



Enhancing HER rate over Pt-TiO₂ nanoparticles under CPI - role of illumination's parameters



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INTRODUCTION

In previous works,^{1,2} our group demonstrated the possibility to enhance H₂ production through HCOOH photocatalytic reforming on metal-TiO₂ nanoparticles under Controlled Periodic Illumination (CPI) in respect to continuous illumination, at the same average incident photon flux.

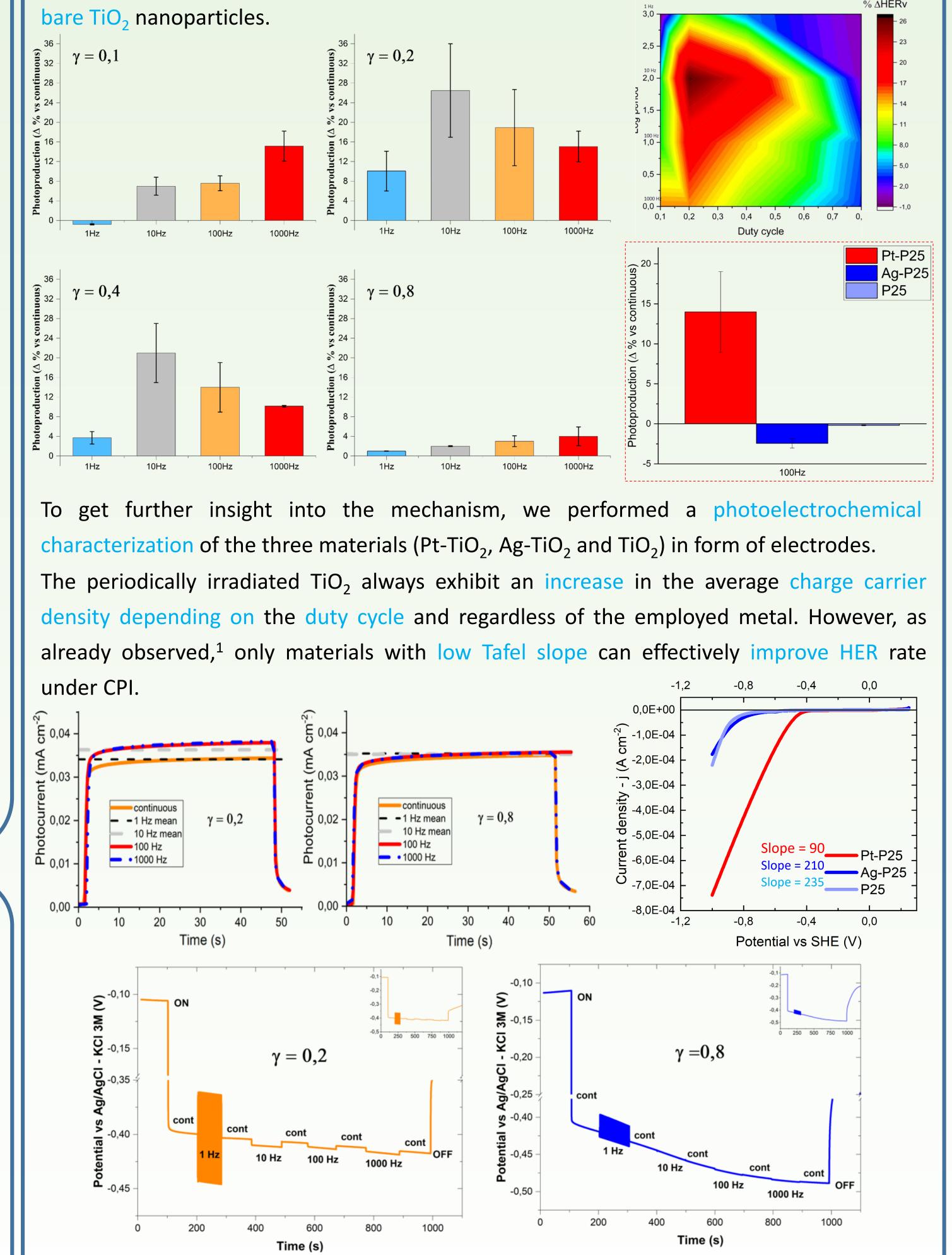
The enhancement was observed only over some specific metals, such as Pt, Pd and Rh. Hydrogen interaction strength (i.e. H absorption/desorption) is strongly dependent on the potential at the metal nanoparticles.³ We observed that CPI induces oscillations in catalyst's potential, however only when some metals are employed

