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Abstract: In this review, we analyzed the history and the past and present trends in photocatalysis research, trying to outline possible scenarios for the future in the medium term. The in-depth analysis of the literature reported here—from a mere bibliometric point of view—allowed us to divide the history of photocatalysis into four different periods characterized by different maturity of the topic and different bibliometric features. The turn of the 20th century saw an explosion in scientific production, which is still continuing now and has reached more than 15,000 papers published each year. Research interest is still growing significantly, and the analysis of different keywords suggests that such growth is substantial and not merely due to "publish or perish" behavior. The analysis of the most-investigated topics in the field of photocatalysis highlighted that, during its history, the focus of the research moved from inorganic oxides to carbon and hybrid materials. Concomitantly, the investigation of the "geography" of photocatalysis allowed us to underline its evolution over the years, with the repositioning of its center of mass from the Atlantic Ocean (USA and Europe) to Asia (China and India). Photocatalysis is active as never before but still awaiting major breakthroughs, which would allow a much broader technological and commercial output. Without such breakthroughs in this decade, the growth in scientific interest could level out or even decrease.

Keywords: photocatalysis; environmental application; bibliometric analysis; future perspectives; photochemical processes; photochemistry

1. Introduction

The solar energy reaching the Earth per year is 3,400,000 EJ, and the corresponding total solar flux is equal to 1.08×10^8 GW. At the distance of the Earth from the Sun, the solar constant is 1367 W m⁻², and as a consequence of the absorption and scattering of the radiation from the atmosphere, the year/day-averaged energy density at the Earth's surface falls down to 170 W m⁻² [1]. The impressive amount of energy reaching our planet as electromagnetic energy from the Sun is the real fuel of the Earth, being at the basis of all of the most important cycles (e.g., hydrologic cycle) and processes allowing the existence of highly evolved forms of life on the Earth's surface. The overall primary energy consumption of the human population is estimated roughly as 6000 GW, between 7000 and 8000 times lower than the energy power reaching the Earth. The conversion of the light reaching the surface into chemicals through the photosynthetic process is at the basis of the development of superior organisms and has contributed to the overall definition of the atmospheric composition.

Due to its great potentialities, the conversion of the sunlight into forms of energy easier to be used and stored (i.e., electricity and/or chemical fuels) has been a *fixed idea* of the scientific community since the beginning. The first systematic and scientific investigation of the chemical processes activated by sunlight was dated 1777 with the studies by the Swedish chemist C. W. Scheel who described the violet-light-activated reduction of silver chloride to metal silver. The first principle of photochemistry was defined by the pioneering



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works by T. von Grotthuss (1817) and J. W. Draper (1841), who established that *only the light absorbed is effective in producing photochemical change* (the so-called Grotthuss–Draper principle). The main limits for a systematic investigation of the photochemical process in the 19th century were the unavailability of both adequate light sources and analytical techniques; as a matter of fact, the only light source at the disposal of the first scientists that systematically faced the potentialities of photochemistry was the light of the Sun (e.g., G. Ciamician in Bologna [2] and G. Lemoin in Paris [3]).

The idea to carry out a photochemical process in the presence of a compound/material able to catalyze the process itself was proposed at the beginning of the 20th century by Plotnikow in 1910 [4] and Landau in 1912–1913 [5,6]. At the same time, in the 1920s, the first approaches to mimic natural photosynthesis, obtaining both organic compounds (formaldehyde and carbohydrates [7]) by the photochemical reaction between carbon dioxide and water and organic nitrogen compounds by the photoactivated reaction between inorganic nitrogen (nitrate [8] or ammonia [9]) and carbon dioxides, were proposed.

Only from the 1940s were the photocatalytic properties of oxides discovered and investigated, often with the practical aim to increase the life of paints, papers and textiles containing titanium dioxide or other white semiconductors as pigments. In this context, the oxidative ability of certain irradiated semiconductors (e.g., ZnO [10–12], Sb₂O₃ [13,14] and TiO₂ [15,16]) has been investigated, and the potential formation of reactive species such as H₂O₂ has been observed [17–19]. An interesting and didactic review regarding the photocatalytic activity of oxides was published by M. C. Markham in 1955 [20].

Especially in the first part of the 20th century, the term "photocatalytic reaction" was often used with a more general meaning of "photochemical reaction"; as an example, in the seminal works by Baly et al. cited above [7–9], the chemical reactions investigated were photochemical processes but not photocatalytic ones, because no catalyst was added in the reaction medium to accelerate the kinetics of the process. In fact, in agreement with the recognized IUPAC definition [21], in a photocatalytic reaction there is always "the presence of a substance—the photocatalyst—that absorbs light and is involved in the chemical transformation of the reaction partners"; furthermore, the photocatalyst must regenerate its chemical composition and properties after each catalytic cycle.

Only from the late 1970s did the photocatalytic processes start to be investigated as effective tools to solve environmental problems. In particular, this was conducted in the field of energetics with the seminal works by Fujishima and Honda on the photo-assisted electro-splitting of water on titanium dioxide electrodes [22] and by Inoue et al. on the photocatalyzed reduction of carbon dioxide on irradiated semiconductors [23] and as an effective tool for environmental remediation especially for (i) the removal of recalcitrant pollutants from water [24–26], (ii) the degradation of air pollutants [27] and (iii) the inactivation of pathogens [28].

The history of photocatalysis can be divided in different ways; as an example, the quite effective division proposed by Serpone and Emeline [29] as a function of the type of materials used to carry out the photocatalytic processes is often cited. In particular, three different generations of photocatalysts have been proposed. The first is composed by pristine inorganic semiconductors (in the beginning mainly ZnO, WO_3 and TiO_2 and then mainly TiO_2 only), of which the photocatalytic properties have been investigated in depth, often with pioneering experimental approaches, with the main aim to give insights into the mechanism of production and transfer of the charge carriers generated inside the material and into the nature of the reactive species operative in the investigated photocatalyzed processes. The second generation of photocatalysts is mainly composed by doped semiconductors synthesized to push the onset of absorption toward longer wavelengths introducing intra-bandgap states, the nature and energetic positioning of which have been deeply investigated, even though the first report of their discovery remained unnoticed for a long time, [30] while a successive report received more recognition (and 16 times more citations!), being the most cited photocatalysis paper of its decade (vide infra) [31]. The doping strategies proposed were based on the introduction of non-metal or metal centers inside the crystallographic structure of the pristine material with the aim to generate color centers (localized states located between the valence and conduction bands [32–34] and/or oxygen vacancies with the formation of centers where the metal cations have different oxidation states, e.g., Ti^{+3} centers in a TiO_2 structure [35,36]). The third and last generation proposed is based on the production of hybrid heterostructures formed by two or more inorganic semiconductors [37–39]. In this case, the dynamics of the charge carriers became extremely complex, especially if the number of the materials composing the hybrids is higher than two. In some cases, these photocatalysts have shown both high efficiency and a better ability to harvest solar irradiation with respect to the pristine and doped semiconductors.

After the synthesis of the new synthetic carbon allotropes (fullerenes [40], carbon nanotubes [41] and graphene [42]) and of their almost infinite modifications, a new class of photocatalysts has been proposed and commonly denoted as metal-free photocatalysts. These materials, either used as is or coupled with inorganic heterostructures with different complexity, have been proposed in all the application fields of the photocatalysis. Recently, some of the authors of this article have questioned if these new materials can be or cannot be considered as the fourth generation of photocatalysts [43]. Overlooking this purely semantic point, it is essential to note that the efforts of the community operating in this field are actually mostly directed at the development and testing of these materials that represent, without any doubts, the most fascinating (or the most in-fashion) topic in the field of photocatalysis.

The aim of the present article is not to give an exhaustive evaluation of the entire literature regarding photocatalysis, since for this purpose some excellent reviews have already been published both recently [44,45] and in the past [46,47]. A fully complete analysis of the actual literature, as underlined on the basis of a detailed bibliometric analysis reported in the following paragraphs, if possible, up to the initial 2000s, is impossible nowadays, due to the impressive number of papers published every year. Using as a keyword "photocatal*", an exhaustive database such as Scopus returns more than 15,000 entries per year, nowadays (see Figure 1). So, to give insights into "photocatalysis" as a scientific topic and to obtain general considerations regarding its past, present and future, we carried out an investigation starting from an in-depth analysis of the most important works published in the first period of the development of this topic and moved gradually to a less detailed evaluation of the literature based on an overall vision of the entire production by using general bibliometric descriptors. With this approach, we propose to divide the entire history of photocatalysis into four blocks, not on the basis of the nature of the photocatalysts investigated, as proposed by Serpone and Emeline and described here above [29], but on the basis of the degree of development of the scientific thinking and of the research regarding the photocatalyzed phenomena.

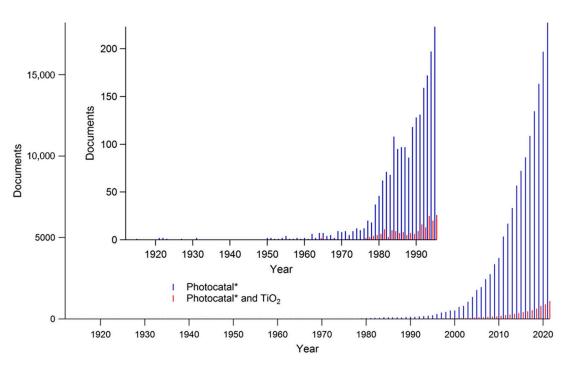


Figure 1. Number of published documents (database: Scopus, 20 September 2022) per year indexed as "*photocatal**" and "*photocatal* * *and TiO*₂". Insert: zoom from 1912 to 1995.

2. Bibliometric Analysis of Photocatalysis as a Scientific Topic: Temporal and Geographic Analysis

To give insight into the field of photocatalysis as a scientific macro-topic, a first tool is the analysis of the number of works published on this theme since the beginning.

The Scopus database (September 2022) has been used as the source of the desired information. Figure 1 shows the number of documents published since 1912 and indexed with "photocatal*"; moreover, what is also reported is (i) the number of reports dedicated to the photocatalytic properties of TiO₂ ("photocatal* AND TiO₂") and (ii) a zoom regarding the data from 1912 to 1995. From 1912 to 1973, the number of articles published per year on photocatalysis was less than 10, from 1974 to 1988 less than 100, from 1989 to 2002 less than 1000 and, finally, from 2003 and 2016 less than 10,000. Since 1974, it an increase in the number of published articles of an order of magnitude every 13/14 years has been observed. From 2016, the number of articles is over 10,000, with a monster number of 18,196 documents in 2021. At the same time, the relative weight of the works on the photocatalytic properties of TiO₂ has changed, being before 2000 almost 10–12% of the overall documents and in the range 4–8% from 2000 to nowadays. This is a clear index of the shift of the attention towards different materials with respect to the semiconductors of the "first generation", even if titanium dioxide continues to be investigated and used as a functional material also in hybrid structures [48,49].

These numbers must be also considered in the light of the overall explosion (with the beginning of the 21st century) of the scientific productivity in states (i.e., China and India) that were marginal before 2000.

From the overlap of the main topics investigated and of the number of published documents, we propose here to divide the history of photocatalysis into four different periods: (i) the *Pioneering years* (until 1980) with less than 50 publications per year mainly dedicated to the fundamentals of photocatalytic phenomena, (ii) the *Maturity* (1981–1995) with less than 300 articles per year related to the better understanding of the operational mechanisms and to the investigation of the potential applications (especially in the field of the removal of pollutants in water), (iii) the *Explosion* (1996–2010) with less than 5000 articles/year dedicated to the most different topics with a peculiar focus on the increment of the quantum yield of the process and the extension of the exploitable spectral windows,

(iv) the *Inflation* (2011–) with an almost complete attention to innovative materials and in particular to hybrid structures with the aim to exploit the properties of the junctions among different inorganic/organic materials, especially for energy applications (H₂ production, CO_2 reduction and artificial photosynthesis). While the knowledge of the most relevant scientific literature on the topic required reading three or four papers per month in the 1980s and early 1990s, at the turn of the century it became a much more difficult task, which eventually turned into an impossible job in the latest 20 years.

