Table 5.

8 Texture profile analysis presented changes for all variables during ripening, in particular in the 9 first 30 days and between 90 and 150 days. In general, hardness, adhesiveness gumminess and 10 chewiness values increased up to 3 days while an inverse behaviour was identified for 11 cohesiveness and resilience parameters.

Hardness, the force necessary to attain a given deformation (Szczesniak, 2002) increased 12 13 particularly at the beginning of ripening (first 30 days), was maintained for up to 90 days of ripening and increased significantly in the last 60 days (about 18 N, from  $25.88 \pm 4.13$  to  $43.15 \pm$ 14 15 5.50). Similar behaviour were also found in gumminess and chewiness properties (respectively defined as the force and energy required to masticate cheese into a uniform state before 16 swallowing). Also adhesiveness, (the work necessary to overcome the attractive forces between 17 18 the cheese and the contact surfaces of the universal testing machine probe) (Tunick, 2000), 19 showed a high increase in the last two months of ripening  $(0.45 \pm 0.15 \text{ mJ})$ . On the other hand, small differences in springiness (measure of the distance recovered by the cheese sample during 20 21 the time between the end of the first bite and start of second bite), were registered throughout the entire ripening period. Instead the parameters of cohesiveness (measure of the strength of the 22 internal bonds of the protein mycelium) (Tunick, 2000), and resilience, (a dimensionless 23 24 parameter which represents the ability of the cheese to regain its original position after the first compression) (Chevanan, Muthukumarappan, Upreti & Metzger, 2006), were characterized by a
 strong decrease in the values between 3 and 30 days, respectively -0.07± 0.01 and -1.51± 0.63.
 The values of these parameters also showed a decrease between 90 and 150 days.

4 Therefore, three distinct phases in texture development took place during Castelmagno PDO
5 cheese ripening: the first between 3 and 30 days, the second, from 30 to 90 days and the third,
6 from 90 to 150 days.

7

#### 8 **4.** Conclusions

9 During the manufacture of Castelmagno PDO cheese, it was possible to detect the conclusion of 10 lactose metabolism with the total conversion of lactose into lactate and the commencement of the 11 primary proteolyses. The volatile profile was characterised by a high level of acids, in particular of 12 hexanoic, octanoic and decanoic acids, which are the primary products of lipolysis metabolism.

During ripening of Castelmagno PDO cheeses, it is possible to observe high degradation of  $\alpha$ casein with an increase of all its degradation products, an evolution of the hydrophilic peptides associated also to the highest concentration of glutamic acid, valine, leucine phenylalanine and lysine. The volatile profiles of Castelmagno PDO cheese during ripening are characterised by a decrease in acid compounds and an increase in ketones and alcohols as a consequence of free fatty acids metabolism. Texture profiles underline an increased of hardness, gumminess, chewiness and adhesiveness properties, and a diminution of cohesiveness.

However, to fully characterize the Castelmagno PDO cheese it will also be necessary to analyze samples of Castelmagno PDO during the summer period, when the producers transfer the herd to grassland where cows eat fresh forage that both directly and indirectly influences the organic acid and volatile profile of the obtained products. At the same time, the producers change the caves, an

act that can influence microbiological effects and, as a consequence, the biochemical pathways of
 Castelmagno PDO cheese during ripening.

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- 14

- 1 Table1: Mean value ± standard deviation for the gross composition of Castelmagno PDO cheese
- 2 during its production and its ripening and result of variance analysis.

	Manuf	acture of ch	leese	Days of ripening					
	Cut of the curd	After 24 hours	Curd after 3 days under whey	3	30	60	90	150	Statistic al signific ance
pH:	$6.57 \pm 0.01^{a}$	$5.06 \pm 0.02^{a}$	$4.71 \pm 0.07^{a}$	$4.74 \pm 0.02^{b}$	$4.71 \pm 0.08^{\rm b}$	$4.80 \pm 0.11^{\circ}$	5.02 ± 0.10 <sup>c</sup>	$4.99 \pm 0.05^{\circ}$	***
Moisture (% w/w):	54.6 ± 1.35 <sup>a</sup>	53.4 ± 1.34 <sup>b</sup>	$52.4 \pm 0.34^{b}$	45.5 ± 0.65°	$40.1 \pm 0.99^{d}$	37.7 ± 0.79 <sup>e</sup>	$35.2 \pm 0.31^{\rm f}$	$33.5 \pm 0.82^{g}$	***
NaCl (% w/w):	$0.08 \pm 0.01^{a}$	$0.07 \pm 0.03^{a}$	$0.08 \pm 0.03^{a}$	$2.14 \pm 0.24^{b}$	$2.50 \pm 0.40^{b,c,d}$	$2.75 \pm 0.54^{ m c,d}$	$2.24 \pm 0.35^{ m b,c}$	$2.69 \pm 0.14^{ m c,d}$	***
Fat (% w/w):	$22.5 \pm 0.44^{a}$	$23.9 \pm 0.21^{b}$	$23.3 \pm 0.46^{b}$	$25.8 \pm 0.76^{\circ}$	$29.0 \pm 1.03^{d}$	$29.9 \pm 1.09^{d}$	31.4 ± 0.57 <sup>e</sup>	31.9 ± 0.79 <sup>e</sup>	***
Protein (% w/w):	$18.44 \pm 0.07^{a}$	$19.52 \pm 0.20^{b}$	$19.97 \pm 0.08^{b}$	20.86± 0.02 <sup>c</sup>	$23.29 \pm 0.02^{d}$	25.97 ± 0.08 <sup>e</sup>	$27.05 \pm 0.09^{\rm e,f}$	$26.67 \pm 0.01^{ m e,f}$	***
pH 4.6-solubleN (% total N)	$8.09 \pm 0.15^{a}$	$7.72 \pm 0.90^{a}$	$5.59 \pm 0.19^{b}$	$5.53 \pm 0.30^{b}$	7.85 ± 0.22 <sup>c</sup>	$9.72 \pm 0.10^{d}$	12.36 ± 0.24 <sup>e</sup>	$\begin{array}{c} 14.95 \pm \\ 0.28^{\mathrm{f}} \end{array}$	***

6 Mean data for the three batches of Castelmagno PDO cheeses analysed in triplicate.

7 a, b, c, d, e, f: Different letters in the same row indicate significant statistical differences (Duncan Test,

p < 0.05).

