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1 **Shape-engineered titanium dioxide nanoparticles (TiO₂-NPs): cytotoxicity and**
2 **genotoxicity in bronchial epithelial cells**

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4 Marta Gea^a, Sara Bonetta^{a*}, Luca Iannarelli^b, Andrea Mario Giovannozzi^b, Valter Maurino^c,
5 Silvia Bonetta^a, Vasile-Dan Hodoroaba^d, Caterina Armato^{a,e}, Andrea Mario Rossi^b, Tiziana
6 Schilirò^a

7

8 ^aDepartment of Public Health and Pediatrics, University of Turin, Piazza Polonia 94, 10126
9 Turin, Italy;

10 ^bQuality of Life Division, National Institute of Metrological Research, Strada delle Cacce 91,
11 10135 Turin, Italy;

12 ^cDepartment of Chemistry, University of Turin, Via Giuria 7, 10125 Turin, Italy;

13 ^dSurface Analysis and Interfacial Chemistry division, Federal Institute for Materials Research
14 & Testing (BAM), 12200 Berlin, Germany;

15 ^eCentre for Sustainable Future Technologies (CSFT@PoliTo), Italian Institute of Technology,
16 Corso Trento 21, 10129 Turin, Italy;

17

18 ***Corresponding author:**

19 Sara Bonetta

20 Department of Public Health and Pediatrics,

21 University of Torino,

22 Piazza Polonia 94, 10126 Turin, Italy,

23 Tel: +390116708192

24 e-mail address: sara.bonetta@unito.it

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26 **List of Abbreviations**

27 BEAS-2B – Human bronchial epithelial cells

28 D_h – Hydrodynamic diameter

29 Fpg – formamidopyrimidine glycosylase

30 NP – Nanoparticle

31 TiO_2 – Titanium dioxide

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51 **Abstract**

52 The aim of this study was to evaluate cytotoxicity (WST-1 assay), LDH release (LDH assay)
53 and genotoxicity (Comet assay) of three engineered TiO₂-NPs with different shapes
54 (bipyramids, rods, platelets) in comparison with two commercial TiO₂-NPs (P25, food grade).
55 After NPs characterization (SEM/T-SEM and DLS), biological effects of NPs were assessed
56 by exposing BEAS-2B cells in the presence of light and in the absence. The cellular uptake of
57 NPs was analyzed using Raman spectroscopy.

58 The cytotoxic effects were mostly slight. After light exposure, using the WST-1 assay, the
59 largest cytotoxicity was observed for rods; P25, bipyramids and platelets showed a similar
60 effect, while no effect was induced by food grade. No LDH release was detected using the
61 LDH assay, confirming the low effect on plasma membrane. Regarding genotoxicity, food
62 grade and platelets induced direct genotoxic effect while P25, food grade and platelets caused
63 oxidative DNA damage. No genotoxic or oxidative DNA damage was induced by bipyramids
64 and rods. In darkness biological effects were overall lower than after light exposure.
65 Considering that only food grade, P25 and platelets (more agglomerated) were internalized by
66 cells, the uptake resulted correlated with genotoxicity.

67 In conclusion, cytotoxicity of NPs was low, influenced by shape as well as by light exposure.
68 Instead, genotoxicity seemed to be influenced by cellular-uptake and aggregation tendency.

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70 **Keywords:** shape-engineered TiO₂ nanoparticles; genotoxic and oxidative damage; Comet
71 assay; cytotoxicity; Raman spectroscopy.

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76 **1. Introduction**

77 Nanoparticles (NPs) are defined as particles having their three dimension in the range of 1 –
78 100 nm (ISO 2015). Actually, many consumer products incorporates NPs. The technological,
79 medical and economic benefits of NPs are considerable, but the presence of nanoparticles in
80 the environment could cause adverse effects to humans. NPs have a greater surface area per
81 mass unit, so they potentially have an increased biological activity compared to fine particles.
82 Moreover, NPs size is comparable to the size of cellular structures, so NPs might potentially
83 emulate biological molecules or interfere physically with biological processes (Magdolenova
84 et al. 2012a).

85 TiO₂ is the oxide of titanium and it has different crystalline structures: anatase, brookite and
86 rutile. Brookite is not produced by industry and is not incorporated in commercial products. In
87 contrast, rutile and anatase are largely used in commercial products (Jovanovic 2015). TiO₂ is
88 one of the most frequently applied NPs and it is in the top five NPs used in consumer products
89 (Shi et al. 2013). TiO₂-NPs produced are used primarily as a pigment owing to their brightness,
90 resistance to discoloration and high refractive index. As a pigment, TiO₂-NPs are incorporated
91 in paints, plastic materials, paper, foods, medical products and cosmetics. Due to its catalytic
92 and photocatalytic properties, TiO₂ is also used as an antimicrobial agent and a catalyst for
93 purification of air and water (Bonetta et al. 2013, Tomankova et al. 2015).

94 TiO₂-NPs could be engineered in terms of shapes and sizes by changing synthesis conditions
95 such as raw material, temperature, acidic and alkaline conditions. Engineered TiO₂-NPs with
96 various shapes (e.g. rods, dots and belts) have been prepared for different applications (Bernard
97 and Curtiss 2005, Sha et al. 2015, Wang et al. 2004). In particular engineered fiber-shaped
98 nanomaterials (i.e. nanowires, nanotubes) are very attractive because they showed higher
99 activity and advantages in photocatalysis, charge transfer and sensing applications due to their

100 structure (Hamilton et al. 2009). However, these new and enhanced properties may also induce
101 higher toxicological effects upon exposure with biological tissues.

102 Humans can be exposed to TiO₂-NPs via three portals of entry: oral (mainly via food
103 consumption), dermal (often through cosmetic and sunscreen applications) and inhalation
104 (mainly under occupational and manufacturing conditions) (Warheit and Donner 2015).

105 Based on the evidence that TiO₂ can induce lung cancer in rats, TiO₂-NPs were classified as
106 possibly carcinogenic to humans (group 2B) by the International Agency for Research on
107 Cancer (IARC 2010). Indeed, the inhalation and instillation of rutile and anatase TiO₂-NPs
108 induced lung tumors (Xu et al. 2010), broncho-alveolar adenomas and cystic keratinizing
109 squamous cell carcinomas (De Matteis et al. 2016; Mohra et al. 2006). TiO₂-NPs were also
110 classified as potential occupational carcinogens by the National Institute for Occupational
111 Safety and Health (NIOSH 2011; Chen et al. 2014).

112 Many *in vitro* studies showed cytotoxicity, genotoxicity and oxidative effects induced by TiO₂-
113 NPs through oxidants generation, inflammation and apoptosis (Jugan et al. 2011, Karlsson et
114 al. 2015, Park et al. 2008, Shi et al. 2010). The potential of NPs to cause DNA damage is an
115 important aspect that needs attention due to possible mutations and carcinogenesis. Physico-
116 chemical characteristics of NPs have an important role in toxicity. Different studies showed
117 that biological effects can be influenced by crystalline structure, size, shape, exterior area,
118 agglomeration/aggregation and surface properties (Bhattacharya et al. 2009, Johnston et al.
119 2009). Some studies revealed that crystalline structure probably influences the induced toxicity,
120 in particular the anatase seems to be more reactive (Sayes et al. 2006) and induces more toxic,
121 genotoxic and inflammatory effects, than the rutile (Falck et al. 2009, Petkovic et al. 2011, Xue
122 et al. 2010). However, other studies gave contradictory results with rutile forms being more
123 toxic than anatase (Gurr et al. 2005, Numano et al. 2014, Uboldi et al. 2016). The effect of
124 agglomeration/aggregation of NPs on toxicity is not well understood yet. In recent studies,

125 some authors demonstrated that agglomeration can influence NPs genotoxicity (Magdolenova
126 et al. 2012b, Prasad et al. 2013).

127 Although physico-chemical properties of NPs can have an important role in the impact on their
128 toxicity, only few studies on shape dependent TiO₂ toxicity has been conducted (Allegri et al.
129 2016, Hamilton et al. 2009, Park et al. 2013). Additional studies are needed to evaluate the role
130 of shape on TiO₂-NPs toxicity in order to produce useful data for assessing the safety of
131 engineered NPs.

132 To address this issue, the aim of this study was to investigate cytotoxicity (WST-1 assay), LDH
133 release (LDH assay) and genotoxicity (Comet assay) of three types of engineered TiO₂-NPs
134 of different shapes (bipyramids, rods and platelet NPs) in BEAS-2B (cells isolated from human
135 bronchial epithelium) in comparison with two commercial types of TiO₂-NPs (P25 and food
136 grade). Since the exposure to TiO₂-NPs mainly occurs through respiratory tract (occupational
137 and manufacturing conditions), human cells of the respiratory system (such as BEAS-2B), were
138 selected as a good cell model for *in vitro* toxicology tests. All the TiO₂-NPs in this study were
139 first physico-chemically characterized, even in different culture media to study their
140 agglomeration state, and then they were biologically evaluated. In order to take into account
141 the photocatalytic properties of the TiO₂-NPs, we investigated the cytotoxicity and
142 genotoxicity on BEAS-2B under light exposure and in darkness. Moreover, a modern
143 application of Raman spectroscopy, the 3D confocal Raman imaging, was used to study the
144 uptake of the NPs within the BEAS-2B cells, as the Raman spectra provide information about
145 both organic molecules and solid NPs simultaneously (Ahlinder et al. 2013).