It is interesting to observe the change in the relative distribution of the affiliations (in terms of country of origin) of the authors for each period. For this reason, the works indexed as "photocatal*" have been divided into the different temporal periods, and the affiliation countries have been compared. Figure 2 shows the relative weight of each country from the *Pioneering years* to nowadays. The first four countries accounted always for more than 50% of all the publications, but the relative positioning changed with the years. The USA moved from the first contributing country until 1980 with more than 36% to the actual third position, with an average weight equal to 8.8%. Japan had a peak of weight in the *Maturity* period (21.5%) that is nowadays reduced to 3.8%. The relative weight of the European countries, which accounted for roughly 20% of the overall publications until 1995, decreased to roughly 12% in the *Explosion* period and to \approx 8% in the *Inflation* period. Conversely, the contribution of China and India that accounted for less than 5% until 1995 is now higher than 45%.

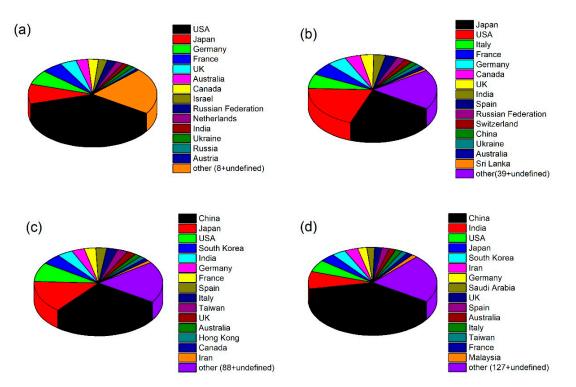


Figure 2. Relative weight of the affiliation country for the documents indexed as "photocatal*" divided into the four periods proposed: (**a**) Pioneering years (until 1980), (**b**) Maturity (1981–1995), (**c**) Explosion (1996–2010) and (**d**) Inflation (2011–2021) (database: Scopus, 20 September 2022).

The papers published in the field of photocatalysis ("photocatalysis" in the title, abstract or keywords) are still increasing every year, and this increase is still exponential, as witnessed by the linear trend observed in the logarithmic plot of Figure 3. The growth was even faster between 1990 and 2000. Searching the database with slightly different keywords (e.g., "photocatalysis and semiconductor" or "semiconductor photocatalysis") leads to very similar trends, although with a different number of published documents.

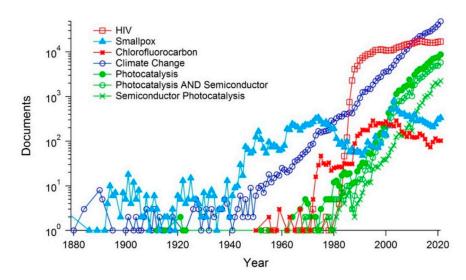


Figure 3. Number of published documents per year as functions of the publication year for several keywords.

Nonetheless, the scientific literature is inflating in many different fields; therefore, we asked ourselves if the growth in the field of photocatalysis is really significant and not just reflecting different habits in the researchers' behavior towards publishing. The best figure to make this assessment is the total number of published documents in the whole scientific literature. Unfortunately, this number cannot be obtained with sufficient accuracy; therefore, we compared photocatalysis with other keywords. Keywords such as "smallpox" and "chlorofluorocarbon" represent fields of research which should have reached their maximum interest in the past, as the last documented smallpox case dates 1979, and the Montreal protocol forbids the further production of chlorofluorocarbons. A significant increase in the published documents in the latest two decades on these two topics would only reflect different habits in the behavior of the researchers, who are becoming more willing to publish and not because of recent breakthroughs in the field. In fact, for "smallpox" and "chlorofluorocarbon" we can observe a decrease in the published documents in recent years.

A keyword such as "HIV" represents a topic which is still researched, as the quest for cures and vaccines is still very active; nevertheless, the insurgence of the disease dates back to the early 1980s. In this case, after a fast growth in the late 1980s, we can observe two phases of linear growth in the early 1990s and 2000s interrupted by two periods of almost constant scientific production in the late 1990s and 2010s.

With "climate change", we wanted to consider a very popular research field. The interest in the topic is not very recent, as since 1954 at least 10 documents have been published every year. Since then, the publications increased exponentially with an almost constant rate. Since 2000, the trends of "climate change" and "photocatalysis" run almost parallel with almost exponential growth. We can therefore conclude that the scientific interest in the field of photocatalysis has been increasing at a steady and impressive rate for 35 years, and researchers are continuing to join the field and give their contributions, positioning photocatalysis among the most important topics in the field of chemistry and material science, especially for its remarkable potentialities for the resolution of worldwide environmental problems.

This comparative analysis between photocatalysis and other selected topics can also be carried out from a geographical point of view; for this reason, we elaborated the data reported in Figure 3, splitting the overall number of published documents as a function of the affiliation country of the authors. Concerning the geography of photocatalysis, which has already been discussed above considering its time evolution, we can highlight that Asian institutions are the most productive (Figure 4e). Among the top-five countries, four are Asian, namely China, India, Japan and South Korea; furthermore, Iran is the eighth most productive country. This trend is similar to that found for an emerging chemistry research project, MXene, for which 8 of the top-12 publishing countries are Asian (Figure 4g). Comparing photocatalysis with the other keywords considered before, the role of North American and European countries becomes predominant. On the one hand, this can be attributed to the recent involvement of Asian countries compared with North America and Europe; on the other hand, the contribution of Western countries seems greater in medicine-related topics compared with chemistry and material science. Among European countries, the United Kingdom (UK) is usually the leading country in almost all the examples here considered, with the notable exception of photocatalysis. In chemistry-related fields, Germany immediately follows the UK, while Italy and France look more competitive in medical research. Photocatalysis appears to be the exception to this rule, with limited contribution from the UK (and USA) and with Germany and Spain as the main European contributors.

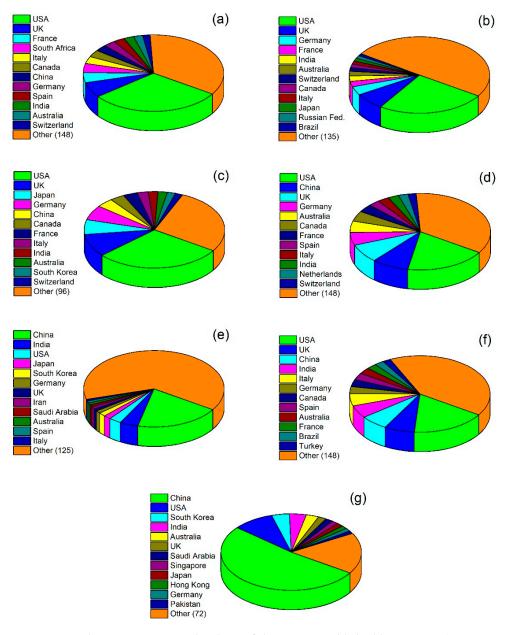


Figure 4. Pie charts representing the share of documents published by country (i.e., researchers affiliated to institutions of each country) by keywords: (a) HIV; (b) smallpox, (c) chlorofluorocarbon; (d) climate change; (e) photocatalysis; (f) COVID-19; (g) MXene.

3. Evolution of the Main Topics in the Field of Photocatalysis

Coming back to the previously defined four periods of photocatalysis, the number of the total publications in each group will be dramatically different, due to the exponential growth of the published documents. Keeping in mind that a comprehensive synthesis of the whole literature has become an impossible task in the last 20 years, we tried to give a critical assessment of the past activities and possibly outline future trends.

We therefore analyzed the 20 most-cited documents, with review articles excluded. Nonetheless, relevant review articles which were cited more than the 20th most-cited paper were considered separately. The publication and the success of review articles are indeed very informative about each period of time considered. This piece of information, together with other data presented in Table 1 and the most relevant keywords, allowed us to associate to each period of time several prominent research goals:

Table 1. Overview of the main bibliometric indexes regarding "photocatalysis" divided into the four temporal blocks.

Years	1800–1980	1981-1995	1996-2010	2011–Present
Name	Pioneering Years	Maturity	Explosion	Inflation
Citation trend last 5 years	Increasing	Decreasing	Decreasing	Increasing
Relevant reviews	4	9	24	61
Citations (% before 2018)	9149 (75%)	13,889 (89%)	61,836 (73%)	36,235 (40%)
Citations/Papers	457	694	3092	1812
Citations ^a (% before 2018)	11,112 (74%)	51,517 (84%)	143,139 (66%)	156,429 (32%)
Citations/Papers ^a	463	1776	3328	1931
Citations 1st	1993	1947	11,488	4816
Citations 20th	145	501	1658	1187
Median citations Median of the	350	589	2053	1461
year of publication	1978	1991	2004.5	2012.5

a: With relevant reviews.

1—Until 1980, the *Pioneering Years*. In this first period of time, 0.07% of the documents were published. The main research interests concerned the mechanism, the demonstration of functionality and the addition of co-catalysts such as metals and RuO₂. The global picture was dominated by TiO₂, but SrTiO₃, CdS and ZnO were also investigated, and they appear also in the top-cited papers. Researchers' attention was devoted also to the production of H₂, CH₄ (photo-Kolbe) and to CO₂ and N₂ reduction. Review papers were general and focused on principles and the mechanism of the process. [50–72]

2—1980–1995, *Maturity*. In total, 0.49% of the documents were published in this period. The research was directed towards the kinetic analysis, metal doping, pollutant removal and structural studies regarding quantum effects on semiconductors. TiO₂ gained a much more prominent role than in the previous period of time. General reviews were joined by more specific reviews considering sub-fields such as environmental applications, water purification and energy applications. [73–99]

3—1996–2010, *Explosion*. In total, 16.6% of the documents appeared in that period of time. The research concerned non-metal doping (C, N, F), heterostructures, the fabrication of Z-schemes with multiple materials and CO_2 utilization. TiO₂ predominance started to be challenged by C_3N_4 and ternary semiconductors such as BiVO₄. Reviews started to represent the majority of top-cited papers,

and they were focused on specific aspects (e.g., surface properties, disinfection, degradation of specific classes of compounds) and on recent advances. [100–138]

4—2010–present, *Inflation*. In total, 82.8% of the documents were published in this most-recent period of time. The research expanded towards plasmon resonance, and carbon-based materials (graphene, C_3N_4 , C quantum dots) became predominant, both alone and in composites. TiO₂ is still the most studied among oxide materials, usually in combination with other materials. Many elements are now considered Mo, Cd, In, Ga, Ag, Ce, W, C, N, F, S, O, Bi, V, Zn, Cu, Fe, ..., even if crustal abundance considerations would suggest limiting the research on abundant and inexpensive elements (vide infra). Reviews represent more than 75% of top-cited papers, and they are mainly focused on recent advances, on sub-topics (Z-scheme, WPS, CO₂...) and on particular material classes (e.g., specific metal oxides). [139–219]

From the analysis of the most-cited documents (Table 1) within each period, several considerations emerge:

- In periods one and two, more than 60% of the most-cited documents were published in the last 5 years considered. This is also due to the exponential growth of the production with time, because there are more recent papers to cite.
- During the *Inflation* period, the median year of publication is 2012.5, very close to the beginning of the period considered (2011). Even though the time available for the most recent articles is limited, the consideration made before is still valid, and the papers published in 2011–2013 are much less than in 2018–2020, as an example. Another important factor could be the difficulty to find, read and cite a specific paper in the gigantic and increasing scientific production. Recent papers could be less cited than expected because they are not known by the whole community
- Citations before 2018 highlight that, besides the obvious trend for the most recent period, papers form the *Pioneering Years* are still popular and reasonably cited, as 1/4 of their citations arrived in the last 5 years, a trend shared with the *Explosion* (1996–2010). Conversely, only 14% of the citations arrived in the last 5 years of the *Maturity* period. Those documents were already very modern, and it is the authors' opinion that they should receive more recognition in the present literature. Unfortunately, it is sometimes preferred to cite more recent papers to give the idea of cutting-edge research, with few references to the noble and pioneering investigations, completely neglecting instead a whole body of relevant and rigorous research, whose size—in terms of published documents—still allows its almost complete knowledge.

Besides the top-cited papers for each single period of time, there are documents which paved the way to new research lines, but, even though their seminal nature has been recognized by a generous amount of citation, usually more than 300–400, they do not appear in the list of the 20 top-cited articles. We identified at least four representative domains:

• Reduction reactions promoted through photocatalysis [220,221]. This field could regain popularity in the near future because of the emerging concern on PFAS pollution, which has recently been assessed as beyond planetary boundary [222]. PFAS are not emitted in significantly larger amounts compared with other pollutants; nevertheless, their inertness makes them extremely persistent in the environment, and their environmental impact and mitigation costs are therefore relevant [223]. Reduction of the C-F bond through semiconductor photocatalysis could represent an effective strategy for their removal from the environment and especially to prevent their dispersion. Furthermore, a better comprehension of the reductive photoactivated processes could be essential for a better comprehension of processes potentially important for energetics, such as the water photosplitting, the production of hydrogen through reforming of organic by-products or residual biomasses, the CO₂ photo-reduction and the artificial photosynthesis [214,224,225]. Moreover, reductive processes activated by irradiated semiconductors have been proposed both in the *Maturity* period and nowadays for the recovery of precious or critical metals from diluted solutions [60,88,226–230].

