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1 **High export of nitrogen and dissolved organic carbon from an Alpine glacier (Indren Glacier, NW**
2 **Italian Alps)**

3

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16 **Abstract**

17 Mountain glaciers can export large amounts of nitrogen (N) and carbon (C) to downstream aquatic ecosystems. To date,
18 however, the number of studies that analysed concentrations and fluxes of N forms and dissolved organic carbon (DOC)
19 from glaciers in the European Alps and worldwide is limited, given the high complexity of data gathering in harsh high-
20 elevation environments. In this work, we rely upon new, unexploited data from field campaigns pursued during 2012–
21 2015 at high elevations (> 3000 m a.s.l.) of the Indren Glacier (NW Italian Alps) to i) develop glacio-hydrological
22 modelling and stream flow estimates within a heavily glacier-fed catchment, ii) provide N forms and DOC
23 concentrations and estimated fluxes in meltwater, and iii) provide possible explanations of cryospheric control upon
24 water chemistry. Water and soil samples were also collected at two lower-elevation sites along the glacial stream to
25 investigate the downstream variability of N forms and DOC. Nitrate-N, dissolved organic nitrogen (DON), and DOC
26 concentrations (0.21 ± 0.12 , 0.19 ± 0.14 , 1.16 ± 0.63 mg L⁻¹, respectively) and yields (220, 210, 1,279 kg km⁻² yr⁻¹,
27 respectively) were among the highest considering other glaciated areas of the globe, probably due to high atmospheric
28 N and C depositions. Limited effect of soil on water characteristics was found and attributed to the reduced soil
29 development in recently deglaciated areas (after the Little Ice Age), thus underlining the role of glacier melting in
30 determining N and C dynamics in high-elevation, Alpine surface waters.

31

32 **Keywords:** Glacier, Dissolved Organic Carbon, Nitrogen, Nitrate, Water chemistry, Soil

33 1 Introduction

34 Runoff from glaciated catchments in mountain regions is an important fresh water resource to downstream areas
35 (Viviroli et al. 2011). Modifications in the glacier storage capacity due to climate change can affect the hydrological
36 cycle (Stahl et al. 2008), and these modifications are expected to affect nitrogen (N) and carbon (C) dynamics in glacial
37 meltwaters (Hood et al. 2015; Milner et al. 2017). Overall, glacier melting and associated changes in habitat conditions,
38 such as channel stability, turbidity, temperature, nutrient loadings, and concentrations of legacy pollutants and trace
39 elements can result in complex ecological modifications in mountain waters (Brighenti et al. 2019). The degree of these
40 effects varies depending on the type and size of the glacier, but even mountain glaciers covering a small proportion of
41 watershed can influence water biogeochemistry and ecosystems (Slemmons et al. 2013).

42 Glacier runoff can act as N source (Milner et al. 2017), concentrated in glacier ice from the atmosphere (Daly and
43 Wania 2005), and originating from microbial ecosystems on and beneath the ice (Wadham et al. 2016). Lakes and
44 streams fed by mountain glaciers have been reported to be enriched in nitrate in comparison to snow/groundwater-fed
45 surface waters (Baron et al. 2009; Slemmons et al. 2017). Several non-mutually exclusive factors (Slemmons et al.
46 2017) could explain the N enrichment of glacier meltwaters such as enhanced nitrification in subglacial environments
47 (Wynn et al. 2007), reduced contact with watershed soils (Brooks et al. 1996), and release through abrasion of old
48 organic N bound in rock (Boyd et al. 2011). Multiple effects of N-enriched glacial meltwater include increased lake
49 primary production (Slemmons and Saros 2012), altered species richness of algal groups (Saros et al. 2010), and
50 modified phytoplankton species composition (Slemmons et al. 2017).

51 Glaciers have been recently recognised as unique ecosystems with potential implications for the global carbon cycle
52 (Hood et al. 2015). Organic carbon (OC) in glaciers originates from recent and past incorporation into the glacial
53 system (Stubbins et al. 2012; Spencer et al. 2014), accumulating from in situ primary production (Hood et al. 2009;
54 Bhatia et al. 2013), and allochthonous terrestrial (Singer et al. 2012) and anthropogenic (Spencer et al. 2014) sources.
55 Globally, mountain glaciers have been estimated to release approximately 0.8 Tera g yr⁻¹ of dissolved organic carbon
56 (DOC) to downstream ecosystems (Li et al. 2018) and the glacier shrinkage can lead to an increase of DOC fluxes
57 reaching downstream ecosystems (Hood et al. 2015). Recent studies have shown that glacier DOC is biologically
58 available, providing a subsidy of bioavailable C to receiving surface waters (Hood et al. 2009; Singer et al. 2012) and
59 sustaining aquatic foodwebs (Fellman et al. 2015).

60 Although it is clear now that glaciers can be fundamental factors in influencing N and C dynamics in surface waters,
61 few studies have been performed on concentrations and fluxes of major N forms (i.e. nitrate and dissolved organic
62 nitrogen) and DOC from mountain glaciers across the globe (e.g. Lafrenière and Sharp 2005; Baron et al. 2009; Li et al.
63 2018), and even fewer in the European Alps (e.g. Tockner et al. 2002; Singer et al. 2012). In this work, seasonal time

64 series of N forms and DOC concentrations and fluxes in the glacial stream originating from the Indren Glacier (NW
65 Italian Alps) are presented. Meltwater samples were collected in the period 2012–2015 and modelled glacier discharge
66 was used to elucidate the role of glacial hydrology on N and C export. Water and soil samples were also collected in the
67 period 2012–2014 at two lower-elevation sites along the glacial stream in order to investigate the downstream
68 variability of N forms and DOC in glacial meltwater.

69

70 **2 Materials and methods**

71 **2.1 Study area**

72 The Indren glacier is a temperate, wet-based mountain glacier (Maggioni et al. 2009), located in the NW Italian Alps, in
73 the Valle d’Aosta Region (Fig. 1), within the LTER (Long term Ecological Research) site, Istituto Mosso. The glacier is
74 in direct contact with a proglacial pond. The glacial stream originates from outflow thereby, and a hydrological station
75 was installed to measure hydrological fluxes (Q1 site, Fig. 1, Tab. 1). The size of the Indren glacier has reduced
76 substantially in the last ~century, retreating by ca. 500 m in the period 1927–2013 (Piccini 2007; Baroni et al. 2014) and
77 losing ca. 45% of the surface area between 1955 and 2006 (from 1.68 to 0.92 km²) (CGI-CNR 1961). Strong glacier
78 recession rates have been reported in other areas of the NW Italian Alps due to the impact of climate change (Giaccone
79 et al., 2015; Colombo et al. 2016a; 2016b). According to the long-term climate series (1928–2013) of the Gressoney-
80 La-Trinité - Lago Gabiet automatic weather station (AWS) (2379 m a.s.l., located ca. 4 km far from the Indren glacier,
81 Tab. S1), in the last ~century maximum air temperatures and number of rainy days significantly increased (+0.015 °C
82 year⁻¹ and +0.17 days year⁻¹, respectively), while fresh snow showed a strong negative trend, although not significant (–
83 1.750 cm year⁻¹) (Fратиanni et al., 2015). For the 2012–2015 period, the Gabiet AWS recorded a mean air temperature of
84 +2.8 °C, yearly cumulated precipitation (including snow water equivalent) of 1052 mm, and snow thickness of 102 cm.
85 Geologically, the study area is dominated by paraschists, with subordinate leucocratic gneisses, metabasites,
86 metagranites, micaschists, and minor bodies of calcschists and marbles.