9 Statistical significance: \*\*\*= P < 0.001; \*\*= P < 0.01; \* = P < 0.05; ns= not significance.

1 Table2: Mean value ± standard deviation of free amino acids composition of Castelmagno PDO

2 cheese (mg/g) during its manufacture and ripening.

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	Manuf	facture of ch	neese	Days of ripening					
	Cut of the curd	After 24 hours	Curd after 3 days under whey	3	30	60	90	150	Statistic al signific ance
Aspartic acid	$0.07 \pm 0.06^{\rm a}$	$0.29\pm 0.05^{a}$	$0.65 \pm 0.19^{a}$	$0.74 \pm 0.03^{a}$	$2.05 \pm 0.55^{b}$	$4.45 \pm 0.25^{\circ}$	$5.24 \pm 0.47^{\circ}$	$7.98 \pm 0.30^{\rm d}$	***
Threonine	$0.02 \pm 0.01^{a}$	$0.15 \pm 0.03^{a}$	$0.44 \pm 0.12^{a,b}$	$0.55 \pm 0.02^{a,b}$	$0.85 \pm 0.20^{b}$	$1.70 \pm 0.29^{\circ}$	$1.90 \pm 0.43^{\circ}$	$2.67 \pm 0.22^{d}$	***
Serine	$0.02 \pm 0.01$	$0.14 \pm 0.05$	$0.38 \pm 0.19$	$0.45 \pm 0.18$	$0.58 \pm 0.22$	$1.00 \pm 0.56$	$1.19 \pm 0.51$	$1.65 \pm 0.72$	ns
Glutamic acid	$0.45 \pm$	$1.55 \pm 0.14^{a,b}$	$2.69 \pm 1.10^{a,b}$	$3.80 \pm 0.00^{a,b}$	$4.02 \pm 1.14^{a,b,c}$	$7.94 \pm 1.62^{b,c}$	$6.74 \pm 1.16^{c,d}$	$11.41 \pm 1.06^{d}$	***
Glycine	0.20 $0.04 \pm 0.03^{a}$	$0.04 \pm 0.02^{a}$	$0.12 \pm 0.03^{a,b}$	$0.09 \pm 0.02^{a,b}$	$0.57 \pm 0.11^{b}$	$1.03 \pm 0.24^{\circ}$	$1.10 \\ 1.58 \pm 0.01^{\circ}$	1.90 2.66 ± 0.51 <sup>d</sup>	***
Alanine	$0.03 \pm 0.01^{a}$	$0.22 \pm 0.01^{a}$	$0.87 \pm 0.18^{\mathrm{a}}$	$1.20 \pm 0.02^{a}$	$1.56 \pm 0.24^{a,b}$	$2.74 \pm 0.15^{b}$	$3.01 \pm 0.09^{b}$	$5.66 \pm 1.10^{\circ}$	***
Cysteine	$0.04 \pm 0.02^{a}$	$0.06 \pm 0.02^{\rm a}$	$0.14 \pm 0.07^{\rm a}$	$0.32 \pm 0.03^{a,b}$	$0.44 \pm 0.01^{ m a,b}$	$0.63 \pm 0.04^{ m b,c}$	$0.91 \pm 0.08^{ m c,d}$	$1.29 \pm 0.66^{d}$	***
Valine	$0.16 \pm 0.05^{a}$	$0.50 \pm 0.04^{\rm a}$	$1.02 \pm 0.28^{a,b}$	$1.76 \pm 0.09^{ m a,b}$	$2.54 \pm 0.09^{b}$	$4.83 \pm 0.05^{\circ}$	$6.28 \pm 0.04^{\circ}$	$9.78 \pm 2.54^{\rm d}$	***
Methionine	$0.02 \pm 0.02^{a}$	$0.13 \pm 0.02^{a}$	$0.27 \pm 0.09^{a}$	$0.80 \pm 0.06^{a}$	$1.43 \pm 0.22^{b}$	$2.91 \pm 0.53^{\circ}$	$3.47 \pm 0.21^{d}$	$4.89 \pm 0.20^{e}$	***
Isoleucine	$0.01 \pm 0.02^{a}$	$0.19 \pm 0.06^{a}$	$0.40 \pm 0.14^{a}$	$0.70 \pm 0.04^{a}$	$1.00 \pm 0.05^{a,b}$	$2.28 \pm 0.30^{b,c}$	$3.37 \pm 0.07^{\circ}$	$5.46 \pm 1.45^{d}$	***
Leucine	$0.12 \pm 0.08^{a}$	$0.70 \pm 0.12^{a,b}$	$2.09 \pm 0.54^{a,b}$	$4.61 \pm 2.81^{b}$	$7.50 \pm 1.36^{\circ}$	$13.57 \pm 1.81^{d}$	$15.99 \pm 1.11^{\circ}$	$20.15 \pm 0.33^{\rm f}$	***
Tyrosine	$0.10 \pm 0.06^{a}$	$0.42 \pm 0.12^{a,b}$	$0.65 \pm 0.27^{b}$	$0.58 \pm 0.05^{a,b}$	$0.80 \pm 0.28^{b}$	$1.33 \pm 0.09^{\circ}$	$1.55 \pm 0.10^{\circ}$	$1.50 \pm 0.07^{\circ}$	***
Phenylalanine	$0.10 \pm 0.07^{ m a}$	$\begin{array}{c} 0.71 \pm \\ 0.14^{\mathrm{a,b}} \end{array}$	$1.81 \pm 0.44^{\rm b}$	$3.35 \pm 1.71^{b}$	$4.82 \pm 0.42^{\circ}$	$7.76 \pm 0.31^{d}$	9.26 ± 0.35 <sup>e</sup>	${\begin{array}{c} 11.88 \pm \\ 1.84^{\rm f} \end{array}}$	***
Histidine	$\begin{array}{c} 0.14 \pm \\ 0.07^{\mathrm{a}} \end{array}$	$0.60 \pm 0.13^{a,b}$	$1.00 \pm 0.25^{\rm b}$	$1.82 \pm 0.09^{b}$	$2.44 \pm 0.08^{\circ}$	$3.31 \pm 0.05^{d}$	$\begin{array}{c} 3.57 \pm \\ 0.23^{d} \end{array}$	$\begin{array}{c} 4.03 \pm \\ 0.83^{d} \end{array}$	***
Lysine	$\begin{array}{c} 0.13 \pm \\ 0.05 \end{array}$	$\begin{array}{c} 0.71 \pm \\ 0.38 \end{array}$	1.15 ± 0.65	$\begin{array}{c} 1.66 \pm \\ 0.56 \end{array}$	$\begin{array}{c} 2.46 \pm \\ 0.41 \end{array}$	$\begin{array}{c} 2.49 \pm \\ 0.80 \end{array}$	4.51 ± 0.03	$\begin{array}{c} 3.98 \pm \\ 0.25 \end{array}$	ns
Arginine	$0.01 \pm 0.01$	ND	$\begin{array}{c} 0.07 \pm \\ 0.09 \end{array}$	$\begin{array}{c} 0.02 \pm \\ 0.01 \end{array}$	$0.05 \pm 0.11$	ND	ND	ND	ns
Proline	$0.11 \pm 0.14^{a}$	$\begin{array}{c} 0.64 \pm \\ 0.18^{\rm a} \end{array}$	$1.03 \pm 0.26^{a}$	$1.03 \pm 0.28^{a}$	$1.05 \pm 0.23^{a}$	$1.19 \pm 1.26^{a}$	$1.41 \pm 0.01^{a}$	3.57 ± 1.15 <sup>b</sup>	***
Total free amino acids	$\begin{array}{c} 1.57 \pm \\ 0.97 \end{array}$	7.05 ± 1.51	14.78± 4.89	23.71 ± 6.09	34.16 ± 5.72	59.42 ± 8.36	69.98 ± 4.90	98.56 ± 14.13	

