Supplementary Material for

Formamide adsorption at the amorphous silica surface: a combined experimental and computational approach

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Detailed structural information for the adducts commented in the main text are provided hereafter (Figures and Tables S1 to S12)



Figure S1. Relaxed geometry of the SiO₂–FA1 model: **(A)** top view; **(B)** side view (along *a* axis); and **(C)** detail of specific interaction and key distances (given in Å). Formamide oxygen is reported in pink, carbon in dark yellow, nitrogen in blue, hydrogen in gray, silicon in bright yellow, SiO₂ oxygen in red.

Table S1. Key structural and spectroscopic features for the SiO₂–FA1 model. Distances are given in Å, frequencies in cm⁻¹. The v(C=O) and δ (NH₂) frequencies have been scaled with separated scale factors.

SiO ₂ –FA1	
d(C=O)	1.235
v(C=O)	1704
δ(NH ₂)	1582



Figure S2. Relaxed geometry of the SiO₂–FA2 model: **(A)** top view; **(B)** side view (along *a* axis); and **(C)** detail of specific interaction and key distances (given in Å). Formamide oxygen is reported in pink, carbon in dark yellow, nitrogen in blue, hydrogen in gray, silicon in bright yellow, SiO₂ oxygen in red.

Table S2. Key structural and spectroscopic features for the SiO₂–FA2 model. Distances are given in Å, frequencies in cm⁻¹. The v(C=O) and δ (NH₂) frequencies have been scaled with separated scale factors.

SiO ₂ –FA2	
d(C=O)	1.226
ν(C=O)	1724
δ(NH ₂)	1591



Figure S3. Relaxed geometry of the SiO₂–FA3 model: **(A)** top view; **(B)** side view (along *a* axis); and **(C)** detail of specific interaction and key distances (given in Å). Formamide oxygen is reported in pink, carbon in dark yellow, nitrogen in blue, hydrogen in gray, silicon in bright yellow, SiO₂ oxygen in red.

Table S3. Key structural and spectroscopic features for the SiO₂–FA3 model. Distances are given in Å, frequencies in cm⁻¹. The v(C=O) and δ (NH₂) frequencies have been scaled with separated scale factors.

SiO ₂ –FA3	
d(C=O)	1.233
v(C=O)	1711
δ(NH ₂)	1600



Figure S4. Relaxed geometry of the SiO₂–FA4 model: **(A)** top view; **(B)** side view (along *a* axis); and **(C)** detail of specific interaction and key distances (given in Å). Formamide oxygen is reported in pink, carbon in dark yellow, nitrogen in blue, hydrogen in gray, silicon in bright yellow, SiO₂ oxygen in red.

Table S4. Key structural and spectroscopic features for the SiO₂–FA4 model. Distances are given in Å, frequencies in cm⁻¹. The v(C=O) and δ (NH₂) frequencies have been scaled with separated scale factors.

SiO ₂ –FA4	
d(C=O)	1.242
ν(C=O)	1684
$\delta(NH_2)$	1603



Figure S5. Relaxed geometry of the SiO₂–FA5 model: **(A)** top view; **(B)** side view (along *a* axis); and **(C)** detail of specific interaction and key distances (given in Å). Formamide oxygen is reported in pink, carbon in dark yellow, nitrogen in blue, hydrogen in gray, silicon in bright yellow, SiO₂ oxygen in red.

Table S5. Key structural and spectroscopic features for the SiO₂–FA5 model. Distances are given in Å, frequencies in cm⁻¹. The v(C=O) and δ (NH₂) frequencies have been scaled with separated scale factors.

SiO ₂ –FA5	
d(C=O)	1.237
v(C=O)	1703
δ(NH ₂)	1585



Figure S6. Relaxed geometry of the SiO₂–2FA1 model: **(A)** top view; **(B)** side view (along *a* axis); and **(C)** detail of specific interaction and key distances (given in Å). Formamide oxygen is reported in pink, carbon in dark yellow, nitrogen in blue, hydrogen in gray, silicon in bright yellow, SiO₂ oxygen in red.

Table S6. Key structural and spectroscopic features for the SiO₂–2FA1 model. Distances are given in Å, frequencies in cm⁻¹. The v(C=O) and δ (NH₂) frequencies have been scaled with separated scale factors.

SiO ₂ –2FA1	
d(C=O)1	1.2418
d(C=O)2	1.2445
v(C=O)1	1676
v(C=O)2	1690
$\delta(NH_2)_1$	1612
δ(NH2)2	1593



Figure S7. Relaxed geometry of the SiO₂–2FA2 model: **(A)** top view; **(B)** side view (along *a* axis); and **(C)** detail of specific interaction and key distances (given in Å). Formamide oxygen is reported in pink, carbon in dark yellow, nitrogen in blue, hydrogen in gray, silicon in bright yellow, SiO₂ oxygen in red.

Table S7. Key structural and spectroscopic features for the SiO₂–2FA2 model. Distances are given in Å, frequencies in cm⁻¹. The v(C=O) and δ (NH₂) frequencies have been scaled with separated scale factors.

SiO ₂ –2FA2	
d(C=O)1	1.241
d(C=O)2	1.237
v(C=O)1	1686
v(C=O)2	1707
$\delta(NH_2)_1$	1592
δ(NH ₂) ₂	1605



Figure S8. Relaxed geometry of the SiO₂–2FA3 model: **(A)** top view; **(B)** side view (along *a* axis); and **(C)** detail of specific interaction and key distances (given in Å). Formamide oxygen is reported in pink, carbon in dark yellow, nitrogen in blue, hydrogen in gray, silicon in bright yellow, SiO₂ oxygen in red.