87

88 **2.2 Modelled and measured hydrological flows**

89 A semi-distributed elevation-belt based version of the *Poly-Hydro* model (Soncini et al. 2017), able to reproduce ice and
90 snow dynamics, evapotranspiration, recharge of groundwater reservoir, discharge formation and routing to the control
91 section, was implemented and tuned (Groppelli et al. 2011; Soncini et al. 2015) (further details in Supporting
92 Information (S), Materials and Methods S1.1). Input data to the model are a digital terrain model-DTM, daily values of
93 precipitation and temperature (and their vertical gradient), and land cover information. In this work, a 2x2m-cell DTM
94 and four AWSs close to the glacier were used (Tab. S1, Materials and Methods S1.1).

95 *Poly-Hydro* tracks the variation of the water content in the ground within one belt W (mm) in two consecutive time
 96 steps ($t, t+\Delta t$), as

$$97 \quad W^{t+\Delta t} = W^t + R\Delta t + M_s\Delta t + M_i\Delta t - ET\Delta t - Q_g\Delta t \quad (1)$$

98 here using the daily time step, R = liquid rain, M_s = snowmelt, M_i = ice melt, ET = actual evapotranspiration, and Q_g =
 99 groundwater discharge, with all parameters expressed in mm d^{-1} . Overland flow Q_s occurs for saturated soil

$$100 \quad Q_s = \begin{cases} W^{t+\Delta t} - W_{Max} & \text{if } W^{t+\Delta t} > W_{Max} \\ 0 & \text{if } W^{t+\Delta t} \leq W_{Max} \end{cases} \quad (2)$$

101 with W_{Max} (mm) greatest potential soil storage. Potential evapotranspiration, however small at this high elevation is
 102 calculated via Hargreaves equation, requiring temperature data (Bocchiola et al. 2010). Groundwater discharge is a
 103 function of soil hydraulic conductivity and water content

$$104 \quad Q_s = K \left(\frac{W}{W_{Max}} \right)^k \quad (3)$$

105 with K (mm d^{-1}) saturated permeability and k power exponent.

106 To depict actual flow variability of this highly glaciated catchment, and to properly tune *Poly-Hydro* model, a flow
 107 gauging station was installed downstream of the proglacial pond, at 3083 m a.s.l. (Fig. 1) in 2012 and 2013. The pond
 108 outlet was surveyed and modelled as a free surface weir, with outlet coefficient calibrated against flow measures,
 109 pursued with a Flow-Tracker® 3D Doppler velocimeter, once per year. Access to the site and operation of the station
 110 was possible in safe conditions during short periods (July 19th–October 31st 2012, and July 24th–September 9th in 2013),
 111 and the station was shut down in winter. The model outputs were tuned against the observed discharges in these periods.

112

113 **2.3 Water and soil sampling, and laboratory analyses**

114 Water samples were collected weekly to monthly during the melting seasons 2012–2015 at the pond outlet (W1 site)
 115 (Fig. 1, Tab.1). In the period 2012–2014 water samples were also collected at two lower-elevation sites along the glacial
 116 stream (W2 and W3 sites) (Fig. 1, Tab.1), selected to describe typical and representative features of the different
 117 pedological conditions around the stream, with vegetated soil cover becoming more widespread at lower elevations
 118 (Tab. 1). In addition, a few water samples were collected during the winter seasons 2012 and 2013. Furthermore, a 300
 119 cm-deep snow profile (ca. 2 km far from the glacier, since the glacier area is hardly accessible) was sampled in April
 120 2015, and 6 snow samples were collected at 50-cm intervals for chemical analyses (for each layer, snow density was
 121 also measured). A rain collector was installed at the end of July 2015 at ca. 2 km distance from the glacier and sampled
 122 when precipitation occurred (5 observations) (Materials and Methods S1.2).

123 Electrical conductivity (EC) was measured using a Crison - Micro CM 2201. The concentration of major anions
 124 (Cl^- , NO_2^- -N, NO_3^- -N, PO_4^{3-} -P, SO_4^{2-} -S) was determined with a Dionex DX-500 (Dionex Corp., California). Major

125 cations (Ca^{2+} , Mg^{2+} , K^+ , Na^+) were determined by flame atomic absorption spectroscopy using a Perkin Elmer AAnalyst
126 400 (Perkin-Elmer Inc., Waltham, Massachusetts). Ammonium ($\text{NH}_4^+\text{-N}$) was determined spectrophotometrically (U-
127 2000, Hitachi, Tokyo, Japan). DOC and total dissolved nitrogen (TDN) were analysed with a VarioTOC (Elementar,
128 Hanau, Germany). Dissolved organic nitrogen (DON) was estimated from the difference between TDN and dissolved
129 inorganic nitrogen ($\text{DIN} = \text{NO}_3^- + \text{NH}_4^+$; NO_2^- was always below the detection limit). The quality of chemical analyses
130 was determined by including method blanks and repeated measurements of certified samples. Analytical precision for
131 major anions was <10%, and for major cations was <5%. Analytical precision for DOC and TDN was <2% (Materials
132 and Methods S1.2).

133 Soil samples were collected close to the water-sampling sites (S1, S2, and S3) (Fig. 1, Tab. 1) during the melting
134 season 2012. Field description of soil profiles was done according to Food and Agriculture Organization-FAO (2006).
135 Soil samples were collected from each genetic horizon (Materials and Methods S1.2). OC and total nitrogen (TN) were
136 analysed by dry combustion with a CN elemental analyzer (CE Instruments NA2100, Rodano, Italy).

137

138 **2.4 Statistical analysis and loads calculation**

139 From 24 March 2012 to 28 September 2015 sampling of water at W1 site (Fig. 1) was carried out on 41 occasions (Fig.
140 2), and corresponding daily discharge rates were obtained from the model output at Q1 site. Since the lowermost
141 average modelled discharge during the melting period (April–October) was $0.05 \text{ m}^3 \text{ s}^{-1}$ (year 2014), this threshold was
142 chosen to discriminate between samples collected during low-flow ($< 0.05 \text{ m}^3 \text{ s}^{-1}$) and high-flow ($\geq 0.05 \text{ m}^3 \text{ s}^{-1}$) periods
143 (Fig. 2, further details in the Results section). This operation was performed in order to investigate the seasonal
144 variations in solute concentrations and their relationships with discharge.

145 The normality of the data was tested using the Shapiro-Wilk test and data were also tested for homogeneity of
146 variance with the Levene's test (Salerno et al. 2016). Differences in samples between low-flow (18 observations) and
147 high-flow (23 observations) periods were evaluated with one-way analysis of variance (ANOVA), if data were normally
148 distributed. When necessary, data were Log transformed to meet the assumptions of normality. If this failed to improve
149 normality, the Kruskal-Wallis test was applied. Significant differences between water sampling sites along the elevation
150 gradient (26 observations) were determined by means of ANOVA followed by Tukey's multiple comparison test. In
151 case of non-normally-distributed data, nonparametric one-way ANOVA using Wilcoxon (Kruskal-Wallis) scores was
152 performed. All analyses were implemented in the R software (R Core Team, 2018) at a significance level of $p < 0.05$.

