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Size resolved metal distribution in the PM matter of the city of Turin (Italy)

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Abstract

A work on the characterization of the air quality in the city of Turin was carried out in different sampling periods, reflecting late summer and winter conditions, including a snow episode during the early 2012 european cold wave.

The concentrations of 16 elements in eight size fractions of the aerosol were determined using inductively coupled plasma-mass spectrometry. The collection was carried out with a Andersen MkII cascade impactor.

The size distribution of elements allowed the identification of three main behavioural types: (a) elements associated with coarse particles (Cd, Cr, Cu, Fe, Mn, Mo, Pt and Sn); (b) elements found within fine particles (As, Co, Pb and V) and (c) elements spread throughout the entire size range (Ni, Pd, Rh and Zn).

Principal Component Analysis allowed to examine the relationships between the inorganic elements and to infer about their origin. Chemometric investigation and assessment of similarity in the distribution led to similar conclusions on the sources.

The concentration of gaseous trace pollutants (O₃, NO_x and VOCs) was determined. The concentrations of these pollutants are scarcely correlated with the metal contents of all the size classes of the PM. The differences found in the O₃, NO₂ and VOCs levels of the two winter campaigns due to the high photochemical reactivity in the period after the snow episode, do not reflect in differences in the metals distribution in the PM. Since PM metals, NO_x and VOC have common sources, this behaviour is due to relevant differences in the transformation and deposition processes.

1 **Keywords**

2 PM₁₀, BTEX, Ozone, Urban air quality, size distribution of elements, principal component analysis

3

4 **Introduction**

5 Air pollution represents one of the greatest concerns of urban environments. In the last decades
6 great attention was paid to Particulate Matter (PM), due to the correlation between fine PM
7 exposure and adverse health effects. The health burden associated to PM air pollution is one of the
8 main environmental health concerns raised by World Health Organization (WHO, 2006). The
9 European Environmental Agency indicates that, with respect to the PM₁₀ and PM_{2.5} thresholds and
10 daily limit values, the Po Valley is one of the most critical areas in Europe (EEA, 2013).

11 Epidemiological and experimental studies have shown that particulate air pollution may induce and
12 aggravate respiratory and cardiovascular diseases. Significant correlations between exposure to
13 ambient air particles and increased morbidity and mortality have been demonstrated (Donaldson et
14 al., 2002, Englert, 2004; Katsouyanni et al., 2001; Krewski and Rainham, 2007). Most of the
15 available studies do not attribute the observed health effects to a particular characteristic of PM.
16 Moreover, the exact physiochemical mechanism by which PM produces adverse effects is still
17 unknown; one hypothesis implicates the oxidative potential of the particles or specific components.
18 In particular, epidemiological studies have found this correlation for fine particle metrics, such as
19 PM_{2.5} (Lall et al., 2005). The pro-inflammatory effect of urban fine and ultrafine PM on human
20 bronchial epithelial cells resulted to be concentration dependent (Ramgolani et al., 2008, Baulig et
21 al, 2009). Short term effects of the fine PM have also been found on daily mortality due to diseases
22 of the cardiocirculatory system (Maté et al., 2010).

23 PM is a complex and heterogeneous mixture, whose composition (particle size distribution and
24 chemical characteristics) changes with time and space and depends on emissions from various
25 sources, atmospheric chemistry and weather conditions. The coarse fraction (2.5-10 µm) comes
26 predominantly from natural sources (geological material, such as fugitive and resuspended dust, and
27 biological material, such as pollen and endotoxins), and its composition changes depending on the
28 geology of the site. The fine fraction (0.1-2.5 µm) is dominated by anthropogenic emissions: a
29 mixture of carbon particles from combustion processes and secondary particles produced by
30 photochemical reactions in the atmosphere (sulphate, nitrate, ammonium). The carbonaceous
31 fraction consists of aggregates of organic and elemental carbon on which transition metals, organic
32 compounds and biological constituents are adsorbed. Penetration in the respiratory system is strictly
33 correlated with the aerodynamic radius. Thus, so concerning the adverse health effects, the

1 distribution of toxic elements and compounds among the various size fractions of the PM is also of
2 primary relevance.

3 Elements are released to the atmosphere from both anthropogenic and natural sources.
4 Anthropogenic sources include fossil fuel combustion, industrial metallurgical processes, vehicle
5 emission and waste incinerations. Natural sources include a variety of processes acting on crustal
6 minerals, such as volcanism, erosion and surface winds, as well as from forest fires and the oceans.
7 Other than the effects on public health, the knowledge of the size distribution and size dependent
8 chemical speciation is also important in identifying the sources and transformation processes during
9 atmospheric transport. Indeed, different sources emit airborne PM with different size distribution,
10 which are deposited at different rates. Particles with diameter in the range of 0.1 – 1.0 μm
11 (accumulation mode) deposit slowly and can be transported far from emission sources with effects
12 on remote areas (Allen et al., 2001). Instead, coarse particles (aerodynamic diameter $> 2.7 \mu\text{m}$) are
13 usually emitted from local sources.

14 In general, as reported above, coarse particles mainly come from road dust re-suspensions, abrasion
15 processes, crustal erosion and sea salts, whereas fine mode particles have been found to mainly
16 derive from anthropogenic sources (combustion processes, high-temperature industrial activities,
17 automotive traffic, etc.) (Handler et al., 2008). Therefore, source apportionment and toxicological
18 information are enhanced by detailed knowledge of the size distribution of aerosol and of its
19 chemical composition.

20 The size distribution of elements within atmospheric particles has been studied over cities around
21 the world and a bimodal distribution of atmospheric particulates has been usually reported, e.g. in
22 Beijing, China (Duan et al., 2012), New York, USA (Song and Gao, 2011), Kanazawa, Japan (Wang
23 et al., 2006), San Roque, Spain (Sánchez de la Campa et al., 2011), Venice, Italy (Toscano et al.,
24 2011, Masiol et al., 2015) and Thessaloniki, Greece (Samara et al. 2006).

25 Concerning the situation in Turin, Italy, time series of the size-fractioned PM_{10} mass concentration
26 were registered. An overview of these was recently published (Casazza et al., 2013), considering
27 winter data of years 1980, 2000 and 2011. These three years, as reported in the paper, can be
28 associated to a known variation of sources characteristics, due to the introduction of new
29 regulations and interventions aimed at air pollution mitigation. In particular, the first data (year
30 1980) are representative of the situation before the introduction of a stricter regulatory approach.
31 The fuel desulphurization and the use of unleaded petrol were introduced just before year 2000,
32 while, since year 2011, it is possible to record the effects of the introduction of regional regulations
33 on domestic heating emission control. The data were considered as representative in the previous
34 analyzed works, since the data values were quite stable along the years, following the same

1 measurement conditions. A reduction of the absolute concentrations of PM is observable. Mean
2 values were recorded of $63 \mu\text{g}/\text{m}^3$ (PM_{10}), $38 \mu\text{g}/\text{m}^3$ ($\text{PM}_{2.5}$) and $22 \mu\text{g}/\text{m}^3$ (PM_1) in year 1980; 59
3 $\mu\text{g}/\text{m}^3$ (PM_{10}), $41 \mu\text{g}/\text{m}^3$ ($\text{PM}_{2.5}$) and $32 \mu\text{g}/\text{m}^3$ (PM_1) in year 2000. Finally we measured a mean
4 value of $45 \mu\text{g}/\text{m}^3$ (PM_{10}), $34 \mu\text{g}/\text{m}^3$ ($\text{PM}_{2.5}$) and $29 \mu\text{g}/\text{m}^3$ (PM_1) in year 2011. The relative errors
5 with respect to the gravimetric data, reported from previous works, fall within a lower 1% and an
6 upper 10% limit. On the other side, the relative distribution of PM_{10} changed. In particular, the
7 percentages of $\text{PM}_{2.5}$ and PM_1 with respect to PM_{10} mass are: 61% ($\text{PM}_{2.5}$) and 35% (PM_1) in year
8 1980; 70% ($\text{PM}_{2.5}$) and 54% (PM_1) in year 2000; 75% ($\text{PM}_{2.5}$) and 65% (PM_1) in year 2011. This
9 means that, while the aerosol emissions have changed as absolute amount, the relative emission of
10 smaller size aerosol particles has increased over the years. It should not be disregarded that these
11 data are referred to stable good weather conditions, when the effects of scavenging are minimal and
12 the emissions are not removed either by transport (wind) or by wet scavenging (precipitations).
13 Thus, the aerosol deposition is limited to gravity effect in proportion to each single particle mass.
14 While we can observe a general relative reduction of the coarse fraction, the PM_1 have increased
15 from 35% to 65% of the total PM_{10} mass concentrations over the 30-years period considered .
16 In order to provide further insights into the source identification and metal-health relationships, we
17 carried out an ambient aerosol sampling focusing on the characteristics of trace metals in different
18 size fractions. Three sets of size-segregated aerosol samples were collected at the centre of the city
19 of Turin, in very different weather conditions. At the same time, VOC, NO_x and O_3 measurements
20 were carried out in order to better assess the relationship between sources, the pollutant chemistry
21 and the deposition processes.
22 The objectives of this study are: 1) to investigate the enrichment levels of selected trace metals and
23 their size distributions, 2) to identify the major categories of sources contributing to the enrichments
24 of trace metals, 3) to assess the influence of weather conditions on the concentrations of trace
25 metals and their size distributions and 4) to verify their correlation with photochemical and gaseous
26 pollutants.

27

28 **Material and methods**

29 *Sampling Location and size segregated PM collection*

30 Turin is a metropolitan area characterized by a high density of residential and commercial premises
31 and a very high volume of vehicular traffic; several industries, including power plants, chemical
32 plants, plastic and metallurgical factories are located on the outskirts.

33 Aerosol samplings were carried out in autumn and winter 2011-2012 in one site localized near the
34 historical centre ($45^\circ 3' 6.412'' \text{ N}$, $7^\circ 40' 51.919'' \text{ E}$) at an elevation of 15 meter with respect to the

1 road level (top roof with no other nearby buildings creating street canyon) using the 8-stages
2 Andersen MkII nonviable cascade impactor with cut-off at: 11.0, 9.0, 5.85, 4.0, 2.7, 1.6, 0.88 and
3 0.54 μm of particle aerodynamic diameter. A back-up filter was mounted after the last stage.
4 Sampling duration for each sample set was 7 days (168 hours), at a flow of 28.3 Lmin^{-1} (actual
5 conditions) as prescribed by the impactor instructions, for a total volume of sampled air of 284 m^3 .
6 The air flow was maintained with a Tecora mod. Charlie (Milan, Italy) air sampling pump. On each
7 stage a quartz tissue filter (83 mm diameter, PALL tissuquartz 2500QAT-UP) was placed as
8 sampling media to collect the size-segregated particulate samples. The backup filters were of the
9 same material but with a diameter of 43 mm. Before sampling the filters were treated with ultrapure
10 concentrated HCl at reflux for two hours, washed several times with ultrapure water and treated at
11 700 $^{\circ}\text{C}$ for one hour, then stored in dessicator over silica gel for 24 hours and weighed (see below).
12 After each sampling, filters were stored in dessicator over silica gel for 24 hours before weighing,
13 then at -18 $^{\circ}\text{C}$ until chemical analysis. Filters were weighed on a Sartorius BP 211D analytical
14 balance with 0.01 mg resolution by using reference masses with the single pan double substitution
15 method. PM mass was obtained as a difference between the filter after and before sampling. The
16 procedure assures uncertainties of 0.01 – 0.03 mg (95% confidence) on PM mass. Given the volume
17 of sampling and the uncertainty on the flow of the sampling pump (5%), the uncertainties on the
18 PM mass concentration are in the range 0.2 – 1.2 $\mu\text{g m}^{-3}$.
19 Weather conditions (wind speed and direction, temperature, barometric pressure, relative humidity
20 and total solar radiation) were registered at the same site.