<sup>4</sup> 

5 Mean data for the three batches of Castelmagno PDO cheeses analysed in triplicate.

6 a, b, c, d, e, f: Different letters in the same row indicate significant statistical differences (Duncan Test,

7 *p*< 0.05).

8 ND: not detected.

9 Statistical significance: \*\*\*= P < 0.001; \*\*= P < 0.01; \* = P < 0.05; ns= not significance.

#### 1 Table 3: Mean value ± standard deviation of organic acids, sugars, diacetyl and acetoin

	Manu	facture of cl	neese		Days of ripening						
	Cut of the curd	After 24 hours	Curd after 3 days under whey	3	30	60	90	150	Statistic al signific ante		
Lactose	$36.65 \pm 2.06^{a}$	6.13 ± 1.71 <sup>b</sup>	$0.72 \pm 0.49^{\circ}$	ND	ND	ND	ND	ND	***		
Glucose	$0.01 \pm 0.01^{a}$	$0.01 \pm 0.01^{b}$	ND	ND	ND	ND	ND	ND	***		
Galactose	$0.10 \pm 0.01^{a}$	$0.11 \pm 0.02^{b}$	$0.12 \pm 0.03^{b}$	$0.03 \pm 0.01^{\circ}$	ND	ND	ND	ND	***		
Lactic acid	$0.90 \pm 0.90^{a}$	$20.31 \pm 2.68^{b,c}$	$31.06 \pm 3.54^{d}$	$30.57\pm$ 2.58 <sup>d</sup>	$24.96 \pm 2.41^{b}$	$19.08 \pm 0.90^{\circ}$	$17.73 \pm 2.07^{\circ}$	$15.28 \pm 1.45^{\circ}$	***		
Diacetyl	$0.04 \pm 0.01^{a}$	$0.20 \pm 0.13^{a}$	$0.66\pm 0.10^{a,b,c}$	$0.52\pm 0.08^{a,b}$	$0.74 \pm 0.25^{a,b,c}$	$1.1/\pm 0.39^{b,c,d}$	$1.39 \pm 0.35^{c,d}$	$1.59 \pm 0.20^{d}$	***		
Acetoin	ND	ND	$0.02 \pm 0.01^{a}$	$0.02 \pm 0.01^{a}$	$0.02 \pm 0.00^{a}$	$0.04 \pm 0.00^{b}$	$0.06 \pm 0.01^{\circ}$	$0.06 \pm 0.02^{d}$	***		
Citric acid	$1.24 \pm 0.04^{a}$	$1.32 \pm 0.11^{b}$	$0.30 \pm 0.17^{\circ}$	$0.03 \pm 0.02^{d}$	$0.03 \pm 0.01^{d}$	$0.04 \pm 0.02^{d}$	$0.07 \pm 0.03^{d}$	$0.09 \pm 0.02^{d}$	***		
Pyruvic acid	$0.04 \pm 0.02^{a}$	$0.12 \pm 0.06^{a}$	$0.43 \pm 0.17^{b}$	$0.41 \pm 0.11^{b}$	$0.12 \pm 0.04^{a}$	$0.04 \pm 0.03^{a}$	$0.03 \pm 0.02^{a}$	$0.02 \pm 0.01^{a}$	***		
Formic acid	$0.03 \pm 0.03$	$\begin{array}{c} 0.04 \pm \\ 0.04 \end{array}$	$\begin{array}{c} 0.02 \pm \\ 0.01 \end{array}$	$\begin{array}{c} 0.03 \pm \\ 0.01 \end{array}$	$\begin{array}{c} 0.02 \pm \\ 0.00 \end{array}$	$\begin{array}{c} 0.02 \pm \\ 0.01 \end{array}$	$\begin{array}{c} 0.02 \pm \\ 0.01 \end{array}$	$\begin{array}{c} 0.02 \pm \\ 0.01 \end{array}$	ns		
Acetic acid	ND	$0.13 \pm 0.05^{a}$	$0.81\pm 0.08^{ m b}$	$0.77 \pm 0.06^{\circ}$	$\begin{array}{c} 0.64 \pm \\ 0.02^{\mathrm{d}} \end{array}$	$0.57 \pm 0.05^{ m d,e}$	$0.51 \pm 0.03^{e}$	$0.50 \pm 0.03^{\rm e}$	***		
Propionic acid	ND	$0.08 \pm 0.02^{\rm a}$	$0.08 \pm 0.03^{a}$	0.09± 0.03 <sup>a</sup>	$0.54 \pm 0.06^{\mathrm{b}}$	$0.83 \pm 0.08^{\circ}$	$1.23 \pm 0.13^{d}$	$1.43 \pm 0.14^{\rm e}$	***		
Oxalic acid	$\begin{array}{c} 0.42 \pm \\ 0.05^{a} \end{array}$	$0.56 \pm 0.07^{\rm b}$	$0.18 \pm 0.09^{\circ}$	$0.03 \pm 0.01^{d}$	ND	ND	$0.01 \pm 0.01^{d}$	$\begin{array}{c} 0.01 \pm \\ 0.01^{d} \end{array}$	***		
Orotic acid	$0.04 \pm 0.02^{a}$	$0.02 \pm 0.01^{b}$	$0.01 \pm 0.01^{\circ}$	0.01 ± 0.01 <sup>c,d</sup>	ND	ND	$0.01 \pm 0.01^{c,d}$	$0.01 \pm 0.01^{c,d}$	***		
Iso-butyric acid	ND	ND	ND	ND	ND	ND	$0.19 \pm 0.28^{\rm a}$	$1.17 \pm 0.09^{b}$	***		
Butyric acid	$0.06 \pm 0.05^{\rm b}$	$0.01 \pm 0.01^{a}$	$0.15 \pm 0.02^{ m c,d}$	$0.14 \pm 0.03^{ m c,d}$	$0.11 \pm 0.02^{b,c}$	$0.15 \pm 0.01^{c,d}$	$0.19 \pm 0.04^{d}$	$\begin{array}{c} 0.19 \pm \\ 0.04^{\mathrm{d}} \end{array}$	***		
Iso-valeric acid	ND	$0.01 \pm 0.01^{a}$	$0.01 \pm 0.01^{a}$	$0.11 \pm 0.06^{a}$	$0.43 \pm 0.09^{b}$	$\begin{array}{c} 0.44 \pm \\ 0.08^{\mathrm{b}} \end{array}$	$0.48 \pm 0.06^{ m b}$	$0.69 \pm 0.28^{b}$	***		
n-Valeric acid	ND	$\begin{array}{c} 0.01 \pm \\ 0.00 \end{array}$	0.01 ± 0.00	ND	ND	ND	$\begin{array}{c} 0.01 \pm \\ 0.02 \end{array}$	$\begin{array}{c} 0.03 \pm \\ 0.02 \end{array}$	ns		
Hippuric acid	0.01 ± 0.01	ND	ND	ND	ND	ND	ND	ND	***		
Uric acid	$\begin{array}{c} 0.02 \pm \\ 0.00^a \end{array}$	$\begin{array}{c} 0.02 \pm \\ 0.00^{a} \end{array}$	$\begin{array}{c} 0.03 \pm \\ 0.00^{\mathrm{b}} \end{array}$	ND	ND	ND	$0.04 \pm 0.01^{\circ}$	$\begin{array}{c} 0.05 \pm \\ 0.01^{d} \end{array}$	***		