153 DOC and N forms loads from the glacier, except $\text{NO}_3^-\text{-N}$, were calculated by multiplying the modelled glacier
154 discharge by the mean analyte concentrations due to the lack of dependence of these analytes upon discharge. The
155 uncertainties associated with the mean concentrations were defined as the standard deviations of analyte concentrations,

156 while the uncertainties associated to the loads also considered the error related to measured vs. modelled discharge and
157 were calculated as the root mean square of the errors of the two estimations (Salerno et al. 2016). Given the dependence
158 of NO_3^- -N upon discharge, the regression equation was used to calculate daily fluxes. Daily fluxes were then
159 aggregated for the entire investigated period and divided by the analysed years to obtain a mean annual load. The daily
160 load uncertainty was calculated as the root mean square of the regression standard error multiplied by the discharge and
161 the measured vs. modelled discharge error. Daily errors were then aggregated for the investigated period divided by the
162 analysed years to obtain a mean annual error.

163

164 **3 Results**

165 **3.1 Meteorological and hydrological conditions**

166 *Poly-Hydro* model provided rather good depiction of the outlet stream flows from the glacier catchment, both in terms
167 of average flows (*Bias* 2%) and flow variability, as expressed by the Nash-Sutcliffe Efficiency coefficient-NSE = 0.65
168 (Tab. S2). The modelled hydrographs showed that, after a low winter base flow, glacier discharge generally began to
169 increase during April-May as driven by spring snowmelt, then peaking in June-July (Fig. 2a, 2b). The late summer-
170 autumn decline (until October) was occasionally interrupted by peaks after intense rainfall events (Fig. 2a, 2b). Average
171 modelled discharges during the melting seasons 2012, 2013, 2014, and 2015 were $0.11 (\pm 0.11)$, $0.09 (\pm 0.10)$, $0.05 (\pm$
172 $0.05)$, and $0.08 (\pm 0.08) \text{ m}^3 \text{ s}^{-1}$, respectively. The overall yearly average stream flow during the four years was estimated
173 at $0.05 (\pm 0.08) \text{ m}^3 \text{ s}^{-1}$.

174

175 **3.2 Discharge-solute relationships, and N and C export**

176 At W1 sampling site, EC was significantly higher during low-flow periods ($52.3 \pm 24.6 \mu\text{S cm}^{-1}$) than during high-flow
177 periods ($21.3 \pm 9.3 \mu\text{S cm}^{-1}$) (Fig. 2c, 3a, Tab. 2). Similar behaviour was found for the dominant major anion (SO_4^{2-} -S)
178 and cation (Ca^{2+}) (Tab. S3). Both ions were significantly higher during low-flow periods (SO_4^{2-} -S = $5.50 \pm 3.82 \text{ mg}$
179 L^{-1} ; Ca^{2+} = $7.23 \pm 4.08 \text{ mg L}^{-1}$) than during high-flow periods (SO_4^{2-} -S = $1.41 \pm 1.16 \text{ mg L}^{-1}$; Ca^{2+} = $2.42 \pm 1.17 \text{ mg}$
180 L^{-1}) (Fig. 2c, 3b, 3c, Tab. 2). The concentrations of the other ions (i.e. Mg^{2+} , K^+ , Na^+ , Cl^-) were strongly correlated with
181 EC ($r > 0.78$; except for Cl^- , $r = 0.62$) (Tab. S4); thus EC and the dominant major ions SO_4^{2-} -S and Ca^{2+} were selected
182 as tracers for discussing the geochemical patterns of the Indren Glacier meltwater. Among the N forms, NO_3^- -N was the
183 main component, representing a high proportion of TDN (47%), followed by DON (42%) and NH_4^+ -N (11%). Similar
184 to EC and major ions, NO_3^- -N was significantly higher during low-flow periods ($0.30 \pm 0.11 \text{ mg L}^{-1}$) than during high-
185 flow periods ($0.15 \pm 0.07 \text{ mg L}^{-1}$) (Fig. 2d, 4a, Tab. 2). In contrast, mean DON concentration did not show statistically
186 significant differences between low-flow ($0.20 \pm 0.15 \text{ mg L}^{-1}$) and high-flow ($0.15 \pm 0.14 \text{ mg L}^{-1}$) periods (Fig. 2d, 4b,

187 Tab. 2), as well as $\text{NH}_4^+\text{-N}$ (not shown). Analogously to DON, mean DOC concentration was similar during low-flow
188 (1.16 ± 0.57) and high-flow (1.15 ± 0.69) periods, not showing any significant difference (Fig. 2d, 4c, Tab. 2). DOC
189 concentration ranged from 0.49 to 3.94 mg L^{-1} , with an average value of $1.16 \pm 0.63 \text{ mg L}^{-1}$.

190 Estimated fluxes of TDN, $\text{NO}_3^-\text{-N}$, DON, $\text{NH}_4^+\text{-N}$ and DOC were 710 (± 364), 315 (± 183), 300 (± 223), 79 (± 57),
191 and 1,829 (± 994) kg yr^{-1} , respectively. This equated to area-weighted yields (analyte flux divided by the contributing
192 area) for TDN, $\text{NO}_3^-\text{-N}$, DON, $\text{NH}_4^+\text{-N}$, and DOC of 496, 220, 210, 55, and 1,279 $\text{kg km}^{-2} \text{ yr}^{-1}$, respectively.

193

194 3.3 N and C in soil, and elevational variations of N forms and DOC in meltwater

195 Soils displayed different properties according to their different elevations. Organo-mineral horizons enriched in OC
196 were observed in soils located at S2 and S3 sites, deglaciated since the Little Ice Age (LIA), while the few-decade-old
197 till (S1) was depleted (Tab. 3). TN was significantly correlated with OC ($r = 0.99$) and had an increasing trend with
198 decreasing elevation, showing the highest content in surface A horizons at S2 and S3. S2 and S3 also showed the
199 presence of vegetation, dominated by *Carex curvula* and other species belonging to the *Caricetum curvulae* association.

200 Along the elevation gradient, no significant variations were found in water $\text{NO}_3^-\text{-N}$, DON, and DOC concentrations
201 (Fig. 5).

202

203 4 Discussion

204 4.2 Loads and possible origin of nitrate and major ions

205 Nitrate-N mean concentration was consistent with concentrations measured in other glaciated sites and high-elevation
206 surface waters (with varying glacial cover) in Europe (e.g. Tockner et al. 2002; Balestrini et al. 2013; Rogora et al.
207 2013). Nitrogen deposition in the above-mentioned areas was higher (10 to 17 $\text{kg N ha}^{-1} \text{ y}^{-1}$) than that measured in our
208 study area (6.6 $\text{kg N ha}^{-1} \text{ y}^{-1}$, as calculated from snow and rain measurements). However, the mean concentrations of
209 $\text{NO}_3^-\text{-N}$ recorded in surface waters were similar, probably because of the lower soil and vegetation cover present in our
210 site, with consequent lower inorganic N retention by soil and vegetation (e.g. Hood et al. 2003). In contrast, $\text{NO}_3^-\text{-N}$
211 concentration and yield (where reported) were higher than those measured in other glaciated environments across the
212 globe such as in Nepal Himalaya (e.g. Lami et al. 2010; Balestrini et al. 2014), Svalbard (e.g. Wynn et al. 2007),
213 Northern Patagonia (Rogora et al. 2008), and Greenland Ice Sheet (Wadham et al. 2016). In these areas, N
214 concentrations and deposition (in snow and rain, where reported) were generally lower than in our site, suggesting an
215 important role of atmospheric source in N dynamics in different glaciated environments. This is further confirmed by
216 the elevated N concentrations observed in surface waters located in watersheds fed by glaciers such as in the Rocky

217 Mountains in Colorado (Williams et al. 2007; Baron et al. 2009; Saros et al. 2010), where elevated N deposition (Fenn
218 et al. 2003; Williams et al. 2009) has been reported due intensive industrial and agricultural activity (Baron et al. 2000).