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150 **2. Materials and methods**

151 **2.1 Synthesis and Preparation of TiO₂ NPs dispersion**

152 Rods and bipyramids TiO₂-NPs were synthesized by the forced hydrolysis of an aqueous
153 solution of TiIV(triethanolamine)₂titanatranne (Ti(TEOAH)₂), using triethanolamine (TEOA)
154 as shape controller; pH of synthesis was adjusted by adding 1 M NaOH solution; details of
155 these procedures were previously reported (Iannarelli et al. 2016, Lavric et al. 2017). The
156 synthesis of platelet NPs was performed with a solvothermal method (Han et al. 2009, Zhang
157 et al. 2012). In a typical synthesis: a precise volume of Ti(OBu)₄ was added in a 150 ml Teflon
158 pot and the desired volume of concentrated hydrofluoric acid was added dropwise under
159 stirring. The Teflon pot was sealed and kept under stirring at high temperature (250°C) for 24h
160 in autoclave. The resulting paste was centrifuged three times and washed with acetone and with
161 water (Milli-Q) to remove the residual organics. The synthesis dispersions were subjected to
162 dialysis process (against ultrapure water, using Spectra/Por dialysis membrane tubing MWCO
163 8–14 kDa) in order to clean the medium. To avoid agglomeration and precipitation,
164 dimethylsulfoxide (DMSO 1% in water) was added to the NPs dispersions (final concentration
165 2.5 mg/ml); the dispersions were homogenized using an ultra-sonication procedure (Iannarelli
166 et al. 2016), few hours before the exposure with cells.

167 The same procedure was employed in the preparation of the dispersion of commercial TiO₂
168 powders, which were the P25 NPs (Evonik), extensively used in toxicity studies (Karlsson et
169 al. 2015, Magdolenova et al. 2014, Valant et al. 2012), and the food grade NPs (Faravelli
170 Group), incorporated in many edible products (Weir et al. 2012).

171 **2.2 Scanning Electron Microscopy (SEM) including Transmission Mode (T-SEM)**

172 The dimensional characterization (size and shape) of TiO₂-NPs was carried out with SEM using
173 a Zeiss Supra 40 instrument (Zeiss) equipped with a Schottky field emitter, the standard
174 secondary electrons, i.e. Everhart-Thornley, detector and a high-resolution In-lens detector.

175 The surface-sensitive In-lens SEM mode better suited to morphological/shape analysis and
176 transmission mode in SEM (T-SEM) better suited for dimensional measurements were applied
177 complementary to the same field of view on the sample.

178 ***2.3 Dynamic Light Scattering (DLS) analysis***

179 Delsa Nano™ C Analyzer (Beckman Coulter) equipped with a 638 nm diode laser and a
180 temperature control was used for the DLS measurements. The laser fluctuation was detected
181 on a photomultiplier tube detector positioned behind the cuvette with an angle of 163°.
182 Hydrodynamic diameters were calculated setting temperature at 25°C, viscosity (η) 0.890 cP
183 and refractive index of water 1.3325. In order to simulate the culture medium conditions, DLS
184 analyses were conducted on dilution of TiO₂ dispersions (1:4) in a 1% DMSO aqueous solution,
185 as reference analysis, and in base RPMI 1640 medium [supplemented with l-glutamine (4 mM)
186 and penicillin-streptomycin (100 U/ml - 100 μ g/ml)] and complete RPMI 1640 medium
187 [supplemented with FBS (10% v/v), l-glutamine (4 mM) and penicillin-streptomycin (100
188 U/ml - 100 μ g/ml)].

189 ***2.4 Raman spectroscopy analysis***

190 The aqueous suspensions of the TiO₂-NPs under investigation were freeze-dried to obtain a
191 solid powder. Raman spectroscopy was used in the analysis of dry TiO₂-NPs powder using a
192 DXR™ Raman Microscope (Thermo Scientific) with a laser wavelength at 532 nm, a laser
193 power of 1 mW and a 10x microscope objective. Spectra were collected in the 50–1800 cm⁻¹
194 spectral region, with a grating resolution of 3.3–3.9 cm⁻¹, exposure time of 1 s and 20 scans in
195 total.

196 ***2.5 Cell culture and exposure***

197 BEAS-2B cells, isolated from human bronchial epithelium, were obtained from the American
198 Type Culture Collection (ATCC® CRL-9609™). BEAS-2B were grown as a monolayer,
199 maintained and treated in complete RPMI 1640 medium [supplemented with FBS (10% v/v),

200 l-glutamine (4 mM) and penicillin-streptomycin (100 U/ml - 100 µg/ml)], at 37°C in a
201 humidified atmosphere containing 5% CO₂.

202 The solution of NPs (2.5 mg/ml, DMSO 1% in water) was vortexed and sonicated (30 min) in
203 order to homogenize the NPs. NPs (5 – 160 µg/ml) were directly pipetted in culture plates
204 containing RPMI 1640 medium and then the cell culture plates were mixed on a shaker (10
205 min). The cells were exposed for 1h under laboratory light and then incubated at 37 °C in
206 darkness (23h) (exposure with light). In order to standardize the exposure with light the cells
207 were exposed in a dark room (obscured by daylight) to a normal laboratory lamp (36W/840
208 Lumilux Cool White-36 W, 3350 lm, 4000 K-supplied from OSRAM lighting AG). The lamp
209 illuminance measured with Quantum photo/radiometer HD 9021 (Delta Ohm) was 289±11 lx.
210 To quantify effects due to the photocatalytic activity of TiO₂, cells were exposed for 24h in
211 darkness (exposure in darkness).

212 After exposure, cytotoxicity and genotoxicity assays were performed.

213 **2.6 Cytotoxicity**

214 Cell viability was assessed using Cell Proliferation Reagent WST-1 (Roche). The assay was
215 performed as previously described by Gea et al. (2018). Briefly, BEAS-2B cells were cultured
216 in 75 cm² flasks and maintained until the cells reached 70 % confluence; cells were then seeded
217 in 24-well plates (5×10⁴ cells/well) and allowed to adhere overnight. After that, culture medium
218 was removed and cells were exposed to NPs (5, 10, 20, 50 and 80 µg/ml, equivalent to 1.3, 2.6,
219 5.2, 13.0, 20.7 µg/cm²) for 24h with light or in darkness (as specified in paragraph 2.5). After
220 exposure, WST-1 was added (50 µl/well) and incubated for 3h (37 °C). After incubation, well
221 contents were centrifuged and the supernatants were transferred in 96-well plate to remove the
222 interference owing to the NPs. The absorbance was measured at 440 nm (Tecan Infinite Reader
223 M200 Pro). Absorbance of unexposed cells was used as negative control. Data were expressed

224 as a percentage of viability. All experiments were performed in quadruplicate (four wells for
225 each experimental condition).

226 As indicator of cell membrane damage, lactate dehydrogenase activity was measured in cell-
227 free culture supernatants using the LDH assay kit (Cytotoxicity Detection Kit PLUS, Roche)
228 modified for NPs exposure. Briefly, BEAS-2B cells were cultured in 75 cm² flasks and
229 maintained until the cells reached 70 % confluence, cells were then seeded in 24-well plates (5
230 × 10⁴ cells/well) and allowed to adhere overnight. After that, culture medium was removed and
231 the cells were exposed to NPs (5, 10, 20, 50 and 80 µg/ml, equivalent to 1.3, 2.6, 5.2, 13.0,
232 20.7 µg/cm²) for 24h with light or in darkness (as specified in paragraph 2.5). After exposure,
233 the contents of each well were centrifuged to remove the interference owing to the NPs. Each
234 supernatant (100 µl) was transferred into 96-well plate, mixed with Reaction Mixture (100
235 µl/well) and incubated for 30 min at 15 – 25 °C. After incubation, Stop Solution (50 µl/well)
236 was added and the absorbance was measured at 490 nm (Tecan Infinite Reader M200 Pro).
237 Absorbance measurement of unexposed cells were used as negative control, while absorbance
238 measurement of unexposed cells lysed with Lysis Solution (Cytotoxicity Detection Kit PLUS,
239 Roche) was used as positive control. Data were expressed as a percentage of LDH release,
240 respect to control cells (100%). All experiments were performed in triplicate (three wells for
241 each experimental condition).

242 **2.7 Genotoxicity**

243 The alkaline Comet assay was used for DNA damage evaluation (direct DNA damage). BEAS-
244 2B cells were cultured in 75 cm² flasks and maintained until the cells reached 70 % confluence;
245 cells were then seeded in 6-well plates (3 × 10⁵ cells/well) and cultured overnight before
246 exposure to NPs. The cells were exposed to different doses of NPs (20, 50, 80, 120 and 160
247 µg/ml, equivalent to 5.2, 13.0, 20.8, 31.2, 41.6 µg/cm²) for 24h with light or in darkness (as
248 specified in paragraph 2.5). Unexposed cells and cells treated with DMSO (1%) were used as

249 negative controls. The alkaline Comet assay was performed according to Tice et al. (2000) after
250 slight modifications (Bonetta et al. 2018). After exposure, cells were washed with base RPMI
251 1640 and PBS, detached using trypsin-EDTA (1x) and cell viability was determined (trypan
252 blue staining). Cells were then centrifuged and mixed with low melting point agarose (0.7%),
253 placed on the slides coated with normal melting agarose (1%) and low melting point agarose
254 added as the top layer. The slides were immersed in lysis solution in the dark overnight (8 mM
255 Tris-HCl, 2.5 M NaCl, 100 mM EDTA disodium salt dihydrate, 1% TRITON X-100 and 10%
256 DMSO, pH 10, 4°C). For the unwinding, the slides were immersed in alkaline electrophoresis
257 buffer (20 min) (1 mM EDTA, 300 mM NaOH, 10% DMSO, pH > 13) and the electrophoresis
258 was carried out in the same buffer (20 min, 1 V/cm and 300 mA). The slides were washed with
259 neutralization buffer (0.4 M Tris-HCl, pH 7.5, 4 °C, 3 min), fixed with ethanol 70% (-20 °C, 5
260 min) and air dried. All steps were performed under yellow light to prevent additional DNA
261 damage. Slides were stained with ethidium bromide (20 µg/ml) and analyzed using a
262 fluorescence microscope (Axioskop HBO 50, Zeiss). The percentage of tail intensity was used
263 to estimate DNA damage. A total of a hundred randomly selected cells per treatment (two gels
264 per slides) were analyzed using the Comet Assay IV software (Perceptive Instruments, Instem).
265 Two independent experiments were performed for each experimental condition.