21 ***NO_x, Volatile Organic Compounds and O₃ analysis***

22 The analysis of NO_x, O₃ and Volatile Organic Compounds (VOC) were carried out directly on site
23 with no intermediate sampling. O₃ and NO_x were monitored continuously by using a HORIBA
24 ambient O₃ monitor mod APOA-360 and HORIBA NO_x monitor mod APOA-370, respectively.
25 VOC were analyzed sampling 1.0 litre of air with an ENTECH mod. 7000 3-stage preconcentration
26 system with water and CO₂ management (microscale purge and trap technique as suggested by the
27 manufacturer). After preconcentration, the VOCs were detected and quantified with an Agilent Gas-
28 chromatograph mod 6890 equipped with a 60 m, 0.32 mm I.D., 1 μm film thickness CP-SIL 5 CB
29 column (Chrompack) and an Agilent quadrupole mass spectrometer mod 5973. The conditions were
30 as follows: carrier gas 1.5 mL/min, column at 35 $^{\circ}\text{C}$ for 5 min, then gradient of 5.5 $^{\circ}\text{C}/\text{min}$ at 145
31 $^{\circ}\text{C}$, then 40 $^{\circ}\text{C}/\text{min}$ at 245 $^{\circ}\text{C}$ (12 min). The analytical procedure is compliant with the EPA TO-15
32 method. Analytical standards were produced by dynamic dilution (Entech 4600A dynamic diluter,
33 Simi Valley CA, USA) with zero grade air of a Restek (Bellefonte PA, USA) TO-15 65 Component

1 Mix (1 ppmv nominal of each component, including CFC-11, 12, 113 and 114). VOC analysis were
2 performed every two hours during the 7 days PM sampling period.

4 ***Determination of Metals in size fractionated PM***

5 The aerosol-loaded filters were digested by a microwave oven (Milestone, MLS-1200 Mega) in 100
6 ml tetrafluoromethoxyl vessels. The digestion mixture was composed of HNO₃ and H₂O₂ at a ratio
7 of 4:1; nitric acid was purified by sub-boiling distillation and hydrogen peroxide was ultra-pure
8 grade (Sigma-Aldrich). The following heating steps were applied: 1 min at 250W, 2 min at 0 W, 5
9 min at 200 W, 5 min at 350 W, 5 min at 550 W and 5 min at 250 W. The resulting solutions were
10 filtered and diluted to 15 ml with Milli-Q (Millipore) ultrapure water (18.2 MΩ cm).

11 As, Cd, Co, Cr, Cu, Fe, Mn, Mo, Ni, Pb, Pd, Pt, Rh, Sn, V and Zn were determined by a magnetic
12 sector inductively coupled plasma mass spectrometer (SF-ICPMS, Thermo Finnigan Element 2).

13 Mass resolution and isotope selection were optimized for each element to ensure resolution of
14 spectral interferences and maximize sensitivity. A minimum of triplicate 180 s analyses on each
15 sample was conducted following a 60 s uptake and stabilization period. Between samples the
16 nebulizer system was rinsed for 2 min with 2% sub-boiling HNO₃, which eliminated carry-over and
17 reconditioned the sampler cone. Sets of instrumental blank and calibration verification checks were
18 run at frequent intervals during the batch sequence. The calibrations were performed with standard
19 solutions prepared in aliquots of sample blanks. Process blanks were incorporated into the
20 dissolution and analytical procedure to assess metal contribution from the filters, bombs, Milli-Q
21 water and purified acids used in this procedure. All signals for samples were obtained after
22 subtraction of their appropriate process blank values. The relative standard deviation for all
23 elements in each sample was always lower than 5 %.

24 NIST SRM 1648a (Urban Particulate Matter) was used to verify that analyte concentrations were
25 within 15% of the expected values before proceeding with sample analysis.

28 **Results and discussion**

29 ***Weather conditions***

30 In this study three sampling periods, namely in October, January and February, were carried out in
31 different weather conditions. Mean meteorological variables are reported in Table 1, Table 2 and
32 Table 3. Hourly data are reported in the supplementary material (Figure S-1, Figure S-2, Figure S-
33 3).

1 The first (October) sampling has been performed when typical late summer weather was present,
2 with clear skies, high temperature, but with strong temperature gradient between day and night and
3 high night relative humidity with the formation of mists. Concerning wind speed and direction, it is
4 noteworthy that it was under 2.5 m/s for the first four days of sampling whereas it increased on the
5 fifth day, when a strong katabatic wind from WNW carried air masses of alpine origin, decreasing
6 both the temperature (5 °C decrease) and the relative humidity (RH) (from 70 to 44%).

7 The second (January) sampling occurred in early winter climatic conditions. The weather conditions
8 were stable, with temperatures below 0 °C only during the night. During this sampling (in the fourth
9 day) a Foehn episode occurred, increasing temperature, atmospheric mixing and lowering relative
10 humidity. The mean RH dropped from about 80% to 55% with minima under 35%.

11 Both katabatic wind and Foehn caused the transport of aged air masses from WNW.

12 The third (February) sampling took place during the early 2012 european cold wave (World
13 Meteorological Organization, 2012). This period started on 28 January and ended on 17 February
14 and was characterized by extremely cold condition. In the nearby of Turin temperature as low as -20
15 °C were registered, with temperatures continuously under 0 °C. This was one of the most cold
16 periods of the last century (being the other in February 1956 and February 1985). During the first
17 days of cold spell a heavy snow episode occurred (January 28 - February 2), scavenging efficiently
18 airborne particulate matter. Therefore, the weather conditions were significantly different with
19 respect to the previous sampling periods. The temperature remained always below zero with stable
20 weather conditions and sunny with very little wind. Low temperatures delayed the snow melting
21 until the end of sampling, greatly increasing the albedo. As a result, the high actinic flux from the
22 snow cover increased the atmospheric photochemical reactivity at ground level, as well as it is
23 known that snowpack can favour photochemical reactions of pollutants adsorbed onto the
24 hydrometeors (e.g. nitrate) (Grannas et al., 2007; Neemann et al., 2014).

25

26 ***O₃, NO_x***

27 The time profiles of NO_x and O₃ are quite different for the three sampling periods, due to relevant
28 differences in the atmospheric conditions and solar radiation, that alter not only the dilution and
29 transport of pollutants, but also the deposition processes and efficiencies. The time profiles with
30 hourly resolution are reported in the supplementary material (Figure S-4, Figure S-5, Figure S-6).

31 The mean concentrations over the entire sampling periods are reported in Table 4. Po valley is the
32 European region with the highest level of photochemical pollution in summer due to the presence of
33 high density of anthropogenic activities and associated emissions (mainly NO_x), along with the
34 presence of relevant emissions of highly reactive biogenic VOCs from the surrounding alpine

1 regions. Furthermore the particular geographic conformation prevents an efficient dispersion of
2 primary pollutants. In summer 2011 more than 30 exceedances of the $120 \mu\text{g m}^{-3}$ (56 ppbv) O_3 EU
3 limit (as 8 hours running average) occurred in >90% of the measuring stations of Po valley (EEA
4 2013). The October data are typical of medium photochemical processing of in situ emitted
5 pollutants (stable air masses), with mean concentration of NO (10 ppbv) under the mean
6 concentration of NO_2 and O_3 (22.9 and 19.3 ppbv, respectively). NO_2 average level exceeded the
7 EU annual average limit (19.5 ppbv or $40 \mu\text{g m}^{-3}$). Two daily peaks of primary pollutants were
8 observed (see also the next section) during rush hours (early morning and late afternoon).
9 Secondary pollutants (NO_2 and O_3) show two daily peaks as well, but broader and with a lag with
10 respect to the peaks of NO. During the katabatic wind episode, the intrusion of aged air masses
11 from WNW and the dilution of the primary pollutants emitted in the urban context cut down the
12 concentration of NO, keeping it constant and relatively high during all the fifth day.

13 The January data are typical of winter days, with no precipitations, stable air masses with low
14 mixing height, strong temperature inversion and low solar radiation. In these conditions the primary
15 pollutants prevail, indeed the O_3 concentrations are very low with quite high NO concentrations. O_3
16 levels increased from under 2 ppbv to 25 ppbv on the third-fourth days due to the Fohen episode,
17 that transported aged air masses from WNW thereby increasing mixing, strongly lowering NO
18 concentration.

19 The February data are somewhat unusual due to the presence of a stable snow cover in the urban
20 environment and temperature constantly under 0°C . During this period the city was permanently
21 snow covered, except for the road surfaces. As a result, at least 70% of the urban area and more than
22 90% of the surrounding rural areas were covered by snow. Mean concentrations of NO, NO_2 and O_3
23 point out to a discrete photochemical processing of primary pollutants, being NO and NO_2 levels
24 very similar. The time profiles of these pollutants show daily cycles very similar to those observed
25 in October, confirming the photochemical conversion, being the O_3 level limited by the higher level
26 of NO_x and by low levels of humidity, but with a manifest daily maximum in late afternoon. The
27 presence of the snow cover affects the surface albedo, atmospheric circulation and dry deposition
28 efficiencies of pollutants. Similar episodes of high winter ozone in the presence of snow cover were
29 registered in close basins (see for example Neeman et al 2014), due to enhanced boundary layer
30 stability and available solar radiation. Indeed the max solar radiation data are very similar to those
31 of October 2011 (Table 1 and Table 3), but in February the albedo of the snow cover makes available
32 also a relevant intensity of reflected radiation. Moreover, the dry deposition velocities of many
33 pollutants are one order of magnitudes lower for snow covers and frozen soils with respect to soil
34 and sand (Galbally I.E. et al. 1980, De Leeuw et al., 1990). Snow photochemistry of pollutant

1 entrapped in the snow flakes during precipitation makes the snow cover a relevant source of
2 secondary pollutants, including NO_x (Grannas et al., 2007; Neemann et al., 2014). Enhanced
3 production of secondary aerosol is thus expected.

4

5 ***Volatile Organic Compounds***

6 During the three sampling periods the concentration of hexane, isooctane, as well as benzene,
7 toluene, xylenes and ethylbenzenes (BTEX) were determined. These compounds are mainly emitted
8 in the exhaust of gasoline powered motor vehicles (Monod et al, 2001; Schauer et al, 2002).
9 Moreover BTEX are precursors of secondary organic aerosol (Kroll et al. 2008, Bahreini et al,
10 2012, Donahue et al. 2009, Ng N.L et al 2007) and the ozone production in urban environment is
11 often under a VOC limited regime. The time profiles of VOC mixing ratios with two hours
12 resolution are reported in the supplementary material (Figure S-7, Figure S-8, Figure S-9). The
13 average concentrations over the entire sampling periods are reported in Table 5. For comparison the
14 mixing ratios of CFC-11, CFC-12 and CFC-113 are also reported. The time profiles of these
15 compounds are flat (data not reported) and the measured concentrations are slightly higher than the
16 average concentrations reported in the northern hemisphere (Bullister, 2014), assuring the quality of
17 the data.

18 The highest concentrations of VOCs were recorded during the January campaign, and they could be
19 ever higher if the Fohen episode during days 4-5 had not took place, whereas, somewhat
20 surprisingly, the lowest concentrations were observed during the February campaign. The October
21 time profiles (Figure S-7) shows two daily peaks in rush hours, highly correlated with NO time
22 profiles and in advance with respect to ozone peaks, indicating local photochemical processing. As
23 stated in the preceding paragraph, the WNW katabatic wind during days 4-5 caused the intrusion of
24 aged air masses and the drop in primary pollutants mixing ratios. During the January campaign the
25 lower solar radiation and stable atmospheric conditions (except in days 4-5 due to the Fohen),
26 favoured accumulation of primary pollutants. VOCs profiles during the first three days show a
27 broad peak during daytime.