219 Nitrate-N concentration in a pond without glacier and surrounded by vegetated soil located in the same area of the
220 Indren Glacier (at 2.5 km distance) was ca. three times lower (Magnani et al. 2017) than NO_3^- -N measured at W1 site,
221 thus confirming the role of mountain glacier catchments in enriching surface waters in NO_3^- in comparison to
222 snow/groundwater-fed surface waters (Slemmons et al. 2017). Indeed, NO_3^- concentrations in surface waters fed by
223 glacial meltwater have been found to be up to 1–2 orders of magnitude higher than in surface waters fed by snow and
224 groundwater (Saros et al. 2010; Warner et al. 2017), with increasing NO_3^- export from glacial areas that has been
225 reported in the last decades (Barnes et al. 2014). Different explanations have been proposed for the higher NO_3^- content
226 in glacial meltwater, such as reduced contact with watershed soils and vegetation (Brooks et al. 1996), flushing of NO_3^-
227 from microbially active sediments (Baron et al. 2009), and high sublimation and evaporation rates on glaciers leading to
228 concentrated N (Daly and Vania, 2005).

229 Only NO_3^- -N, among the N forms and DOC, was positively correlated to the major ions and negatively correlated to
230 glacier discharge, showing higher concentration in meltwater during low-flow periods (Fig. 2d, 4a, Tab. 2, S4). This
231 occurrence suggests that NO_3^- -N derived mainly from subglacial environment. Indeed, during the winter and early
232 melting season (November-April), low discharge in association with higher EC values, and SO_4^{2-} -S and Ca^{2+}
233 concentrations (Fig. 2c, 3) indicate that streamwater chemistry mainly derived from solute-rich subglacial water (winter
234 season) in association with solute-enriched snowmelt due to leaching of ions from the snowpack (early melting season)
235 (Stachnik et al. 2016). Progressive ionic increases from high- to low-flow periods during the melting season (Fig. 2c, 3)
236 suggest that a decreasing ablation rate in the catchment reduced the dilution of solutes by meltwater, resulting in
237 favourable conditions for mineral dissolution beneath the glacier (Stachnik et al. 2016). This is further confirmed by the
238 general behaviour of the major ions, which showed an inverse dependence upon glacier discharge (Tab. S4). In this
239 regard, solute concentrations have been observed to be often inversely related with glacier runoff (Tranter 2006; Yde et
240 al. 2014; Stachnik et al. 2016), with both hydrologic and chemical controls that exist under glaciated conditions (Tranter
241 et al. 2002).

242 The occurrence of meltwater with longer residence time, routed more slowly through a subglacial distributed
243 drainage system (Tranter et al. 1997; Brown et al. 2006) during low-flow periods is further confirmed by the elevated
244 SO_4^{2-} -S concentration. In particular, S-ratio [$\text{SO}_4^{2-} / (\text{HCO}_3^- + \text{SO}_4^{2-})$] and Ca-ratio ($\text{Ca}^{2+} / \text{SO}_4^{2-}$) (where units of
245 concentrations are expressed in equivalents per litre and HCO_3^- was calculated as the difference between cations and
246 anions since it was not measured in this study) were calculated. The higher S-ratio during low-flow periods ($0.68 \pm$
247 0.18) with respect to high-flow periods (0.49 ± 0.16) and the lower Ca-ratio during low-flow periods (1.2 ± 0.47) in

248 comparison to high-flow periods (1.7 ± 0.73) highlighted an enhanced sulphide oxidation and sulphate release. Sulphate
249 can be considered an indicator of distributed type drainage beneath glaciers (Brown et al. 2006; Wadham et al. 2010)
250 and its increase, in most glacial environments, is an effect of increasing residence times and rock/water interaction in
251 subglacial environment (Cooper et al. 2002), where bedrock comminution releases highly reactive sulphide-rich
252 minerals (common in the study area) to meltwater (Tranter et al. 1993). Moreover, high sulphate concentrations in
253 subglacial meltwaters can also be a consequence of microbially-mediated anoxic sulphide oxidation (Tranter et al.
254 2002). The maximum sulphate concentration that can be generated from sulphide oxidation using oxygen-saturated
255 waters at 0 °C is ca. 400 $\mu\text{eq L}^{-1}$ (Tranter et al. 2002). In our case, sulphate concentrations in bulk runoff up to ca. 1050
256 $\mu\text{eq L}^{-1}$ during low-flow periods, with a number of samples above 400 $\mu\text{eq L}^{-1}$, could indicate meltwater drainage from
257 basal regions where additional sulphate contributions may be generated via microbially-mediated anoxic sulphide
258 oxidation (Tranter et al. 2002; Bhatia et al. 2010). Thus, during high-flow periods, most meltwater (including relevant
259 snowmelt contribution) could be intercepted by crevasses and moulins, and drained via major englacial or subglacial
260 conduits with little interaction with subglacial sediments (cf. Hubbard and Nienow, 1997). During low-flow periods,
261 some surface-derived runoff and meltwater produced by basal melting could drain by a more extensive distributed
262 drainage system that allowed more significant water-sediment interaction, with an important contribution of
263 microbially-derived processes to weathering. This hydrological component likely constituted a significant contribution
264 to runoff under recession flow conditions and early in the melt season before the development of major drainage
265 channels (cf. Lafrenière and Sharp 2004).

266 Microbial activity could contribute to NO_3^- export from the Indren Glacier catchment, thanks to enhanced
267 nitrification from microbial populations in subglacial environment (Wynn et al. 2007; Wadham et al. 2016). Indeed,
268 active microbes have been found in cryospheric settings spanning from arctic to alpine environments (Mader et al.
269 2006; Williams et al. 2007). Some studies reported that in high-mountain areas also fine sediments under talus deposits
270 are capable of supporting microbial activity, thus contributing in exporting NO_3^- into surface waters (e.g. Williams et
271 al. 1997). However, debris deposits in the W1 site catchment were of minor importance (3.7% of the catchment) and
272 mainly constituted by coarse ablation till, with glacier (64.1%) and steep bedrock (32%) that were the predominant land
273 cover types (Tab. 1). Thus, NO_3^- -N contribution from debris deposits outside the glacier body was likely limited. A
274 further evidence of the subglacial origin of NO_3^- -N derives from the observation of the NO_3^- -N: Cl^- ratios, which were
275 higher in meltwater (2.3) than in rain (1.4) and snow (0.4). Assuming that Cl^- behaved conservatively, then the
276 difference in the NO_3^- -N: Cl^- ratio between the glacial meltwater and the other potential N sources can be considered an
277 indication of the effect of catchment processes on NO_3^- -N export (Lafrenière and Sharp 2005).