#### 2 concentrations of Castelmagno PDO (mg/g) cheese during its manufacture and ripening.

3 Mean data for the three batches of Castelmagno PDO cheeses analysed in triplicate.

4 a, b, c, d, e: Different letters in the same row indicate significant statistical differences (Duncan Test, p <

5 0.05).

6 ND: not detected.

7 Statistical significance: \*\*\*= P < 0.00; \*\*= P < 0.01; \* = P < 0.05; ns= not significance.

## 1 Table 4: Mean value ± standard deviation of volatile compound concentrations of Castelmagno

## 2 PDO ( $\mu$ g/kg) cheese during its manufacture and ripening.

3

		I	Manufactu	re of cheese			Days of r	ipening	
Compounds	L RI w	Cut of the curd	After 24 hours	Curd after 3 days under whey	3	30	60	90	150
Acids									
Acetic acid	14 38	$28.97 \pm 38.79$	17.94 ± 8.61	124.53 ± 64.67	136.21 ± 13.29	248.66 ± 92.53	136.43 ± 71.41	103.20 ± 24.08	142.06 ± 60.07
acid	15 28	ND	ND	ND	ND	$4.85 \pm 4.93$	12.75 ± 14.34	4.95	21.77 ± 7.87
Isobutyric acid	15 60	$0.18 \ \pm 0.32$	ND	ND	$0.48\pm\ 0.05$	$3.43\pm~0.58$	$2.70\pm2.73$	1.17 ± 0.60	6.76 ± 4.44
Butyric acid	16 20	141.32 ± 82.97	8.02 ± 2.57	178.29 ± 89.78	166.79 ± 8.63	397.15 ± 115.75	235.34 ± 119.82	162.74 ± 50.05	278.45 ± 165.54
Isovaleric acid	16 65	$1.43 \hspace{0.1cm} \pm \hspace{0.1cm} 2.48$	0.60 ± 0.17	$1.90\pm0.73$	$1.18\pm\ 020$	16.33 ± 3.82	$\begin{array}{c} 10.30 \pm \\ 9.50 \end{array}$	4.90 ± 1.79	59.35 ± 46.83
Valeric acid	17 36	$2.11 \pm 1.46$	$0.26 \pm 0.04$	$0.75\pm0.33$	$0.52\pm\ 0.05$	$2.97 \pm \ 0.61$	$1.69\pm0.51$	$1.49 \pm 0.65$	$3.62\pm2.30$
Hexanoic acid	18 42	328.85 ± 187.73	309.78 ± 7.13	780.34 ± 343.41	541.58 ± 176.69	1341.57 ± 678.09	1001.54 ±357.95	733.01 ± 251.50	$\begin{array}{r} 1094.37 \pm \\ 698.89 \end{array}$
Heptanoic acid	19 50	10.19 ± 6.34	2.59 ± 0.69	$4.25\pm2.35$	$1.58\pm\ 0.59$	11.21 ± 5.38	$9.36 \pm 5.46$	$\begin{array}{c} 6.93 \pm \\ 4.68 \end{array}$	15.25 ± 12.85
Octanoic acid	20 56	$522.77 \pm 248.85$	516.19 ± 16.28	155.89 ± 80.77	102.97 ± 38.57	$637.05 \pm 198.77$	$509.07 \pm 285.30$	380.47 ± 163.66	$720.76 \pm 517.91$
Nonanoic acid	21 63	19.00 ± 11.23	4.03 ± 0.94	$9.13 \pm 2.73$	$1.47\pm2.31$	11.71 ± 3.41	$\begin{array}{c} 10.02 \pm \\ 4.77 \end{array}$	9.47 ± 6.06	$\begin{array}{c} 22.43 \pm \\ 18.38 \end{array}$
Decanoic acid	22 68	594.32 ± 335.82	$280.39 \pm 26.15$	$259.90 \pm 114.41$	128.06 ± 59.15	$692.78 \pm 203.65$	570.27 ± 329.06	448.79 ± 197.36	842.10 ±553.94
Undecanoic acid	23 78	14.66 ± 13.42	0.43 ± 0.46	ND	ND	$1.94 \pm \ 3.36$	ND	7.61 ± 5.79	15.18 ± 19.91
Benzoic acid	-	$3.44 \hspace{0.1cm} \pm \hspace{0.1cm} 2.53$	1.30 ± 0.41	$5.41 \pm 2.05$	$3.95\pm~0.63$	$8.56 \pm 1.21$	$4.78\pm3.03$	3.88 ± 1.99	9.61 ± 15.60
Dodecanoic acid	-	246.21± 154.89	20.77 ± 8.41	$\begin{array}{c} 54.22 \pm \\ 26.77 \end{array}$	18.86 ± 16.25	188.25 ± 44.75	164.06 ± 106.76	$148.29 \pm 83.84$	290.84 ± 218.32
Tetradecanoi c acid	-	$1.72\pm2.99$	ND	ND	ND	ND	ND	5.71 ± 6.30	$1.65\pm2.85$
Total		$1915.17 \pm 1089.80$	1162.30 ± 71.86	1574.62 ±728.01	1103.65 ± 316.41	3566.45 ± 1356.85	2668.30 ± 1310.64	$2029.33 \pm 803.30$	3524.21 ± 2345.72
rercentage (%) <sup>x</sup>		93.35	86.99	85.48	88.13	75.89	70.70	76.20	78.70
Esters									
Ethyl acetate	88 0	$0.13\pm0.14$	$\begin{array}{c} 0.30 \pm \\ 0.07 \end{array}$	$1.12\pm0.59$	$0.60\pm0.18$	$2.05 \pm 1.06$	$1.88 \pm 1.81$	$\begin{array}{c} 0.97 \pm \\ 0.21 \end{array}$	$1.42\pm0.04$
Methyl butanoate	97 8	$0.03\pm0.05$	0.22 ± 0.20	$0.96\pm0.55$	ND	ND	1.14 ± 1.25	$\begin{array}{c} 0.30 \pm \\ 0.53 \end{array}$	$1.03 \pm 1.04$
Ethyl butanoate	10 30	$0.72\pm0.21$	1.34 ± 1.72	$2.27\pm0.66$	$1.09\pm0.30$	$6.89 \pm 4.26$	11.52 ± 15.82	5.92 ± 1.37	$7.76\pm3.89$
Butyl butanoate	11 20	ND	ND	ND	ND	ND	ND	$0.22 \pm 0.20$	$0.35\pm0.60$