28 The snow cover during February, slowing down the deposition rates of O₃ and increasing the
29 availability of solar radiation at ground, enhanced photochemical reactivity at ground/top roofs
30 level. The time profiles of O₃, NO_x and VOCs were very similar to typical photochemical smog
31 episodes. Thus a relevant production of secondary organic aerosol is expected.

32

33 ***Mass distribution of the PM***

1 Figure 1 shows the gravimetric size distribution of airborne particulate matter collected in February
2 sampling. The other mass distributions for October and January samplings were not determined.
3 PM₁₀ distribution is bimodal, with modes around 5 μm and 1 μm, indicating an input from several
4 sources (vehicular traffic and/or fossil fuel combustion processes, as well as secondary formation)
5 and typical of urban aerosols (Allen et al., 2001; Keywood et al., 1999; Taiwo et al., 2014). It is also
6 present a just outlined mode for PM greater than 10 μm. However, the accumulation mode (around
7 1 μm) is by far the highest. Indeed, by integrating data we note that PM_{2.5} represents already 80% of
8 the total mass of PM collected; this percentage increases to 92% with PM₁₀ while it is only 8% for
9 the largest PM. These results are comparable with PM₁₀ concentrations determined by the
10 monitoring stations of the Italian Regional Agency for Environmental Protection (ARPA) for the
11 same sampling period.

12 This pattern is likely due to the meteorological conditions present during sampling period
13 (temperature constantly under 0°C, low wind speed, presence of a snow cover) reducing vertical
14 circulation, thereby lowering the mixing layer thickness, reducing deposition rates and enhancing
15 production of secondary aerosols, thus promoting accumulation of fine suspended particles at
16 ground level. Moreover, gas to particle conversion of emissions from motor vehicles and from
17 household heating systems is stimulated at low temperature (Mulawa et al., 1997).

18 19 ***Size distribution of elements associated to time integrated PM samples***

20 The average elemental concentrations, expressed in ng/m³, measured in the three sampling periods
21 (October 2011, January 2011 and February 2011) in Turin are reported in Table 6, together with
22 those detected in other Italian and European towns (Vecchi et al., 2007; Dongarrà et al., 2007;
23 Bocca et al., 2003; Toscano et al., 2011; Johansson et al., 2009; Heal et al., 2005; Querol et al.,
24 2001). Most of the elements present lower concentrations in Turin than in other considered urban
25 areas; this is in part due to the decreasing trend associated to the recent European directives for the
26 control of emissions into the atmosphere have been applied for longer times. Taking into account
27 the great variability present in atmospheric conditions, it is however possible to state that the
28 element concentrations in Turin atmospheric particulate matter are similar to those of the other
29 considered towns.

30 The highest metal concentrations in atmospheric PM samples were reached by Cr (0.83 ng/m³), Zn
31 (32.7 ng/m³), Cd (1.85 ng/m³), Sn (30.2 ng/m³) and Pt (0.03 ng/m³) in October sample; Co (0.06
32 ng/m³), Ni (1.55 ng/m³), Cu (2.52 ng/m³), As (0.13 ng/m³), Mo (0.34 ng/m³) and Pb (2.25 ng/m³) in
33 January sample; and V (0.30 ng/m³), Mn (3.10 ng/m³), Fe (207 ng/m³), Rh (0.02 ng/m³) and Pd
34 (0.03 ng/m³) in February sample.

1 The elements having a higher variability among the three sampling periods were vanadium, cobalt,
2 arsenic, rhodium, cadmium and tin, i.e. those that are most likely emitted by combustion processes,
3 with the exception of rhodium. The high variability of this last element, instead, could be an artifact
4 due to its low concentration, close to the instrumental detection limit (0.5 pg/m^3).

5 Figure 2 shows the size distributions of elements for the three sampling periods carried out.

6 First of all, it can be noted that there is a greater difference in element size distribution between the
7 first sampling (October) and the last two (January and February), despite the last sampling having
8 been performed in the days immediately following a heavy snowfall. Evidently the difference
9 existing among the sources influencing the element size distribution of Turin airborne particulate
10 matter in autumn and winter seasons is more significant than the scavenging effect of a snowfall on
11 PM. Indeed, elements, such as Co, V and partially Ni, typical markers of fossil fuels combustion
12 processes, show a larger mode in the finest fraction ($< 1.6 \text{ }\mu\text{m}$) in winter than in autumn. This
13 behaviour can be ascribed both to higher emissions from the combustion of fossil fuels in heating
14 systems and to persistent thermal inversions at ground level occurring in Northern Italy during the
15 winter, as reported in various studies (Marcazzan et al., 2001; Toscano et al., 2011; Malandrino et
16 al., 2013). Conversely, zinc and rhodium present larger mode in the fine fraction ($< 2.7 \text{ }\mu\text{m}$) in
17 autumn; this can probably be attributed to an increased concentration of fine particles emitted into
18 atmosphere from vehicular traffic. Many studies stated that zinc in the air is generated by road dust
19 resuspended by vehicular movement, vehicle exhaust emissions and tyre wear (Pakkanen et al.,
20 2001; Birmili et al., 2006). Rh, instead, together with Pd and Pt, the so-called Platinum-Group
21 Elements (PGEs), is contained in the catalytic converters for automotive traction. Three way
22 catalytic (TWC) converters for Otto cycle engines contain only Pd and Rh in ratios from 2:1 to 5:1.
23 Diesel Oxidation Catalysts (DOC) and Nitrogen Storage Catalysts (NSC) for diesel engines contain
24 principally Pt and Pd (10 to 1-2 and 2:1, respectively) with only traces of Rh. These elements are
25 emitted primarily as particulate matter in consequence of the thermal and mechanical losses of the
26 catalytic material when the engine is working. For this reason, traffic can be considered the main
27 source of PGEs contamination of urban areas (Bocca et al., 2003; Artelt et al., 2000). To confirm
28 this explanation it is worth to note that, if we consider only the Pt size distribution in the fine
29 fraction, also this element presents a higher concentration in the $< 2.7 \text{ }\mu\text{m}$ fraction in the autumn
30 period.

31 A possible explanation for Zn and Rh seasonal size distribution is the greater influence of road dust
32 and vehicular traffic on the fine fraction in autumn owing to the lower humidity in the air in this
33 season than in winter; this can induce a greater suspension and persistence of the particles emitted
34 by vehicles into the air. This assumption is supported by relative humidity (RH%) values recorded

1 in the sampling periods: they are lower in October (average RH% = 63) than in January and
2 February (average RH% = 73 and 76 respectively).

3 The PM₁₀ concentrations of Pt (20 pg m⁻³) and Pd (26 pg m⁻³) in the Turin urban atmosphere are
4 within the ranges (1.3–23 pg m⁻³ for Pt and 0.8–51.4 pg m⁻³ for Pd) reported in urban atmosphere
5 sampled globally since 1998 to 2003 (Rauch et al., 2005). Although Pd and Pt are often below
6 detection limits, these results are in agreement with the data collected by Rauch et al. (2005)
7 showing typical Pt/Pd ranges of 0.2 – 2.9 in urban air particles. Instead, the PM₁₀ concentration of
8 Rh (9 pg m⁻³) exceeds these limits (the highest value reported in Rauch et al. (2005) is 4.2 pg m⁻³).
9 A possible explanation to this behaviour can be due to a higher contribution of this element in
10 atmosphere from crustal sources particularly in winter period; indeed, this element is characterized
11 by a high mode in coarse fraction in both winter samplings.

12 Overall, on the basis of the size distributions, elements could be divided into three groups. Group 1
13 elements are mostly concentrated in coarse mode (2.7 – 11 μm), including Cd, Cr, Cu, Fe, Mn, Mo,
14 Pt and Sn; Group 2 elements show higher concentrations in accumulation mode (< 2.7 μm),
15 including As, Co, Pb and V; Group 3 elements present several modes spread throughout the size
16 distribution including Ni, Pd, Rh and Zn. This behaviour is generally in line with the findings
17 reported by other authors (Wang et al., 2006; Duan et al., 2012, Samara et al., 2006); however, the
18 size distribution of some elements was different in the three sampling periods, as reported above.

19 Enrichment Factors (EFs) of elements in aerosol relative to Earth's crustal composition were
20 calculated according to:

21

$$22 \quad EF = (C_{Me} / C_{Fe})_{\text{sample}} / (C_{Me} / C_{Fe})_{\text{Crust}}$$

23

24 where $(C_{Me}/C_{Fe})_{\text{sample}}$ is the concentration ratio of the metal and Fe in the investigated sample and
25 $(C_{Me}/C_{Fe})_{\text{Crust}}$ is the same ratio in Earth upper-crust as reported by Wedepohl (1995). We used
26 crustal Fe level as a reference since soil is considered to be the major source of Fe in aerosol. These
27 values were employed to assess the extent of non-crustal contributions to the elemental
28 concentration levels for the three groups. However, it must be mentioned that EFs only provide
29 qualitative information on element sources because of the wide variation in concentrations in the
30 upper crust (Contini et al., 2010).

31 As shown in Figure 3, Group 1 elements present different enrichment factors indicating the lack of
32 a common source for all. Indeed, Cr, Fe and Mn show low EFs in both coarse and fine mode.
33 Therefore they are mainly originated from crustal sources. Their percentages in the coarse fraction
34 ranged from 69 % (Mn) to 79 % (Cr) as average values, but they were higher in October and

1 January (from 77 % for Mn to 92 % for Fe) than in February (from 49 % for Mn to 73 % for Cr),
2 likely to indicate a greater effect of the snowfalls on PM having coarser mode ($> 2.7 \mu\text{m}$) and
3 crustal origin.

4 Instead, the EFs of the other Group 1 elements (Cd, Cu, Mo, Pt and Sn) are high both in coarse and
5 (mainly) in accumulation mode, suggesting that their sources are mainly anthropogenic activities
6 and, at lesser degree, soil dust. In particular, these elements might mainly come from the re-
7 suspended soil containing previously deposited PM and vehicle exhaust emissions.

8 Group 2 elements show a higher concentration in fine fraction particularly in winter and, therefore,
9 they should be related to fossil fuel combustion processes, likely in household and industrial heating
10 systems. It is interesting to note that their EFs are not high both in coarse and in accumulation
11 mode, but they increase by approximately an order of magnitude between the two fractions (e.g. for
12 vanadium, the EFs for coarse and fine fractions are 0.3 and 3.6 respectively) to indicate a possible
13 anthropogenic contribution in fine fraction. In more detail, As and Pb are characterized by
14 homogeneous size distribution and similar EFs over time, whereas the size distribution and the EFs
15 for V and Co change over time, more precisely their EFs are higher in January sampling and their
16 concentrations in the accumulation fraction are clearly greater in winter than in autumn. This
17 scenario allows us to assume that these elements derive from different combustion sources. Many
18 researchers have indicated multiple sources for Pb in airborne particulate matter: vehicle emission,
19 steel manufacturing, waste disposal and incineration, non-ferrous metal production, coal and oil
20 combustion (Mishra et al., 2004; Al-Masri et al., 2006; Lim et al., 2010). Regarding As, Yli-Tuomi
21 et al. (2003) reported the use of this element as a marker for non-ferrous metal refinery smelting
22 processes. Moreover Chueinta et al. (2000) assigned As as marker for coal combustion and V for oil
23 combustion. Ni, V and Co are widely used as markers for the combustion of heating fuel (Vallius et
24 al., 2005).