278 In some occasions, NO_3^- -N concentration was found to be high with relatively high discharge towards the end of the
279 melt season. This occurred after heavy rainfall events (e.g. August-September 2012 and 2015, 3 main events) (Fig. 2,
280 4a). In this case, it is possible that, during rainfall events, water which passed through the glacier flushed water that was
281 potentially stored for some time at the glacier bed, thereby flushing nitrate-rich meltwater from subglacial environment
282 (cf. Theakstone and Knudsen 1996). This is further corroborated by the fact that, during these events, also EC and major
283 ions (especially sulphate) (Fig. 2, 3) increased.

284

285 **4.2 Loads and possible origin of DOC and DON**

286 The DOC average value ($1.16 \pm 0.63 \text{ mg L}^{-1}$) in glacial meltwater was approximately three and two times higher
287 than the average global values for mountain glaciers reported by Hood et al. (2015) (0.37 mg L^{-1}) and Li et al. (2018)
288 (0.54 mg L^{-1}), respectively. This resulted in a high yield ($1,279 \text{ kg km}^{-2} \text{ yr}^{-1}$) confirming the high DOC export
289 estimates in the European Alps recently proposed by Li et al. (2018), and far higher than previous average regional
290 estimate for the European Alps ($170 \text{ kg km}^{-2} \text{ yr}^{-1}$) (Singer et al. 2012) and other alpine glaciers (e.g. Roseg and
291 Tschierwa glaciers, Switzerland, $360 \text{ kg km}^{-2} \text{ yr}^{-1}$) (Tockner et al. 2002). This high DOC export could derive from
292 atmospheric deposition (cf. Stibal et al. 2008; Stubbins et al. 2012), considering the location of the Indren Glacier, close
293 to the Po Valley, a European atmospheric pollution hotspot from where air masses can be transported to the NW Alps
294 (Diémoz et al. 2019). DOC concentration in snow in the study area ($0.77 \pm 0.15 \text{ mg L}^{-1}$) was generally higher than
295 values reported for other glaciated areas, with average values ranging from 0.06 to 0.33 mg L^{-1} (e.g. Stubbins et al.
296 2012; Spencer et al. 2014; Yan et al. 2016; Musilova et al. 2017). DOC deposition in the Indren Glacier area was 15.5
297 $\text{kg C ha}^{-1} \text{ y}^{-1}$, with rain contributing 43% ($6.6 \text{ kg C ha}^{-1} \text{ y}^{-1}$) of the total DOC deposition, which highlighted its
298 relevance in C dynamics in the investigated area. Unfortunately, there are few quantitative studies of OC in atmospheric
299 deposition and even fewer for atmospheric deposition in mountainous environments (Mladenov et al. 2012). However,
300 average DOC concentrations in precipitation in the Turin area and other locations in polluted southern Europe have
301 been found to be between 1.01 and 2.50 mg L^{-1} (Iavorivska et al. 2016), which is consistent with the concentration
302 measured in meltwater and precipitation in this study. Moreover, recent studies reported a number of dust-deposition
303 events in snow- and glacier-covered area of the European Alps (Gabbi et al. 2015; Di Mauro et al. 2019). Thus, a
304 further source of DOC could be linked to dust depositions, since atmospheric aerosol transport is one vector for the
305 delivery of C to alpine areas (Mladenov et al. 2012).

306 Although DON measurements in glacial meltwater are sparse, DON concentration ($0.19 \pm 0.14 \text{ mg L}^{-1}$) and yield
307 ($210 \text{ kg km}^{-2} \text{ yr}^{-1}$) in this study were higher than other values reported for few glaciated catchments in Europe (Tockner
308 et al. 2002) and for Greenland Ice Sheet (Wadham et al. 2016), with similar values found in Alaska (Hood and Scott

309 2008). High N deposition in the study area and site-specific characteristics (e.g. absence of developed soils in the Indren
310 Glacier catchment that could promote mineralisation processes of DON) could explain this occurrence.

311 In contrast to NO_3^- -N and major ions, DON and DOC concentrations were similar during low- and high-flow
312 periods, showing no dependence upon glacier discharge (Fig. 2d, 4b, 4c, Tab. 2, S4). Thus, this occurrence points
313 towards a different origin of DON and DOC with respect to NO_3^- -N (and major ions). Previous research suggested that
314 DON and DOC concentrations could derive from different pools of DON (Wadham et al. 2016) and DOC (Bhatia et al.
315 2013; Spencer et al. 2014) within glaciers.

316 The acquisition of DON could occur both in surface and basal ecosystems (Wadham et al. 2016), with the presence
317 of a significant N-rich component to dissolved organic matter exported from glacier ecosystems in runoff (Hood et al.
318 2009; Bhatia et al. 2013). Analogously, DOC could be released from both surface and basal glacial zones. Previous
319 studies have attributed DOC origin in surface waters during the ablation period to DOC leached from wet snow and/or
320 the glacier surface in the supraglacial ecosystems (Stibal et al. 2012a; Spencer et al. 2014). Cryoconite holes were
321 observed to be widely distributed in the ablation area of the Indren Glacier and supraglacial streams transporting
322 organo-mineral material from cryoconite holes to the glacier front were numerous. DOC could originate also at the
323 glacier base, where C is derived from the underlying bedrock, sediments, and ice (Bhatia et al. 2010). Indeed, several
324 authors reported the presence of OC in subglacial environment (Foght et al. 2004; Skidmore et al. 2005; Stibal et al.
325 2012b), with basal ice showing higher DOC concentrations with respect to surface-englacial ice and cryoconite holes on
326 numerous mountain glaciers (Hood et al. 2015). Interestingly, Colombo et al. (2019) reported that the fluorescence
327 properties of DOC in the Indren proglacial pond were characterised by a high seasonal variability during the melting
328 season 2015. The authors hypothesised that beside an aquatic microbial source of DOC, a source of terrestrially-derived
329 DOC might also exist in the glacier catchment. However, due the fact that W1 site was close to the glacier front and OC
330 in the coarse deposit at S1 site (Fig. 1, Tab. 1) was low (0.14%, Tab. 3), the contribution from the leaching of DOC
331 derived from proglacial detritus areas (cf. McClelland et al. 2014) to meltwater was likely negligible. In addition,
332 developed soils and vegetal material (i.e. plant debris) were absent in the proglacial area. Therefore, it is possible that
333 the re-arrangement and flushing of the subglacial drainage system could cause transient variations in the fluorescence
334 properties of DOC, implying the possible coexistence of sources of terrestrially- and microbially-derived DOC in the
335 subglacial environment (cf. Bhatia et al. 2010). Thus, the OC in the Indren Glacier is probably an amalgamation of
336 supraglacial and subglacial sources (with different microbial and terrestrial origin) (cf. Bhatia et al. 2010; 2013),
337 although further investigations are necessary to properly identify the DOC sources in the Indren Glacier catchment and
338 their fractional contribution to DOC export.