- Surface modification by fluorination [231,232] was found to deeply modify TiO₂ behavior in F⁻-containing solutions, [233] because of its strong adsorption on {001} facets and defective sites [234]. This led to a new field of investigation concerning the engineering of TiO₂ nanoparticles, as fluorides could act as powerful shape controllers during hydrothermal synthesis [235]. The resulting nanoplatelets now find applications in different fields, e.g., nanometrology [236].
- Different crystalline facets' reactivity demonstrated by Ohno and coworkers [237], which elegantly demonstrated how {101} facets in anatase and {110} facets in rutile are able to more efficiently trap electrons and therefore promote reduction reactions, whereas {001} facets in anatase and {011} in rutile preferentially trap photoholes and therefore promote oxidation reactions, e.g., PbO₂ deposition from aqueous Pb²⁺ in the original work. This concept has been confirmed and exploited at several reprises, and it represents one of the main strategies to improve the efficiency of the photocatalytic process [238,239].
- Mechanistic studies of semiconductor photocatalysis. Even nowadays, several decades after the pioneering investigations in the field, errors and misconceptions around the working mechanism behind semiconductor photocatalysis are still present and widespread in the specialized literature. One of the most common examples is the interpretation in terms of substrate adsorption of the non-linear growth of substrate removal rate as a function of its concentration. Even if the Langmuir–Hinshelwood isotherm is still frequently reported to justify such a behavior, this explanation has no real physical significance, as elegantly observed by Emeline et al. [240] and formally demonstrated through quantitative kinetic modeling at several reprises [241–246].

4. Present Situation and Future Challenges

To analyze the actual hot topics in the field of photocatalysis, the following approach was adopted. By taking into account the five years 2017–2021, we selected the top 15 mostcited articles (excluding the reviews), and we categorized these articles as a function of the keywords that better described each paper. This approach is able to select the topics on which the scientific community paid more attention that could not necessarily be the topics essential to go ahead both in the comprehension of the phenomena (from a mechanistic point of view) and in the technological exploitation of the process. Furthermore, this classification of the actual main topics has been split in an analysis devoted to the photocatalytic process and phenomena (Figure 5a) and in another one dedicated to the investigated materials (Figure 5b).

Regarding the photocatalytic process investigated, currently the main attention is related to the photocatalytic production of hydrogen, while both the photocatalytic reduction of CO_2 and the traditional removal of pollutants are investigated at the same level and roughly in half of the cases with respect to those dedicated to the H₂ generation. The photocatalytic production of ammonia from molecular nitrogen continues to be a niche topic mainly for the intrinsic difficulty to activate the stable triple covalent bond of N₂.

From the point of view of the mechanism and of the general features of the photocatalytic phenomena, the main attention is driven by the construction and properties of heterostructures (in hybrid photocatalysts) where Z- and S-schemes [247] are operative with the aim to optimize the quantum yield of the process, maximizing the charge carrier separation. At the same time, huge attention is placed on the production of visible-active photocatalysts. On the contrary, topics such as plasmonic photocatalysis, photocatalytic disinfection, the green synthesis of complex compounds and molecular oxygen activation have intercepted a lower number of citations.

On the side of the most-investigated photocatalytic materials, a plethora of compounds is now investigated, and only a limited number of materials are significantly more frequently used. $g-C_3N_4$ is currently the most-studied material, while the role of graphene

and graphene oxide (GO) as functional materials is nowadays dramatically decreased with respect to the impressive attention generated after its discovery and first applications. From the point of view of the most-studied "classic" semiconductors, TiO₂, WO₃, CdS, bismuth-based semiconductors and ZnO are the most investigated. Emerging materials are the MXene materials, quantum dots, the metal organic framework (MOF) and 2D phosphorus (black phosphorous and phosphorene).

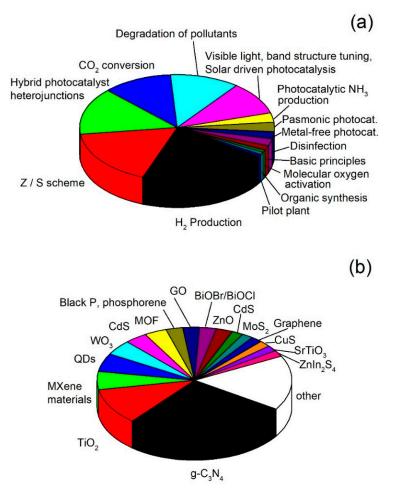


Figure 5. Evaluation of the main topics (**a**) and most-investigated materials (**b**) in the field of photocatalysis in the period 2017–2021.

Next to these more investigated, fascinating and in-fashion topics, there is in our opinion a forgotten topic that is actually far from the overall attention but is actually of pivotal importance. This is the investigation of the photocatalytic technologies in real contexts and at a dimension higher than the lab-scale. Proof of this scarce (almost null) attention comes from the comparison of the works published in the field of photocatalysis dedicated to tests at pilot-plant scale. Of the 158,950 documents found on the Scopus database (September 2022) by using as a keyword "photocatal*", only 300 works (reviews included) are indexed as "photocatal* AND pilot-plant". For an absolute mature scientific theme with important potential applications such as photocatalysis, this scarce attention to the real applicative aspects can for sure be considered a bottleneck in its development. Interestingly, only three countries (Spain, Italy and Portugal) contributed more than 2/3 to the production of these 300 documents. This geographic "anomaly" can be closely related to the scarce diffusion in the world of big infrastructures dedicated to solar photochemistry. In this light, the Plataforma Solar of Almeria (Spain), inserted in the framework of the European infrastructure of the research, is a praiseworthy exception.

5. Conclusions and Perspectives

If we observe the topic of photocatalysis from the beginning to nowadays, this sector will appear with a high level of scientific development thanks especially to the contributions of the Maturity period (1981–1995) from some consolidated research groups located mainly in the USA, Japan and Europe (Italy, France, Germany, the UK and Spain) that mainly clarified from a theoretical point of view the operative mechanism of the photocatalytic processes. On the contrary, the exploitation of these processes in technological applications is still far off, both for energetic applications (e.g., water photosplitting, reforming of organic compounds and photocatalytic CO₂ reduction) and for the removal of recalcitrant pollutants. The only exception is the application of the photocatalytic process for the air depuration/disinfection. Photocatalytic cements are currently on the market [248,249] together with photocatalytic devices able to promote the degradation of gaseous pollutants or deactivation of pathogens [250,251], often with the concomitant application of the photocatalytic technology with ozone and/or UV photolysis (the market of these devices was boosted during the COVID-19 pandemic for obvious reasons). Moreover, on the market there are actually some examples of (i) superwettability and antifogging devices (e.g., mirrors for the automotive sector) based on the covering of a surface with photocatalytic coatings becoming superhydrophilic under irradiation; and (ii) antimicrobial surfaces able to abate the proliferation of pathogens activating the photocatalytic production or reactive species. In particular, investigations of the photocatalytic technologies in large pilot plants are scarce, and this hinders a clear evaluation of all the real potentialities of the photocatalytic technologies.

Interesting and demonstrative is the fact that despite the impressive number of published articles per year and the gigantism of the topic, the Nobel Prize in Chemistry has never been awarded to scientists that have operated in this field. This can be also due to (i) the fact that the scientific knowledge on this topic has been created more with a diffuse activity of different groups of research than by the work of a single scientist, (ii) the presence of only niche applications for photocatalysis has not had impressive consequences on society (as an example, the broad application in everyday life of lithium batteries was for sure at the basis of the appointment of the Nobel prize in Chemistry in 2019 [252,253]).

From our analysis, three main possible scenarios for photocatalysis can be envisaged in the medium term:

- 1. Reduced involvement and reduced scientific production as a consequence of the development of alternative technologies. Examples could be (i) the production of hybrid technologies able to very efficiently convert the sunlight into electricity through photovoltaics and to store this energy in super electrochemical capacitors (and/or other energy storage systems), making useless the photocatalytic production of high energetic vectors such as H₂; (ii) the development of very efficient and scarce energy-demand membrane technologies able to easily remove pollutants from water, creating a market for photocatalytic technologies for the removal of biorecalcitrant pollutants; (iii) for the specific application of effluents decontamination and disinfection, the exponential growing of renewables energies for the production of electricity-based processes that have substantially reduced their environmental impact.
- 2. Almost constant involvement of the researchers, as the research on photocatalysis continues without any major breakthrough and the sector maintains good relevance in the fields of chemistry and material science, reaching a physiological limit in the number of publications mainly related to the limits and rules of the editorial scientific markets.
- 3. Continued increase in the involvement of new research groups attracted by one or more breakthroughs within the traditional research lines in photocatalysis (e.g., synthesis of catalysts with a quantum yield near one operative in the visible spectrum) or outside it (e.g., production of UV irradiation systems at dramatically low cost and with very low energy requests or increment of the cost of competitor technologies

of photocatalysis). In this way the photocatalytic technologies could reach a high technology readiness level (TRL), paying back the impressive scientific efforts carried out—since the beginning—on this topic from generations of scientists.

A conclusive remark must be made on the importance to develop photocatalytic technologies that take into account the real availability of the elements in the earth's crust [254]. The suggestion is always to focus on materials and technologies not based on the use of critical raw materials: for photocatalytic technologies to be a successful "story", they cannot be in competition with other technologies for the same elements (especially if critical [255]) and must not be limited by the crustal availability of the elements essential for their implementation.

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References

- 1. Breeze, P. Solar Power. In Power Generation Technologies, 3rd ed.; Newnes: London, UK, 2019; p. 462.
- 2. Ciamician, G. Sur le actions chimiques de lumière. Bull. Soc. Chim. Fr. 1908, IV, i–xxvii.
- 3. Lemoine, G. Etudes quantitatives sur l'action chimique de la lumière pour òa décomposition mutuelle de l'acide oxalique et du chlorure ferrique. *Ann. Chim. Phys. (Paris)* **1895**, *VI*, 433–540.
- 4. Plotnikow, J. Textbook of Photochemistry; Verlag von Willhelm Knapp: Berlin, Germany, 1910; p. 72.
- 5. Landau, M. Le phénomène de la photocatalyse. Compt. Rend. 1913, 156, 1894–1896.
- 6. Landau, M. Action of ultraviolet rays on lactic acid. Compt. Rend. 1912, 152, 1308–1309.
- Baly, E.C.C.; Heilbron, I.M.; Barker, W.F. Phtocatalysis. Part I. The Synthesis of Fmmaldehyde and Carbohydrates from Carbon Dioxide and Water. J. Chem. Soc. Trans. 1921, 119, 1025–1035. [CrossRef]
- Baly, E.C.C.; Heilbron, I.M.; Hudson, D.P. Photocatalysis. Part II. The Photosynthesis of Nitrogen Compounds from Nitrates and Carbon Dioxide. J. Chem. Soc. Trans. 1922, 121, 1078–1088. [CrossRef]
- Baly, E.C.C.; Heilbron, I.M.; Stern, H.J. Photocatalysis. Part III. The Photosynthesis of Naturally Occurring Xitrogen Compounds from Carbon Dioxide and Ammonia. J. Chem. Soc. Trans. 1923, 123, 185–197. [CrossRef]
- 10. Markham, M.C.; Hannan, M.C.; Paternostro, R.M.; Rose, C.B. Oxidation of Alcohols Catalyzed by Zinc Oxide and Light. J. Am. Chem. Soc. 1958, 80, 5394–5397. [CrossRef]
- Markham, M.C.; Hannan, M.C.; Evans, S.W. Factors Influencing the Oxidation of Phenols, Catalyzed by Zinc Oxide and Light. J. Am. Chem. Soc. 1954, 76, 820–823. [CrossRef]
- 12. Schwab, G.-M.; Noller, H.; Steinbach, F.; Venugopalan, M. Oxidation of Carbon Monoxide and Methyl Alcohol photosensitized in the Gas Phase by Zinc Oxide. *Nature* **1962**, *193*, 774–775. [CrossRef]
- Markham, M.C.; Hannan, M.C.; Lin, L.; Coffey, C.; Jones, B. Photochemical properties of antimony trioxide. J. Phys. Chem. 1958, 62, 989–992. [CrossRef]
- 14. Cohn, G.; Goodeve, C.F. The photochemistry of antimony trioxide. Trans. Faraday Soc. 1940, 35, 433–440. [CrossRef]
- 15. Weyl, W.A.; Förland, T. Photochemistry of rutile. Ind. Eng. Chem. 1950, 42, 257–263. [CrossRef]
- 16. Jacobsen, A.E. Titanium dioxide pigments Correlation between Photochemical Reactivity and Chalking. *Ind. Eng. Chem.* **1949**, *41*, 523–526. [CrossRef]
- 17. Vail, C.B.; Holmquist, J.P.; White, L., Jr. Function of Organic Material in the Photochemical Formation of Hydrogen Peroxide at Zinc Oxide Surfaces. J. Am. Chem. Soc. **1954**, 76, 624–625. [CrossRef]
- 18. Rubin, T.R.; Calvert, J.G.; Rankin, G.T.; MacNevin, W. Photochemical Synthesis of Hydrogen Peroxide at Zinc Oxide Surface. *J. Am. Chem. Soc.* **1953**, *75*, 2850–2853. [CrossRef]
- Calvert, J.G.; Theurer, K.; Rankin, G.T.; MacNevin, W.M. A Study of the Mechanism of the Photochemical Synthesis of Hydrogen Peroxide at Zinc Oxide Surfaces. J. Am. Chem. Soc. 1954, 76, 2575–2578. [CrossRef]
- 20. Markham, M.C. Photocatalytic properties of oxides. J. Chem. Educ. 1955, 32, 540–543. [CrossRef]
- Braslavsky, S.E.; Braun, A.M.; Cassano, A.E.; Emeline, A.V.; Litter, M.I.; Palmisano, L.; Parmon, V.N.; Serpone, N. Glossary of terms used in photocatalysis and radiation catalysis (IUPAC Recommendations 2011). *Pure Appl. Chem.* 2011, *83*, 931–1014. [CrossRef]
- 22. Fujishima, A.; Honda, K. Electrochemical Photolysis of Water at a Semiconductor Electrode. Nature 1972, 238, 37–38. [CrossRef]