Butyl 3- methyl-	12	ND	ND	ND	$0.07 \pm 0.12$	1.95 ± 0.50	1.90 ± 2.10	$1.09 \pm 0.31$	1.61 ± 0.68
butanoate Ethyl hexanoate	57 12 26	$0.95 \pm 0.87$	3.09 ±	$22.89 \pm 13.20$	$16.11 \pm 1.51$	$120.11 \pm 5.30$	$111.11 \pm 10.86$	75.50 ±	89.12 ± 44.10
<i>n</i> -Hexyl acetate	12 65	ND	ND	ND	ND	ND	$0.25 \pm 0.29$	$0.26 \pm 0.17$	$0.52 \pm 0.13$
Propyl hexanoate	13 12	ND	ND	ND	ND	$8.09\pm\ 8.15$	12.21 ± 13.79	15.11 ± 8.63	14.85 ± 5.03
Ethyl heptanoate	13 25	ND	$\begin{array}{c} 0.22 \pm \\ 0.38 \end{array}$	ND	ND	$0.40\pm~0.68$	$0.90 \pm 1.56$	$\begin{array}{c} 0.25 \pm \\ 0.22 \end{array}$	$0.78\pm0.87$
Ethyl octanoate	13 36	$2.96 \pm 2.04$	$\begin{array}{r} 4.47 \pm \\ 6.60 \end{array}$	$1.87\pm0.70$	$1.25\pm0.65$	36.98 ± 45.36	44.73 ± 69.44	22.45 ± 9.58	17.96 ± 6.98
caproate	13 64	ND	ND	ND	ND	0.07 ± 0.13	$0.05\pm0.08$	$0.23 \pm 0.05$	$0.49\pm0.32$
acetate Methyl	13 66 13	ND	ND 1 46 +	ND	ND	ND	1.09 ± 1.88	$0.23 \pm 0.23$ 1 15 +	$0.32 \pm 0.28$
octanoate <i>n</i> -Hexyl	13 80 14	0.33 ± 0.57	2.53	ND	ND	0.81 ± 1.41	3.04 ± 5.27	2.00 0.69 ±	1.48 ± 1.54
butanoate Propyl	08 15	ND	ND	ND	ND	$0.43 \pm 0.52$	$0.70 \pm 0.56$	0.14 3.07 ±	$0.93 \pm 0.30$
octanoate Ethyl	13 15	0.19 + 0.19	ND	ND	ND	0.99 + 1.71	2.07 ± 5.85	2.13 0.08 ±	2.51 ± 1.18
nonanoate Methyl	29 15	2.83 ± 1.52	2.10 ±	2.91 ± 1.48	0.66 ± 0.14	12.01 ±	11.72 ±	3.48 10.50 ± 4.85	17.94 ±
<i>n</i> -Hexyl hexanoate	00 16 05	ND	ND	ND	ND	2.45 ± 1.32	$1.89 \pm 0.74$	4.85 1.72 ± 0.59	$2.83 \pm 1.63$
Ethyl decanoate	16 36	6.71 ± 7.03	5.84 ± 10.12	3.09 ± 2.69	$2.26 \pm 1.33$	$\begin{array}{c} 56.20 \pm \\ 63.10 \end{array}$	52.21 ± 80.43	46.14 ± 30.37	65.86 ± 72.29
Ethyl 9- decenoate	16 87	$0.82\pm0.43$	$0.53 \pm 0.91$	ND	ND	$5.88 \pm \ 6.49$	$5.86 \pm 8.68$	5.47 ± 3.46	$7.79 \pm 9.55$
Methyl undecanoate	16 94	ND	ND	ND	ND	$0.06\pm\ 0.10$	ND	$0.03 \pm 0.05$	$0.23\pm0.41$
Propyl decanoate	17 24	ND	ND	ND	ND	$5.79\pm\ 5.08$	$3.51 \pm 4.76$	3.38 ± 2.21	$3.61 \pm 1.85$
undecanoate Methyl	17 41 17	ND	$0.07 \pm 0.12$ 0.42 +	ND	ND	$0.55 \pm 0.95$	ND	$0.43 \pm 0.78$ 1 09 +	ND
dodecanoate 2-Phenyl	17 99 18	0.67 ± 1.16	0.73	ND	ND	ND	ND	1.88 $3.54 \pm$	ND
ethyl acetate Propyl	08 19	ND	ND	ND	$0.05 \pm 0.08$	$9.18 \pm 0.55$	$2.78 \pm 4.82$	1.15 0.74 ±	$2.44 \pm 2.11$
dodecanoate Ethyl	27	ND	0.17 +	ND	ND	0.37 ± 0.32	$0.00 \pm 1.04$	0.19 5.62 +	$0.46 \pm 0.31$
tetradecanoat e	20 52	ND	0.29	ND	ND	5.37 ± 1.62	0.10 ± 0.18	4.00	$1.06 \pm 1.04$
tetradecanoat	20 08	ND	$\begin{array}{c} 0.08 \pm \\ 0.14 \end{array}$	ND	ND	ND	ND	$\begin{array}{c} 0.40 \pm \\ 0.44 \end{array}$	$0.33\pm0.35$
Total	00	16.33 ± 14.21	20.31 ± 31.31	35.10 ± 19.88	22.09 ± 4.31	280.03 ± 167.93	271.88 ± 238.28	208.559 ± 80.71	243.72 ± 172.18
Percentage (%)		0.80	1.52	1.91	1.76	5.96	7.20	7.83	5.44