339

340 4.2 DOC and N variation along the elevation gradient

341 Along the elevation gradient, the % of soil cover increased downstream, reaching a value close to 6% at S3 site (Tab.
342 1), but without any relevant effect on NO_3^- -N concentration (Fig. 5a). This could be attributed to the slope steepness
343 and to the fact that the area was recently deglaciated (LIA) with soils showing little development (soil age ca. 170
344 years), covering a small % of the basin area. A number of studies have demonstrated that, some physical features of the
345 catchment strongly influence the chemical composition of surface water at high elevations (e.g. Clow and Sueker 2000;
346 Balestrini et al. 2013). Balestrini et al. (2013) reported that in some North American and European high-elevation
347 catchments, NO_3^- concentrations in surface waters are inversely related to the areal extensions of developed soils,
348 indicating the fundamental role that the soil biological community plays in the retention and loss of N and therefore the
349 strict connection between soil and waters in mountain remote ecosystems. However, Clow and Sueker (2000) in high-
350 elevation basins in the Colorado Rocky Mountains found how minimal soil development limits chemical weathering
351 and N uptake. Limited soil development in the study area can be explained by the fact that soil evolution, although
352 rather quick in the Alps, could be influenced by frost disturbance and slope instabilities that are able to impede or delay
353 pedogenesis (Haugland 2004).

354 OC and TN concentrations in soil increased at lower elevation (Tab. 3), without any corresponding increase in DOC
355 and DON in stream water (Fig. 5b, 5c). Several authors reported water DOC increases with increasing distance from the
356 glacier fronts (e.g. Li et al. 2018) and decreasing glaciated areas (e.g. Hood and Scott 2008), due to the loss of terrestrial
357 DOC from unglaciated terrains, implying a strong relationship between soil C pool and DOC export to surface water
358 (Lafrenière and Sharp 2004). Moreover, previous studies reported similar DON yields among catchments with different
359 glacier covers, suggesting that glacial runoff is an important contributor to DON (Hood and Scott 2008). In our site, the
360 higher (although not significant) DOC concentrations at the glacier front and the similar DON concentrations along the
361 elevation gradient point towards the limited effect of soil on water characteristics due to its reduced surface area but, at
362 the same time, underline the importance of the glacier melting in determining N and C dynamics in surface water in this
363 high-elevation, remote environment.

364

365 Conclusion

366 Concentrations and estimated fluxes of N forms and DOC from the high-elevation Indren Glacier catchment were
367 analysed in the period 2012–2015, and concentrations of these analytes in the glacial stream were further investigated
368 along an elevation gradient during 2012–2014. NO_3^- -N, as well as EC and major ions, showed a negative relationship
369 with modelled glacier discharge, suggesting a possible subglacial source. DON and DOC were not dependent upon
370 discharge, entailing a potential origin from distinct DON and DOC pools within the glacier. Meltwater concentrations

371 and yields for NO_3^- -N ($0.21 \pm 0.12 \text{ mg L}^{-1}$, $220 \text{ kg km}^{-2} \text{ yr}^{-1}$), DON ($0.19 \pm 0.14 \text{ mg L}^{-1}$, $210 \text{ kg km}^{-2} \text{ yr}^{-1}$), and DOC
372 ($1.16 \pm 0.63 \text{ mg L}^{-1}$, $1,279 \text{ kg km}^{-2} \text{ yr}^{-1}$), were among the highest reported for other glaciated areas in the world,
373 probably due to high atmospheric N and C deposition. Minor changes in water NO_3^- -N, DON, and DOC concentrations
374 along the elevation gradient suggested a limited effect of soil on water characteristics due to its reduced surface area and
375 little development but, at the same time, underlined the importance of the glacier melting in determining N forms and
376 DOC concentrations in the glacial stream. Further work is needed to improve our understanding of the origin of N
377 forms and DOC in the Indren Glacier meltwater. However, our results contribute to clarifying the little investigated, and
378 yet very important role of fast melting cryospheric features in the Alps, and worldwide, in shaping water quantity and
379 quality downstream under global warming.