- 23. Inoue, T.; Fujishima, A.; Konishi, S.; Honda, K. Photoelectrocatalytic reduction of carbon dioxide in aqueous suspensions of semiconductor powders. *Nature* **1979**, 277, 637–638. [CrossRef]
- 24. Carey, J.H.; Lawrence, J.; Tosine, H.M. Photodechlorination of PCB's in the presence of titanium dioxide in aqueous suspensions. *Bull. Environ. Contam. Toxicol.* **1976**, *16*, 697–701. [CrossRef] [PubMed]
- Pruden, A.L.; Ollis, D.F. Degradation of Chloroform by Photoassisted Heterogeneous Catalysis in Dilute Aqueous Suspensions of Titanium Dioxide. *Environ. Sci. Technol.* 1983, 17, 628–631. [CrossRef] [PubMed]
- 26. Barbeni, M.; Pramauro, E.; Pelizzetti, E.; Borgarello, E.; Serpone, N. Photodegradation of pentachlorophenol catalyzed by semiconductor particles. *Chemosphere* **1985**, *14*, 195–208. [CrossRef]
- 27. Peral, J.; Ollis, D.F. Heterogeneous photocatalytic oxidation of gas-phase organics for air purification: Acetone, 1-butanol, butyraldehyde, formaldehyde, and m-xylene oxidation. *J. Catal.* **1992**, *136*, 554–565. [CrossRef]
- 28. Ireland, J.C.; Klostermann, P.; Rice, E.W.; Clark, R.M. Inactivation of Escherichia coli by Titanium Dioxide Photocatalytic Oxidation. *Appl. Environ. Microbiol.* **1993**, *59*, 1668–1670. [CrossRef]
- 29. Serpone, N.; Emeline, A.V. Semiconductor Photocatalysis—Past, Present, and Future Outlook. J. Phys. Chem. Lett. 2012, 3, 673–677. [CrossRef]
- 30. Sato, S. Photocatalytic activity of NOx-doped TiO₂ in the visible light region. *Chem. Phys. Lett.* **1986**, *123*, *126–128*. [CrossRef]
- Asahi, R.; Morikawa, T.; Ohwaki, T.; Aoki, K.; Taga, Y. Visible-Light Photocatalysis in Nitrogen-Doped Titanium Oxides. *Science* 2001, 293, 269–271. [CrossRef]
- 32. Biedrzycki, J.; Livraghi, S.; Giamello, E.; Agnoli, S.; Granozzi, G. Fluorine- and niobium-doped TiO₂: Chemical and spectroscopic properties of polycrystalline n-type-doped anatase. *J. Phys.Chem. C* **2014**, *118*, 8462–8473. [CrossRef]
- 33. Czoska, A.M.; Livraghi, S.; Chiesa, M.; Giamello, E.; Agnoli, S.; Granozzi, G.; Finazzi, E.; Di Valentiny, C.; Pacchioni, G. The nature of defects in fluorine-doped TiO₂. *J. Phys. Chem. C* 2008, *112*, 8951–8956. [CrossRef]
- 34. Di Valentin, C.; Finazzi, E.; Pacchioni, G.; Selloni, A.; Livraghi, S.; Paganini, M.C.; Giamello, E. N-doped TiO₂: Theory and experiment. *Chem. Phys.* **2007**, *339*, 44–56. [CrossRef]
- 35. Diebold, U. The surface science of titanium dioxide. Surf. Sci. Rep. 2003, 48, 53–229. [CrossRef]
- Liu, N.; Schneider, C.; Freitag, D.; Hartmann, M.; Venkatesan, U.; Müller, J.; Spiecker, E.; Schmuki, P. Black TiO₂ Nanotubes: Cocatalyst-Free Open-Circuit Hydrogen Generation. *Nano Lett.* 2014, 14, 3309–3313. [CrossRef] [PubMed]
- Garcia-Muñoz, P.; Fresno, F.; Ivanez, J.; Robert, D.; Keller, N. Activity enhancement pathways in LaFeO₃@TiO₂ heterojunction photocatalysts for visible and solar light driven degradation of myclobutanil pesticide in water. *J. Hazard. Mater.* 2020, 400, 123099. [CrossRef] [PubMed]
- 38. Ling, C.; Yue, C.; Yuan, R.; Qiu, J.; Liu, F.Q.; Zhu, J.J. Enhanced removal of sulfamethoxazole by a novel composite of TiO₂ nanocrystals in situ wrapped-Bi₂O₄ microrods under simulated solar irradiation. *Chem. Eng. J.* **2020**, *384*, 123278. [CrossRef]
- Tobajas, M.; Belver, C.; Rodriguez, J.J. Degradation of emerging pollutants in water under solar irradiation using novel TiO₂-ZnO/clay nanoarchitectures. *Chem. Eng. J.* 2017, 309, 596–606. [CrossRef]
- 40. Kroto, H.W.; Heath, J.R.; O'Brien, S.C.; Curl, R.F.; Smalley, R.E. C60: Buckminsterfullerene. Nature 1985, 318, 162–163. [CrossRef]
- 41. Iijima, S. Helical Microtubules of Graphitic Carbon. Nature 1991, 354, 56–58. [CrossRef]
- Novoselov, K.S.; Geim, A.K.; Morozov, S.V.; Jiang, D.; Zhang, Y.; Dubonos, S.V.; Grigorieva, I.V.; Firsov, A.A. Electric Field Effect in Atomically Thin Carbon Films. *Science* 2004, 306, 666–669. [CrossRef]
- 43. Minella, M.; Sordello, F.; Minero, C. Graphitic carbon nitride-based metal-free photocatalyst. In *Material Science in Photocatalysis*; García-López, E.I., Palmisano, L., Eds.; Elsevier: Amsterdam, The Netherlands, 2021; pp. 449–484.
- Malato, S.; Miralles-Cuevas, S.; Cabrera-Reina, A. Solar photocatalysis for water decontamination and disinfection (2017–2020). Photochemistry 2022, 49, 236–269.
- 45. Zhang, L.; Zhang, J.; Yu, H.; Yu, J. Emerging S-Scheme Photocatalyst. Adv. Mater. 2022, 34, 2107668. [CrossRef] [PubMed]
- Fujishima, A.; Fujishima, A.; Rao, T.N.; Tryk, D.A. Titanium dioxide photocatalysis. J. Photochem. Photobiol. C-Photochem. Rev. 2000, 1, 2000. [CrossRef]
- 47. Carp, O.; Huisman, C.L.; Reller, A. Photoinduced reactivity of titanium dioxide. *Prog. Solid State Chem.* 2004, 32, 33–177. [CrossRef]
- Som, I.; Roy, M. Recent development on titania-based nanomaterial for photocatalytic CO₂ reduction: A review. J. Alloys Compd. 2022, 918, 165533. [CrossRef]
- Nasir, A.; Khalid, S.; Yasin, T.; Mazare, A. A Review on the Progress and Future of TiO₂/Graphene Photocatalysts. *Energies* 2022, 15, 6248. [CrossRef]
- 50. Morrison, S.R.; Freund, T. Chemical role of holes and electrons in ZnO photocatalysis. J. Chem. Phys. 1967, 47, 1543–1551. [CrossRef]
- 51. Gerischer, H. Charge transfer processes at semiconductor-electrolyte interfaces in connection with problems of catalysis. *Surf. Sci.* **1969**, *18*, 97–122. [CrossRef]
- 52. Bickley, R.I.; Stone, F.S. Photoadsorption and photocatalysis at rutile surfaces. I. Photoadsorption of oxygen. *J. Catal.* **1973**, *31*, 389–397. [CrossRef]
- 53. Bickley, R.I.; Munuera, G.; Stone, F.S. Photoadsorption and photocatalysis at rutile surfaces. II. Photocatalytic oxidation of isopropanol. *J. Catal.* **1973**, *31*, 398–407. [CrossRef]

- 54. Mollers, F.; Tolle, H.J.; Memming, R. On the Origin of the Photocatalytic Deposition of Noble Metals on TiO₂. *J. Electrochem. Soc.* **1974**, 121, 1160–1167. [CrossRef]
- 55. Frank, S.N.; Bard, A.J. Heterogeneous photocatalytic oxidation of cyanide and sulfite in aqueous solutions at semiconductor powders. *J. Phys. Chem.* **1977**, *81*, 1484–1488. [CrossRef]
- Frank, S.N.; Bard, A.J. Heterogeneous Photocatalytic Oxidation of Cyanide Ion in Aqueous Solutions at TiO₂ Powder. J. Am. Chem. Soc. 1977, 99, 303–304. [CrossRef]
- 57. Schrauzer, G.N.; Guth, T.D. Photolysis of Water and Photoreduction of Nitrogen on Titanium Dioxide. J. Am. Chem. Soc. 1977, 99, 7189–7193. [CrossRef]
- Watanabe, T.; Takizawa, T.; Honda, K. Photocatalysis through excitation of adsorbates. 1. Highly efficient N-deethylation of rhodamine B adsorbed to CdS. J. Phys. Chem. 1977, 81, 1845–1851. [CrossRef]
- 59. Janusz, J.M.; Berson, J.A. Heterogeneous Photocatalytic Synthesis of Methane from Acetic Acid—New Kolbe Reaction Pathway. J. Am. Chem. Soc. **1978**, 100, 2239–2240.
- Kraeutler, B.; Bard, A.J. Heterogeneous Photocatalytic Preparation of Supported Catalysts. Photodeposition of Platinum on TiO₂ Powder and Other Substrates. J. Am. Chem. Soc. **1978**, 100, 4317–4318. [CrossRef]
- 61. Kraeutler, B.; Bard, A.J. Heterogeneous Photocatalytic Decomposition of Saturated Carboxylic Acids on TiO₂ Powder. Decarboxylative Route to Alkanes. *J. Am. Chem. Soc.* **1978**, *100*, 5985–5992. [CrossRef]
- 62. Maruska, H.P.; Ghosh, A.K. Photocatalytic decomposition of water at semiconductor electrodes. *Sol. Energy* **1978**, *20*, 443–458. [CrossRef]
- 63. Takizawa, T.; Watanabe, T.; Honda, K. Photocatalysis through excitation of adsorbates. 2. A comparative study of rhodamine B and methylene blue on cadmium sulfide. *J. Phys. Chem.* **1978**, *82*, 1391–1396. [CrossRef]
- 64. Bard, A.J. Photoelectrochemistry and heterogeneous photo-catalysis at semiconductors. J. Photochem. 1979, 10, 59–75. [CrossRef]