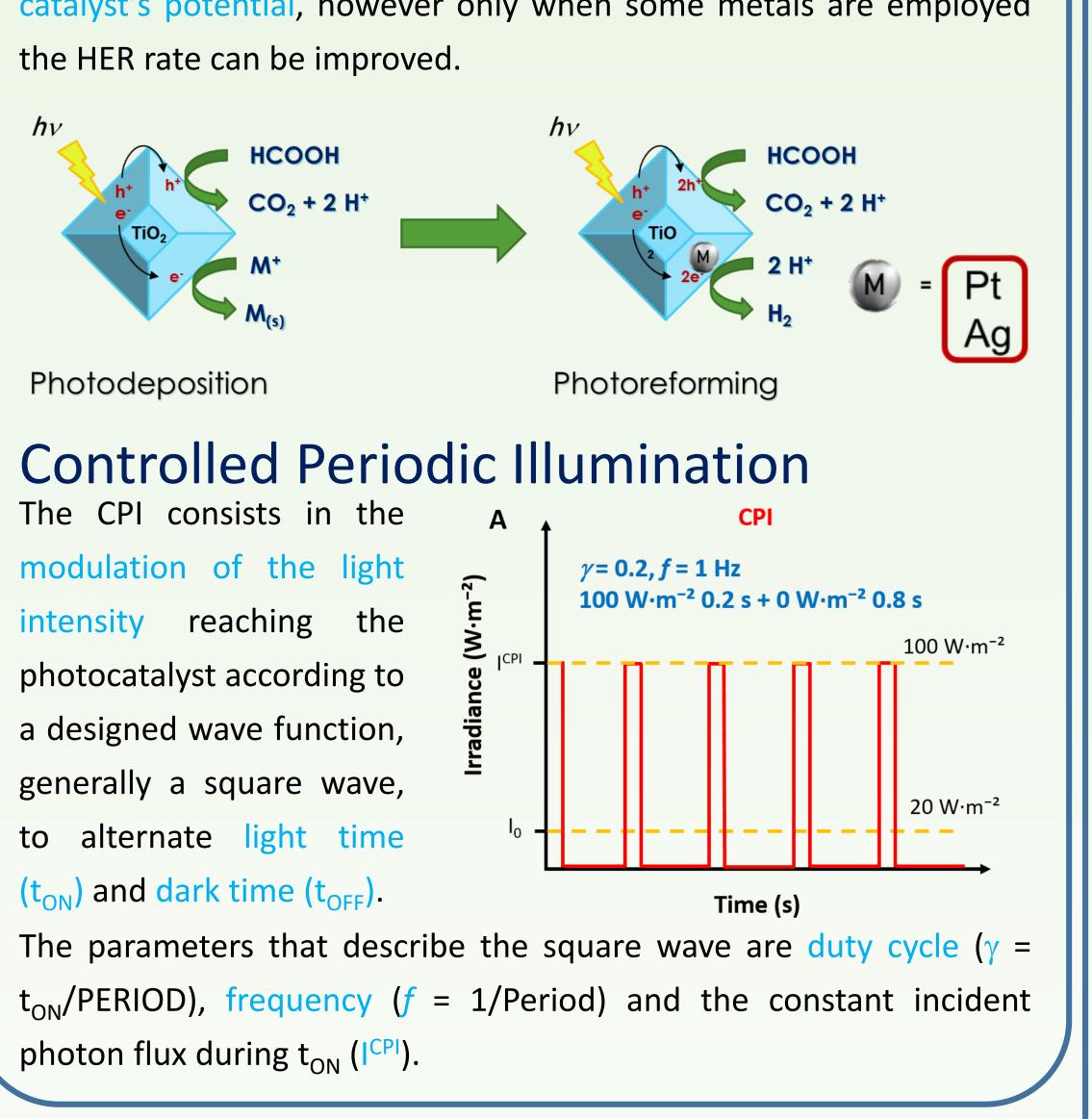
RESULTS

The Pt-TiO₂ catalyst was tested at four different duty cycles and at frequencies from 1 Hz to 1 kHz.

In almost all cases, an increment in HER rate relative to continuous illumination was observed. This result is particularly pronounced above 1 Hz and for intermediate duty cycle ($\gamma = 0.2$, $\gamma = 0.4$).