380

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390

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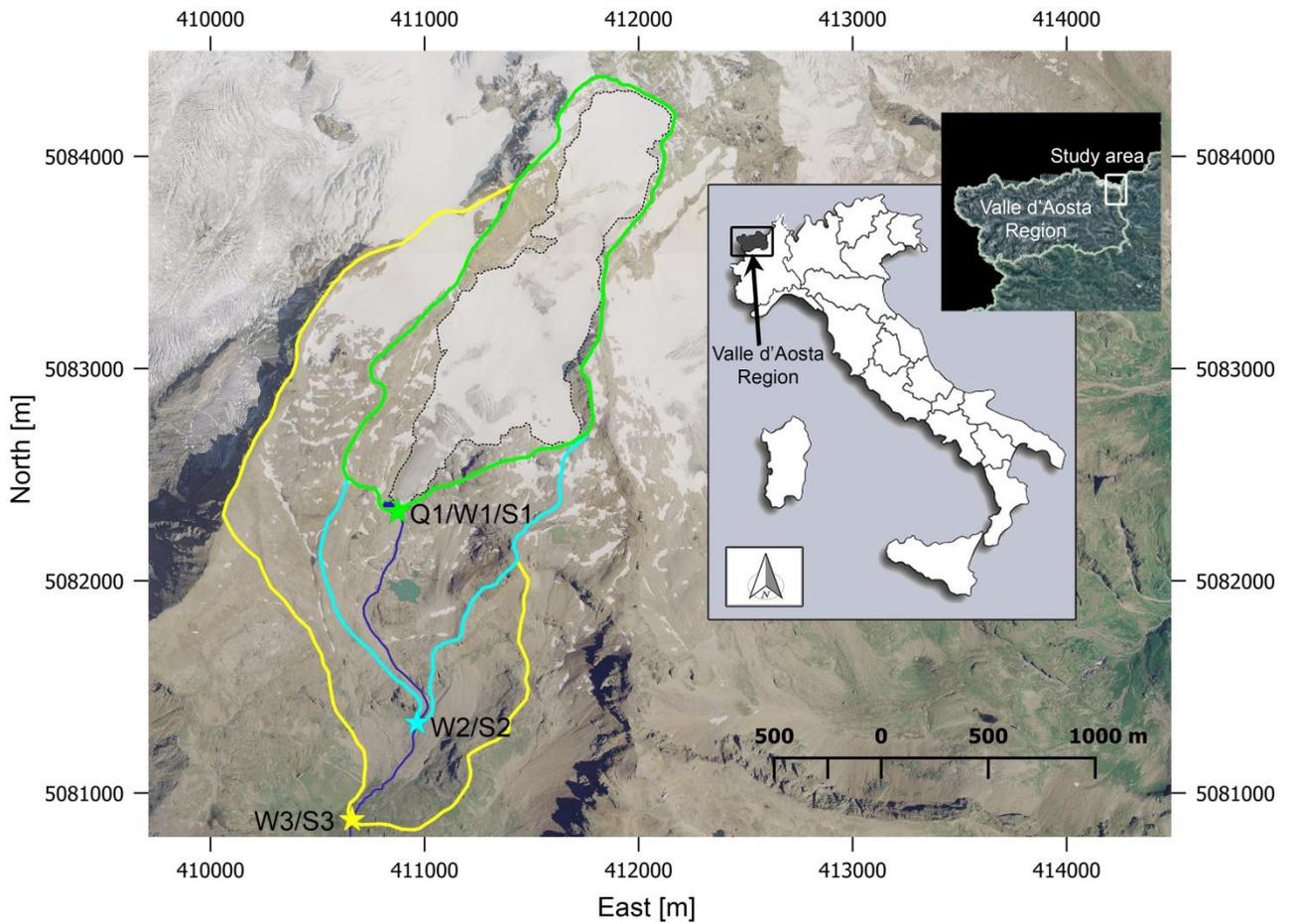
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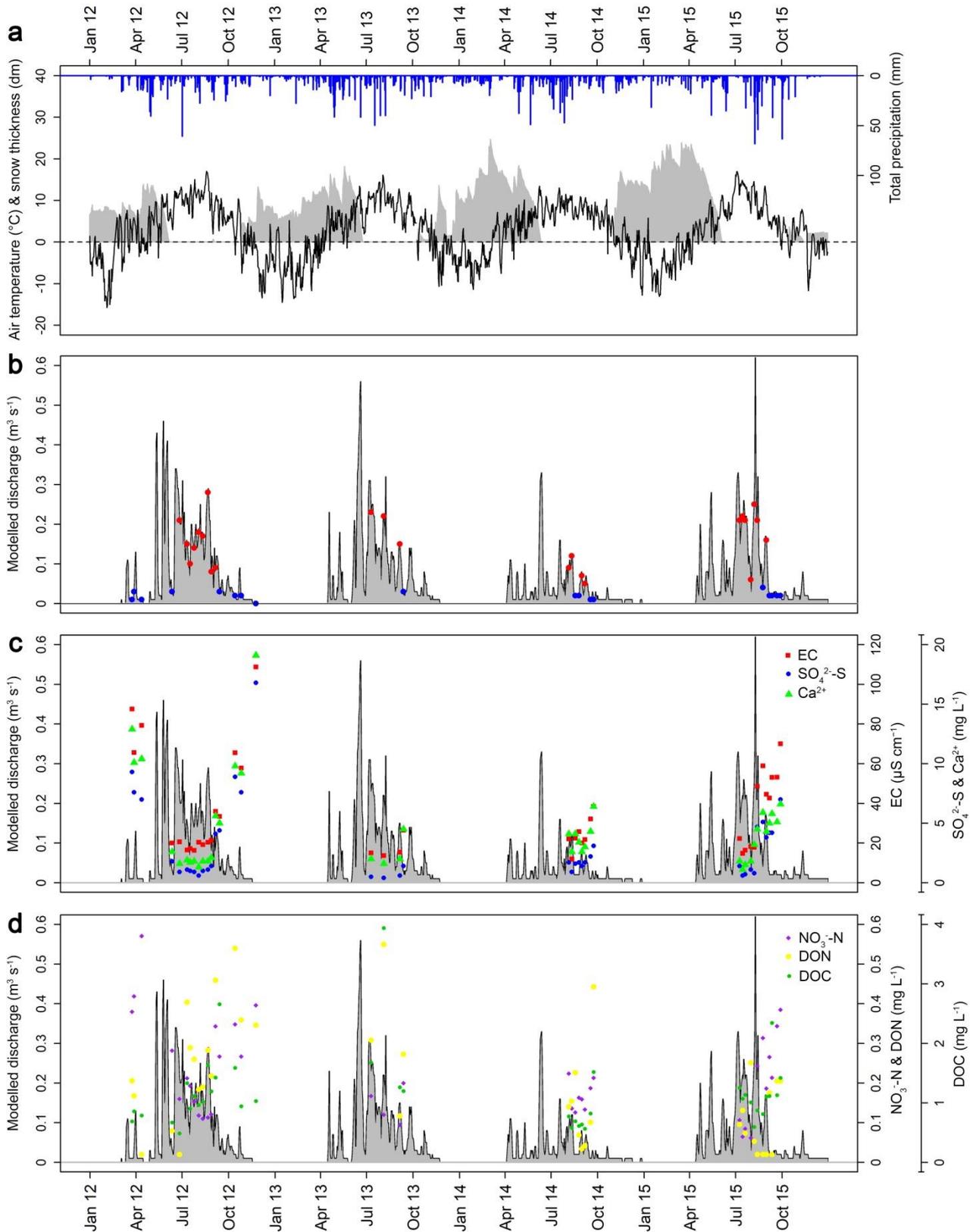
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580 **Figures and tables**



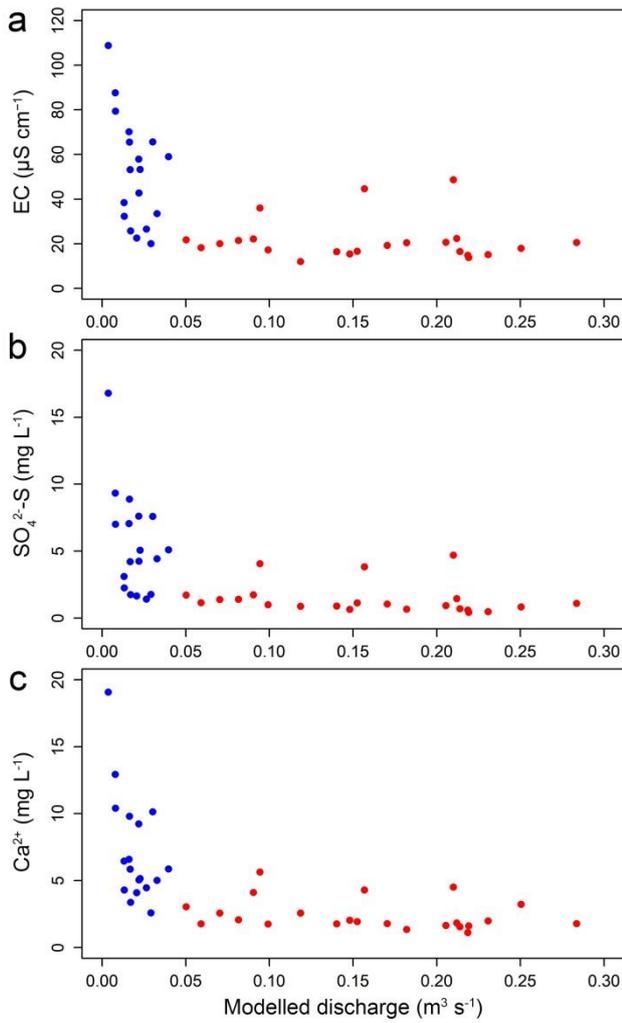
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582 Fig. 1 Location of the study area in Italy and in the Valle d'Aosta Region (www.pcn.minambiente.it), and aerial
583 overview of the study area (orthoimage year 2006) (coordinate system WGS84 / UTM 32N). Black dotted polygon
584 indicates the glacier surface. Stars refer to sampling sites along the glacial stream. Polygons refer to sampling site
585 basins: green - Q1/W1/S1, cyan - W2/S2, and yellow - W3/S3. Details of sampling sites are reported in Table 1.