 Jaeger, C.D.; Bard, A.J. Spin trapping and electron spin resonance detection of radical intermediates in the photodecomposition of water at TiO₂ particulate systems. J. Phys. Chem. 1979, 83, 3146–3152. [CrossRef]

- Reiche, H.; Dunn, W.W.; Bard, A.J. Heterogeneous photocatalytic and photosynthetic deposition of copper on TiO₂ and WO₃ powders. *J. Phys. Chem.* 1979, *83*, 2248–2251. [CrossRef]
- 67. Bard, A.J. Photoelectrochemistry. Science 1980, 207, 139–144. [CrossRef]
- 68. Domen, K.; Naito, S.; Soma, M.; Onishi, T.; Tamaru, K. Photocatalytic decomposition of water vapour on an NiO-SrTiO₃ catalyst. *J. Chem. Soc. Chem. Commun.* **1980**, *12*, 543–544. [CrossRef]
- Izumi, I.; Dunn, W.W.; Wilbourn, K.O.; Fan, F.R.F.; Bard, A.J. Heterogeneous photocatalytic oxidation of hydrocarbons on platinized TiO₂ powders. *J. Phys. Chem.* 1980, 84, 3207–3210. [CrossRef]
- 70. Kawai, T.; Sakata, T. Conversion of carbohydrate into hydrogen fuel by a photocatalytic process. *Nature* **1980**, *286*, 474–476. [CrossRef]
- 71. Kawai, T.; Sakata, T. Photocatalytic hydrogen production from liquid methanol and water. J. Chem. Soc. Chem. Commun. 1980, 15, 694–695. [CrossRef]
- Kawai, T.; Sakata, T. Photocatalytic decomposition of gaseous water over TiO₂ and TiO₂–RuO₂ surfaces. *Chem. Phys. Lett.* 1980, 72, 87–89. [CrossRef]
- 73. Watanabe, T.; Honda, K. Measurement of the extinction coefficient of the methyl viologen cation radical and the efficiency of its formation by semiconductor photocatalysis. *J. Phys. Chem.* **1982**, *86*, 2617–2619. [CrossRef]
- 74. Shilov, A.E. Energy Resources through Photochemistry and Catalysis; Academic Press: New York, NY, USA, 1983.
- 75. Anpo, M.; Shima, T.; Kodama, S.; Kubokawa, Y. Photocatalytic hydrogenation of CH₃CCH with H₂O on small-particle TiO₂: Size quantization effects and reaction intermediates. *J. Phys. Chem.* **1987**, *91*, 4305–4310. [CrossRef]
- 76. Matthews, R.W. Photooxidation of organic impurities in water using thin films of titanium dioxide. *J. Phys. Chem.* **1987**, *91*, 3328–3333. [CrossRef]
- Al-Ekabi, H.; Serpone, N. Kinetic studies in heterogeneous photocatalysis. 1. Photocatalytic degradation of chlorinated phenols in aerated aqueous solutions over TiO₂ supported on a glass matrix. *J. Phys. Chem.* **1988**, *92*, 5726–5731. [CrossRef]
- Kormann, C.; Bahnemann, D.W.; Hoffmann, M.R. Preparation and characterization of quantum-size titanium dioxide. *J. Phys. Chem.* 1988, 92, 5196–5201. [CrossRef]
- 79. Kormann, C.; Bahnemann, D.W.; Hoffmann, M.R. Photocatalytic production of H₂O₂ and organic peroxides in aqueous suspensions of TiO₂, ZnO, and desert sand. *Environ. Sci. Technol.* **1988**, *22*, 798–806. [CrossRef]
- 80. Matthews, R.W. Kinetics of photocatalytic oxidation of organic solutes over titanium dioxide. J. Catal. 1988, 111, 264–272. [CrossRef]
- Abdullah, M. Effects of common inorganic anions on rates of photocatalytic oxidation of organic carbon over illuminated titanium dioxide. J. Phys. Chem. 1990, 94, 6820–6825. [CrossRef]
- 82. Turchi, C.S.; Ollis, D.F. Photocatalytic degradation of organic water contaminants: Mechanisms involving hydroxyl radical attack. *J. Catal.* **1990**, *122*, 178–192. [CrossRef]
- Bickley, R.I.; Gonzalez-Carreno, T.; Lees, J.S.; Palmisano, L.; Tilley, R.J.D. A structural investigation of titanium dioxide photocatalysts. J. Solid State Chem. 1991, 92, 178–190. [CrossRef]
- Gerischer, H.; Heller, A. The role of oxygen in photooxidation of organic molecules on semiconductor particles. *J. Phys. Chem.* 1991, 95, 5261–5267. [CrossRef]

- Kormann, C.; Hoffmann, M.R.; Bahnemann, D.W. Photolysis of Chloroform and Other Organic Molecules in Aqueous TiO₂ Suspensions. *Environ. Sci. Technol.* 1991, 25, 494–500. [CrossRef]
- 86. Ollis, D.F.; Pelizzetti, E.; Serpone, N. Destruction of water contaminants. Environ. Sci. Technol. 1991, 25, 1522–1529. [CrossRef]
- 87. Fox, M.A.; Dulay, M.T. Heterogeneous Photocatalysis. Chem. Rev. 1993, 93, 341–357. [CrossRef]
- Mills, A.; Davies, R.H.; Worsley, D. Water purification by semiconductor photocatalysis. *Chem. Soc. Rev.* 1993, 22, 417–425. [CrossRef]
- Vinodgopal, K.; Hotchandani, S.; Kamat, P.V. Electrochemically assisted photocatalysis. TiO₂ particulate film electrodes for photocatalytic degradation of 4-chlorophenol. *J. Phys. Chem.* 1993, 97, 9040–9044. [CrossRef]
- Weller, H. Colloidal Semiconductor Q-Particles: Chemistry in the Transition Region Between Solid State and Molecules. *Angew. Chem. Int. Ed. Engl.* 1993, 32, 32–41. [CrossRef]
- 91. Wold, A. Photocatalytic Properties of TiO₂. Chem. Mater. 1993, 5, 280–283. [CrossRef]
- 92. Yamaguchi, T. Application of ZrO2 as a catalyst and a catalyst support. Catal. Today 1994, 20, 199–217. [CrossRef]
- 93. Bamwenda, G.R.; Tsubota, S.; Nakamura, T.; Haruta, M. Photoassisted hydrogen production from a water-ethanol solution: A comparison of activities of AuTiO₂ and PtTiO₂. *J. Photochem. Photobiol. A* **1995**, *89*, 177–189. [CrossRef]
- 94. Fernández, A.; Lassaletta, G.; Jiménez, V.M.; Justo, A.; González-Elipe, A.R.; Herrmann, J.M.; Tahiri, H.; Ait-Ichou, Y. Preparation and characterization of TiO₂ photocatalysts supported on various rigid supports (glass, quartz and stainless steel). Comparative studies of photocatalytic activity in water purification. *Appl. Catal. B Environ.* **1995**, *7*, 49–63. [CrossRef]
- 95. Hoffmann, M.R.; Martin, S.T.; Choi, W.; Bahnemann, D.W. Environmental Applications of Semiconductor Photocatalysis. *Chem. Rev.* **1995**, 95, 69–96. [CrossRef]
- Linsebigler, A.L.; Lu, G.; Yates, J.T., Jr. Photocatalysis on TiO₂ Surfaces: Principles, Mechanisms, and Selected Results. *Chem. Rev.* 1995, 95, 735–758. [CrossRef]
- Obee, T.N.; Brown, R.T. TiO₂ Photocatalysis for Indoor Air Applications: Effects of Humidity and Trace Contaminant Levels on the Oxidation Rates of Formaldehyde, Toluene, and 1,3-Butadiene. *Environ. Sci. Technol.* **1995**, *29*, 1223–1231. [CrossRef]
- Serpone, N.; Maruthamuthu, P.; Pichat, P.; Pelizzetti, E.; Hidaka, H. Exploiting the interparticle electron transfer process in the photocatalysed oxidation of phenol, 2-chlorophenol and pentachlorophenol: Chemical evidence for electron and hole transfer between coupled semiconductors. J. Photochem. Photobiol. A 1995, 85, 247–255. [CrossRef]
- 99. Vinodgopal, K.; Kamat, P.V. Enhanced Rates of Photocatalytic Degradation of an Azo Dye Using SnO₂/TiO₂ Coupled Semiconductor Thin Films. *Environ. Sci. Technol.* **1995**, *29*, 841–845. [CrossRef]
- 100. Mills, A.; Le Hunte, S. An overview of semiconductor photocatalysis. J. Photochem. Photobiol. A-Chem. 1997, 108, 1–35. [CrossRef]
- 101. Andreozzi, R.; Caprio, V.; Insola, A.; Marotta, R. Advanced oxidation processes (AOP) for water purification and recovery. *Catal. Today* **1999**, *53*, 51–59. [CrossRef]
- 102. Herrmann, J.M. Heterogeneous photocatalysis: Fundamentals and applications to the removal of various types of aqueous pollutants. *Catal. Today* **1999**, *53*, 115–129. [CrossRef]
- 103. Kudo, A.; Omori, K.; Kato, H. A novel aqueous process for preparation of crystal form-controlled and highly crystalline BiVO₄ powder from layered vanadates at room temperature and its photocatalytic and photophysical properties. *J. Am. Chem. Soc.* 1999, 121, 11459–11467. [CrossRef]
- Yong, X.; Schoonen, M.A.A. The absolute energy positions of conduction and valence bands of selected semiconducting minerals. *Am. Mineral.* 2000, *85*, 3–4, 543–556.
- 105. Houas, A.; Lachheb, H.; Ksibi, M.; Elaloui, E.; Guillard, C.; Herrmann, J.M. Photocatalytic degradation pathway of methylene blue in water. *Appl. Catal. B Environ.* **2001**, *31*, 145–157. [CrossRef]
- Zou, Z.; Ye, J.; Sayama, K.; Arakawa, H. Direct splitting of water under visible light irradiation with an oxide semiconductor photocatalyst. *Nature* 2001, 414, 625–627. [CrossRef] [PubMed]
- Kamat, P.V. Photophysical, photochemical and photocatalytic aspects of metal nanoparticles. J. Phys. Chem. B 2002, 106, 7729–7744.
 [CrossRef]
- Yu, J.C.; Yu, J.; Ho, W.; Jiang, Z.; Zhang, L. Effects of F⁻ doping on the photocatalytic activity and microstructures of nanocrystalline TiO₂ powders. *Chem. Mater.* 2002, 14, 3808–3816. [CrossRef]
- Hurum, D.C.; Agrios, A.G.; Gray, K.A.; Rajh, T.; Thurnauer, M.C. Explaining the enhanced photocatalytic activity of Degussa P25 mixed-phase TiO₂ using EPR. J. Phys. Chem. B 2003, 107, 4545–4549. [CrossRef]
- Irie, H.; Watanabe, Y.; Hashimoto, K. Nitrogen-concentration dependence on photocatalytic activity of TiO_{2-x}N_x powders. J. Phys. Chem. B 2003, 107, 5483–5486. [CrossRef]
- Sakthivel, S.; Kisch, H. Daylight Photocatalysis by Carbon-Modified Titanium Dioxide. Angew. Chem. Int. Ed. Engl. 2003, 42, 4908–4911. [CrossRef]
- 112. Konstantinou, I.K.; Albanis, T.A. TiO₂-assisted photocatalytic degradation of azo dyes in aqueous solution: Kinetic and mechanistic investigations: A review. *Appl. Catal. B Environ.* **2004**, *49*, 1–14. [CrossRef]
- 113. Pera-Titus, M.; García-Molina, V.; Baños, M.A.; Giménez, J.; Esplugas, S. Degradation of chlorophenols by means of advanced oxidation processes: A general review. *Appl. Catal. B Environ.* **2004**, *47*, 219–256. [CrossRef]
- Subramanian, V.; Wolf, E.E.; Kamat, P.V. Catalysis with TiO₂/Gold Nanocomposites. Effect of Metal Particle Size on the Fermi Level Equilibration. J. Am. Chem. Soc. 2004, 126, 4943–4950. [CrossRef]