25 Taking this into account it is possible to assume that these elements are emitted in fossil fuel
26 combustion processes by coal and heavy oil inside the industrial complexes (Pb and As) and by
27 heating oil in household heating systems (V and Co).

28 Ni, Pd, Rh and Zn, instead, present a bimodal distribution indicating an input from several sources
29 (local traffic and/or fossil fuel combustion processes). The EFs for Pd, Rh and Zn are very high both
30 in coarse and in accumulation mode: this indicates a predominance of anthropogenic sources for
31 these elements in both fractions. Taking into account that traffic is the main source of pollution by
32 PGEs in populated areas and that the presence of zinc in the air is often associated to vehicular
33 traffic, resuspended road dust, vehicle exhaust emission and tyre wear, it is possible to suppose a
34 common origin for these elements influencing both coarse and accumulation fraction: vehicular

1 traffic and resuspended road dust, containing previously deposited PM. With regard to nickel, even
2 if the size distribution of this element is bimodal, it presents a high EF only in accumulation mode.
3 Therefore, it is possible to suppose that it is emitted by multiple sources: fossil fuel combustion in
4 motor vehicles and in household heating systems to a greater degree and road dust to a lesser
5 degree.

6 Overall, As, Co, Ni, Pb, V and Zn were found in large proportions in particle size below 0.88 μm
7 and could therefore deposit mainly into the alveoli region of human respiratory system. The
8 presence of some among the more toxic elements in the finer PM sizes in all the samples indicates
9 that Turin urban atmospheric PM is capable of causing both proximal and distal contamination, and
10 individual particles are small enough to be easily inhaled into deep lung zones (Romanazzi et al,
11 2014).

12

13 ***Chemometric data analysis***

14 In an effort to better identify the possible sources that influence the chemical composition of PM_{10}
15 and to evidence the impact of weather conditions on the concentrations of trace metals and their size
16 distributions, a chemometric treatment of the experimental data was carried out through the
17 application of a well-known statistical multivariate analysis technique: principal component
18 analysis (PCA). It is an established method for aerosol analysis after its first application to aerosol
19 source apportionment (Thurston and Spengler, 1985). Its strength is that it is based on the evolution
20 of data collected in a specific site, and an a priori knowledge about sources is not required. This
21 data elaboration has been performed with XIStat 2013.2.04 software package.

22 The chemometric study was carried out considering the total data set (27 samples with 15
23 variables). Only the chemical species with concentrations above the detection limit in more than
24 70% of the samples were included in the dataset; therefore Rh was excluded from the multivariate
25 analysis. A value generated applying NIPALS (Nonlinear Iterative Partial Least Squares) method
26 was inserted for concentrations below the detection limit in order to thoroughly apply PCA and
27 HCA without losing any data and, at last, the data set was auto-scaled.

28 Figure 4 a and b show respectively the score and loading plot obtained by PCA. Taking into account
29 the scree plot, we have reported only the first two principal components that contain a total variance
30 of about 50%; all the others PCs contain a low, and therefore not significant, percentage of
31 information. The numbers from 0 to 8 reported beside sampling date in each sample indicate its size
32 range, respectively from 11 to $< 0.54 \mu\text{m}$.

33 As all the samples having size range less to $2.7 \mu\text{m}$ have negative values on PC1, it is possible to
34 hypothesize that this component is related to the size distribution of Turin airborne particulate

1 matter. Instead, as almost all samples collected in February take positive values on PC2, with the
2 exception of the first two samples (Feb 0 and Feb 1), relating to the atmospheric particles having
3 greater diameter, and all the samples collected in October take negative values on PC2, with the
4 exception of the first one (Oct 0), we can hypothesize that the second component is related to
5 sampling period. Overall we observe a differentiation between the samples collected in winter
6 (January and February) and in autumn period (October) and, within each of the two clusters,
7 between the samples having size range less to 2.7 μm and the other ones. This distribution permits
8 to affirm that the element composition of airborne particulate matter of Turin depends on the
9 sampling period: the samples collected in winter are influenced by most of the variables (elements)
10 considered, suggesting a higher input of pollutants due to thermal inversion, to a larger use of
11 household heating systems and to the fact that the conversion of emissions from motor vehicles into
12 particulate matter is stimulated at low temperature (Mulawa et al., 1997). Moreover a clear
13 separation between coarse and fine particles is evident in all the samplings, but it is to be noted that
14 the particle size distribution in the two winter samplings is very similar and differs from that
15 characterizing the autumnal sampling. Furthermore the fine particles collected in winter are
16 characterized by high concentrations of Group 2 elements (As, Co, Ni, Pb and V), which are
17 considered markers of the fossil fuel combustion source, whereas the coarse particles are much
18 more influenced by iron and manganese, typical crustal elements, and by Cu, Mo, Pd and Pt,
19 elements usually connected with vehicular emissions and road dust. Instead, the fine particles
20 collected in autumn are characterized by low loadings for most of the elements considered with the
21 exception of the Oct 8 sample, influenced by the variable identified by the vector of arsenic. This
22 behaviour is a further confirmation of the influence of sampling period discussed previously and of
23 the different source of As, likely linked to fossil fuel combustion processes inside industrial
24 complexes. Finally, the coarse particles of autumn sampling are more scattered than those collected
25 in winter. In particular the Oct 0 sample is strongly influenced by crustal elements (i.e. Fe and Mn)
26 indicating likely a greater input of soil dust transported by wind also from long distance. The other
27 samples are aligned with the direction of Pd, Cd and Sn vectors, having a unquestionable
28 anthropogenic origin, possibly due to vehicular traffic.

29 Regarding variables, it is interesting to note that all the variables related to fossil fuel combustion
30 processes, i.e. Ni, V, Co, Pb and As, are represented by vectors having direction opposite, relative to
31 PC1, to those of most of the other elements to confirm their clearly different sources. Moreover it is
32 possible to observe that the correlations between V and Co and between Pb and As are the highest
33 (Pearson correlation coefficients for $\alpha = 0.05$ are 0.83 and 0.68 respectively) among those
34 calculated for the above reported elements: this confirms different combustion sources for these

1 elements, i.e. household heating systems for V and Ni and industrial combustion processes for Pb
2 and As. Finally the correlation between Co and Ni (Pearson correlation coefficient for $\alpha = 0.05$ is
3 0.60) is possibly due to a combustion source, but the content in Ni is also related to traffic. A
4 correlation between Mn and Fe is evident, suggesting a common origin due to the input of soil dust
5 transported by wind. The correlation among Pt, Pd, Mo and Cu could indicate a common source
6 linked to the emissions by vehicular traffic and road dust. Chromium is partially correlated with
7 both the crustal and anthropogenic elements, suggesting an input from several sources (local traffic
8 and soil dust). The correlation between Cd and Sn is very likely reflecting their anthropogenic
9 origin, possibly due to industrial emissions and/or vehicular traffic. Finally the direction of the
10 vector representing the variable zinc is more difficult to interpret because it is not correlated with
11 other elements: this is likely due to the fact that a well-defined trend is not observable for this
12 element, since it can be emitted by multiple sources (e.g. industrial, road dust and traffic).

13

14 **Conclusions**

15 The concentrations of 16 elements (As, Cd, Co, Cr, Cu, Fe, Mn, Mo, Ni, Pb, Pd, Pt, Rh, Sn, V and
16 Zn) in eight size fractions of urban airborne particulate matter collected in Turin (Italy) were
17 determined during two sampling periods reflecting late summer and winter weather conditions. A
18 third winter measurement period took place after the early 2012 european cold wave, with very
19 unusual conditions (persistent snow cover, temperatures constantly under 0 °C, clear sky and stable
20 atmosphere). The size distribution of elements in all the sampling periods allowed the identification
21 of three main behavioural types: (a) elements found mainly within coarse particles (Cd, Cr, Cu, Fe,
22 Mn, Mo, Pt and Sn); (b) elements found mainly within fine particles (As, Co, Pb and V) and (c)
23 elements with several modes spread throughout the entire size range (Ni, Pd, Rh and Zn).

24 The clustering of the scores obtained by Principal Component Analysis indicates that the element
25 distribution as a function of the PM aerodynamic diameter in Turin strongly depends on the
26 sampling period. Moreover the chemometric analysis confirms that, for elements investigated, it is
27 more evident a separation between the two winter samplings and the autumnal sampling, proving
28 that the metal particle size distribution is less influenced by particular weather phenomena like the
29 strong snowfall happened during February sampling than other atmospheric pollutants. Finally the
30 correlations observed between the inorganic elements permit to infer about their origin.

31 The concentration levels gaseous trace pollutants, namely O₃, NO_x and VOCs, are scarcely
32 correlated with the metal contents of all the size classes of the PM. The differences found in the O₃,
33 NO₂ and VOCs levels of the two winter campaigns due to the relevant photochemical reactivity in
34 the period after the snow episode, do not reflect in differences in the metals distribution in the PM.

1 Since PM metals, NO_x and VOC have common sources, this behaviour can be ascribed to relevant
2 differences in the transformation and deposition processes.

3

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6

7 **Supplementary Data**

8 Additional information as noted in text. Supplementary Information associated with this article can
9 be found, in the online version, at <http://xxxxxxxxxxxxxx>

10

11

12 **References**

13

14 Allen A.G., Nemitz E., Shi J.P., Harrison R.M., Greenwood J.C., 2001. Size distribution of trace
15 metals in atmospheric aerosols in the United Kingdom. *Atmos. Environ.* 35, 4581-4591.

16

17 Al-Masri M.S., Al-Kharfan K., Al-Shamali K., 2006. Speciation of Pb, Cu, and Zn determined by
18 sequential extraction for identification of air pollution sources in Syria. *Atmos. Environ.* 40, 753–
19 761.

20

21 Artelt S., Levsen K., König P., Rosner G., 2000. in *Anthropogenic Platinum-Group Element*
22 *Emissions and Their Impact on Man and Environment*, ed. F. Alt and F. Zereini, Springer-Verlag,
23 Berlin, 33.

24

25 Bahreini R., Middlebrook A. M., de Gouw J. A., Warneke C., Trainer M., Brock C. A., Stark H.,
26 Brown S. S., Dube W. P., Gilman J. B., Hall K., Holloway J. S., Kuster W. C., Perring A. E., Prevot
27 A. S. H., Schwarz J. P., Spackman J. R., Szidat S., Wagner N. L., Weber R. J., Zotter P., Parrish D.
28 D., 2012. Gasoline emissions dominate over diesel in formation of secondary organic aerosol mass.
29 *Geophys. Res. Lett.* 39, L06805.

30

31 Baulig A., Singh S., Marchand A., Schins R., Baroukic R., Garlatti M., Marano F., Baeza-Squiban
32 A., 2009. Role of Paris PM_{2.5} components in the pro-inflammatory response induced in airway
33 epithelial cells. *Toxicology* 261, 126-135.