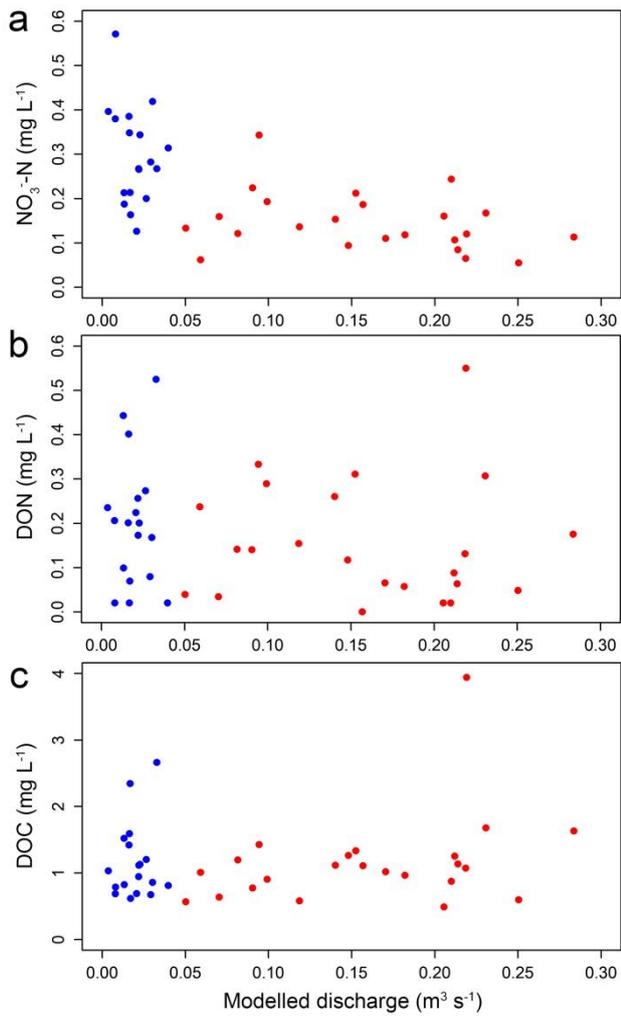
586

587 Fig. 2 (a) Mean daily air temperature (black line), snow thickness (grey polygons), and total precipitation (blue
 588 histograms) (Gabiet AWS, Tab. S1). (b) Modelled glacier discharge at Q1 site (grey polygons with black contour). Blue
 589 and red dots indicate the sampling dates during low- and high-flow periods, respectively. Time series of relation
 590 between modelled glacier discharge and (c) EC / $\text{SO}_4^{2-}\text{-S}$ / Ca^{2+} , and (d) $\text{NO}_3^- \text{-N}$ / DON / DOC.



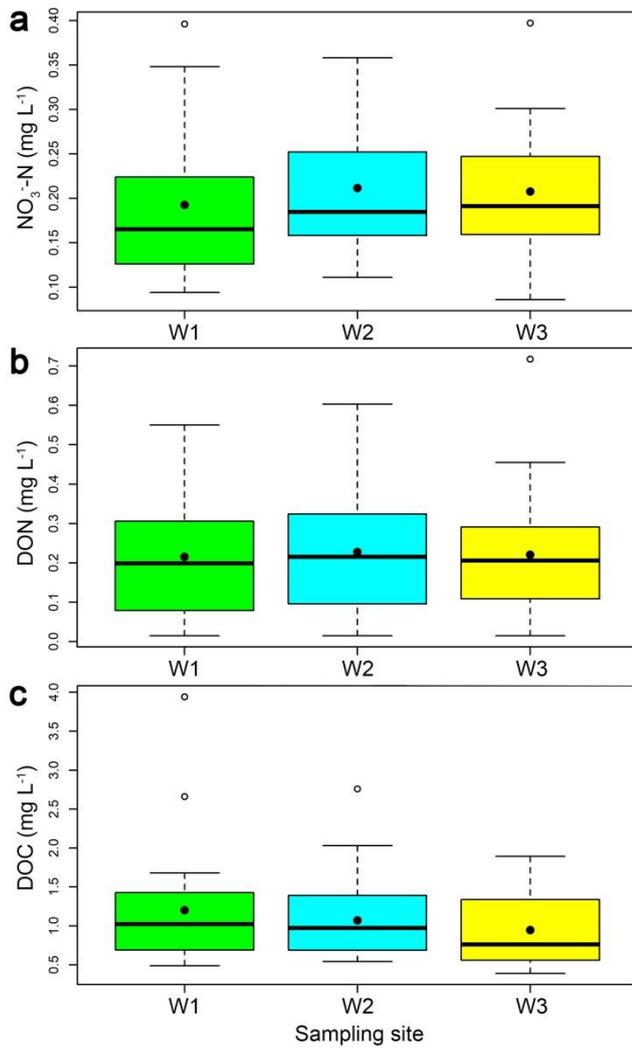
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592 Fig. 3 Modelled glacier discharge versus (a) EC, and (b) SO₄²⁻⁻-S and (c) Ca²⁺ concentrations. Blue and red dots indicate
 593 the sampling dates during low- and high-flow periods, respectively.



594

595 Fig. 4 Modelled glacier discharge versus (a) NO₃⁻-N, (b) DON, and (c) DOC concentrations. Blue and red dots indicate
 596 the sampling dates during low- and high-flow periods, respectively.



597

598 Fig. 5 (a) NO₃⁻-N, (b) DON, and (c) DOC concentrations in the glacial stream along the elevation gradient (n = 26).

599 Black dots indicate the mean. Sampling site locations and relative basins are reported in Figure 1. Further details about

600 the sampling sites are reported in Table 1.

Parameter	Indren Glacier	Parameter	Q1/W1/S1 sampling site	W2/S2 sampling site	W3/S3 sampling site
Elevation range (m a.s.l.)	3083 – 4081	Elevation (m a.s.l.)	3083	2695	2605
Area, year 2006 (km ²)	0.92	Basin area (km ²)	1.43	2.12	3.59
Length / width (m)	2310 / 740	Distance from the glacier (km)	-	1	1.4
		Basin land cover	Glacier: 64.1%	Glacier: 43.1%	Glacier: 31%
			Bedrock: 32%	Bedrock: 38.9%	Bedrock: 29.2%
			Coarse sediment: 3.7%	Coarse sediment: 16.9%	Coarse sediment: 33.7%
			Soil: 0%	Soil: 0.4%	Soil: 5.6%
			Water surface: 0.2%	Water surface: 0.7%	Water surface: 0.4%

601 Table 1 Morphometric characteristics of the Indren Glacier, and morphometric and land cover characteristics of the sampling sites. The relevant maps are reported in Figure 1.

Parameter	Low-flow period	High-flow period
EC ($\mu\text{S cm}^{-1}$, 20 °C)	52.3 \pm 24.6	21.3 \pm 9.3
SO ₄ ²⁻ -S (mg L ⁻¹)	5.50 \pm 3.82	1.41 \pm 1.16
Ca ²⁺ (mg L ⁻¹)	7.23 \pm 4.08	2.42 \pm 1.17
NO ₃ ⁻ -N (mg L ⁻¹)	0.30 \pm 0.11	0.15 \pm 0.07
DON (mg L ⁻¹)	0.20 \pm 0.15	0.15 \pm 0.14
DOC (mg L ⁻¹)	1.16 \pm 0.57	1.15 \pm 0.69

602 Table 2 EC, and SO₄²⁻-S, Ca²⁺, NO₃⁻-N, DON, and DOC concentrations during low- and high-flow periods. Mean \pm
603 standard deviation are reported.

Sampling site	Horizon	Depth (cm)	OC (%)	TN (%)
S1	C	0-10	0.14	0.05
S2	A1	0-2	3.86	0.27
	A2	2-10	0.96	0.07
	AC	10-20+	0.35	0.02
S3	A	0-2	5.52	0.37
	CA	2-7	0.87	0.05
	2CB	7-20+	0.22	0.02

608 Table 3 Soil characteristics across the elevation gradient. Locations and details of sampling sites are reported in Figure
609 1 and Table 1.