- 115. Hashimoto, K.; Irie, H.; Fujishima, A. TiO₂ photocatalysis: A historical overview and future prospects. *Jpn. J. Appl. Phys.* **2005**, *44*, 8269–8285. [CrossRef]
- 116. Tian, Y.; Tatsuma, T. Mechanisms and applications of plasmon-induced charge separation at TiO₂ films loaded with gold nanoparticles. *J. Am. Chem. Soc.* **2005**, 127, 7632–7637. [CrossRef] [PubMed]
- Maeda, K.; Teramura, K.; Lu, D.; Takata, T.; Saito, N.; Inoue, Y.; Domen, K. Photocatalyst releasing hydrogen from water. *Nature* 2006, 440, 295. [CrossRef] [PubMed]
- Mor, G.K.; Varghese, O.K.; Paulose, M.; Shankar, K.; Grimes, C.A. A review on highly ordered, vertically oriented TiO₂ nanotube arrays: Fabrication, material properties, and solar energy applications. *Sol. Energy Mater. Sol. Cells* 2006, 90, 2011–2075. [CrossRef]
- Park, J.H.; Kim, S.; Bard, A.J. Novel carbon-doped TiO₂ nanotube arrays with high aspect ratios for efficient solar water splitting. *Nano Lett.* 2006, *6*, 24–28. [CrossRef] [PubMed]
- 120. Chen, X.; Mao, S.S. Titanium dioxide nanomaterials: Synthesis, properties, modifications and applications. *Chem. Rev.* 2007, 107, 2891–2959. [CrossRef]
- 121. Kamat, P.V. Meeting the clean energy demand: Nanostructure architectures for solar energy conversion. J. Phys. Chem. C 2007, 111, 2834–2860. [CrossRef]
- Ni, M.; Leung, M.K.H.; Leung, D.Y.C.; Sumathy, K. A review and recent developments in photocatalytic water-splitting using TiO₂ for hydrogen production. *Renew. Sustain. Energy Rev.* 2007, 11, 401–425. [CrossRef]
- 123. Fujishima, A.; Zhang, X.; Tryk, D.A. TiO₂ photocatalysis and related surface phenomena. *Surf. Sci. Rep.* **2008**, *63*, 515–582. [CrossRef]
- 124. Gaya, U.I.; Abdullah, A.H. Heterogeneous photocatalytic degradation of organic contaminants over titanium dioxide: A review of fundamentals, progress and problems. *J. Photochem. Photobiol. C-Photochem. Rev.* **2008**, *9*, 1–12. [CrossRef]
- 125. Li, Q.; Mahendra, S.; Lyon, D.Y.; Brunet, L.; Liga, M.V.; Li, D.; Alvarez, P.J.J. Antimicrobial nanomaterials for water disinfection and microbial control: Potential applications and implications. *Water Res.* **2008**, *42*, 4591–4602. [CrossRef] [PubMed]
- Williams, G.; Seger, B.; Kamt, P.V. TiO₂-graphene nanocomposites. UV-assisted photocatalytic reduction of graphene oxide. ACS Nano 2008, 2, 1487–1491. [CrossRef] [PubMed]
- 127. Kudo, A.; Miseki, Y. Heterogeneous photocatalyst materials for water splitting. *Chem. Soc. Rev.* 2009, *38*, 253–278. [CrossRef] [PubMed]
- Malato, S.; Fernández-Ibáñez, P.; Maldonado, M.I.; Blanco, J.; Gernjak, W. Decontamination and disinfection of water by solar photocatalysis: Recent overview and trends. *Catal. Today* 2009, 147, 1–59. [CrossRef]
- 129. Wang, X.; Maeda, K.; Thomas, A.; Takanabe, K.; Xin, G.; Carlsson, J.M.; Domen, K.; Antonietti, M. A metal-free polymeric photocatalyst for hydrogen production from water under visible light. *Nat. Mater.* **2009**, *8*, 76–80. [CrossRef]
- Yan, S.C.; Li, Z.S.; Zou, Z.G. Photodegradation performance of g-C₃N₄ fabricated by directly heating melamine. *Langmuir* 2009, 25, 10397–10401. [CrossRef]
- 131. Chen, X.; Shen, S.; Guo, L.; Mao, S.S. Semiconductor-based photocatalytic hydrogen generation. *Chem. Rev.* **2010**, *110*, 6503–6570. [CrossRef]
- Chong, M.N.; Jin, B.; Chow, C.W.K.; Saint, C. Recent developments in photocatalytic water treatment technology: A review. Water Res. 2010, 44, 2997–3027. [CrossRef]
- 133. Li, H.; He, X.; Kang, Z.; Huang, H.; Liu, Y.; Liu, J.; Lian, S.; Tsang, C.H.A.; Yang, X.; Lee, S.T. Water-soluble fluorescent carbon quantum dots and photocatalyst design. *Angew. Chem. Int. Ed. Engl.* **2010**, *49*, 4430–4434. [CrossRef]
- Liu, G.; Niu, P.; Sun, C.; Smith, S.C.; Chen, Z.; Lu, G.Q.; Cheng, H.M. Unique electronic structure induced high photoreactivity of sulfur-doped graphitic C₃N₄. *J. Am. Chem. Soc.* 2010, *132*, 11642–11648. [CrossRef]
- 135. Maeda, K.; Domen, K. Photocatalytic water splitting: Recent progress and future challenges. J. Phys. Chem. Lett. 2010, 1, 2655–2661. [CrossRef]
- Yi, Z.; Ye, J.; Kikugawa, N.; Kako, T.; Ouyang, S.; Stuart-Williams, H.; Yang, H.; Cao, J.; Luo, W.; Li, Z.; et al. An orthophosphate semiconductor with photooxidation properties under visible-light irradiation. *Nat. Mater.* 2010, *9*, 559–564. [CrossRef] [PubMed]
- 137. Yoon, T.P.; Ischay, M.A.; Du, J. Visible light photocatalysis as a greener approach to photochemical synthesis. *Nat. Chem.* **2010**, *2*, 527–532. [CrossRef] [PubMed]
- Zhang, H.; Lv, X.; Li, Y.; Wang, Y.; Li, J. P25-graphene composite as a high performance photocatalyst. ACS Nano 2010, 4, 380–386.
 [CrossRef]
- 139. Barakat, M.A. New trends in removing heavy metals from industrial wastewater. Arab. J. Chem. 2011, 4, 361–377. [CrossRef]
- 140. Bi, Y.; Ouyang, S.; Umezawa, N.; Cao, J.; Ye, J. Facet effect of single-crystalline Ag₃PO₄ sub-microcrystals on photocatalytic properties. *J. Am. Chem. Soc.* **2011**, *133*, 6490–6492. [CrossRef]
- 141. Chen, X.; Liu, L.; Yu, P.Y.; Mao, S.S. Increasing solar absorption for photocatalysis with black hydrogenated titanium dioxide nanocrystals. *Science* **2011**, *331*, 746–750. [CrossRef]
- 142. Hanaor, D.A.H.; Sorrell, C.C. Review of the anatase to rutile phase transformation. J. Mater. Sci. 2011, 46, 855–874. [CrossRef]
- 143. Henderson, M.A. A surface science perspective on TiO₂ photocatalysis. Surf. Sci. Rep. 2011, 66, 185–297. [CrossRef]
- 144. Kumar, S.G.; Devi, L.G. Review on modified TiO₂ photocatalysis under UV/visible light: Selected results and related mechanisms on interfacial charge carrier transfer dynamics. *J. Phys. Chem. A* **2011**, *115*, 13211–13241. [CrossRef]
- 145. Li, Q.; Guo, B.; Yu, J.; Ran, J.; Zhang, B.; Yan, H.; Gong, J.R. Highly efficient visible-light-driven photocatalytic hydrogen production of CdS-cluster-decorated graphene nanosheets. *J. Am. Chem. Soc.* **2011**, *133*, 10878–10884. [CrossRef] [PubMed]

- 146. Linic, S.; Christopher, P.; Ingram, D.B. Plasmonic-metal nanostructures for efficient conversion of solar to chemical energy. *Nat. Mater.* **2011**, *10*, 911–921. [CrossRef] [PubMed]
- 147. Liu, J.; Zhang, T.; Wang, Z.; Dawson, G.; Chen, W. Simple pyrolysis of urea into graphitic carbon nitride with recyclable adsorption and photocatalytic activity. J. Mater. Chem. 2011, 21, 14398–14401. [CrossRef]
- Oller, I.; Malato, S.; Sánchez-Pérez, J.A. Combination of Advanced Oxidation Processes and biological treatments for wastewater decontamination-A review. *Sci. Total Environ.* 2011, 409, 4141–4166. [CrossRef] [PubMed]
- Roy, P.; Berger, S.; Schmuki, P. TiO₂ nanotubes: Synthesis and applications. *Angew. Chem. Int. Ed. Engl.* 2011, 50, 2904–2939.
 [CrossRef] [PubMed]
- Wang, G.; Wang, H.; Ling, Y.; Tang, Y.; Yang, X.; Fitzmorris, R.C.; Wang, C.; Zhang, J.Z.; Li, Y. Hydrogen-treated TiO₂ nanowire arrays for photoelectrochemical water splitting. *Nano Lett.* 2011, *11*, 3026–3033. [CrossRef]
- Xiang, Q.; Yu, J.; Jaroniec, M. Preparation and enhanced visible-light photocatalytic H₂- production activity of graphene/C₃N₄ composites. J. Phys. Chem. C 2011, 115, 7355–7363. [CrossRef]
- 152. Huang, X.; Qi, X.; Boey, F.; Zhang, H. Graphene-based composites. Chem. Soc. Rev. 2012, 41, 666–686. [CrossRef]
- 153. Kubacka, A.; Fernández-García, M.; Colón, G. Advanced nanoarchitectures for solar photocatalytic applications. *Chem. Rev.* 2012, 112, 1555–1614. [CrossRef]
- 154. Li, H.; Kang, Z.; Liu, Y.; Lee, S.T. Carbon nanodots: Synthesis, properties and applications. *J. Mater. Chem.* **2012**, *22*, 24230–24253. [CrossRef]
- 155. Nakata, K.; Fujishima, A. TiO₂ photocatalysis: Design and applications. *J. Photochem. Photobiol. C-Photochem. Rev.* **2012**, *13*, 169–189. [CrossRef]
- 156. Naldoni, A.; Allieta, M.; Santangelo, S.; Marelli, M.; Fabbri, F.; Cappelli, S.; Bianchi, C.L.; Psaro, R.; Dal Santo, V. Effect of nature and location of defects on bandgap narrowing in black TiO₂ nanoparticles. *J. Am. Chem. Soc.* 2012, 134, 7600–7603. [CrossRef] [PubMed]
- 157. Niu, P.; Zhang, L.; Liu, G.; Cheng, H.M. Graphene-like carbon nitride nanosheets for improved photocatalytic activities. *Adv. Funct. Mater.* **2012**, *22*, 4763–4770. [CrossRef]
- 158. Pelaez, M.; Nolan, N.T.; Pillai, S.C.; Seery, M.K.; Falaras, P.; Kontos, A.G.; Dunlop, P.S.M.; Hamilton, J.W.J.; Byrne, J.A.; O'Shea, K.; et al. A review on the visible light active titanium dioxide photocatalysts for environmental applications. *Appl. Catal. B Environ.* 2012, 125, 331–349. [CrossRef]
- 159. Tong, H.; Ouyang, S.; Bi, Y.; Umezawa, N.; Oshikiri, M.; Ye, J. Nano-photocatalytic materials: Possibilities and challenges. *Adv. Mater.* **2012**, *24*, 229–251. [CrossRef]
- 160. Wang, X.; Blechert, S.; Antonietti, M. Polymeric graphitic carbon nitride for heterogeneous photocatalysis. *ACS Catal.* **2012**, *2*, 1596–1606. [CrossRef]
- 161. Wang, Y.; Wang, X.; Antonietti, M. Polymeric graphitic carbon nitride as a heterogeneous organocatalyst: From photochemistry to multipurpose catalysis to sustainable chemistry. *Angew. Chem. Int. Ed. Engl.* **2012**, *51*, 68–89. [CrossRef]
- 162. Xiang, Q.; Yu, J.; Jaroniec, M. Graphene-based semiconductor photocatalysts. Chem. Soc. Rev. 2012, 41, 782–796. [CrossRef]
- Xiang, Q.; Yu, J.; Jaroniec, M. Synergetic effect of MoS₂ and graphene as cocatalysts for enhanced photocatalytic H₂ production activity of TiO 2 nanoparticles. J. Am. Chem. Soc. 2012, 134, 6575–6578. [CrossRef]
- 164. Xu, P.; Zeng, G.M.; Huang, D.L.; Feng, C.L.; Hu, S.; Zhao, M.H.; Lai, C.; Wei, Z.; Huang, C.; Xie, G.X.; et al. Use of iron oxide nanomaterials in wastewater treatment: A review. *Sci. Total Environ.* **2012**, *424*, 1–10. [CrossRef]
- 165. Xuan, J.; Xiao, W.J. Visible-light photoredox catalysis. Angew. Chem. Int. Ed. Engl. 2012, 51, 6828–6838. [CrossRef] [PubMed]
- Zhang, Z.; Yates, J.T. Band bending in semiconductors: Chemical and physical consequences at surfaces and interfaces. *Chem. Rev.* 2012, 112, 5520–5551. [CrossRef] [PubMed]
- 167. Zheng, Y.; Liu, J.; Liang, J.; Jaroniec, M.; Qiao, S.Z. Graphitic carbon nitride materials: Controllable synthesis and applications in fuel cells and photocatalysis. *Energy Environ. Sci.* **2012**, *5*, 6717–6731. [CrossRef]
- Habisreutinger, S.N.; Schmidt-Mende, L.; Stolarczyk, J.K. Photocatalytic reduction of CO₂ on TiO₂ and other semiconductors. *Angew. Chem. Int. Ed. Engl.* 2013, 52, 7372–7408. [CrossRef] [PubMed]
- Kondratenko, E.V.; Mul, G.; Baltrusaitis, J.; Larrazábal, G.O.; Pérez-Ramírez, J. Status and perspectives of CO₂ conversion into fuels and chemicals by catalytic, photocatalytic and electrocatalytic processes. *Energy Environ. Sci.* 2013, 6, 3112–3135. [CrossRef]
- 170. Li, R.; Zhang, F.; Wang, D.; Yang, J.; Li, M.; Zhu, J.; Zhou, X.; Han, H.; Li, C. Spatial separation of photogenerated electrons and holes among {010} and {110} crystal facets of BiVO₄. *Nat. Commun.* **2013**, *4*, 1432. [CrossRef]
- 171. Osterloh, F.E. Inorganic nanostructures for photoelectrochemical and photocatalytic water splitting. *Chem. Soc. Rev.* 2013, 42, 2294–2320. [CrossRef]
- 172. Pan, X.; Yang, M.Q.; Fu, X.; Zhang, N.; Xu, Y.J. Defective TiO₂ with oxygen vacancies: Synthesis, properties and photocatalytic applications. *Nanoscale* 2013, 5, 3601–3614. [CrossRef]
- 173. Qu, X.; Alvarez, P.J.J.; Li, Q. Applications of nanotechnology in water and wastewater treatment. *Water Res.* 2013, 47, 3931–3946. [CrossRef]
- 174. Scanlon, D.O.; Dunnill, C.W.; Buckeridge, J.; Shevlin, S.A.; Logsdail, A.J.; Woodley, S.M.; Catlow, C.R.A.; Powell, M.J.; Palgrave, R.G.; Parkin, I.P.; et al. Band alignment of rutile and anatase TiO₂. *Nat. Mater.* **2013**, *12*, 798–801. [CrossRef]
- 175. Yang, J.; Wang, D.; Han, H.; Li, C. Roles of cocatalysts in photocatalysis and photoelectrocatalysis. *Acc. Chem. Res.* 2013, 46, 1900–1909. [CrossRef] [PubMed]