1
2 Birmili W., Allen A.G., Bary F., Harrison R.M., 2006. Trace metal concentrations and water
3 solubility in size-fractionated atmospheric particles and influence of road traffic. *Environ. Sci.*
4 *Technol.* 40, 1144-1153.
5
6 Bocca B., Petrucci F., Alimonti A., Caroli S., 2003. Traffic-related platinum and rhodium
7 concentrations in the atmosphere of Rome. *J. Environ. Monit.* 5, 563–568.
8
9 Bullister J. 2014, Atmospheric CFC-11, CFC-12, CFC-113, CCl4 and SF6 Histories (1910-2014).
10 http://cdiac.ornl.gov/oceans/new_atmCFC.html. Last accessed 18 february 2015
11
12 Casazza M., Gilli G., Piano A., Alessio S., 2013. Thirty-years assessment of size-fractionated
13 particle mass concentrations in a polluted urban area and its implications for regulatory framework.
14 *J. Environ. Accounting and Management* 1, 259-267.
15
16 Chueinta W., Hopke P.K., Paatero P., 2000. Investigation of sources of atmospheric aerosol at urban
17 and suburban residential areas in Thailand by positive matrix factorization. *Atmos. Environ.* 34,
18 3319–3329.
19
20 Contini D., Genga A., Cesari D., SicilianoM., Donateo A., Bove M.C., Gualcito, M.R., 2010.
21 Characterization and source apportionment of PM10 in an urban background site in Lecce. *Atmos.*
22 *Res.* 95, 40–54.
23
24 De Leeuw F.A.A.M., Van Rheineck Leyssius H.J., 1990 Modeling Study of SO_x and NO_x Transport
25 during the January 1985 smog episode, *Wat. Air Soil Pollution* 51, 357-371.
26
27 Donaldson K., Brown D., Clouter A., Duffin R., MacNee W., Renwick L., Tran L., Stone V, 2002.
28 The pulmonary toxicology of ultrafine particles. *J. Aerosol Med.* 15, 213-220.
29
30 Donahue N.M., Robinson A.L., Pandis S.N., 2009. Atmospheric organic particulate matter: From
31 smoke to secondary organic aerosol. *Atmos Environ* 43, 94-106
32
33 Dongarrà G., Manno E., Varrica D., Vultaggio M. 2007. Mass levels, crustal component and trace
34 elements in PM10 in Palermo, Italy, *Atmos. Environ.* 41, 7977–7986.

1
2 Duan J., Tan J., Wang S., Hao J., Chai F., 2012. Size distributions and sources of elements in
3 particulate matter at curbside, urban and rural sites in Beijing. *J. Environ. Sci.* 24, 87-94.
4
5 Englert N., 2004. Fine particles and human health – a review of epidemiological studies. *Toxicol.*
6 *Lett.* 149, 235-242.
7
8 EEA (European Environment Agency), 2013. Air quality in Europe – 2013 report 9/2013.
9 Copenhagen: European Environment Agency, ISBN 978-92-9213-406-8, ISSN 1725-9177,
10 doi:10.2800/92843. <http://www.eea.europa.eu/publications/air-quality-in-europe-2013>, last accessed
11 16 february 2015.
12
13 Galbally I.E., Roy C.R., 1980. Destruction of ozone at the earth's surface. *Quarterly J. Royal*
14 *Meteorol. Soc.* 106, 599-620
15
16 Grannas A.M., Jones A.E., Dibb J., Ammann M., Anastasio C., Beine H.J., Bergin M., Bottenheim
17 J., Boxe C.S., Carver G., Chen G., Crawford J.H., Dominé F., Frey M.M., Guzmán M.I., Heard D.E.,
18 Helmig D., Hoffmann M.R., Honrath R.E., Huey L.G., Hutterli M., Jacobi H.W., Klán P., Lefer B.,
19 McConnell J., Plane J., Sander R., Savarino J., Shepson P.B., Simpson W.R., Sodeau J.R., von
20 Glasow R., Weller R., Wolff E.W., Zhu T., 2007. An overview of snow photochemistry: evidence,
21 mechanisms and impacts, *Atmos. Chem. Phys.*, 7, 4329–4373.
22
23 Handler M., Puls C., Zbiral J., Marr I., Puxbaum H., Limbeck A., 2008. Size and composition of
24 particulate emissions from motor vehicles in the Kaisermuhlen-Tunnel, Vienna. *Atmos. Environ.* 42,
25 2173-2186.
26
27 Heal M.R., Hibbs L.R., Agius R.M., Berland I.J., 2005. Total and water-soluble metal content of
28 urban background PM10, PM2.5 and black smoke in Edinburgh, UK. *Atmos. Environ.* 39, 1417–
29 1430.
30
31 Johansson C., Norman M., Burman L., 2009. Road traffic emission factors for heavy metals. *Atmos.*
32 *Environ.* 43, 4681–4688.
33

1 Katsouyanni K., Touloumi G., Samoli E., 2001. Confounding and effect modification in the short-
2 term effects of ambient particles on total mortality: results from 29 European cities within the
3 ALPHEA2 Project. *Epidemiology* 12, 521-531.
4

5 Keywood M.D., Ayers G.P., Gras J.L, Gillett R.W., Cohen D.D., 1999. Relationships between size
6 segregated mass concentration data and ultrafine particle number concentrations in urban areas.
7 *Atmos. Environ.* 33, 2907-2913
8

9 Krewski D., Rainham D., 2007. Ambient Air Pollution and Population Health: Overview. *J. Toxicol.*
10 *Env. Heal. A*, 70, 275–283.
11

12 Kroll J.H., Seinfeld J.H., 2008. Chemistry of secondary organic aerosol: Formation and evolution of
13 low-volatility organics in the atmosphere. *Atmos. Environ.* 42, 3593–3624
14

15 Lall R., Kendall M., Ito K., Thurston G.D., 2005. Estimation of historical PM_{2.5} exposures for
16 health effects assessment. *Atmos. Environ.*, 38, 5217-5226.
17

18 Lim J.M., Lee J.H., Moon J.H., Chung Y.S., Kim K.H., 2010. Source apportionment of PM₁₀ at a
19 small industrial area using Positive Matrix Factorization. *Atmos. Res.* 95, 88–100.
20

21 Malandrino M., Di Martino M., Ghiotti G, Geobaldo F., Grosa M.M., Giacomino A., Abollino O.,
22 2013. Inter-annual and seasonal variability in PM₁₀ samples monitored in the city of Turin (Italy)
23 from 2002 to 2005. *Microchem. J.* 107, 76–85.
24

25 Marcazzan G.M., Vaccaro S., Valli G, Vecchi R., 2001. Characterization of PM₁₀ and PM_{2.5}
26 particulate matter in the ambient air of Milan (Italy). *Atmos. Environ.* 35, 4639–4650.
27

28 Masiol M., Squizzato S., Ceccato D., Pavoni B., 2015. The size distribution of chemical elements of
29 atmospheric aerosol at semi-rural coastal site in Venice (Italy). The role of atmospheric circulation.
30 *Chemosphere*, 119, 400-406.
31

32 Maté T., Guaita R., Pichiule M., Linares C., Diaz J., 2010. Short-term effect of fine particulate
33 matter (PM_{2.5}) on daily mortality due to diseases of the circulatory system in Madrid (Spain). *Sci.*
34 *Tot. Env.*, 408, 5750-5757.

1
2 Mishra V.K., Kim K.H., Kang C.H., Choi K.C., 2004. Winter time sources and distribution of
3 airborne lead in Korea. *Atmos. Environ.* 38, 2653–2664.
4
5 Monod A., Sive B.C., Avino P., Chen T., Blake D.R., Rowland F.S., 2001. Monoaromatic
6 compounds in ambient air of various cities: a focus on correlations between the xylenes and
7 ethylbenzene. *Atmos. Environ.* 35, 135-149.
8
9 Mulawa P.A., Cadle S.H., Knapp K., Zweidinger R., Snow R., Lucas R., Goldbach J., 1997. Effect
10 of ambient temperature and E-10 fuel on primary exhaust particulate matter emissions from light-
11 duty vehicles. *Environ. Sci. Technol.* 31, 1302–1307.
12
13 Neemann E.M., Crosman E.T., Horel J.D., Avey L., 2014. Simulations of a cold-air pool associated
14 with elevated wintertime ozone in the Uintah Basin, Utah. *Atmos. Chem. Phys. Discuss.*, 14,
15 15953–16000.
16
17 Ng, N.L., Kroll, J.H., Chan, A.W.H., Chhabra, P.S., Flagan, R.C., Seinfeld, J.H., 2007. Secondary
18 organic aerosol formation from m-xylene, toluene, and benzene. *Atmospheric Chemistry and
19 Physics* 7, 3909–3922.
20
21 Pakkanen T., Loukkola K., Kohonen C., Aurela M., Mäkelä T., Hillamo R., Aarnio P., Koskentalo
22 T., Kousa A., Maenhaut W., 2001. Sources and chemical composition of atmospheric fine and
23 coarse particles in the Helsinki area. *Atmos. Environ.* 35, 5381–5391.
24
25 Querol X., Alastuey A., Rodriguez S., Plana F., Ruiz C.R., Cots N., Massagué, G., Puig, O., 2001.
26 PM10 and PM2.5 source apportionment in the Barcelona Metropolitan area, Catalonia, Spain.
27 *Atmos. Environ.* 35, 6407–6419.
28
29 Rauch S., Hemond H., Peucker-Ehrenbrink B., Ek K.H., Morrison G., 2005. Platinum group
30 element concentrations and osmium isotopic composition in urban airborne particles from Boston,
31 Massachusetts. *Environ. Sci. Technol.* 39, 9464-9470.
32

1 Ramgolan K., Chevaillier S., Marano F., Baeza-Squiban A., Martinon L., 2008. Proinflammatory
2 effect of fine and ultrafine particulate matter using size-resolved urban aerosols from Paris.
3 *Chemosphere* 72, 1340-1346.
4

5 Romanazzi V., Casazza M., Malandrino M., Maurino V., Piano A., Schilirò T., Gilli G., 2014. PM10
6 size distribution of metals and environmental-sanitary risk analysis in the city of Torino.
7 *Chemosphere* 112, 210–216.
8

9 Samara C., Voutsas D., 2006. Size distribution of airborne particulated matter and associated heavy
10 metals in the roadside environment. *Chemosphere* 59, 1197-1206.
11

12 Sánchez de la Campa A.M., Moreno T., de la Rosa J., Alastuey A., Querol X., 2011. Size
13 distribution and chemical composition of metalliferous stack emissions in the San Roque petroleum
14 refinery complex, southern Spain. *J. Hazard. Mater.* 190, 713-722.
15

16 Schauer J.J., Kleeman M.J., Cass G.R., Simoneit B.R., 2002. Measurement of Emissions from Air
17 Pollution Sources. 5. C1-C32 Organic Compounds from Gasoline-Powered Motor Vehicles.
18 *Environ. Sci. Technol.* 36, 1169-1180
19

20 Song F., Gao Y., 2011. Size distributions of trace elements associated with ambient particular matter
21 in the affinity of a major highway in the New Jersey-New York metropolitan area. *Atmos. Environ.*
22 45, 6714-6723.
23

24 Taiwo A.M., Beddows D.C.S., Shi Z., Harrison R.M., 2014. Mass and size distribution of
25 particulate matter components: comparison of an industrial site and an urban background site. *Sci.*
26 *Tot. Env.* 475, 29-38.
27

28 Thurston G.D., Spengler J.D., 1985. A quantitative assessment of source contributions to inhalable
29 particulate matter pollution in metropolitan Boston. *Atmos. Environ.* 19, 9–25.
30

31 Toscano G., Moret I., Gambaro A., Barbante C., Capodaglio G., 2011. Distribution and seasonal
32 variability of trace elements in atmospheric particulate in the Venice Lagoon. *Chemosphere*, 85,
33 1518-1524.
34

- 1 Vallius M., Janssen N.A.H., Heirich J., Hoek G., Ruuskanen J., Cyrys J., Grieken R.V., Hartog J.J.,
2 Kreyling W.G., Pekkanen J., 2005. Sources and elemental composition of ambient PM_{2.5} in three
3 European cities. *Sci. Tot. Env.* 337, 147–162.
- 4
- 5 Vecchi R., Marcazzan G., Valli G., 2007. A study on nighttime-daytime PM₁₀ concentration and
6 elemental composition in relation to atmospheric dispersion in the urban area of Milan (Italy),
7 *Atmos. Environ.*, 41, 2136–2144.
- 8
- 9 Wang X.L., Sato T., Xing B.S., 2006. Size distribution and anthropogenic sources apportionment of
10 airborne trace metals in Kanazawa, Japan. *Chemosphere*, 65, 2440–2448.
- 11
- 12 Wedepohl, K.H., 1995. The composition of the continental crust. *Geochim. Cosmochim. Acta* 59,
13 1217–1232.
- 14
- 15 World Health Organization Europe, 2006. *Air Quality Guidelines -Global Update 2005*. Geneva:
16 World Health Organization.
- 17
- 18 World Meteorological Organization, 2012. Cold spell in Europe and Asia in late winter 2011/2012.
19 https://www.wmo.int/pages/mediacentre/news/documents/dwd_2012_report.pdf. Accessed 15
20 February 2015.
- 21
- 22 Yli-Tuomi, T., Venditte, L.,Hpoke, P.K., Basunia,M.S., Landsberger, S., Viisanen, Y., Paatero, J.,
23 2003. Composition of the Finnish Arctic aerosol: collection and analysis of historic filter samples.
24 *Atmos. Environ.* 37, 2355–2364.