266 Genotoxic effect (direct DNA damage) was evaluated comparing cells exposed to NPs with
267 control cells (DMSO 1%).

268 The formamidopyrimidine glycosylase (Fpg)-modified Comet assay was performed for DNA
269 damage evaluation (direct + indirect DNA damage) as reported in Bonetta et al. (2009) with
270 slight modification (Gea et al. 2018). The test was carried out as described for the alkaline
271 Comet assay but, after lysis, the slides were washed with Fpg Buffer (5 min for three times)
272 (40 mM Hepes, 0.1M KCl, 0.5 mM EDTA disodium salt dihydrate, 0.2 mg/ml bovine serum

273 albumin, pH 8). Then, each gel was incubated with 1 unit of Fpg enzyme (*Escherichia coli*)
274 (TREVIGEN) at 37°C for 30 min. Procedure control slides were incubated with buffer only.
275 Cells treated with DMSO (1%) and enzyme were used as negative controls. Two independent
276 experiments were performed for each experimental condition.

277 The DNA damage (direct + indirect DNA damage) was evaluated comparing cells exposed to
278 NPs with control cells (DMSO 1% +Fpg).

279 The oxidative damage was calculated subtracting the mean tail intensity (%) in enzyme-treated
280 cells (+Fpg) from the relative mean tail intensity (%) in enzyme-untreated cells (-Fpg).

281 **2.8 3D confocal micro-Raman imaging spectroscopy**

282 Raman grade Calcium fluoride (CaF₂) windows (Crystran Tachnology srl) were employed as
283 alternative substrate instead of standard plastic substrates for cells growing due to the low
284 toxicity and almost absent background signals (Kann et al. 2015). The BEAS-2B cells were
285 cultured overnight in 6-well plates on a CaF₂ substrate (3×10^5 cells/well) before exposure to
286 NPs. Cells were treated with NPs (80 µg/ml, 24h). After exposure, cells were washed twice
287 with PBS and fixed with 3 ml of methanol. CaF₂ substrates were dried and stained with Giemsa
288 dye (4% Giemsa's azur eosin methylene blue solution, 4% Sorensen buffer 0.067 M pH 6.8, 8
289 min at room temperature), then washed twice with distilled water and dried. Giemsa staining
290 is one of the standard procedures in histology, useful to evidence morphological cells features,
291 such as cell nuclei, which appear in various shades of red/purple, and the cytoplasm, which
292 appears blue.

293 3D confocal micro-Raman imaging spectroscopy of BEAS-2B cells was conducted with a
294 DXR™xi Raman Imaging Microscope (Thermo Scientific) using a laser wavelength at 532
295 nm, a 1 mW laser power, a 100X microscope objective and a motorized stage with a 1 µm of
296 step size and a 1 µm offset. Spectra were collected in the 50–3500 cm⁻¹ spectral region with a
297 grating resolution of 5 cm⁻¹, an exposure time of 0.025 s and 5 scans in total. 3D Raman images

298 were reconstructed taking the Raman peaks at 1600 cm^{-1} of methylene blue and the E_g band at
299 144 cm^{-1} of the TiO_2 -NPs, respectively. Each cell was investigated at different focal planes and
300 a chemical image was obtained by the combination of the $\nu(\text{C-C})$ ring at 1600 cm^{-1} of the
301 methylene blue and the E_g band at 144 cm^{-1} of the TiO_2 -NPs. Since methylene blue is contained
302 in the Giemsa stain and it is widely distributed into the fixed cells, its signals were considered
303 representative of the entire volume of the cells. As far as the tracking of the NPs are concerned,
304 the E_g band at 143 cm^{-1} is the most intense signal in the molecular fingerprint of the anatase
305 TiO_2 and the region between 50 cm^{-1} and 400 cm^{-1} in the Raman spectrum is usually free of
306 the vibrational bands of biological species. Therefore, this signal was selected to sensitively
307 locate the TiO_2 -NPs inside the cells. Image J software was used in the development of the 3D
308 chemical images both for cells and TiO_2 -NPs, which were superimposed using a Solidworks®
309 2016 Cad based software. 3D Raman chemical images are presented using a color meshwork
310 i.e. blue for cell tissues and red for TiO_2 agglomerates.

311 **2.9 Statistical analysis**

312 IBM SPSS software (ver. 24.0) was used to perform statistical analysis. The results of WST-
313 1, LDH and Comet assays are presented as the mean \pm standard deviation. Differences between
314 exposed and control cells were tested by ANOVA followed by the post hoc Dunnett's test
315 procedure. Differences between light and dark exposure were tested by ANOVA, followed by
316 the post hoc Tukey's test procedure. Data were considered statistically different for a p-value
317 less than 0.05.

318

319 **3. Results**

320 **3.1 Raman characterization of NPs and size distribution**

321 In order to establish a relationship among the physico-chemical features of NPs and their ability
322 to induce a toxic effect, well-defined and controlled protocols were developed for the

323 production of engineered anatase TiO₂-NPs with different shapes. All the NPs produced in this
324 study were first characterized with a SEM equipped with a transmission-unit for T-SEM, which
325 provided information both on the shape and the size of the constituent NPs (Fig. 1a-e). The Fig.
326 1 and Table 1 show shapes and particle size of commercial TiO₂-NPs and fabricated engineered
327 TiO₂-NPs.

328 These NPs were also characterized by Dynamic Light Scattering (DLS) as a quick method for
329 sizing and determining the state of NP agglomeration. For each kind of sample, the
330 agglomeration in 1% DMSO aqueous solution, in base RPMI [supplemented with l-glutamine
331 (4 mM) and penicillin-streptomycin (100 U/ml - 100 µg/ml)] and complete RPMI
332 [supplemented with FBS (10% v/v), l-glutamine (4 mM) and penicillin-streptomycin (100
333 U/ml - 100 µg/ml)] (Fig. 1f-j) were compared. In all the TiO₂ materials considered for this
334 study, the agglomeration state increase in base RPMI, while the size distribution in DMSO and
335 in complete RPMI is quite similar.

336 The crystalline composition of the TiO₂-NPs, analyzed by Raman spectroscopy, showed a
337 typical fingerprint of the anatase TiO₂ (Fig. S.1) with the characteristic phonon bands E_g at
338 143 cm⁻¹, E_g at 197 cm⁻¹, A_{1g} at 397 cm⁻¹, B_{1g} at 515 cm⁻¹ and E_g at 639 cm⁻¹ for all the
339 investigated NPs. Since P25 is a known mixture of anatase and rutile (5:1), with also a small
340 amount of amorphous TiO₂ (Ohtani et al. 2010), its Raman spectrum still retains all the typical
341 anatase Raman bands but it also contains two small shoulders at 450 cm⁻¹ and 600 cm⁻¹, which
342 were assigned to the E_g and A_{1g} phonon bands, respectively, of rutile (Tompsett et al. 1995).

343 All the physiochemical properties of the TiO₂-NPs under study such as shape, particle size,
344 hydrodynamic diameter in different liquid media and the crystalline phase are summarized in
345 Table 1.

346 **3.2 Cytotoxicity**

347 The results of the effects of different TiO₂-NPs concentration on cell viability (WST-1 assay)
348 are reported in Fig. 2a (exposure with light) and in Fig. 2b (exposure in darkness).
349 In general, a low cytotoxic effect was observed at the tested doses both in the exposure with
350 light and in the exposure in darkness. The observed viability ranged from 102.8 to 88.4% for
351 the exposure with light and from 99.6 to 87.4% for the exposure in darkness.
352 Considering the exposure with light, the commercial P25 induced a slight decrease in viability
353 starting from the doses of 50 µg/ml ($p < 0.05$) while no cytotoxic effects were observed for the
354 other commercial NPs (food grade) at the tested concentrations. As far as engineered NPs are
355 concerned, bipyramids and platelet NPs induced the same cytotoxic effect of commercial P25
356 NPs; on the contrary, rods is the NP shape with higher cytotoxic effect showing a viability
357 decrease already starting from 10 µg/ml ($p < 0.05$ or $p < 0.001$).
358 Considering the exposure in darkness, a lower cytotoxic effect was observed for commercial
359 P25 NPs with respect to light exposure because a slight decrease in viability was observed for
360 P25 NPs only at the highest dose (80 µg/ml) ($p < 0.05$). As reported after exposure with light,
361 no cytotoxic effect was observed for the other commercial NPs (food grade). About engineered
362 NPs, the exposure in darkness did not modify the cytotoxic effect of bipyramids NPs resulting
363 in a viability reduction starting from the dose of 50 µg/ml ($p < 0.001$) as reported in the
364 experiment with light. In contrast, in the darkness, rods NPs showed a lower cytotoxic effect
365 than observed with light because a slight decrease in viability was observed for rod NPs only
366 starting from the dose of 20 µg/ml ($p < 0.05$). As during the exposure with light, platelet NPs
367 induced a decrease in viability; the cytotoxic effect was significant starting from a less dose
368 (10 µg/ml, $p < 0.05$) than in the experiment with light (50 µg/ml).
369 The results of the effects of different TiO₂-NPs concentration on LDH release has been reported
370 in Fig. 2c (exposure with light) and in Fig. 2d (exposure in darkness).

371 No significant LDH release was detected using LDH assay in both exposure protocols (with
372 light or in darkness), confirming the low cytotoxic effect evidenced by WST-1 assay.