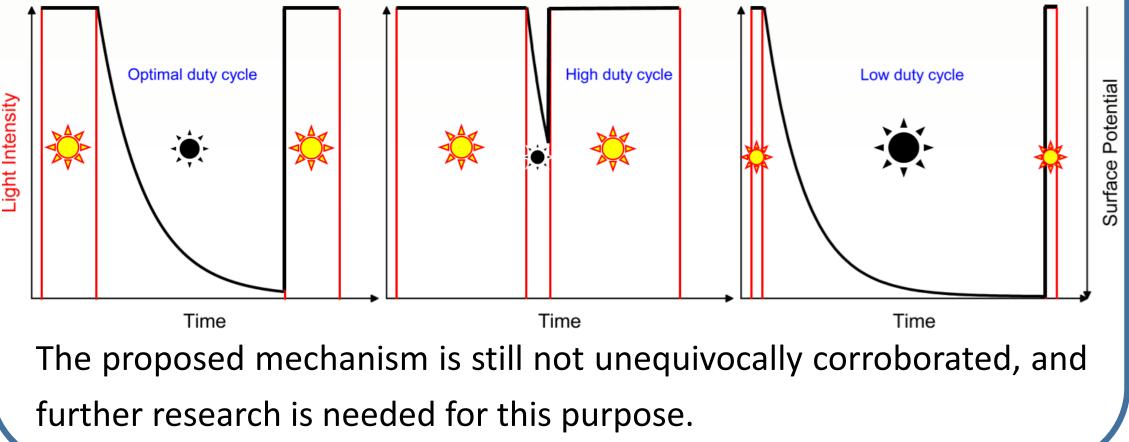
As a comparison, it wasn't possible to observe any increment in HER rate over Ag-TiO₂ and





CONCLUSIONS

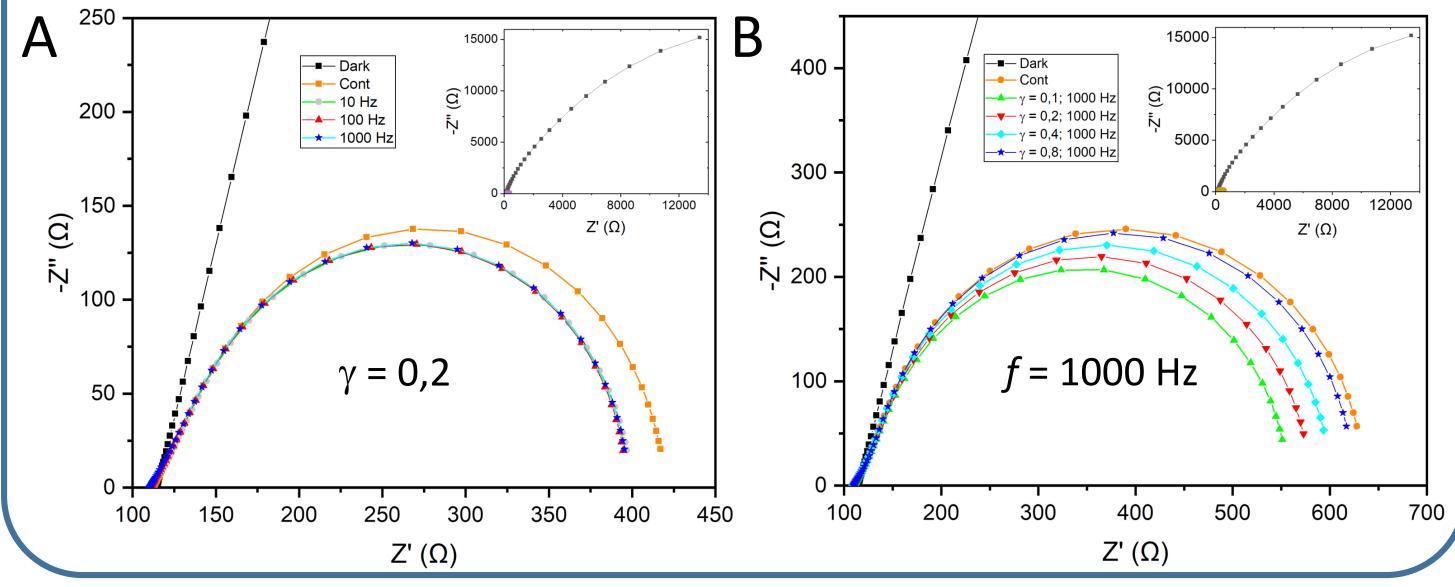
In this work we have demonstrated the dependence of HER rate under CPI on frequency and duty cycle on a Pt-TiO2 catalyst, which is also reflected in the current density increase and the OCP decay. We suppose this behaviour to be related to the oscillating catalyst's surface potential during CPI, that enable the system to overcome the optimum efficiency achievable in static conditions, in analogy with the theory of surface catalytic resonance.⁴



REFERENCES

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To further investigate the charge carriers dynamic under different illumination conditions, EIS measurements were performed at different frequencies for a given duty cycle (A), and at different duty cycles for a given frequency (B). Results clearly show that the electron transfer to the electrolyte is improved under CPI and for low duty cycles.



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