27 **Table and Figure Captions**

28

29 **Table 1.** Daily mean micrometeorological variables during the first sampling period (3-10 October
30 2011). The weather was characterized by stable atmosphere, clear sky (cloud cover 2/8 on 6
31 October). Due to high nocturnal RH, in the days 03-06 October during the night and in the morning
32 the visibility was low due to fog formation at ground. On 7 October (hours 96-120) an episode of
33 katabatic wind from W-NW, caused RH and temperature reduction, with a slow rebound in the
34 subsequent days, and a increase in visibility.

1

2 **Table 2.** Daily mean micrometeorological variables during the second sampling period (17-24
3 January 2012). The weather was characterized by stable atmosphere, clear sky. Fog formation at
4 ground during the first three days. On 20-21 January an episode of Foehn (W-NW) occurred;
5 temperatures, mixing, transport and visibility increased.

6

7 **Table 3.** Daily mean micrometeorological variables during the third sampling period (03-10
8 February 2012), after snow. The weather was characterized by stable atmosphere, clear sky, except
9 on 07 February. Fog formation at ground.

10

11 **Table 4.** Mean concentrations and standard deviation of NO, NO₂ and O₃ during the three sampling
12 periods. EU limits are reported for comparison (8-hour daily average limit for O₃ (120 µg m⁻³,
13 corresponding to 56 ppbv), annual average for NO₂ (40 µg m⁻³, corresponding to 19.5 ppbv)).

14

15 **Table 5.** Mean concentrations and their standard deviation of Hexane, Isooctane, BTEX and
16 selected CFCs during the three sampling periods.

17

18 **Table 6.** Mean concentration in ng m⁻³ measured in the three sampling periods (October 2011,
19 January 2011 and February 2011) in Turin city and mean concentration observed in some Italian
20 and European urban areas.

21

22 **Figure 1.** Size resolved gravimetric distribution of airborne particulate matter sampled in February
23 with a eight stages Andersen MkII nonviable cascade impactor.

24

25 **Figure 2.** Element size distributions during October (A), January (B) and February (C) sampling
26 periods.

27

28 **Figure 3.** Enrichment Factors (EFs) of the elements calculated as average values for three sampling
29 periods considered. Coarse fraction: 2.7 – 11 µm; fine fraction: < 2.7 µm.

30

31 **Figure 4.** Score (a) and loading (b) plot on PC1-PC2 for all sampling periods.

Tables And Figures

Day [dd/mm/yyyy]	Mean Pressure [hPa]	Temperature (T_{\min} - T_{\max}) [C°]	RH (RH_{\min} - RH_{\max}) [%]	Mean wind speed (max speed) [m/s]	Max Total solar radiation [W/m ²]	Visibility
03/10/2011	1023	21.4 (16.2 - 27.8)	68 (48 - 86)	1.1 (2.5)	497	Low
04/10/2011	1021	21.6 (16.5 - 28.3)	69 (46 - 88)	1.1 (2.8)	510	Low
05/10/2011	1020	21.2 (16.5 - 27.3)	72 (53 - 85)	1.2 (2.5)	467	Low
06/10/2011	1015	21.3 (17.9 - 25.3)	70 (55 - 85)	1.7 (3.7)	436	Low
07/10/2011	1009	18.9 (12.8 - 22.0)	44 (21 - 87)	3.5 (11.4)	559	Good
08/10/2011	1012	15.4 (8.9 - 21.3)	44 (31 - 66)	2.8 (9.8)	583	Good
09/10/2011	1017	14.0 (8.4 - 20.8)	55 (36 - 79)	1.2 (3.7)	519	Good
10/10/2011	1020	15.1 (9.2 - 22.4)	57 (37 - 78)	1.4 (3.2)	541	Good

2

3 **Table 1. Daily mean micrometeorological variables during the first sampling period (3-10 october 2011).**
4 **The weather was characterized by stable atmosphere, clear sky (cloud cover 2/8 on 6 october). Due to**
5 **high nocturnal RH, in the days 03-06 october during the night and in the morning the visibility was low**
6 **due to fog formation at ground. On 7 october (hours 96-120) an episode of katabatic wind from W-NW,**
7 **caused RH and temperature reduction, with a slow rebound in the subsequent days, and a increase in**
8 **visibility.**

9

1

Day [dd/mm/yyyy]	Mean Pressure [hPa]	Temperature (T_{\min} - T_{\max}) [C°]	RH (RH_{\min} - RH_{\max}) [%]	Mean wind speed (max speed) [m/s]	Max Total solar radiation [W/m ²]	Visibility
17/01/12	1026	-0.5 (-4.3 – 5.5)	81 (61 - 87)	0.8 (2.3)	371	Low
18/01/12	1030	-0.5 (-4.2 – 4.4)	80 (46 - 88)	0.8 (2.0)	295	Low
19/01/12	1026	0.6 (-3.0 – 4.6)	76 (65 - 83)	0.6 (2.3)	219	Low
20/01/12	1014	7.9 (-0.9 – 17.0)	55 (30 - 86)	4.1 (11.0)	386	Good
21/01/12	1013	8.3 (3.9 – 13.4)	57 (37 - 82)	1.7 (11.4)	298	Good
22/01/12	1009	7.8 (2.7 - 15.3)	70 (46 - 86)	0.9 (3.8)	352	Good
23/01/12	1012	5.3 (2.6 – 10.6)	82 (63 - 89)	0.9 (3.0)	282	Good
24/01/12	1014	4.8 (1.9 – 11.8)	79 (53 - 89)	1.0 (12.5)	355	Good

2

3 **Table 2. Daily mean micrometeorological variables during the second sampling period (17-24 January**
4 **2012). The weather was characterized by stable atmosphere, clear sky. Fog formation at ground during**
5 **the first three days. On 20-21 January an episode of Foehn (W-NW) occurred; temperatures, mixing,**
6 **transport and visibility increased.**

7

1

Day [dd/mm/yyyy]	Mean Pressure [hPa]	Temperature (T _{min} - T _{max}) [C°]	RH (RH _{min} - RH _{max}) [%]	Mean wind speed (max speed) [m/s]	Max Total solar radiation [W/m ²]	Visibility
03/02/2012	1018	-2.5 (-5.1 – 0.0)	72 (56 - 82)	0.7 (3.2)	408	Low
04/02/2012	1019	-4.6 (-8.0 – -1.1)	72 (60 - 82)	0.9 (2.4)	232	Low
05/02/2012	1022	-4.8 (-8.6 – 0.0)	72 (59 - 82)	1.0 (2.5)	418	Low
06/02/2012	1017	-6.8 (-12.3 – -0.1)	73 (52 - 83)	1.1 (4.5)	516	Low
07/02/2012	1014	-5.2 (-12.5 – -0.8)	71 (55 - 84)	1.1 (5.0)	329	Low
08/02/2012	1024	1.1 (-1.8 – 4.8)	70 (48 - 87)	0.8 (3.2)	427	Low
09/02/2012	1019	-1.7 (-7.2 – 2.5)	76 (59 - 86)	0.8 (3.0)	486	Low
10/02/2012	1021	-1.9 (-3.6 – -0.2)	72 (48 - 85)	1.3 (4.1)	399	Low

2

3 **Table 3. Daily mean micrometeorological variables during the third sampling period (03-10 February**
4 **2012), after snow. The weather was characterized by stable atmosphere, clear sky, except on 07 February.**
5 **Fog formation at ground.**

6

	Mean Concentrations and their Standard Deviation over the entire sampling periods, ppbv			EU Limits, ppbv
	October	January	February	
NO	10.1 ± 16.3	87.3 ± 63.5	49.5 ± 42.2	
NO ₂	22.9 ± 14.7	40.5 ± 15.3	48.5 ± 12.5	19.5
O ₃	19.3 ± 15.9	3.3 ± 5.2	5.7 ± 7.1	56.0

7 **Table 4. Mean concentrations and their standard deviations of NO, NO₂ and O₃ during the three sampling**
8 **periods. EU limits are reported for comparison (8-hour daily average limit for O₃ (120 µg m⁻³,**
9 **corresponding to 56 ppbv), annual average for NO₂ (40 µg m⁻³, corresponding to 19.5 ppbv))**

1

	Mean Concentrations and their Standard Deviations over the entire sampling periods, ppbv		
	October	January	February
Hexane	0.24 ± 0.15	0.24 ± 0.20	0.08 ± 0.04
Isooctane	0.12 ± 0.10	0.12 ± 0.09	0.05 ± 0.05
Benzene	0.72 ± 0.57	2.20 ± 1.49	1.15 ± 0.55
Toluene	2.50 ± 2.04	3.42 ± 2.50	1.44 ± 0.87
Ethylbenzene	0.20 ± 0.16	0.30 ± 0.27	0.11 ± 0.08
m and p-Xylene	0.56 ± 0.47	0.81 ± 0.24	0.30 ± 0.29
o-Xylene	0.26 ± 0.22	0.35 ± 0.31	0.12 ± 0.09
CFC12	0.62 ± 0.09	0.60 ± 0.03	0.59 ± 0.05
CFC11	0.29 ± 0.04	0.30 ± 0.02	0.27 ± 0.01
CFC113	0.09 ± 0.005	0.09 ± 0.003	0.09 ± 0.004

2

3 **Table 5. Mean concentrations and their standard deviations of Hexane, Isooctane, BTEX and selected CFCs**
4 **during the three sampling periods.**

5

	As	Co	Cd	Cr	Cu	Fe	Mn	Mo	Ni	Pb	Pd	Pt	Rh	Sn	V	Zn
October	0.05	0.02	1.85	0.83	2.44	105	2.50	0.28	1.17	1.53	0.03	0.03	0.003	30.2	0.08	32.7
January	0.03	0.04	0.35	0.73	1.88	207	3.10	0.22	1.05	1.88	0.03	0.02	0.016	5.32	0.30	17.8
February	0.13	0.06	0.80	0.54	2.52	110	2.22	0.34	1.55	2.25	0.02	0.02	0.007	7.82	0.27	21.0
Milan ^a	-	-	-	13	72	1830	45		10	71						180
Palermo ^b	1.5	0.3		6.5	49	496	12	3.9	5.5	18					20	48
Rome ^c												0.02	0.004			
Venice ^d	3		1.9		8	247	14		6	16					6	47
Stockholm ^e	0.88	0.15	0.11	2.3	7.7		5.5	1.6	2.3	3.4				14	1.6	17
Edinburgh ^f	0.37		0.34	1.60	4.93	183	2.94		3.43	14.1					1.14	13.3
Barcelona ^g				6.0	74		24		7.0	149					13	250

6

7 ^a Vecchi et al. (2007).

8 ^b Dongarrà et al. (2007).