373 **3.3 Genotoxicity**

374 The results of genotoxic effect and oxidative DNA damage induced by different concentration
375 of NPs are reported in Fig. 3.

376 Considering the exposure with laboratory light, no genotoxic effect was showed in enzyme
377 untreated cells (direct DNA damage) for commercial P25 NPs (Fig. 3a). On the other hand, a
378 dose-dependent increase of DNA damage was observed for these NPs in enzyme treated cells
379 (direct and indirect DNA damage) respect to the control cells ($p<0.05$ or $p<0.001$), with the
380 exception of the last dose (160 $\mu\text{g/ml}$) that induced a DNA damage equal to 80 $\mu\text{g/ml}$. A
381 significant oxidative damage was observed for P25 NPs starting from 50 $\mu\text{g/ml}$ ($p<0.05$ or
382 $p<0.001$). The results obtained with the other commercial NPs (food grade)(Fig. 3b) showed
383 the presence of a significant dose-response DNA damage both in enzyme untreated cells and
384 in enzyme treated cells starting from 50 $\mu\text{g/ml}$. Moreover, the difference between the two
385 effects resulted significant starting from 50 $\mu\text{g/ml}$ ($p<0.05$ or $p<0.001$) highlighting an
386 oxidative damage induced by food grade NPs.

387 Respect to commercial NPs, engineered NPs showed a lower extent of DNA damage. In
388 particular, neither genotoxic effect nor oxidative damage were observed for engineered
389 bipyramids and rods NPs (Fig. 3c,d). Platelet NPs induced a significant DNA damage respect
390 to the control cells ($p<0.05$ or $p<0.001$) both in enzyme untreated cells and in enzyme treated
391 cells and they induced a significant oxidative DNA damage starting from 80 $\mu\text{g/ml}$ ($p<0.001$)
392 (Fig. 3e). However in contrast with commercial NPs (food grade), a dose-response of the
393 effects were not observed.

394 As demonstrated by other authors (Kalsson 2010, Karlsson et al. 2015), an interference during
395 the scoring of the assay was detected in particular at the higher doses of P25 and platelet NPs,

396 indeed nanoparticles with some autofluorescence were visible in the comets “head” and the
397 stained DNA appeared faded. The interference probably caused the loss of concentration-
398 dependent increase in DNA direct and oxidative damage observed for the higher doses. The
399 phenomenon could be explained also considering that base oxidation is hard to measure
400 accurately when there are a lot of strand breaks, because the Comet assay becomes saturated
401 (Collins et al. 2017).

402 In order to evaluate the role of the light on the genotoxic and oxidative damage induced by
403 commercial and engineered NPs, the highest doses (80, 120, 160 $\mu\text{g/ml}$) of NPs that showed a
404 genotoxic effect (P25, food grade and platelet NPs) were tested in darkness (24h).

405 Considering the exposure in darkness, no genotoxic effect was observed for commercial P25
406 NPs in enzyme untreated cells (direct DNA damage) (Fig. 3f) as reported in the experiment
407 with light (Fig. 3a). However, in the enzyme treated cells a dose-response DNA damage (direct
408 and indirect DNA damage) was observed with respect to control cells ($p < 0.05$ or $p < 0.001$), but
409 oxidative DNA damage was lower than in the experiment with light ($p < 0.05$ or $p < 0.001$). The
410 commercial food grade NPs induced a significant dose-response DNA damage both in enzyme
411 untreated cells and in enzyme treated cells ($p < 0.001$ and $p < 0.05$ respectively) (Fig. 3g).
412 However, the DNA damage resulted in both cases lower than in the experiment with light
413 $p < 0.05$ or $p < 0.001$) and an oxidative damage was induced only at the highest dose (160 $\mu\text{g/ml}$)
414 ($p < 0.05$).

415 With regard to engineered NPs, platelet NPs induced a significant DNA damage with respect
416 to the control cells ($p < 0.05$ or $p < 0.001$) both in enzyme untreated cells and in enzyme treated
417 cells (Fig. 3h). However, while the DNA damage in enzyme untreated cells was equivalent to
418 the DNA damage induced in the experiment with light (Fig. 3e), a decrease of DNA damage
419 in enzyme treated cells was observed, resulting in no oxidative damage induced by platelet NPs
420 in darkness (Fig. 3h).

421 **3.4 Confocal micro-Raman spectroscopy**

422 The confocal micro-Raman imaging spectroscopy was used in order to evaluate qualitatively
423 the presence/absence of different types of TiO₂-NPs inside the cells. 3D chemical images are
424 built by superimposing the different maps of each cell at their corresponding focal planes and
425 they are presented using a color meshwork i.e. blue for cell tissues and red for TiO₂
426 agglomerates. At least five cells were analyzed to provide statistically significant results. As
427 the sections of Fig. 4 show, the uptake of the TiO₂-NPs by the cells was mainly demonstrated
428 for P25, food grade and platelet NPs (Fig. 4a,b,c) while no TiO₂ signal was registered inside
429 the cells for bypiramids and rods (Fig. 4d,e).

430

431 **4. Discussion**

432 Many *in vitro* studies have been conducted to investigate cytotoxicity/genotoxicity of TiO₂-
433 NPs but the results are often conflicting and employed doses were sometimes high (Valant et
434 al., 2012; Chen et al., 2014; Magdolenova et al., 2014; Karlsson et al., 2015; Moller et al.,
435 2015a). The aim of this study was to investigate the cytotoxicity and genotoxicity of three
436 different shapes of TiO₂-NPs and to compare them with two commercially available TiO₂-NPs.
437 The issues taken into account for this study were: i) the physico-chemical properties of the
438 particles (shape, particle size, agglomeration state in culture media, crystalline phase) that can
439 influence biological effects, ii) the ability of the particles to induce cytotoxicity and
440 genotoxicity, iii) the increase of the toxicological effects under light exposure due to the
441 photocatalytic activity of TiO₂ and iv) the uptake of the NPs by human cells.

442 In the present study, the cytotoxicity assays were selected in order to reduce the interference of
443 NPs with the assays (interference with optical detection methods, ability to convert the
444 substrates). Moreover, as suggested by other studies (Wilhelmi et al., 2012; Guadagnini et al.,
445 2015; Popescu et al., 2015), the assays processes were optimized for evaluation of NPs and

446 (before the assessment of NPs cytotoxicity) relevant controls, assessing particles for their
447 ability to interfere with the assays, were conducted (data not shown).

448 For genotoxicity evaluation, a literature revision (on application of genotoxicity assays testing
449 NPs) was made before the application of Comet assay (Karlsson 2010; Magdolenova et al.,
450 2012a; Karlsson et al., 2015; Cowie et al., 2015; Moller et al., 2015b; Huk et al., 2015). To
451 ensure the correct evaluation of DNA damage two independent experiment were performed for
452 each experimental condition. The analysis of each comet was made using the Comet Assay IV
453 and the automatic evaluation proposed by the software was carefully checked by an operator.

454 Published results on toxicity of TiO₂-NPs show high variability. Reasons for this variability
455 include physico-chemical characteristics of NPs, different methods to prepare NPs dispersions,
456 differences in NPs size and dispersion stability, and different exposure protocols (Charles et
457 al., 2018). The characteristics of NPs dispersion can be influenced by medium components,
458 such as serum proteins, and by NPs properties (size, shape, surface charge, surface coating etc.)
459 (Huk et al. 2015). According to the study of Prasad et al. (2013), the present results showed
460 that in all the TiO₂-NPs dispersions, the agglomeration state increases in base RPMI (without
461 serum), while the size distribution in DMSO and in complete RPMI medium (with serum) is
462 quite similar. The different agglomeration state is probably due to the ability of metal oxide
463 NPs to adsorb proteins onto their surface, forming a “protein corona” which favors less
464 agglomeration in complete medium, which contains more proteins (Prasad et al. 2013).
465 Considering the results obtained, complete medium was selected as cytotoxicity/genotoxicity
466 assay medium.

467 The viability of BEAS-2B treated with commercial and engineered TiO₂-NPs after exposure
468 with light or in darkness was assessed using the WST-1 assay.

469 Commercial TiO₂-NPs induced low (P25) or no viability reduction (food grade) detected by
470 WST-1 assay; these results are in agreement with some reports on commercial TiO₂-NPs

471 (Bhattacharya et al. 2009, Falck et al. 2009). Previous studies that investigated the cytotoxicity
472 of commercial P25 on BEAS-2B showed that only 100 $\mu\text{g/ml}$ of commercial P25 NPs produced
473 a viability decrease after 24h exposure (Prasad et al. 2013). Fewer studies have been performed
474 using commercial food grade TiO_2 -NPs. Proquin et al. (2017) tested these NPs on different cell
475 lines: on Caco-2, they observed cytotoxicity, while on HCT116 they did not observe any
476 cytotoxic effect up to the concentration of 100 $\mu\text{g/cm}^2$. The result obtained on HCT116 was in
477 accordance with the low cytotoxic effect induced by commercial food grade TiO_2 -NPs detected
478 in the present study. Recently, the scientific community have produced reference NPs, which
479 have been well characterized. Di Bucchianico et al. (2016) assessed cytotoxic effects of some
480 of these NPs (anatase 50-150 nm, anatase 5-8 nm, rutile 20-28 nm) in BEAS-2B cells and,
481 according to the present results, showed in general no or low effects at the tested doses (2-100
482 $\mu\text{g/ml}$).

483 On the contrary, other studies showed that commercial TiO_2 -NPs induced higher cytotoxicity
484 on BEAS-2B (Shi et al. 2010, Ursini et al. 2014). In particular, Park et al. (2008) found that
485 exposure of BEAS-2B cells to commercial P25 (5-40 $\mu\text{g/ml}$) for 24h led to significant cell
486 death, both in a time- and concentration-dependent manner.