Ketones

Acetone	81 0	$0.81\pm0.38$	$0.32 \pm 0.14$	$0.40\pm0.10$	$0.73\pm0.32$	$6.81 \pm \ 6.85$	$4.72\pm\ 5.04$	1.28 ± 1.14	$2.42 \pm 1.04$
2-Butanone	89 6	$0.17\pm0.17$	0.24 ± 0.21	$0.55\pm0.29$	$0.53\pm0.37$	$32.00 \pm 26.60$	73.64 ± 85.31	58.17 ± 10.68	46.88 ± 25.99
2-Pentanone	96 5	$0.22\pm0.24$	ND	$2.75\pm3.15$	0.94 ± 1.33	82.05 ± 9.89	$54.72 \pm 76.05$	13.97 ± 14.25	21.27 ± 15.86
Diacetyl	97 0	$3.97 \pm 6.56$	ND	ND	$2.75 \pm 1.62$	$0.67 \pm \ 1.16$	ND	ND	ND
2-Hexanone	10 69	ND	ND	ND	ND	$0.43 \pm 0.74$	$1.74\pm2.75$	$0.28 \pm 0.49$	$0.81\pm0.18$
2-Heptanone	11 71	$0.68 \pm 1.18$	ND	ND	$1.60\pm0.80$	25.03 ± 14.39	77.63 ± 13.89	13.05 ± 5.19	41.36 ± 23.75
Acetoin	12 70	44.37 ± 75.69	$10.82 \pm 6.82$	27.76 ± 23.59	27.04 ± 15.52	$11.06 \pm 6.17$	$3.86 \pm 2.40$	1.84 ± 1.05	3.25 ± 2.59
2-Octanone	12 74	ND	ND	ND	ND	$0.70\pm~063$	$1.85 \pm 2.64$	$0.38 \pm 0.12$	$1.16\pm0.70$
2-Nonanone	13 79	$2.27 \pm 1.21$	$\begin{array}{c} 0.10 \pm \\ 0.18 \end{array}$	$0.99\pm0.13$	$1.03\pm0.20$	$34.20 \pm 25.70$	65.44 ± 82.77	24.41 ± 10.56	57.71 ± 26.26
8-Nonen-2- one	14 32	$0.11 \pm 0.19$	ND	ND	ND	$5.99\pm\ 5.93$	$11.00 \pm 14.40$	3.18 ± 1.16	$\begin{array}{c} 10.11 \pm \\ 4.88 \end{array}$
3,5-Octadien- 2-one Bonzul	15 06	$1.88 \pm 1.09$	3.66 ± 1.65	10.48 ± 4.43	ND	ND	ND	ND	ND
methyl ketone	17 17	ND	ND	ND	ND	$0.23 \pm 0.40$	ND	ND	$0.42\pm0.46$
2- Tridecanone	18 09	ND	ND	ND	ND	$256.62 \pm 44.48$	$0.76 \pm 1.14$	ND	$2.09\pm2.97$
Total Percentage		54.47 ± 86.70	15.14 ± 9.00	$42.92 \pm 31.68$	34.62 ± 20.16	455.78 ± 142.94	525.26 ± 341.69	$208.72 \pm 69.99$	316.12 ± 146.04
(%)		2.66	1.13	2.33	2.76	9.70	13.92	7.84	7.06
Aldehydes									
Acetaldehyde	-	ND	$0.35 \pm 0.25$	$1.61 \pm 1.79$	$2.79 \pm 1.61$	$4.24 \pm 1.79$	$2.62\pm0.36$	$0.20 \pm 0.17$	$1.11 \pm 1.36$
3-Methyl butanal	91 1	$0.54\pm0.87$	$0.25 \pm 0.24$	$1.68\pm0.95$	$0.56\pm0.15$	$1.20\pm\ 0.87$	$0.21\pm\ 0.24$	0.14 ± 0.12	$0.70\pm0.61$
Hexanal	10 71	$1.31\pm0.60$	26.43 ± 10.52	$\begin{array}{c} 30.20 \pm \\ 26.51 \end{array}$	$1.51\pm0.17$	$0.98\pm\ 0.85$	$0.06\pm\ 0.10$	0.27 ± 0.47	$0.42\pm0.72$
Heptanal	11 71	ND	4.81 ± 4.66	12.43 ± 9.10	ND	ND	ND	ND	ND
<i>trans</i> -2- Hexenal	12 03	ND	2.42 ± 1.20	$2.80 \pm 2.51$	ND	4.11 ± 7.12	ND	ND	ND
Octanal	12 72	ND	4.13 ± 3.09	ND	ND	ND	ND	ND	ND
2-Heptenal	13 07	ND	1.96 ± 1.78	$2.90\pm2.19$	ND	ND	ND	ND	ND
Nonanal	13 83	ND	6.57 ± 2.97	ND	$0.15\pm0.27$	ND	ND	ND	ND
2-Octenal	14 12	ND	4.56 ± 2.37	$5.03 \pm 4.77$	ND	ND	ND	ND	ND
Benzaldehyd e	15 04	$0.37\pm0.35$	$0.80 \pm 0.27$	$1.39 \pm 1.72$	ND	ND	ND	ND	$0.59 \pm 1.03$
2-Nonenal	15 24	$1.06\pm0.96$	19.03 ± 11.40	24.93 ± 22.92	$0.62\pm0.15$	ND	$0.04 \hspace{0.1in} \pm 0.07$	$0.08 \pm 0.14$	$0.09\pm0.15$
Decanal	17 67	ND	ND	ND	ND	$0.88 \pm 1.53$	ND	0.12 ± 0.21	$0.61\pm0.15$
Total		3 28 + 2 78	71.31 ± 38.75	82.96 ± 72.47	5 64 + 2 34	11.42 ± 12.16	$3.00 \pm 0.91$	0.81 ±	$352 \pm 402$
Percentage		5.20 ± 2.70	50.75	12.47	5.04 ± 2.54	12.10	5.00 - 0.91		5.52 ± 4.02