- 176. Yang, S.; Gong, Y.; Zhang, J.; Zhan, L.; Ma, L.; Fang, Z.; Vajtai, R.; Wang, X.; Ajayan, P.M. Exfoliated graphitic carbon nitride nanosheets as efficient catalysts for hydrogen evolution under visible light. *Adv. Mater.* 2013, 25, 2452–2456. [CrossRef]
- 177. Aresta, M.; Dibenedetto, A.; Angelini, A. Catalysis for the valorization of exhaust carbon: From CO₂ to chemicals, materials, and fuels. technological use of CO₂. *Chem. Rev.* **2014**, *114*, 1709–1742. [CrossRef] [PubMed]
- Clavero, C. Plasmon-induced hot-electron generation at nanoparticle/metal-oxide interfaces for photovoltaic and photocatalytic devices. *Nat. Photonics* 2014, *8*, 95–103. [CrossRef]
- 179. Dincer, I.; Acar, C. Review and evaluation of hydrogen production methods for better sustainability. *Int. J. Hydrogen Energy* **2014**, 40, 11094–11111. [CrossRef]
- Hisatomi, T.; Kubota, J.; Domen, K. Recent advances in semiconductors for photocatalytic and photoelectrochemical water splitting. *Chem. Soc. Rev.* 2014, 43, 7520–7535. [CrossRef]
- Kolodziejczak-Radzimska, A.; Jesionowski, T. Zinc oxide-from synthesis to application: A review. *Materials* 2014, 7, 2833–2881.
 [CrossRef]
- 182. Ma, Y.; Wang, X.; Jia, Y.; Chen, X.; Han, H.; Li, C. Titanium dioxide-based nanomaterials for photocatalytic fuel generations. *Chem. Rev.* **2014**, *114*, 9987–10043. [CrossRef]
- 183. Oturan, M.A.; Aaron, J.J. Advanced oxidation processes in water/wastewater treatment: Principles and applications. A review. *Crit. Rev. Environ. Sci. Technol.* 2014, 44, 2577–2641. [CrossRef]
- Ran, J.; Zhang, J.; Yu, J.; Jaroniec, M.; Qiao, S.Z. Earth-abundant cocatalysts for semiconductor-based photocatalytic water splitting. *Chem. Soc. Rev.* 2014, 43, 7787–7812. [CrossRef]
- 185. Schultz, D.M.; Yoon, T.P. Solar synthesis: Prospects in visible light photocatalysis. Science 2014, 343, 6174. [CrossRef] [PubMed]
- Wang, C.C.; Li, J.R.; Lv, X.L.; Zhang, Y.Q.; Guo, G. Photocatalytic organic pollutants degradation in metal-organic frameworks. Energy Environ. Sci. 2014, 7, 2831–2867. [CrossRef]
- 187. Wang, H.; Zhang, L.; Chen, Z.; Hu, J.; Li, S.; Wang, Z.; Liu, J.; Wang, X. Semiconductor heterojunction photocatalysts: Design, construction, and photocatalytic performances. *Chem. Soc. Rev.* **2014**, *43*, 5234–5244. [CrossRef] [PubMed]
- 188. Yu, J.; Low, J.; Xiao, W.; Zhou, P.; Jaroniec, M. Enhanced photocatalytic CO₂-Reduction activity of anatase TiO₂ by Coexposed {001} and {101} facets. *J. Am. Chem. Soc.* **2014**, *136*, 8839–8842. [CrossRef] [PubMed]
- Zhang, T.; Lin, W. Metal-organic frameworks for artificial photosynthesis and photocatalysis. *Chem. Soc. Rev.* 2014, 43, 5982–5993.
 [CrossRef]
- 190. Zhou, P.; Yu, J.; Jaroniec, M. All-solid-state Z-scheme photocatalytic systems. Adv. Mater. 2014, 26, 4920–4935. [CrossRef]
- 191. Bonaccorso, F.; Colombo, L.; Yu, G.; Stoller, M.; Tozzini, V.; Ferrari, A.C.; Ruoff, R.S.; Pellegrini, V. Graphene, related twodimensional crystals, and hybrid systems for energy conversion and storage. *Science* **2015**, 347, 6217. [CrossRef]
- 192. Brillas, E.; Martínez-Huitle, C.A. Decontamination of wastewaters containing synthetic organic dyes by electrochemical methods. An updated review. *Appl. Catal. B Environ.* **2015**, *166–167*, 603–643. [CrossRef]
- Brongersma, M.L.; Halas, N.J.; Nordlander, P. Plasmon-induced hot carrier science and technology. *Nat. Nanotechnol.* 2015, 10, 25–34. [CrossRef]
- 194. Cao, S.; Low, J.; Yu, J.; Jaroniec, M. Polymeric Photocatalysts Based on Graphitic Carbon Nitride. *Adv. Mater.* **2015**, *27*, 2150–2176. [CrossRef]
- 195. Li, X.; Yu, J.; Low, J.; Fang, Y.; Xiao, J.; Chen, X. Engineering heterogeneous semiconductors for solar water splitting. *J. Mater. Chem. A* 2015, *3*, 2485–2534. [CrossRef]
- 196. Lim, S.Y.; Shen, W.; Gao, Z. Carbon quantum dots and their applications. Chem. Soc. Rev. 2015, 44, 362–381. [CrossRef] [PubMed]
- 197. Liu, J.; Liu, Y.; Liu, N.; Han, Y.; Zhang, X.; Huang, H.; Lifshitz, Y.; Lee, S.T.; Zhong, J.; Kang, Z. Metal-free efficient photocatalyst for stable visible water splitting via a two–electron pathway. *Science* **2015**, *347*, 970–974. [CrossRef] [PubMed]
- 198. Moniz, S.J.A.; Shevlin, S.A.; Martin, D.J.; Guo, Z.X.; Tang, J. Visible-light driven heterojunction photocatalysts for water splitting-a critical review. *Energy Environ. Sci.* 2015, *8*, 731–759. [CrossRef]
- 199. Sirelkhatim, A.; Mahmud, S.; Seeni, A.; Kaus, N.H.M.; Ann, L.C.; Bakhori, S.K.M.; Hasan, H.; Mohamad, D. Review on zinc oxide nanoparticles: Antibacterial activity and toxicity mechanism. *Nano-Micro Lett.* **2015**, *7*, 219–242. [CrossRef]
- 200. Zhao, Z.; Sun, Y.; Dong, F. Graphitic carbon nitride based nanocomposites: A review. Nanoscale 2015, 7, 15–37. [CrossRef]
- 201. Zou, X.; Zhang, Y. Noble metal-free hydrogen evolution catalysts for water splitting. *Chem. Soc. Rev.* 2015, 44, 5148–5180. [CrossRef]
- 202. Gawande, M.; Goswami, A.; Felpin, F.X.; Asefa, T.; Huang, X.; Silva, R.; Zou, X.; Zboril, R.; Varma, R.S. Cu and Cu-Based Nanoparticles: Synthesis and Applications in Catalysis. *Chem. Rev.* **2016**, *116*, 3722–3811. [CrossRef]
- Georgakilas, V.; Tiwari, J.N.; Kemp, K.C.; Perman, J.A.; Bourlinos, A.B.; Kim, K.S.; Zboril, R. Noncovalent Functionalization of Graphene and Graphene Oxide for Energy Materials, Biosensing, Catalytic, and Biomedical Applications. *Chem. Rev.* 2016, 116, 5464–5519. [CrossRef]
- 204. Lee, K.M.; Lai, C.W.; Ngai, K.S.; Juan, J.C. Recent developments of zinc oxide based photocatalyst in water treatment technology: A review. Water Res. 2016, 88, 428–448. [CrossRef]
- 205. Li, X.; Yu, J.; Jaroniec, M. Hierarchical photocatalysts. Chem. Soc. Rev. 2016, 45, 2603–2636. [CrossRef]
- Montini, T.; Melchionna, M.; Monai, M.; Fornasiero, P. Fundamentals and Catalytic Applications of CeO₂-Based Materials. *Chem. Rev.* 2016, 116, 5987–6041. [CrossRef] [PubMed]