487 The data of present study demonstrated that cytotoxicity was slightly affected by light
488 exposure, which induced an increase of cellular damage after incubation with commercial P25
489 and engineered rods. The influence of light exposure on cytotoxicity was also observed in other
490 studies (Vevers and Jha 2008, Reeves et al. 2008). Differently from P25 and rods, exposure to
491 platelet NPs induced higher cytotoxicity in darkness than after light exposure; the mechanism
492 that led to this result is not clear.

493 Comparing the results of cytotoxicity (WST-1 assay) and LDH release, the first showed low
494 cytotoxic effect at the doses tested, while the second did not show any cytotoxicity in both
495 exposure protocols. The discrepancy between cytotoxicity (WST-1) and LDH release data

496 suggests that the viability reduction may be caused by apoptosis, a cell death pathway in which
497 the plasma membrane is maintained, as observed in other studies (Schilirò et al. 2015). This is
498 in accordance with previous studies, which demonstrated that TiO₂-NPs could cause apoptosis
499 in BEAS-2B cells (Park et al. 2008, Shi et al. 2010). The observed discrepancy could be also
500 explained considering that the tested compounds (TiO₂-NPs) could induce an effect on the
501 intracellular activity (mitochondria activity) without causing plasma membrane breakage, as
502 observed by other authors (Weyermann et al. 2005, Fotakis and Timbrell 2006).

503 Results of Comet assay in presence of light and in darkness showed a significant DNA damage
504 induced by commercial P25 and food grade NPs and engineered platelet NPs, while no
505 genotoxicity was observed with the other engineered NPs (bipyramids and rods).

506 Considering that the uptake of NPs could involve interactions of NPs with DNA, the observed
507 genotoxic effect could be related to the presence of P25, food grade and platelet NPs into the
508 BEAS-2B as observed by other authors (Bhattacharya et al. 2009, Park et al. 2008).

509 In the present study, the higher uptake of P25, food grade and platelet NPs seemed to be related
510 with higher agglomeration tendency (higher measure of hydrodynamic diameter) (table 1). In
511 particular, the engineered platelet NPs were the most agglomerated (platelet shape could
512 probably promote more agglomeration than the other shapes) and commercial P25 and food
513 grade were more agglomerated than the other engineered NPs (bipyramids and rods). The
514 variation in cellular uptake could be due to agglomeration tendency because NPs that form
515 large agglomerates, differently from NPs that form smaller ones, precipitate at the bottom of
516 the cell culture wells, increasing the real amount of NPs to which cells are exposed
517 (Magdolenova et al. 2012b). Cells exposed to more NPs could probably internalize more NPs.
518 Then, in the present study, the agglomeration tendency does not seem to have prevented the
519 uptake of NPs in the cells, in accordance with the study of Ahlinder et al. (2013).

520 The major uptake of P25, food grade and platelet NPs could be related with higher genotoxic
521 effect considering that, after penetration into the cells, NPs may have direct access to DNA via
522 transport into the nucleus and/or during mitosis when the dissolution of nuclear membrane
523 occurs. NPs interacting directly with DNA could cause DNA breakage (Magdolenova et al.,
524 2014). Moreover, NPs, after penetration into the cells, can enhance the permeability of the
525 lysosomal membrane, inducing the release of DNases and so causing genotoxic effects
526 (Karlsson et al., 2010). Finally, accumulation of NPs within cells can cause aggregates of NPs
527 that deform nucleus inducing DNA damage (Di Virgilio et al., 2010).

528 In order to quantify effects due to the photocatalytic activity of TiO₂, the highest doses (80,
529 120, 160 µg/ml) of NPs that showed a genotoxic effect were tested also in darkness (24h).
530 Results obtained in this study showed that light exposure induced additional indirect
531 genotoxicity, demonstrating a higher oxidative potential of TiO₂-NPs after exposure with light.
532 The presence of light increased DNA oxidative damage probably due to the photocatalytic
533 activity of TiO₂-NPs, which caused an increase of NPs ability to produce radicals. In particular,
534 based on previous studies, the anatase crystal structure of TiO₂ (the same used in the present
535 study) seems to be the most catalytic/photocatalytic crystalline structure of TiO₂ and seems to
536 be activated under both ultraviolet and visible light (Warheit and Donner 2015). A recent study
537 (De Matteis et al. 2016) demonstrated that, in particular using anatase, light is a dominant factor
538 to induce oxidative stress and toxic effects. Also Gerloff et al. (2009) showed the increase of
539 oxidative genotoxic effects induced by TiO₂-NPs (80%/20% anatase-rutile) in the presence of
540 interior light.

541 However, an oxidative damage (although low) was observed in the present study also in
542 darkness as reported in the study of Gurr et al. (2005) that demonstrated that in darkness TiO₂-
543 NPs can induce oxidative DNA damage. On the contrary, Karlsson et al. (2008) and Gerloff et

544 al. (2009) found that TiO₂-NPs (mixture of rutile and anatase) in darkness did not show
545 oxidative DNA damage using the Fpg-modified Comet assay.

546 Moreover, the results obtained in this study highlight that only food grade and platelet NPs
547 induced direct genotoxicity. However, while for food grade NPs the direct genotoxic effect
548 remains the same both after exposure with light and in the darkness, for the commercial food
549 grade NPs, the direct damage was higher in presence of light than in darkness. This result agree
550 with the study of Gopalan et al. (2009); they suggest that TiO₂ (anatase 40 – 70 nm range) is
551 capable of inducing higher direct genotoxic effects after simultaneous irradiation with UV,
552 respect to genotoxicity induced in darkness. The increase of direct DNA damage after exposure
553 with light attested by Gopalan et al. (2009) and detected for food grade NPs, remain to be
554 explained. A possible mechanism that may lead to this effect could be related to the potential
555 interaction of TiO₂-NPs with proteins involved in DNA repair, as demonstrated by Jugan et al.
556 (2011). Genotoxicity is not only linked to the level of DNA damage but also to the type of
557 lesions generated and their capacity to be repaired. NPs exposure in presence of light could
558 influence activity of proteins such as repair enzymes, resulting in DNA damage not repaired or
559 misrepaired (Magdolenova et al. 2014). Then, the exposure with light may have caused
560 inactivation of repair enzymes, inducing a higher direct genotoxic effect induced by food grade
561 NPs after exposure with light respect to exposure in darkness.

562 In conclusion, the results of this study showed that the cytotoxicity was overall low (WST-1
563 assay) and was influenced by the NP shape as well as by light exposure. According to the low
564 cytotoxic effect, no LDH release was detected using the LDH assay.

565 Instead, genotoxicity seemed to be influenced by the cellular-uptake and the aggregation
566 tendency of TiO₂-NPs. These two aspects are probably related to different physico-chemical
567 characteristics of NPs, such as the shape. Moreover, the presence of light enhanced the
568 genotoxic effect of some NPs primarily increasing the oxidative stress.

569 Although more studies have to be performed in order to assess the potential toxicity of
570 engineered NPs, the results of this preliminary study showed that engineered NPs did not
571 induced a high cytotoxic/genotoxic effect compared to the other commercial TiO₂-NPs, so they
572 could be used for future technological applications. The results of this study are important
573 considering that engineered NPs, due to their peculiar characteristics, could support and
574 improve TiO₂-NPs applications in different areas such as energy (i.e. use of engineered TiO₂-
575 NPs in dye-sensitized solar cells), environment (i.e. application of engineered TiO₂-NPs as
576 photocatalyst for the abatement of air and water pollutants) and health (i.e. use of engineered
577 TiO₂ –NPs for the production of nanostructured coatings of orthopedic and dental prostheses
578 exhibiting optimized interfacial properties).

579

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583

584 **Competing interests**

585 The authors declare that they have no competing interests.

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589 **References**

590 Ahlinder, L., Ekstrand – Hammarstrom, B., Geladi, P., Osterlund, L., 2013. Large uptake of
591 titania and iron oxide nanoparticle in the nucleus of lung epithelial cells as measured by raman
592 imaging and multivariate classification. *Biophys. J.*, 105, 310 – 319. doi:
593 10.1016/j.bpj.2013.06.017

594

595 Allegri, M., Bianchi, M.G., Chiu, M., Varet, G., Costa, A.L., Ortelli, S., Blosi, M., Bussolati,
596 O., Poland, C.A., Bergamaschi, E., 2016. Shape-related toxicity of titanium dioxide nanofibres.
597 PLoS ONE, 11(3), 1 – 21. doi: 10.1371/journal.pone.0151365.

598

599 Bernard, A.S., Curtiss, L.A., 2005. Prediction of TiO₂ nanoparticle phase and shape transitions
600 controlled by surface chemistry. Nano. Lett., 5: 1261 – 1266. doi: 10.1021/nl050355m.

601

602 Bhattacharya, K., Davoren, M., Boertz, J., Schins, R.P., Hoffmann, E., Dopp, E., 2009.
603 Titanium dioxide nanoparticles induce oxidative stress and DNA-adduct formation but not
604 DNA-breakage in human lung cells. Part. Fibre. Toxicol., 6, 17 – 27. doi: 10.1186/1743-8977-
605 6-17.

606

607 Bonetta, S., Bonetta, S., Motta, F., Strini, A., Carraro, E., 2013. Photocatalytic bacterial
608 inactivation by TiO₂-coated surfaces. AMB Express., 3: 59 – 66. doi: 10.1186/2191-0855-3-
609 59.