Alcohols									
Ethanol	93 0	$8.19 \pm 8.15$	42.99 ± 42.87	57.18 ± 39.87	27.83 ± 13.12	30.18 ± 7.50	62.37 ± 59.46	45.95 ± 39.57	157.26 ± 106.51
2-Butanol	10 25 10	ND	ND	ND	ND	$9.84 \pm \ 6.95$	$35.95 \pm 45.99$	$50.30 \pm 12.31$	$40.14 \pm 6.98$
Isobutanol	97	$0.14\pm0.24$	ND	ND	$1.44 \pm 0.43$	$2.08 \pm 1.61$	$1.26 \pm 0.81$	$0.80 \pm 0.49$	$0.96 \pm 0.99$
2-Pentanol	11 23	ND	ND	ND	0.11 ± 0.19	33.58 ± 9.21	40.97 ± 47.22	$14.61 \pm 10.42$	$16.14 \pm 11.63$
1-Butanol	11 44	ND	ND	ND	0.53 ± 0.09	1.92 ± 0.93	2.29 ± 1.95	$3.59 \pm 0.83$	3.61 ± 1.40
Isopentanol	12 06	6.94 ± 12.03	$2.38 \pm 2.06$	$\begin{array}{r} 14.79 \pm \\ 6.68 \end{array}$	$10.64 \pm 2.22$	47.35 ± 24.66	20.95 ± 19.26	$13.40 \pm 4.88$	17.66 ± 8.69
2-Heptanol	13 20	ND	ND	ND	ND	$32.84 \pm 32.10$	36.63 ± 44.28	$\begin{array}{r} 15.57 \pm \\ 7.88 \end{array}$	$22.58 \pm 4.29$
1-Hexanol	13 51	$0.36\pm0.34$	$1.28 \pm 0.55$	$3.30 \pm 1.31$	$1.80\pm\ 0.58$	$6.19\pm\ 3.99$	$6.07\pm2.05$	8.56 ± 3.34	8.47 ± 2.76
2-Nonanol	15 19	ND	ND	ND	ND	45.74 ± 70.44	23.83 ± 22.64	15.98 ± 2.02	$\begin{array}{r} 28.60 \pm \\ 7.87 \end{array}$
2,3- butanediol <i>d</i> , <i>l</i>	15 35	6.65 ± 11.52	ND	ND	$1.70 \pm 1.05$	$9.52\pm~3.83$	$7.55\pm7.81$	$\begin{array}{c} 4.04 \pm \\ 1.81 \end{array}$	$5.08\pm0.91$
1-Octanol	15 55	ND	2.17 ± 1.99	ND	$0.07\pm0.12$	ND	$0.19\pm0.19$	$0.34 \pm 0.12$	$0.78 \pm 0.33$
butanediol meso	15 73	13.63 ± 23.61	5.51 ± 3.92	$\begin{array}{c} 14.28 \pm \\ 5.18 \end{array}$	19.43 ± 1.60	40.47 ± 20.40	15.11 ± 6.43	4.95 ± 4.31	5.69 ± 3.53
Benzyl alcohol	18 83	ND	ND	ND	ND	$1.43 \pm 1.60$	$0.53\pm0.92$	ND	ND
Total Percentage		35.92 ± 55.89	54.31 ± 51.39	89.56 ± 53.05	$\begin{array}{c} 63.55 \pm \\ 19.40 \end{array}$	261.15 ± 183.21	253.71 ± 259.01	178.09 ± 87.97	306.96 ± 155.89
(%)		1.75	4.07	4.86	5.06	5.56	6.72	6.69	6.85
Lactones									
δ- Octalactone	16 09	$2.63\ \pm 0.99$	$\begin{array}{c} 0.86 \pm \\ 0.30 \end{array}$	$3.90\pm2.26$	$1.81\pm\ 0.51$	3.73 ± 1.67	$2.68 \pm 1.24$	$\begin{array}{c} 1.90 \pm \\ 0.69 \end{array}$	$2.88 \pm 1.50$
δ- Decalactone	21 57	8.81 ±4.35	6.65 ± 3.29	$3.61\pm 6.26$	12.58 ± 11.14	$\begin{array}{c} 80.09 \pm \\ 48.88 \end{array}$	23.07 ± 6.18	$\begin{array}{c} 16.62 \pm \\ 0.52 \end{array}$	$\begin{array}{r} 41.87 \pm \\ 42.80 \end{array}$
δ- Dodecalacton		8.26 ± 6.37	3.82 ± 4.67	4.23 ± 1.62	1.51 ± 1.84	15.94 ± 7.84	$9.77 \pm 5.82$	8.20 ± 3.74	13.78 ± 12.17
Total	-	19.70 ± 11.71	11.32 ± 8.26	11.74 ± 10.13	15.91 ± 13.49	$99.76 \pm 58.40$	35.52 ± 13.25	26.72 ± 4.95	58.54 ± 56.46
Percentage (%)		0.96	0.85	0.64	1.27	2.12	0.94	1.00	1.31
Hydrocarbon s									
Decane	10 00	$0.62 \ \pm 0.22$	$0.45 \pm 0.04$	$1.89\pm0.77$	$1.45 \pm 0.10$	$2.21 \pm 1.04$	$1.35\pm0.73$	1.50 ± 1.20	$0.24\pm0.28$
Undecane	11 00	ND	ND	ND	$2.26 \pm 1.32$	$0.86\pm\ 0.89$	ND	ND	$0.08 \pm 0.14$
Dodecane	12 00	ND	ND	ND	$0.36\pm\ 0.37$	$2.87 \pm \ 3.23$	$2.77\pm2.16$	1.83 ± 0.89	$0.79\pm0.95$
Total Percentage		$0.62\pm0.22$	$\begin{array}{c} 0.45 \pm \\ 0.04 \end{array}$	$1.89\pm0.77$	$4.07 \pm 1.78$	$6.04 \pm 5.33$	4.12 ± 2.89	3.33 ± 2.09	1.11 ± 1.36
(%)		0.03	0.03	0.10	0.32	0.13	0.11	0.12	0.02