Table S8. Key structural and spectroscopic features for the SiO₂–2FA3 model. Distances are given in Å, frequencies in cm⁻¹. The v(C=O) and δ (NH₂) frequencies have been scaled with separated scale factors.

SiO ₂ –2FA3	
d(C=O)1	1.238
d(C=O)2	1.244
v(C=O)1	1673
v(C=O)2	1698
$\delta(NH_2)_1$	1582
δ(NH ₂) ₂	1595



Figure S9. Relaxed geometry of the SiO₂–Dim1 model: **(A)** top view; **(B)** side view (along *a* axis); and **(C)** detail of specific interaction and key distances (given in Å). Formamide oxygen is reported in pink, carbon in dark yellow, nitrogen in blue, hydrogen in gray, silicon in bright yellow, SiO₂ oxygen in red.

Table S9. Key structural and spectroscopic features for the SiO₂–Dim1 model. Distances are given in Å, frequencies in cm⁻¹. The v(C=O) and δ (NH₂) frequencies have been scaled with separated scale factors.

SiO ₂ –Dim1	
d(C=O)1	1.245
d(C=O)2	1.240
v(C=O)1	1703
v(C=O)2	1683
$\delta(NH_2)_1$	1598
δ(NH ₂) ₂	1580



Figure S10. Relaxed geometry of the SiO₂–Dim2 model: **(A)** top view; **(B)** side view (along *a* axis); and **(C)** detail of specific interaction and key distances (given in Å). Formamide oxygen is reported in pink, carbon in dark yellow, nitrogen in blue, hydrogen in gray, silicon in bright yellow, SiO₂ oxygen in red.

Table S10. Key structural and spectroscopic features for the SiO₂–Dim2 model. Distances are given in Å, frequencies in cm⁻¹. The v(C=O) and δ (NH₂) frequencies have been scaled with separated scale factors.

SiO ₂ –Dim2	
d(C=O)1	1.240
d(C=O)2	1.245
v(C=O)1	1697
v(C=O)2	1672
$\delta(NH_2)_1$	1612
δ (NH ₂) ₂	1594



Figure S11. Relaxed geometry of the SiO₂–Dim3 model: **(A)** top view; **(B)** side view (along *a* axis); and **(C)** detail of specific interaction and key distances (given in Å). Formamide oxygen is reported in pink, carbon in dark yellow, nitrogen in blue, hydrogen in gray, silicon in bright yellow, SiO₂ oxygen in red.

Table S11. Key structural and spectroscopic features for the SiO₂–Dim3 model. Distances are given in Å, frequencies in cm⁻¹. The v(C=O) and δ (NH₂) frequencies have been scaled with separated scale factors.

SiO ₂ –Dim3	
d(C=O)1	1.254
d(C=O)2	1.240
v(C=O)1	1708
v(C=O)2	1678
$\delta(NH_2)_1$	1624
δ(NH ₂) ₂	1613



Figure S12. Relaxed geometry of the SiO₂–Dim4 model: **(A)** top view; **(B)** side view (along *a* axis); and **(C)** detail of specific interaction and key distances (given in Å). Formamide oxygen is reported in pink, carbon in dark yellow, nitrogen in blue, hydrogen in gray, silicon in bright yellow, SiO₂ oxygen in red.

Table S12. Key structural and spectroscopic features for the SiO₂–Dim4 model. Distances are given in Å, frequencies in cm⁻¹. The v(C=O) and δ (NH₂) frequencies have been scaled with separated scale factors.

SiO ₂ –Dim4	
d(C=O)1	1.248
d(C=O)2	1.244
v(C=O)1	1701
v(C=O)2	1669
$\delta(NH_2)_1$	1593
δ(NH2)2	1586

The energies of adsorption of two formamide molecules to give directly the HC models were computed. The following process has been considered:

 $SiO_2 + 2$ formamide $\rightarrow SiO_2 - 2$ formamide (S1)

The energetic parameter calculated accordingly are reported in Table S13.

Table S13. Adsorption energies (ΔE^c , with explicit dispersive contributions reported in brackets), enthalpies (ΔH^c) and Gibbs free energies (ΔG^c) for the SiO₂-formamide adducts high coverage models generated by double adsorption over the pristine SiO₂ surface. The energy values are calculated according to the reaction reported in eq. 2. All the reported energy values are expressed in kJmol⁻¹ have been BSSE corrected through the counterpoise method. The Bolzmann population (p) calculated from the ΔG^c at STP conditions (298.15 K, 1 atm) is reported too.

Model	ΔE^{c} (ΔE disp)	∆Hc	ΔGc	р
SiO ₂ -2FA1	-150.7 (-55.8)	-136.8	-35.6	0.00
SiO ₂ -2FA2	-149.4 (-47.8)	-135.9	-40.2	0.01
SiO ₂ -2FA3	-160.4 (-44.3)	-146.7	-51.8	0.91
SiO2-Dim1	-139.8 (-51.6)	-126.7	-26.6	0.00
SiO ₂ -Dim2	-166.5 (-51.5)	-151.4	-45.8	0.08
SiO ₂ -Dim3	-144.7 (-46.7)	-130.2	-31.5	0.00
SiO ₂ -Dim4	-151.6 (-48.7)	-136.1	-35.3	0.00