610

611 Bonetta, S., Bonetta, S., Schilirò, T., Ceretti, E., Feretti, D., Covolo, L., Vannini, S., Villarini,
612 M., Moretti, M., Verani, M., Carducci, A., Bagordo, F., De Donno, A., Bonizzoni, S., Bonetti,
613 A., Pignata, C., Carraro, E., Gelatti, U., MAPEC_LIFE Study Group, Gilli, G., Romanazzi, V.,
614 Gea, M., Festa, A., Viola, G.C.V., Zani, C., Zerbini, I., Donato, F., Monarca, S., Fatigoni, C.,
615 Levorato, S., Salvatori, T., Donzelli, G., Palomba, G., Casini, B., De Giorgi, M., Devoti, G.,
616 Grassi, T., Idolo, A., Panico, A., Serio, F., Furia, C., Colombi, P., 2019. Mutagenic and
617 genotoxic effects induced by PM0.5 of different Italian towns in human cells and bacteria: The

618 MAPEC_LIFE study. Environ. Pollut., 245:1124 – 1135. doi:
619 <https://doi.org/10.1016/j.envpol.2018.11.017>.
620
621 Bonetta, S., Giannotti, V., Bonetta, S., Gosetti, F., Oddone, M., Carraro, E., 2009. DNA
622 damage in A549 cells exposed to different extracts of PM2.5 from industrial, urban and
623 highway sites. Chemosphere, 77, 1030 – 1034. doi: 10.1016/j.chemosphere.2009.07.076.
624
625 Charles, S., Jomini, S., Fessard, V., Bigorgne-Vizade, E., Rousselle, C., Michel, C., 2018.
626 Assessment of the in vitro genotoxicity of TiO₂ nanoparticles in a regulatory context.
627 Nanotoxicology, 12(4): 357 – 374. doi: 10.1080/17435390.2018.1451567.
628
629 Chen, T., Yan, J., Li, Y., 2014. Genotoxicity of titanium dioxide nanoparticles. J. Food. Drug.
630 Anal., 22, 95 – 104. doi: 10.1016/j.jfda.2014.01.008.
631
632 Collins, A., El Yamani, N., Dusinska, M., 2017. Sensitive detection of DNA oxidation damage
633 induced by nanomaterials. Free Radic. Biol. Med., 107, 69 – 76. doi:
634 [10.1016/j.freeradbiomed.2017.02.001](https://doi.org/10.1016/j.freeradbiomed.2017.02.001).
635
636 Cowie, H., Magdolenova, Z., Saunders, M., Drlickova, M., Carreira, S. C., Kenzaoui, B. H.,
637 Gombau, L., Guadagnini, L., Lorenzo, Y., Walker, L., Fjellsbo, L. M., Huk, H., Rinna, A., Tran
638 L., Volkova, K., Boland, S., Jullierat- Jeanneret, L., Marano, F., Collins, A. R., Dusinska, M.
639 2015. Suitability of human and mammalian cells of different origin for the assessment of
640 genotoxicity of metal and polymeric engineered nanoparticles. Nanotoxicology 9: 57 – 65. doi:
641 [10.3109/17435390.2014.940407](https://doi.org/10.3109/17435390.2014.940407).
642

643 De Matteis, V., Cascione, M., Brunetti, V., Toma, C.C., Rinaldi, R., 2016. Toxicity assessment
644 of anatase and rutile titanium dioxide nanoparticles: the role of degradation in different pH
645 conditions and light exposure. *Toxicol. in Vitro*, 37, 201 – 210. doi: 10.1016/j.tiv.2016.09.010.
646

647 Di Bucchianico, S., Cappellini, F., Le Bihanic, F., Zhang, Y., Dreij, K., Karlsson, H.L., 2017.
648 Genotoxicity of TiO₂ nanoparticle assessed by mini-gel Comet assay and micronucleus scoring
649 with flow cytometry. *Mutagenesis*, 32(1), 127 – 137. doi: 10.1093/mutage/gew030.
650

651 Di Virgilio, A.L., Reigosa, M., Arnal, P.M., Fernández Lorenzo de Mele, M., 2010.
652 Comparative study of the cytotoxic and genotoxic effects of titanium oxide and aluminium
653 oxide nanoparticles in Chinese hamster ovary (CHO-K1) cells. *J. Hazard. Mater.*, 177, 711 –
654 718. doi: 10.1016/j.jhazmat.2009.12.089.
655

656 Falck, G.C.M., Lindberg, H.K., Suhonen, S., Vippola, M., Vanhala, E., Catalan, J., Savolainen,
657 K., Norppa, H., 2009. Genotoxic effects of nanosized and fine TiO₂. *Hum. Exp. Toxicol.*, 28(6
658 – 7), 339 – 352. doi: <https://doi.org/10.1177/0960327109105163>.
659

660 Fotakis, G., Timbrell, J.A., 2006. In vitro cytotoxicity assays: comparison of LDH, neutral red,
661 MTT and protein assay in hepatoma cell lines following exposure to cadmium chloride.
662 *Toxicol. Lett.*, 160: 171 – 177. doi: 10.1016/j.toxlet.2005.07.001.
663

664 Gea, M., Schilirò, T., Iacomussi, P., Degan, R., Bonetta, S., Gilli, G., 2018. Cytotoxicity and
665 genotoxicity of light emitted by incandescent, halogen, and LED bulbs on ARPE-19 and
666 BEAS-2B cell lines. *J. Toxicol. Environ. Health A*, 81 (19): 998–1014. doi:
667 10.1080/15287394.2018.1510350.

668

669 Gerloff, K., Albrecht, C., Boots, A.W., Förster, I., Schins, R.P.F., 2009. Cytotoxicity and
670 oxidative DNA damage by nanoparticles in human intestinal Caco-2 cells. *Nanotoxicology*,
671 3(4), 355–364. doi: <https://doi.org/10.3109/17435390903276933>.

672

673 Gopalan, R.C., Osman, I.F., Amani, A., De Matas, M., Anderson, D., 2009. The effect of zinc
674 oxide and titanium dioxide nanoparticles in the Comet assay with UVA photo activation of
675 human sperm and lymphocytes. *Nanotoxicology*, 3(1), 33 – 39. doi:
676 <https://doi.org/10.1080/17435390802596456>.

677

678 Guadagnini, R., Halamoda Kenzaoui, B., Walker, L., Pojana, G., Magdolenova, Z., Bilanicova,
679 D., Saunders, M., Juillerat-Jeanneret, L., Marcomini, A., Huk, A., Dusinska, M., Fjellsbo,
680 L.M., Marano, F., Boland, S., 2015. Toxicity screenings of nanomaterials: challenges due to
681 interference with assay processes and components of classic in vitro tests. *Nanotoxicology*, 9:
682 13 – 24. doi: 10.3109/17435390.2013.829590.

683

684 Gurr, J.R., Wang, A.S., Chen, C.H., Jan, K.Y., 2005. Ultrafine titanium dioxide particles in the
685 absence of photoactivation can induce oxidative damage to human bronchial epithelial cells.
686 *Toxicology*, 213, 66 – 73. doi: 10.1016/j.tox.2005.05.007.

687

688 Hamilton, R.F., Wu, N., Porter, D., Buford, M., Wolfarth, M., Holian, A., 2009. Particle length-
689 dependent titanium dioxide nanomaterials toxicity and bioactivity. *Part. Fibre. Toxicol.*, 6, 35
690 – 46. doi: <https://doi.org/10.1186/1743-8977-6-35>.

691

692 Han, X., Kuang, Q., Jin, M., Xie, Z., Zheng, L., 2009. Synthesis of titania nanosheets with a
693 high percentage of exposed (001) facets and related photocatalytic properties. *J. Am. Chem.*
694 *Soc.*, 131 (9), 3152 – 3153. doi: 10.1021/ja8092373.

695

696 Huk, A., Collins, A.R., El Yamani, N., Porredon, C., Azqueta, A., de Lapuente, J., Dusinska,
697 M., 2015. Critical factors to be considered when testing nanomaterials for genotoxicity with
698 the comet assay. *Mutagenesis*, 30, 85 – 88. doi: 10.1093/mutage/geu077.

699

700 Iannarelli, L., Giovannozzi, A.M., Morelli, F., Viscotti, F., Bigini, P., Maurino, V., Spoto, G.,
701 Martra, G., Ortel, E., Hodoroaba, V., Rossi, A.M., Diomede, L., 2016. Shape engineered TiO₂
702 nanoparticles in *Caenorhabditis elegans*: a Raman imaging based approach to assist tissue-
703 specific toxicological studies. *R.S.C. Adv.*, 6, 70501 – 70509. doi:
704 <https://doi.org/10.1039/c6ra09686g>.

705

706 IARC, 2010. Carbon Black, Titanium Dioxide, and Talc. *IARC Monogr. Eval. Carcinog. Risks*
707 *Hum.*, 93, 1 – 413.

708

709 International Standards Organisation (ISO), 2015. *Nanotechnologies -- Vocabulary -- Part 2:*
710 *Nano-objects*. ISO 80004-2, Geneva, Switzerland.

711

712 Johnston, H.J., Hutchison, G.R., Christensen, F.M., Peters, S., Hankin, S., Stone, V., 2009.
713 Identification of the mechanisms that drive the toxicity of TiO₂ particulates: the contribution
714 of physicochemical characteristics. *Part. Fibre. Toxicol.*, 6, 33. doi:
715 <https://doi.org/10.1186/1743-8977-6-33>.