9 ^c Bocca et al. (2003)

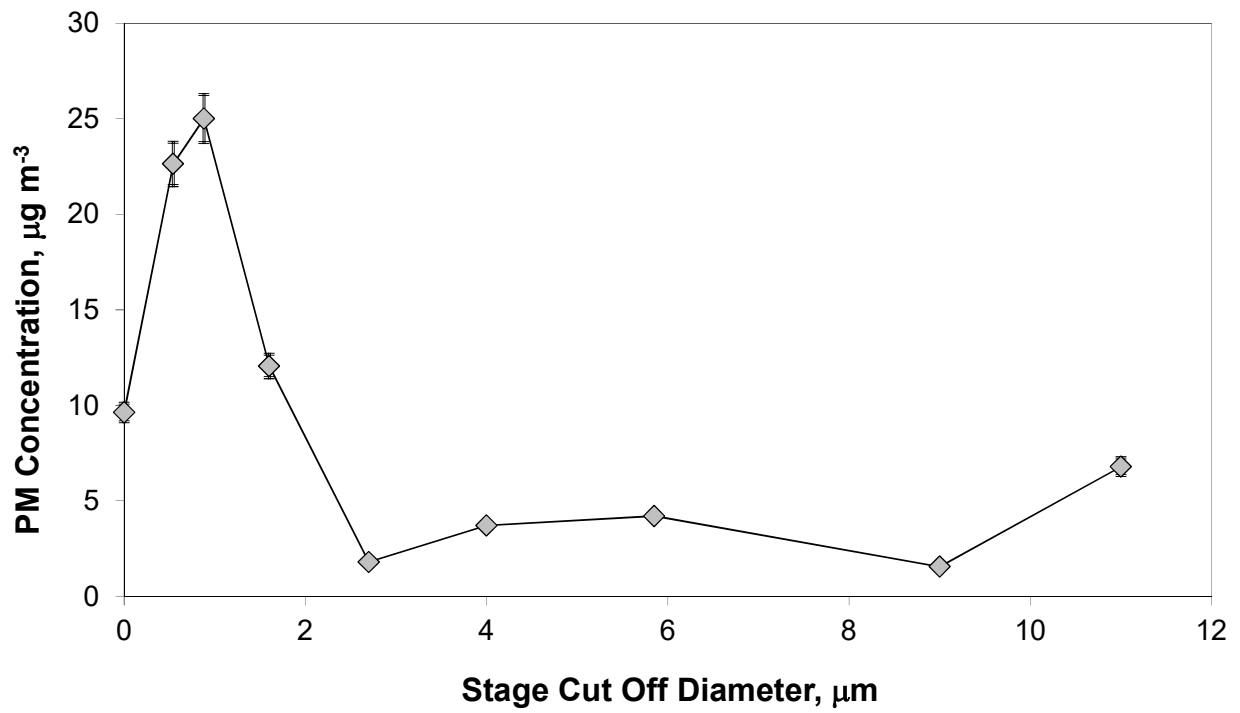
10 ^d Toscano et al. (2011).

11 ^e Johansson et al. (2009).

12 ^f Heal et al. (2005).

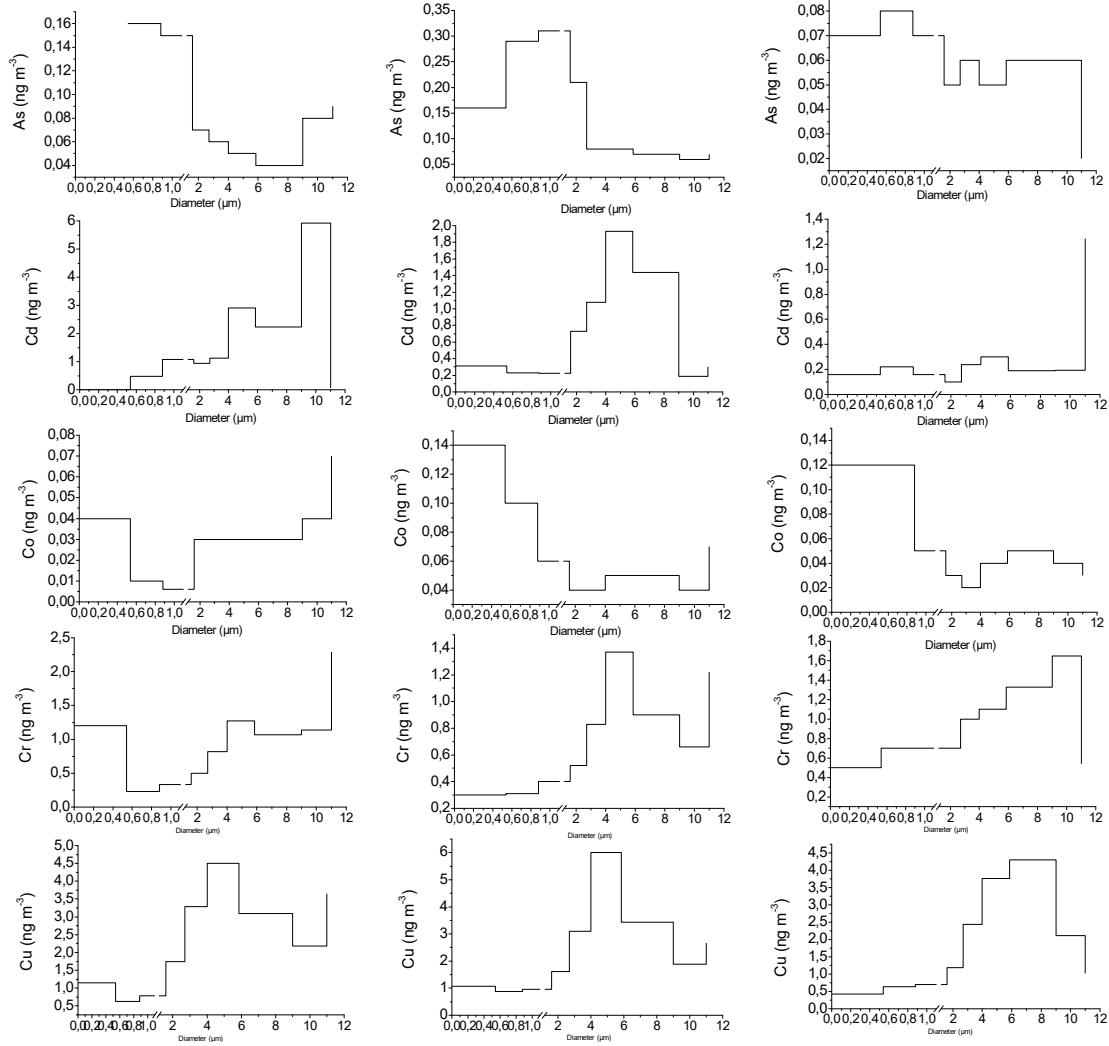
13 ^g Querol et al. (2001).

14 **Table 6. Mean concentration in ng m⁻³ measured in the three sampling periods (October 2011, January**
15 **2011 and February 2011) in Turin city and mean concentration observed in some Italian and European**
16 **urban areas.**

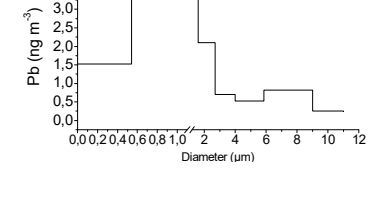
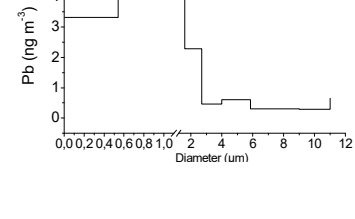
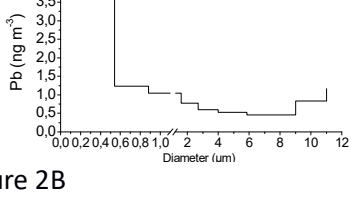
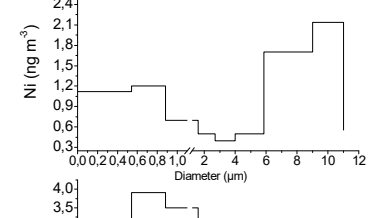
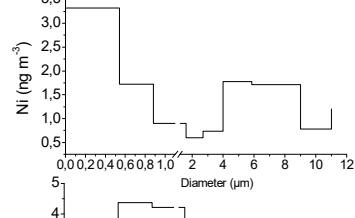
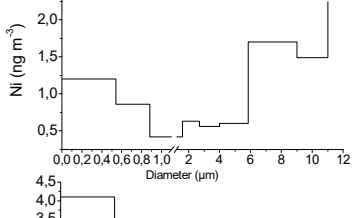
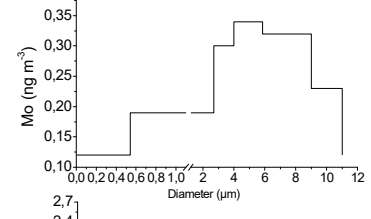
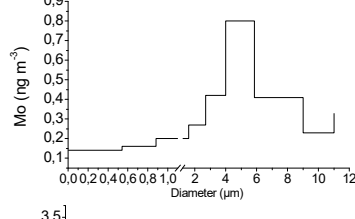
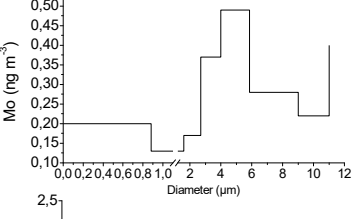
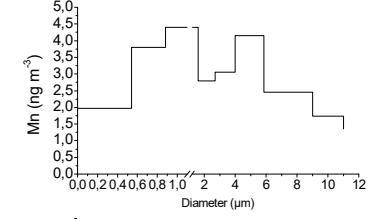
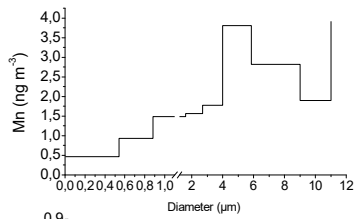
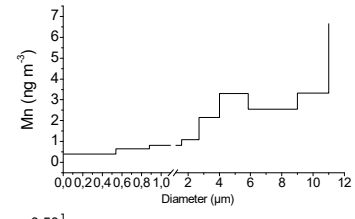
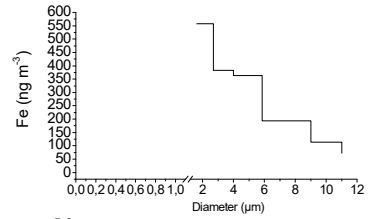
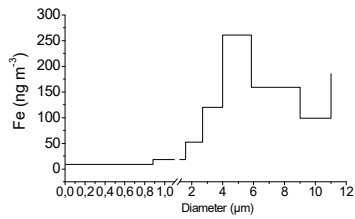
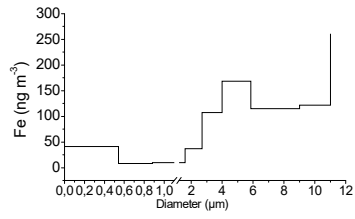