- 207. Ong, W.J.; Tan, L.L.; Ng, Y.H.; Yong, S.T.; Chai, S.P. Graphitic Carbon Nitride (g-C₃N₄)-Based Photocatalysts for Artificial Photosynthesis and Environmental Remediation: Are We a Step Closer to Achieving Sustainability? *Chem. Rev.* 2016, 116, 7159–7329. [CrossRef] [PubMed]
- Huang, Y.B.; Liang, J.; Wang, X.S.; Cao, R. Multifunctional metal-organic framework catalysts: Synergistic catalysis and tandem reactions. *Chem. Soc. Rev.* 2017, 46, 126–157. [CrossRef] [PubMed]
- Low, J.; Yu, J.; Jaroniec, M.; Wageh, S.; Al-Ghamdi, A.A. Heterojunction Photocatalysts. Adv. Mater. 2017, 29, 1601694. [CrossRef]
 [PubMed]
- 210. Nosaka, Y.; Nosaka, A.Y. Generation and Detection of Reactive Oxygen Species in Photocatalysis. *Chem. Rev.* 2017, 117, 11302–11336. [CrossRef]
- 211. Tan, C.; Cao, X.; Wu, X.J.; He, Q.; Yang, J.; Zhang, X.; Chen, J.; Zhao, W.; Han, S.; Nam, G.H.; et al. Recent Advances in Ultrathin Two-Dimensional Nanomaterials. *Chem. Rev.* 2017, 117, 6225–6331. [CrossRef]
- 212. Wen, J.; Xie, J.; Chen, X.; Li, X. A review on g-C₃N₄-based photocatalysts. Appl. Surf. Sci. 2017, 391, 72–123. [CrossRef]
- Yu, H.; Shi, R.; Zhao, Y.; Bian, T.; Zhao, Y.; Zhou, C.; Waterhouse, G.I.N.; Wu, L.Z.; Tung, C.H.; Zhang, T. Alkali-Assisted Synthesis of Nitrogen Deficient Graphitic Carbon Nitride with Tunable Band Structures for Efficient Visible-Light-Driven Hydrogen Evolution. *Adv. Mater.* 2017, 29, 1605148. [CrossRef]
- 214. Fu, J.; Yu, J.; Jiang, C.; Cheng, B. g-C₃N₄-Based Heterostructured Photocatalysts. Adv. Energy Mater. 2018, 8, 1701503. [CrossRef]
- Liu, L.; Corma, A. Metal Catalysts for Heterogeneous Catalysis: From Single Atoms to Nanoclusters and Nanoparticles. *Chem. Rev.* 2018, 118, 4981–5079. [CrossRef] [PubMed]
- Miklos, D.B.; Remy, C.; Jekel, M.; Linden, K.G.; Drewes, J.E.; Hübner, U. Evaluation of advanced oxidation processes for water and wastewater treatment—A critical review. *Water Res.* 2018, 139, 118–131. [CrossRef] [PubMed]
- 217. Yuan, S.; Feng, L.; Wang, K.; Pang, J.; Bosch, M.; Lollar, C.; Sun, Y.; Qin, J.; Yang, X.; Zhang, P.; et al. Stable Metal-Organic Frameworks: Design, Synthesis, and Applications. *Adv. Mater.* **2018**, *30*, 1704303. [CrossRef] [PubMed]
- Fu, J.; Xu, Q.; Low, J.; Jiang, C.; Yu, J. Ultrathin 2D/2D WO₃/g-C₃N₄ step-scheme H₂-production photocatalyst. *Appl. Catal. B Environ.* 2019, 243, 556–565. [CrossRef]
- Wu, H.; Tan, H.L.; Toe, C.Y.; Scott, J.; Wang, L.; Amal, R.; Ng, Y.H. Photocatalytic and Photoelectrochemical Systems: Similarities and Differences. *Adv. Mater.* 2020, 32, 1904717. [CrossRef]
- Minero, C.; Piccinini, P.; Calza, P.; Pelizzetti, E. Photocatalytic reduction/oxidation processes occurring at the carbon and nitrogen of tetranitromethane. *New J. Chem.* 1996, 20, 1159–1164.
- Calza, P.; Minero, C.; Pelizzetti, E. Photocatalytically Assisted Hydrolysis of Chlorinated Methanes under Anaerobic Conditions. Environ. Sci. Technol. 1997, 31, 2198–2203. [CrossRef]
- Cousins, I.T.; Johansson, J.H.; Salter, M.E.; Sha, B.; Scheringer, M. Outside the Safe Operating Space of a New Planetary Boundary for Per- and Polyfluoroalkyl Substances (PFAS). *Environ. Sci. Technol.* 2022, 56, 11172–11179. [CrossRef]
- 223. Obsekov, V.; Kahn, L.G.; Trasande, L. Leveraging. Systematic Reviews to Explore Disease Burden and Costs of Per- and Polyfluoroalkyl Substance Exposures in the United States. *Expo. Health* **2022**. [CrossRef]
- Kärkäs, M.D.; Verho, O.; Johnston, E.V.; Åkermark, B. Artificial Photosynthesis: Molecular Systems for Catalytic Water Oxidation. *Chem. Rev.* 2014, 114, 11863–12001. [CrossRef]
- Tao, X.; Zhao, Y.; Wang, S.; Li, C.; Li, R. Recent advances and perspectives for solar-driven water splitting using particulate photocatalysts. *Chem. Soc. Rev.* 2022, *51*, 3561–3608. [CrossRef] [PubMed]
- Baba, R.; Nakabayashi, S.; Fujishima, A.; Honda, K. Investigation of the mechanism of hydrogen evolution during photocatalytic water decomposition on metal-loaded semiconductor powders. J. Phys. Chem. 1985, 89, 1902–1905. [CrossRef]
- 227. Borgarello, E.; Serpone, N.; Emo, G.; Harris, R.; Pelizzetti, E.; Minero, C. Light-Induced Reduction of Rhodium(III) and Palladium(II) on Titanium Dioxide Dispersions and the Selective Photochemical Separation and Recovery of Gold(III), Platinum(IV), and Rhodium(III) in Chloride Media. *Inorg. Chem.* **1986**, 25, 4499–4503. [CrossRef]
- Curran, J.S.; Domenech, J.; Jaffrezic-Renault, N.; Philippe, R. Kinetics and Mechanism of Platinum Deposition by Photoelectrolysis in Illuminated Suspensions of Semiconducting Titanium Dioxide. J. Phys. Chem. 1985, 89, 957–963. [CrossRef]
- 229. Litter, M.I. Heterogeneous photocatalysis Transition metal ions in photocatalytic systems. *Appl. Catal. B Environ.* **1999**, 23, 89–114. [CrossRef]
- Chen, Y.; Xu, M.; Wen, J.; Wan, Y.; Zhao, Q.; Cao, X.; Ding, Y.; Wang, Z.L.; Li, H.; Bian, Z. Selective recovery of precious metals through photocatalysis. *Nat. Sustain.* 2021, *4*, 618–626. [CrossRef]
- Minero, C.; Mariella, G.; Maurino, V.; Pelizzetti, E. Photocatalytic Transformation of Organic Compounds in the Presence of Inorganic Anions. 1. Hydroxyl-Mediated and Direct Electron-Transfer Reactions of Phenol on a Titanium Dioxide–Fluoride System. *Langmuir* 2000, 16, 2632–2641. [CrossRef]
- Minero, C.; Mariella, G.; Maurino, V.; Vione, D.; Pelizzetti, E. Photocatalytic transformation of organic compounds in the presence of inorganic ions. 2. Competitive reactions of phenol and alcohols on a titanium dioxide-fluoride system. *Langmuir* 2000, 16, 8964–8972. [CrossRef]
- Monllor-Satoca, D.; Gómez, R. Electrochemical Method for Studying the Kinetics of Electron Recombination and Transfer Reactions in Heterogeneous Photocatalysis: The Effect of Fluorination on TiO₂ Nanoporous Layers. J. Phys. Chem. C 2007, 112, 139–147. [CrossRef]

- Minella, M.; Faga, M.G.; Maurino, V.; Minero, C.; Pelizzetti, E.; Coluccia, S.; Martra, G. Effect of fluorination on the surface properties of titania P25 powder: An FTIR study. *Langmuir* 2010, 26, 2521–2527. [CrossRef]
- 235. Taguchi, T.; Saito, Y.; Sarukawa, K.; Ohno, T.; Matsumura, M. Formation of new crystal faces on TiO₂ particles by treatment with aqueous HF solution or hot sulfuric acid. *New J. Chem.* **2003**, *27*, 1304–1306. [CrossRef]
- Maurino, V.; Pellegrino, F.; Picotto, G.B.; Ribotta, L. Quantitative three-dimensional characterization of critical sizes of nonspherical TiO₂ nanoparticles by using atomic force microscopy. *Ultramicroscopy* 2022, 234, 113480. [CrossRef] [PubMed]
- Ohno, T.; Sarukawa, K.; Matsumura, M. Crystal faces of rutile and anatase TiO₂ particles and their roles in photocatalytic reactions. *New J. Chem.* 2002, 26, 1167–1170. [CrossRef]
- D'Arienzo, M.; Dozzi, M.V.; Redaelli, M.; Di Credico, B.; Morazzoni, F.; Scotti, R.; Polizzi, S. Crystal Surfaces and Fate of Photogenerated Defects in Shape-Controlled Anatase Nanocrystals: Drawing Useful Relations to Improve the H₂ Yield in Methanol Photosteam Reforming. J. Phys. Chem. C 2015, 119, 12385–12393. [CrossRef]
- Pellegrino, F.; Sordello, F.; Mino, L.; Minero, C.; Hodoroaba, V.-D.; Martra, G.; Maurino, V. Formic Acid Photoreforming for Hydrogen Production on Shape-Controlled Anatase TiO₂ Nanoparticles: Assessment of the Role of Fluorides, {101}/{001} Surfaces Ratio, and Platinization. ACS Catal. 2019, 9, 6692–6697. [CrossRef]
- Emeline, A.V.; Ryabchuk, V.K.; Serpone, N. Dogmas and Misconceptions in Heterogeneous Photocatalysis. Some Enlightened Reflections. J. Phys. Chem. B 2005, 109, 18515–18521. [CrossRef]
- 241. Minero, C. A rigorous kinetic approach to model primary oxidative steps of photocatalytic degradations. *Sol. Energy Mater. Sol. Cells* **1995**, *38*, 421–430. [CrossRef]
- 242. Minero, C. Kinetic analysis of photoinduced reactions at the water semiconductor interface. *Catal. Today* **1999**, *54*, 205–216. [CrossRef]
- 243. Minero, C.; Vione, D. A quantitative evalution of the photocatalytic performance of TiO₂ slurries. *Appl. Catal. B Environ.* **2006**, 67, 257–269. [CrossRef]
- 244. Camera-Roda, G.; Augugliaro, V.; Cardillo, A.G.; Loddo, V.; Palmisano, L.; Parrino, F.; Santarelli, F. A reaction engineering approach to kinetic analysis of photocatalytic reactions in slurry systems. *Catal. Today* **2016**, *259*, 87–96. [CrossRef]
- Brandi, R.J.; Alfano, O.M.; Cassano, A.E. Evaluation of Radiation Absorption in Slurry Photocatalytic Reactors. 1. Assessment of Methods in Use and New Proposal. *Environ. Sci. Technol.* 2000, 34, 2623–2630. [CrossRef]
- 246. Brandi, R.J.; Alfano, O.M.; Cassano, A.E. Evaluation of Radiation Absorption in Slurry Photocatalytic Reactors. 2. Experimental Verification of the Proposed Method. *Environ. Sci. Technol.* 2000, *34*, 2631–2639. [CrossRef]
- 247. Xu, Q.; Zhang, L.; Cheng, B.; Fan, J.; Yu, J. S-Scheme Heterojunction Photocatalyst. Chem 2020, 6, 1543–1559. [CrossRef]
- 248. Hamdany, A.H.; Satyanaga, A.; Zhang, D.; Kim, Y.; Kim, J.R. Photocatalytic Cementitious Material for Eco-Efficient Construction— A Systematic Literature Review. *Appl. Sci.* **2022**, *12*, 8741. [CrossRef]
- 249. Dudek, D.; Janus, M. Photoactive Cements: A Review. Materials 2022, 15, 5407. [CrossRef] [PubMed]
- 250. Minella, M.; Minero, C. Evaluation of gas / solid photocatalytic performance for the removal of VOCs at ppb and sub-ppb levels. *Chemosphere* **2021**, 272, 129636. [CrossRef] [PubMed]
- Zacarías, S.M.; Manassero, A.; Pirola, S.; Alfano, O.M.; Satuf, M.L. Design and performance evaluation of a photocatalytic reactor for indoor air disinfection. *Environ. Sci. Pollut. Res.* 2021, 28, 23859–23867. [CrossRef]
- Available online: https://www.nobelprize.org/uploads/2019/10/advanced-chemistryprize2019.pdf (accessed on 16 September 2022).
- 253. Manthiram, A. A reflection on lithium-ion battery cathode chemistry. Nat. Commun. 2020, 11, 1550. [CrossRef]
- 254. Suess, H.E.; Urey, H.C. Abundances of the Elements. Rev. Mod. Phys. 1956, 28, 1956. [CrossRef]
- 255. Rhodes, C.J. Endangered elements, critical raw materials and conflict minerals. Sci. Prog. 2019, 102, 304–350. [CrossRef]