716

717 Jovanovic, B., 2015. Critical review of public health regulations of titanium dioxide, a human
718 food additive. *Integr. Environ. Assess. Manag.*, 11, 10 – 20. doi: 10.1002/ieam.1571.
719

720 Jugan, M.L., Barillet, S., Simon-Deckers, A., Herlin-Boime, N., Sauvaigo, S., Douki, T.,
721 Carriere, M., 2011. Titanium dioxide nanoparticles exhibit genotoxicity and impair DNA repair
722 activity in A549 cells. *Nanotoxicology*, 6(5), 501 – 513. doi: 10.3109/17435390.2011.587903.
723

724 Kann, B., Offerhaus, H.L., Windbergs, M., Otto, C., 2015. Raman microscopy for cellular
725 investigations – from single cell imaging to drug carrier uptake visualization. *Adv. Drug. Deliv.*
726 *Rev.*, 89, 71 – 90. doi: 10.1016/j.addr.2015.02.006.
727

728 Karlsson, H.L., Cronholm, P., Gustafsson, J., Möller, L., 2008. Copper oxide nanoparticles are
729 highly toxic: a comparison between metal oxide nanoparticles and carbon nanotubes. *Chem.*
730 *Res. Toxicol.*, 21, 1726 – 1732. doi: 10.1021/tx800064j.
731

732 Karlsson, H.L., Di Bucchianico, S., Collins, A., Dusinska, M., 2015. Can the Comet Assay be
733 used reliably to detect nanoparticle-induced genotoxicity? *Environ. Mol. Mutagen.*, 56, 82 –
734 96. doi: 10.1002/em.21933.
735

736 Karlsson, H.L., 2010. The Comet assay in nanotoxicology research. *Anal. Bioanal. Chem.*, 398,
737 651-666. doi: 10.1007/s00216-010-3977-0.
738

739 Lavric, V., Isopescu, R., Maurino, V., Pellegrino, F., Pellutiè, L., Ortel, E., Hodoroaba, A.,
740 2017. New model for nano-TiO₂ crystal birth and growth in hydrothermal treatment using an

741 oriented attachment approach. *Cryst. Growth Des.*, 17 (11), 5640–5651. doi:
742 10.1021/acs.cgd.7b00302

743

744 Magdolenova, Z., Bilanicova, D., Pojana, G., Fjellsbo, L.M., Hudecova, A., Hasplova, K.,
745 Marcomini, A., Dusinska, M., 2012b. Impact of agglomeration and different dispersion of
746 titanium dioxide nanoparticles on the human related in vitro cytotoxicity and genotoxicity. *J.*
747 *Environ. Monit.*, 14, 455 – 464. doi: 10.1039/c2em10746e.

748

749 Magdolenova, Z., Collins, A., Kumar, A., Dhawan, A., Stone, V., Dusinska, M., 2014.
750 Mechanisms of genotoxicity. A review of in vitro and in vivo studies with engineered
751 nanoparticles. *Nanotoxicology*, 8(3), 233 – 278. doi: 10.3109/17435390.2013.773464.

752

753 Magdolenova, Z., Lorenzo, Y., Collins, A., Dusinska, M., 2012a. Can standard genotoxicity
754 tests be applied to nanoparticles? *J. Toxicol. Environ. Health A*, 75, 13 – 15. doi:
755 <https://doi.org/10.1080/15287394.2012.690326>.

756

757 Mohra, U., Ernst, H., Roller, M., Pott, F., 2006. Pulmonary tumor types induced in Wistar rats
758 of the so-called 19-dust study. *Exp. Toxicol. Pathol.*, 58(1), 13 – 20. doi:
759 10.1016/j.etp.2006.06.001

760

761 Moller, P., Hemmingsen, J. G., Jensen, D. M., Danielsen, P. H., Karottki, D. G., Jantzen, K.,
762 Roursgaard, M., Cao, Y., Kermanizadeh, A., Klingberg, H., Christophersen, D.V., Hersoug,
763 L., Loft, S., 2015a. Applications of the comet assay in particle toxicology: air pollution and
764 engineered nanomaterials exposure. *Mutagenesis* 30, 67 – 83. doi: 10.1093/mutage/geu035.

765

766 Moller, P., Jensen, D. M., Christophersen, D. V., Kermanizadeh, A., Jacobsen, N. R.,
767 Hemmingsen, J. G., Danielsen, P. H., Karotki, D. G., Roursgaard, M., Cao, Y., Jantzen, K.,
768 Klinberg, H., Hersoug, L., Loft, S., 2015b. Measurement of oxidative DNA damage to DNA
769 in nanomaterial exposed cells and animals, *Environ. Mol. Mutagen.*, 56, 97 – 110. doi:
770 10.1002/em.21899.
771
772 National Institute for Occupational Safety and Health (NIOSH), 2011. Occupational Exposure
773 to Titanium Dioxide. In *Current Intelligence Bulletin 63*. Cincinnati: National Institute for
774 Occupational Safety and Health.
775
776 Numano, T., Xu J., Futakuchi, M., Fukamachi, K., Alexander, D. B., Furukawa, F., Kanno, J.,
777 Hirose, A. Tsuda, H., Suzuim, M., 2014. Comparative study of toxic effects of anatase and
778 rutile type nanosized titanium dioxide particles in vivo and in vitro. *Asian Pac. J. Cancer Prev.*,
779 15 (2), 929 – 935. doi: <http://dx.doi.org/10.7314/APJCP.2014.15.2.92>
780
781 Ohtani, B., Prieto-Mahaney, O.O., Li, D., Abe, R., 2010. What is Degussa (Evonik) P25?
782 Crystalline composition analysis, reconstruction from isolated pure particles and photocatalytic
783 activity test. *J. Photochem. Photobiol. A*, 216, 179 – 182. doi:
784 <http://dx.doi.org/10.1016/j.jphotochem.2010.07.024>.
785
786 Park, E.J., Lee, G., Shim, H., Kim, J., Cho, M., Kim, D., 2013. Comparison of toxicity of
787 different nanorod-type TiO₂ polymorphs in vivo and in vitro. *J. Appl. Toxicol.*, 34, 357 – 366.
788 doi: 10.1002/jat.2932.
789

790 Park, E.J., Yi, J., Chung, Y.H., Ryu, D.Y., Choi, J., Park, K., 2008. Oxidative stress and
791 apoptosis induced by titanium dioxide nanoparticles in cultured BEAS-2B cells. *Toxicol. Lett.*,
792 180(3), 222 – 229. doi: 10.1016/j.toxlet.2008.06.869.

793

794 Petkovic, J., Zegura, B., Stevanovic, M., Drnovsek, N., Uskokovic, D., Novak, S., Filipic, M.,
795 2011. DNA damage and alterations in expression of DNA damage responsive genes induced
796 by TiO₂ nanoparticles in human hepatoma HepG2 cells. *Nanotoxicology*, 5, 341 – 353. doi:
797 10.3109/17435390.2010.507316.

798

799 Popescu, T., Lupu, A. R., Raditoiu, V., Purcar, V., Teodorescu, V. S., 2015. On the
800 photocatalytic reduction of MTT tetrazolium salt on the surface of TiO₂ nanoparticles:
801 formazan production kinetics and mechanism, *J. Colloid Interface Sci.*, 457: 108 – 120. doi:
802 10.1016/j.jcis.2015.07.005.

803

804 Prasad, R.Y., Wallace, K., Daniel, K.M., Tennant, A.H., Zucker, R.M., Strickland, J., Dreher,
805 K., Klingerman, A.,D., Blackman, C.F., De Marini, D.M., 2013. Effect of treatment media on
806 the agglomeration of titanium dioxide nanoparticles: impact on genotoxicity, cellular
807 interaction, and cell cycle. *A.C.S. Nano*, 7, 1929 – 1942. doi: 10.1021/nn302280n.

808

809 Proquin, H., Rodriguez-Ibarra, C., Moonen, C.G.J., Urrutia Ortega, I.M., Briede, J.J., de Kok,
810 T.M., van Loveren, H., Chirino, Y.I., 2017. Titanium dioxide food additive (E171) induces
811 ROS formation and genotoxicity: contribution of micro and nano-sized fractions. *Mutagenesis*,
812 32, 139 – 149. doi: 10.1093/mutage/gew051.

813

814 Reeves, J. F., Davies, S. J., Dodd, N. J. F., Jha, A. N., 2008. Hydroxyl radicals ($\bullet\text{OH}$) are
815 associated with titanium dioxide (TiO_2) nanoparticle-induced cytotoxicity and oxidative DNA
816 damage in fish cells. *Mutat Res*, 640 (1-2), 113 – 22. doi: 10.1016/j.mrfmmm.2007.12.010.
817 Epub 2007 Dec 31.

818

819 Sayes, C.M., Wahi, R., Kurian, P.A., Liu, Y., West, J.L., Ausman, K.D., Warheit, D.B., Colvin,
820 V.L., 2006. Correlating nanoscale titania structure with toxicity: a cytotoxicity and
821 inflammatory response study with human dermal fibroblast and human lung epithelial cells.
822 *Toxicol. Sci.*, 92,174 – 185. doi: 10.1093/toxsci/kfj197.

823

824 Schilirò, T., Bonetta, S., Alessandria, L., Gianotti, V., Carraro, E., Gilli, G., 2015. PM10 in a
825 background urban site: chemical characteristics and biological effects. *Environ. Toxicol.*
826 *Pharmacol.*, 39(2), 833 – 844. doi: <https://doi.org/10.1016/j.etap.2015.02.008>.

827

828 Sha, B., Gao, W., Cui, W., Wang, L., Xu, F., 2015. The potential health challenges of TiO_2
829 nanomaterials. *J. Appl. Toxicol.*, 35, 1086 – 1101. doi: 10.1002/jat.3193.

830

831 Shi, H., Magaye, R., Castranova, V., Zhao, J., 2013. Titanium dioxide nanoparticles: a review
832 of current toxicological data. *Part. Fibre. Toxicol.*, 10, 15. doi: 10.1186/1743-8977-10-15.

833

834 Shi, Y., Wang, F., He, J., Yadav, S., Wang, H., 2010. Titanium dioxide nanoparticles cause
835 apoptosis in BEAS-2B cells through caspase 8/t-Bid-independent mitochondrial pathway.
836 *Toxicol. Lett.*, 196, 21 – 27. doi: 10.1016/j.toxlet.2010.03.014.