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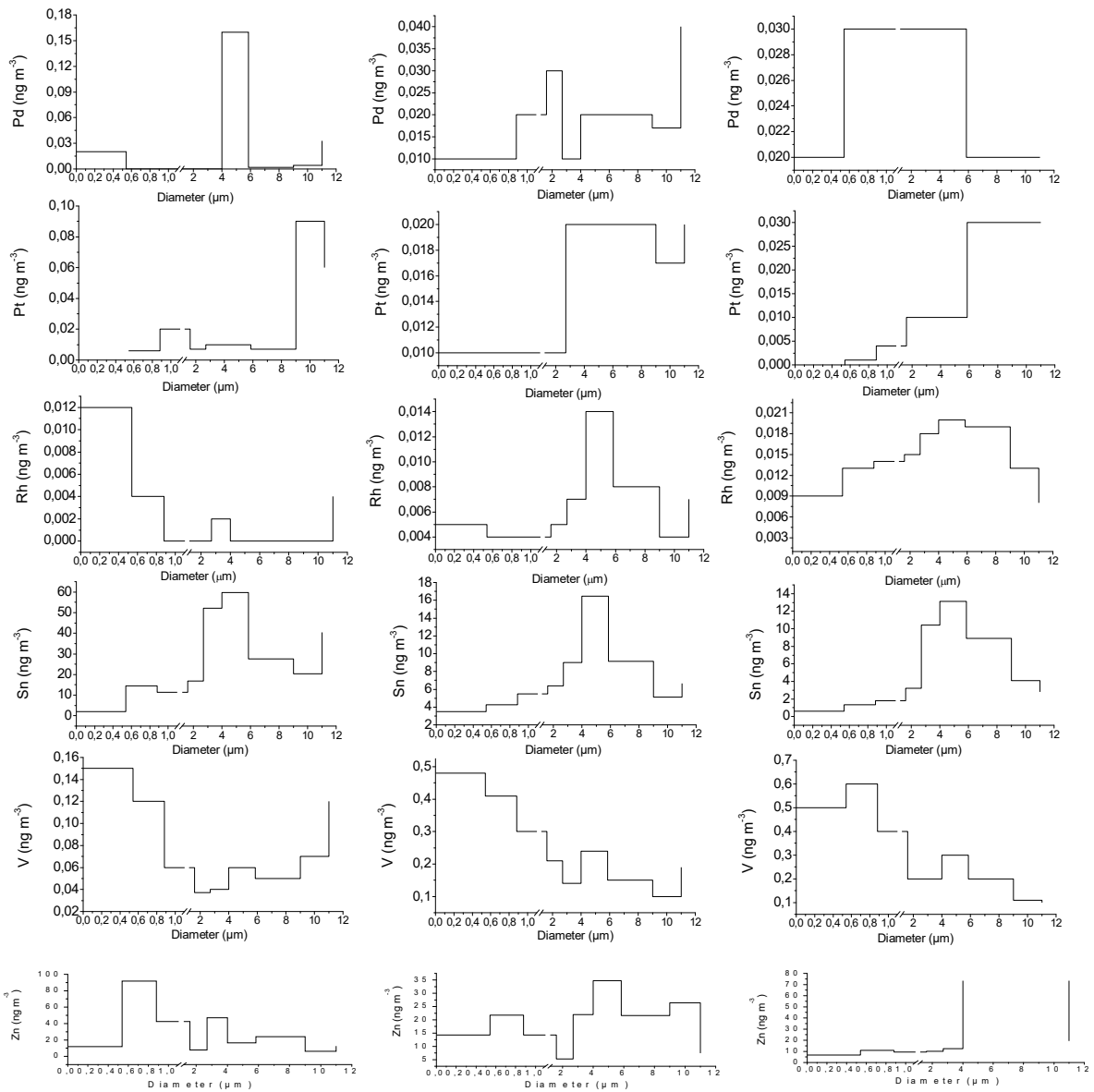
2 **Figure 1. Size resolved gravimetric distribution of airborne particulate matter sampled in February with a**
3 **eight stages Andersen MkII nonviable cascade impactor.**



1
2 Figure 2A



1
2 Figure 2B



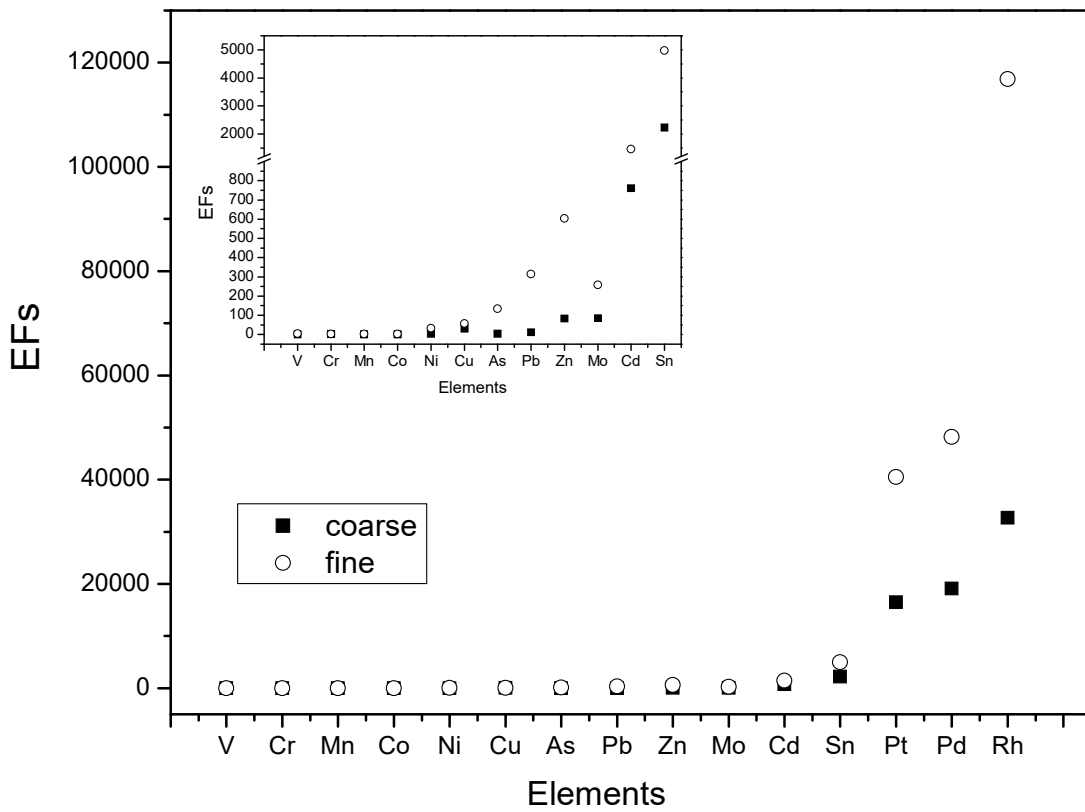
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2

3 Figure 2C

4 Figure 2. Element size distributions during October (A), January (B) and February (C) sampling periods.

5

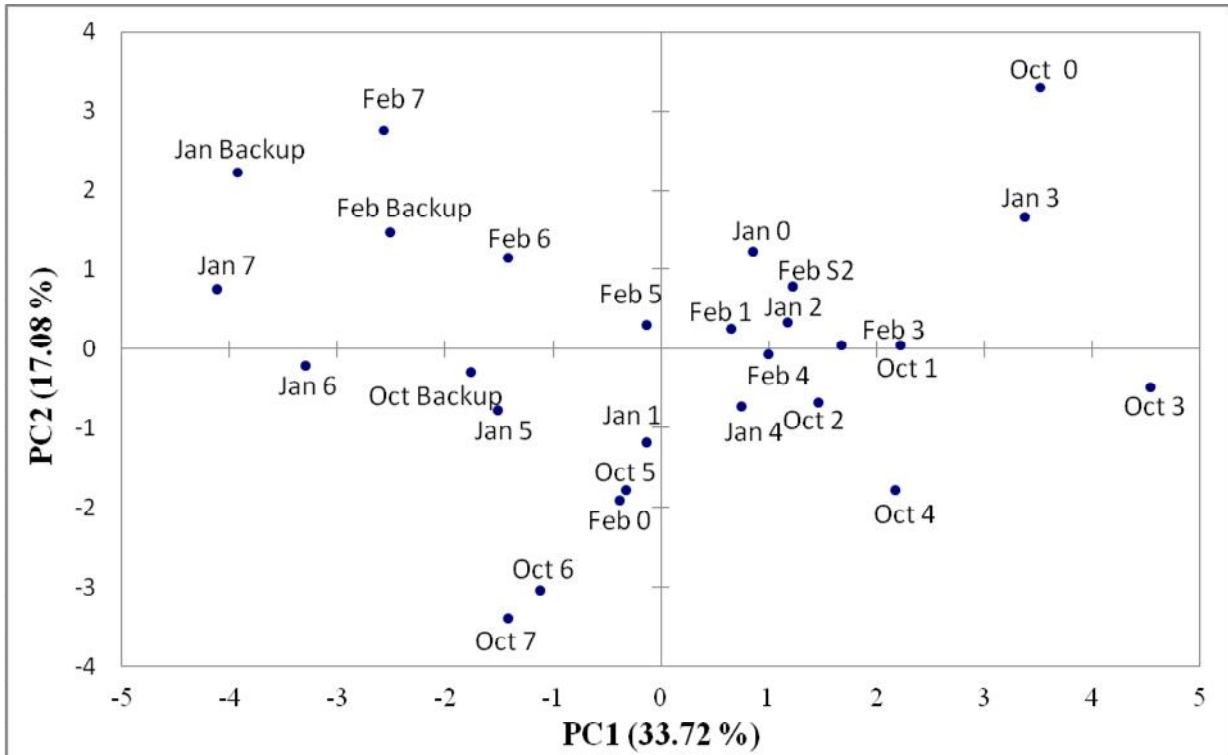


1

2 **Figure 3. Enrichment Factors (EFs) of the elements calculated as average values for three sampling**
 3 **periods considered. Coarse fraction: 2.7 – 11 μm ; fine fraction: < 2.7 μm .**

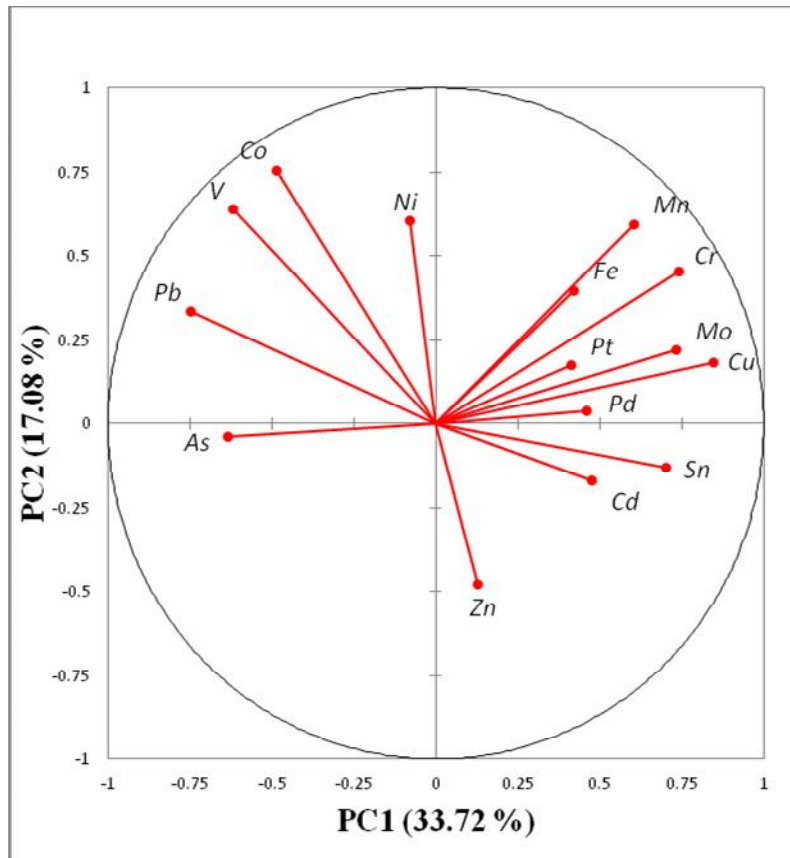
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a)



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b)



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4

5 Figure 4. Score (a) and loading (b) plot on PC1-PC2 for all sampling periods.

6