Other compounds Phenolic compounds									
Dimethoxybe	17 40	ND	ND	ND	ND	$0.80 \pm 1.38$	ND	$\begin{array}{c} 0.32 \pm \\ 0.55 \end{array}$	$1.12\pm1.50$
Phenol	20 00	ND	ND	ND	$0.26\pm\ 0.02$	$0.21\pm0.20$	ND	ND	$8.86 \pm 9.74$
Terpenes Limonene	11 81	ND	ND	ND	ND	$0.10 \pm 0.17$	ND	ND	0.70 ± 1.22
compounds N,N-Dibutyl formamide	17 73	0.18 ± 0.30	ND	ND	ND	ND	ND	ND	ND
alcohol Phenyl ethyl alcohol S-	19 07	5.84 ± 9.56	0.98 ± 0.13	$3.35 \pm 1.61$	2.51 ± 0.73	17.78 ± 6.63	12.49 ± 13.39	7.44 ± 2.36	10.21 ± 4.14
<b>compounds</b> Dimethyl disulfide		ND	ND	ND	ND	ND	ND	ND	2.21 ± 2.87
Total		$6.01 \pm 9.87$	0.98 ± 0.13	$3.35 \pm 1.61$	$2.78\pm0.75$	$\begin{array}{r} 18.89 \pm \\ 8.38 \end{array}$	12.49 ± 13.39	7.76 ± 2.91	23.10 ± 19.47
Percentage (%)		0.29	0.07	0.18	0.22	0.40	0.33	0.29	0.52

2 Mean data for the three batches of Castelmagno PDO cheese analysed in triplicate

3 <sup>w</sup> LRI: linear retention index.

4 ND: not detected.

<sup>z</sup> Percentage (%): percentage of volatile compounds of each chemical group in each step of manufacture
and ripening analysed.

### 1 Table 5: Mean value ± standard deviation of TPA parameters of Castelmagno PDO cheese during

### 2 its ripening.

		Days of Ripening										
	3	30	60	90	150	Statistical significance						
Hardness (N)	$12.44\pm2.79^{a}$	$23.21\pm4.46^{b}$	$24.07\pm3.36^{b}$	$25.88\pm4.13^{b}$	$43.15\pm5.50^{\rm c}$	***						
Cohesiveness (-)	$0.28\pm0.06^{a}$	$0.21\pm0.05^{\text{b}}$	$0.25\pm0.04^{\rm a}$	$0.20\pm0.06^{\text{b}}$	$0.17\pm0.03^{\text{b}}$	***						
Adhesiveness (mJ)	- $0.04 \pm 0.04^{a}$	$\textbf{-0.21} \pm 0.13^{b}$	$\textbf{-0.13} \pm 0.08^{a,b}$	$\textbf{-0.12} \pm 0.05^{a,b}$	$\textbf{-0.57} \pm 0.20^{c}$	***						
Gumminess (N)	$3.51 \pm 1.34^{a}$	$4.91 \pm 1.75^{a,b}$	$6.09 \pm 1.41^{\text{b,c}}$	$5.24 \pm 1.70^{b}$	$7.51 \pm 1.77^{\rm c}$	***						
Springiness (mm)	$3.27\pm0.23^a$	$3.31\pm0.23^a$	$3.58\pm0.18^{b}$	$3.57\pm0.24^{b}$	$3.18\pm0.42^{\rm a}$	***						
Chewiness (mJ)	$11.53\pm4.40^a$	$16.41\pm6.34^{a,b}$	$21.91\pm5.29^{b,c}$	$18.92\pm6.47^{b,c}$	$24.00\pm7.69^{\rm c}$	***						
Resilience (-)	$3.15\pm2.26^{a}$	$1.64 \pm 1.63^{b}$	$1.63 \pm 1.15^{\rm b}$	$1.41\pm1.30^{b,c}$	$0.79\pm0.31^{\rm c}$	***						

5 Mean data for the three batches of Castelmagno PDO cheese analysed in quintuplicate

6 a, b, c: Different letters in the same row indicate significant statistical differences (Duncan Test, p < 0.05).

7 Statistical significance: \*\*\*= P < 0.001; \*\*=P < 0.01; \* = P < 0.05; ns= not significance.





2 Figure 1: Example of a TPA profile of Castelmagno PDO cheese.





Figure 2: Urea-polyacrylamide gel electrophoretrograms of Castelmagno PDO cheese (batch A)
during its production (A2= cut of curd; A3=after 24hours; A4= curd after 3 days under whey) and its

- 4 ripening (A5= 3 days of ripening; A6= 30 days of ripening; A7= 60 days of ripening; A8= 90 days of
- 5 ripening; A9= 150 days of ripening). Lane st = Na-caseinate.



Figure 3: RP-HPLC chromatograms of the pH 4.6-soluble fraction of Castemagno PDO cheese
(batchA) during the manufacture (A2= cut of the curd; A3= curd after 24 hours; A4=curd after 3 days
under the whey) and the ripening (A5= 3 days of ripening; A6= 30 days of ripening; A7= 60 days of
ripening; A8=90 days of ripening; A9=150 days of ripening).