837

838 Tice, R.R., Agurell, E., Anderson, D., Burlison, B., Hartmann, A., Kobayashi, H., Miyamae,
839 Y., Rojas, E., Ryu, J.C., Sasaki, Y.F., 2000. Single cell gel/Comet assay: guidelines for in
840 vitro and in vivo genetic toxicology testing. *Environ. Mol. Mutagen.*, 35, 206 – 221. Doi:
841 [https://doi.org/10.1002/\(SICI\)1098-2280\(2000\)35:3<206::AID-EM8>3.0.CO;2-J](https://doi.org/10.1002/(SICI)1098-2280(2000)35:3<206::AID-EM8>3.0.CO;2-J).
842

843 Tomankova, K., Horakova, J., Harvanova, M., Malina, L., Soukupova, J., Hradilova, S.,
844 Kejllova, K., Malohlava, J., Licman, L., Dvorakova, M., Jirova, D., Kolarova, H., 2015.
845 Cytotoxicity, cell uptake and microscopic analysis of titanium dioxide and silver nanoparticles
846 in vitro. *Food Chem. Toxicol.*, 82, 106 – 115. doi: 10.1016/j.fct.2015.03.027.
847

848 Tompsett, G.A., Bowmaker, G.A., Cooney, R.P., Metson, J.B., Rodgers, K.A., Seakins, J.M.,
849 1995. The Raman spectrum of brookite, TiO₂ (Pbc, Z = 8). *J. Raman Spectrosc.*, 26, 57 – 62.
850 doi: <https://doi.org/10.1002/jrs.1250260110>.
851

852 Uboldi, C., Urban, P., Gilliland, D., Bajak, E., Valsami-Jones, E., Ponti, J., Rossi, F., 2016.
853 Role of the crystalline form of titanium dioxide nanoparticles: rutile, and not anatase, induces
854 toxic effects in Balb/3T3 mouse fibroblasts. *Toxicol. in Vitro*, 31: 137 – 145. doi:
855 10.1016/j.tiv.2015.11.005.
856

857 Ursini, C.L., Cavallo, D., Fresegna, A.M., Ciervo, A., Maiello, R., Tassone, P., Buresti, G.,
858 Casciardi, S., Iavicoli, S., 2014. Evaluation of cytotoxic, genotoxic and inflammatory response
859 in human alveolar and bronchial epithelial cells exposed to titanium dioxide nanoparticles. *J.*
860 *Appl. Toxicol.*, 34, 1209 – 1219. doi: 10.1002/jat.3038.
861

862 Valant, J., Iavicoli, I., Drobne, D., 2012. The importance of a validated standard methodology
863 to define in vitro toxicity of nano- TiO₂. *Protoplasma*, 9, 493 – 502. doi: 10.1007/s00709-011-
864 0320-3.

865

866 Vevers, W. F., Jha, A. N., 2008. Genotoxic and cytotoxic potential of titanium dioxide (TiO₂)
867 nanoparticles on fish cells in vitro. *Ecotoxicology*, 17(5), 410 – 420. doi: 10.1007/s10646-008-
868 0226-9. Epub 2008 May 20.

869

870 Wang, W., Gu, B., Liang, L., Hamilton, W.A., Wesolowski, D.J., 2004. Synthesis of rutile (α -
871 TiO₂) nanocrystals with controlled size and shape by low-temperature hydrolysis: effects of
872 solvent composition. *J. Phys. Chem. B*, 108, 14789 – 14792. doi: 10.1021/jp0470952.

873

874 Warheit, D.B., Donner, E.M., 2015. Risk assessment strategies for nanoscale and fine-sized
875 titanium dioxide particles: recognizing hazard and exposure issues. *Food Chem. Toxicol.*, 85,
876 138 – 147. doi: 10.1016/j.fct.2015.07.001.

877

878 Weir, A., Westerhoff, P., Fabricius, L., Von Goetz, N., 2012. Titanium dioxide nanoparticles
879 in food and personal care products. *Environ. Sci. Technol.*, 46(4), 2242 – 2250. doi:
880 10.1021/es204168d.

881

882 Weyermann, J., Lochmann, D., Zimmer, A., 2005. A practical note on the use of cytotoxicity
883 assays. *Int. J. Pharm.*, 288: 369 – 376. doi: 10.1016/j.ijpharm.2004.09.018.

884

885 Wilhelmi, V., Fischer, U., Van Berlo, D., Schulze-Osthoff, K., Schins, R.P.F., Albrecht, C.,
886 2012. Evaluation of apoptosis induced by nanoparticles and fine particles in RAW 264.7

887 macrophages: facts and artefacts. *Toxicol. in Vitro*, 26: 323 – 334. doi:
888 10.1016/j.tiv.2011.12.006.

889

890 Xu, J., Futakuchi, M., Iigo, M., Fukamachi, K., Alexander, D.B., Shimizu, H., Sakai, Y.,
891 Tamano, S., Furukawa, F., Uchino, T., Tokunaga, H., Nishimura, T., Hirose, A., Kanno, J.,
892 Tsuda, H., 2010. Involvement of macrophage inflammatory protein 1a (MIP1a) in promotion
893 of rat lung and mammary carcinogenic activity of nanoscale titanium dioxide particles
894 administered by intra-pulmonary spraying. *Carcinogenesis*, 31, 5927 – 5935. doi:
895 10.1093/carcin/bgq029.

896

897 Xue, C., Wu, J., Lan, F., Liu, W., Yang, X., Zeng, F., Xu, H., 2010. Nano titanium dioxide
898 induces the generation of ROS and potential damage in HaCaT cells under UVA irradiation. *J.*
899 *Nanosci. Nanotechnol.*, 10, 8500 – 8507. doi: 10.1166/jnn.2010.2682.

900

901 Zhang, J., Wang, J., Zhao, Z., Yu, T., Feng, J., Yuan, Y., Tang, Z., Liu, Y., Li, Z., Zou, Z.,
902 2012. Reconstruction of the (001) surface of TiO₂ nanosheets induced by the fluorine-
903 surfactant removal process under UV-irradiation for dye-sensitized solar cells. *Phys. Chem.*
904 *Chem. Phys.*, 14 (14), 4763–4769. doi: 10.1039/c2cp24039d.

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912 **Table 1**

Sample	Particle size (nm)	D_h DMSO (nm)	D_h RPMI Base (nm)	D_h RPMI Complete (nm)	Crystalline Phase
P25	20 ± 5 quasi-spherical	107 ± 31	722 ± 246	121 ± 37	Anatase:Rutile (5:1)
Food grade	150 ± 50 undefined shape	184 ± 61	278 ± 54	184 ± 55	Anatase
Bipyramids	50 ± 9* (aspect ratio 3:2)	66 ± 20	259 ± 46	88 ± 24	Anatase
Rods	108 ± 47* (aspect ratio 1:5)	36 ± 12	1500 ± 471	39 ± 17	Anatase
Platelets	75 ± 25* (aspect ratio 8:1)	233 ± 70	281 ± 83	250 ± 82	Anatase

913

914 Table 1. Physico-chemical properties of the TiO₂-NPs samples. Data are presented as mean ±
915 standard deviation of 500 NPs for the particle size and 5 measurements for the hydrodynamic
916 diameter (D_h) of each sample. *The particle size was calculated along the major axis of the
917 NPs.

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926 **Figure captions**

927 Figure 1. SEM In-lens micrographs: (a) P25, (b) food grade, (c) bipyramids; (e) platelet NPs.
928 T-SEM micrograph of rods (d). DLS analyses, normalized by volume distribution (f-j): (f) P25,
929 (g) food grade, (h) bipyramids, (i) rods and (j) platelet NPs, suspensions in DMSO 1% (black
930 line), RPMI base (red line) and RPMI complete (blue line).

931 Figure 2. Cytotoxicity measured with WST-1 (a,b) and LDH release (c,d) of BEAS-2B cells
932 exposed to different concentrations (5–80 $\mu\text{g/ml}$) of commercial and engineered NPs. Control
933 level is at 100%. Data represent effects detected after exposure with laboratory light (a,c) and
934 in darkness (b,d). Data represent the mean % of the different wells, bars represent standard
935 deviation. $\ast = p < 0.05$ $\S = p < 0.001$; vs control cells (C-) according to ANOVA test, followed by
936 Dunnett's test.

937 Figure 3. Effect of BEAS-2B cells exposure to commercial and engineered NPs. AC (-Fpg) =
938 alkaline Comet assay (direct DNA damage); MC (+Fpg) = Fpg-modified Comet assay (direct
939 + indirect DNA damage). Ox = oxidative DNA damage (tail intensity (%) in enzyme-treated
940 cells - tail intensity (%) in enzyme-untreated cells). Exposure with laboratory light (a-e): (a)
941 P25, (b) food grade, (c) bipyramids, (d) rods, (e) platelet NPs; exposure in darkness (f-h): (f)
942 P25, (g) food grade, (h) platelet NPs. Data represent the mean % of tail intensity; bars represent
943 standard deviation of two independent experiments for each experimental condition. $\ast = p < 0.05$
944 $\S = p < 0.001$ DNA damage vs control cells (C-). a= $p < 0.05$ b= $p < 0.001$ oxidative DNA damage
945 vs control cells (C-). According to ANOVA test, followed by Dunnett's test.

946 Figure 4. 3D confocal micro-Raman imaging of BEAS-2B cells after exposure to commercial
947 and engineered NPs. Top views (optical and 3D Raman) and 3D Raman sections are shown
948 from the left to the right: (a) P25, (b) food grade, (c) platelet NPs, (d) bipyramids, (e) rods. 3D
949 chemical images are built by superimposing the different maps of each cell at their
950 corresponding focal planes and they are presented using a color meshwork i.e. blue for cell

951 tissues (methylene blue $\nu(\text{C-C})$ ring at 1600 cm^{-1}) and red for TiO_2 agglomerates (Eg band at
952 144 cm^{-1} of the anatase TiO_2).